BEFORE THE ILLINOIS POLLUTION CONTROL BOARD

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In the Matter of:

PROPOSED AMENDMENTS TO GROUNDWATER QUALITY (35 ILL. ADM. CODE 620) R 2022-018

(Rulemaking – Public Water Supply)

NOTICE OF FILING

To: ALL PARTIES ON THE SERVICE LIST

PLEASE TAKE NOTICE that I have today electronically filed with the Office of the Clerk of the Illinois Pollution Control Board, the **PRE-FILED ANSWERS OF ERIC BALLENGER ON BEHALF OF NATIONAL WASTE & RECYCLING ASSOCIATION**, copies of which are hereby served upon you.

Dated: November 23, 2022

By /s/ Claire A. Manning

BROWN, HAY & STEPHENS, LLP

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BEFORE THE ILLINOIS POLLUTION CONTROL BOARD

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In the Matter of:

PROPOSED AMENDMENTS TO GROUNDWATER QUALITY (35 ILL. ADM. CODE 620) R 2022-018

(Rulemaking – Public Water Supply)

PRE-FILED ANSWERS OF ERIC BALLENGER ON BEHALF OF NATIONAL WASTE & RECYCLING ASSOCIATION

QUESTIONS FROM THE ILLINOIS POLLUTION CONTROL BOARD

23. On page 2, you state that it should be understood by the regulating agencies and the Board that landfills are receivers of PFAS, not users or producers. Please comment on whether most of the contaminants in landfill leachate are derived from wastes received by the landfills and not produced by the landfills.

ANSWER: Waste companies provide a public service by disposing of waste created by the public in landfills that have been constructed in accordance with regulatory standards long considered safe – utilizing regulatory design standards that include liners and leachate collection systems. PFAS contaminants in landfill leachate would derive from the legally authorized waste received by the landfill and disposed of therein, which includes waste with PFAS-containing compounds. This includes many common household products, food packaging, commercial waste, WWTP biosolids, and many other common MSW Landfill waste streams.

See Attachment A, Letter to USEPA in Docket ID No. EPA-HQ-OLEM-2019-0341, Nov. 7, 2022.

See Attachment B, Letter to USEPA from NWRA and Solid Waste Association of North America, in Docket ID No. EPA-HQ-OLEM-2019-0341, Nov. 7, 2022.

- 24. On page 2, you state, "this will affect 807 sites as well as "greenfield" sites all the way through post-closure of currently active facilities."
 - a. Regarding Part 807 facilities, please clarify whether you are referring to landfills or all types of waste disposal facilities regulated under that Part.

ANSWER: The concern throughout my comments is that we do not know how IEPA intends to implement these new strict groundwater standards in the context of the landfill regulations, and we have no idea or control over when or whether the IEPA might seek to change the landfill regulations to address these concerns. Thus, we are forced to address the issues based upon our experience with IEPA implementing other Board-promulgated groundwater standards at landfill sites.

Our further concern here is that given the very strict standards proposed, and the ubiquitous nature of PFAS, the proposed standards may not be achievable and/or may reflect background conditions unrelated to possible landfill releases – forcing environmental violations without properly assessing actual environmental or public health risk in the context of landfill operations.

As to the Board's specific question, I am referring to all permitted landfills that have groundwater monitoring obligations – recognizing of course that any landfill still regulated under Part 807 has long ago closed and, while it has different obligations than newer landfills regulated pursuant to Part 814, some old waste units continue to be regulated under Part 807 and still have groundwater monitoring obligations as the IEPA has not released those areas from post-closure care.

b. If you are referring to landfills, please comment on whether landfills in the State that are still being regulated under Part 807 or they generally regulated under Parts 813 and 814.

ANSWER: See above.

c. Please explain what you mean by "greenfields" in the above statement.

<u>ANSWER</u>: Greenfields is a term used to describe the site upon which a new landfill facility may be located.

- 25. On page 2, you state, "data reported by others in various studies and sample results for our landfills in other states indicate PFAS will be detected in landfill leachate especially at such proposed conservative low detection limits."
 - *d. Please submit the studies you mention above and PFAS sampling data from your landfills in other states into the record.*

<u>ANSWER</u>: See Attachment C, Michigan Waste & Recycling Association Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent. March 1, 2019.

See Attachment D, North Carolina Collective Study Report, March 10, 2020.

e. In what states are your landfills located where PFAS were sampled? Do these states require monitoring of PFAS constituents?

ANSWER: Where sampling has occurred, it was generally upon the request of a POTW who is accepting leachate from a nearby landfill or by a request of the relevant regulatory agency. I am aware that limited sampling of leachate and/or groundwater monitoring for PFAS has occurred in New Hampshire, Vermont, Michigan, North Carolina, and California, but I am not aware that it has been mandated by state regulation.

f. Please clarify whether the PFAS data reported in various studies attribute the presence of PFAS to the waste generating the leachate or to leaching of PFAS from monitoring systems, sampling and/or laboratory equipment.

ANSWER: Most PFAS in leachate comes from source materials (i.e., waste) but most of that PFAS is sequestered in (remains in) landfills. Yet, some PFAS results from contamination in sampling. Various state sampling Guidance (e.g., Michigan) advise numerous protocols in sampling since PFAS can be introduced from monitoring and sampling equipment or procedures because these compounds are present in so many products and even can be present in rainfall.

See Attachment E, <u>WasteAdvantage</u> article, November 2, 2020.

26. On page 3 you state because of PFAS background conditions landfills would be required to perform multiple sampling events of upgradient wells and potentially all wells if intrawell statistical values are permitted. Please comment on whether this is the case for any contaminant that is detected in the landfill leachate not just PFAS.

ANSWER: Since PFAS compounds are so ubiquitous, and potentially in the well materials or sampling equipment, whether in the upgradient well or within the pumping mechanism, the upgradient well and pumping mechanism will have to be fully investigated – and that would not be true for other parameters where there is no concern related to PFAS in the well or sampling instrument itself.

- 27. Also on page 3, you state that dedicated sampling systems may include materials with PFAS that have nothing to do with impacts from the facility.
 - a. Please clarify whether PFAS detected in groundwater monitoring wells may be leaching from the sampling systems as opposed to coming from the waste.

<u>ANSWER</u>: Yes, see response 25.f. There are many sampling sources that might contribute to PFAS detections in monitored groundwater.

See Attachment F, Best Practices for Optimizing PFAS ANALYSIS, Shimadzu

See Attachment G, An Equipment Manufacturer's Perspective on Regulatory Guidance and Ambiguity on PFAS in Groundwater Sampling, QED Environmental Systems, Inc. 2020.

b. If so, what's the basis for your statement? Have there been any studies done to indicate that well monitoring systems contribute significant amounts of PFAS in relation to the amounts leaching from the waste disposed in the landfill? If there are, please submit them into the record.

<u>ANSWER</u>: What is a "significant" contributor in the context of PFAS monitoring remains to be seen, given the very stringent nature of the proposed and developing regulations. I am aware that studies are ongoing. See above answer for guidance offered by equipment manufacturers and laboratories.

- 28. On page 3, you repeat your concerns regarding contamination associated with lab or sampling equipment with respect to analysis of PFAS in landfill leachate.
 - a. Is it your position that any analysis of PFAS in leachate or groundwater samples would be suspect because of contamination from sampling or lab equipment?

ANSWER: See previous answers.

b. If so, do you have any alternatives for protecting groundwater from potential PFAS contamination from landfills?

ANSWER: Neither the proposed regulations nor the existing landfill regulations address cross contamination from laboratory or sampling or well sources. Those should be addressed.

Further, we believe the federally derived landfill regulations, which require synthetic liners and leachate collection systems, are effective in containing PFAS in landfills. Nonetheless, given the ubiquitous nature of PFAS, more research is required as to the actual potentiality of PFAS-related groundwater contamination from landfills or landfill leachate before the Board adopts such a stringent standard, based upon toxicology that is relevant to human (i.e., infant) consumption, that would then be used, as it has been used historically, to require strict compliance at sources that have no immediate link to human consumption. We certainly support strict standards for drinking water, but we would urge a better understanding of actual risks to human health and the environment prior to adopting such a strict standard as a general groundwater standard applicable to all potential sources throughout Illinois – making them immediately subject to enforcement for any detections above the limit.

One of the approaches taken by other states is to require screening sampling of PFAS at groundwater near landfills prior to determining if further screening or regulatory monitoring is required, consistent with potential receptors.

Another approach we would urge is that the Board delay applicability of the PFAS groundwater standard as an enforceable standard as to landfills until it has had an opportunity is to review the landfill regulations in a public hearing and make whatever revisions might be necessary to allow for a reasonable and environmentally protective, but economically reasonable and technically feasible, approach to these ubiquitous emerging contaminants as it relates to landfills. As is, the IEPA has not committed when or whether it will seek to amend the landfill regulations to address the waste industry's concerns.

29. Regarding ground water impact assessment (GIA) at landfills, you state that the current modeling requirements have the potential to be substantially affected and become unreasonably complicated. Please comment on whether the Board's Part 811 landfill regulations could be modified to accommodate concerns regarding application of GIA provisions to PFAS.

ANSWER: Yes, revising or removing the GIA provisions, and/or making them inapplicable to PFAS compounds, and potentially other compounds, would be a welcome change. My understanding is that the GIA provisions (not required in other states) were designed to predict whether a landfill will fail (i.e., leak). Yet, in my years of experience in Illinois there is no Subtitle D landfill designed cell that I've worked on that has leaked (i.e., caused groundwater contamination from leachate). Given that the IEPA's implementation of the GIA provisions requires the contaminant transport model to presume the most conservative input parameters (i.e., provide the highest predicted model concentration), the landfill industry has significant reason to believe that the GIA model will fail when inputting the PFAS compounds – and will not be an accurate predictor of PFAS contamination from leachate. A failed GIA model will halt development of new or expanded landfills – and may have other adverse permit repercussions.

The IEPA's current GIA implementation methods are designed to obtain absolute results, based upon overly conservative presumptions, which contribute to the complexity of the GIA. A passing or failing model may be determined by a concentration of one part per billion. Prior to implementation of the proposed rules as standards applicable in the Board's solid waste rules, a thorough evaluation of impacts to the GIA (35 IAC 811.317) should be conducted. By reference, impacts to the GIA will also affect the Assessment of Potential Groundwater Impact defined in 35 IAC 811.319(c) and Corrective Action Measures Assessment provided in Section 811.324.

- 30. On page 4 regarding treatment of landfill leachate at publicly owned treatment works (POTWs), you state that there is a significant risk that POTWs will begin to refuse landfill leachate due to concerns about PFAS.
 - a. Please comment on whether you are aware of any specific POTW in the states you operate that currently do not accept landfill leachate for treatment.

ANSWER: Bloomington-Normal Water Reclamation District has advised that it will discontinue accepting leachate from McLean County Landfill, effective January 1, 2023.

b. Are you aware of any state or federal PFAS surface water quality standards or NPDES (National Pollution Discharge Elimination System) permit limits that may cause POTWs to refuse acceptance of landfill leachate containing PFAS?

ANSWER: I understand that Michigan has begun to add PFAS compliance limits for certain POTWs during permit renewals.

- 31. On Page 5, you state that landfills monitor water bearing units that are not potable water sources and we believe that setting potential "drinking water limits", i.e., Class I limits, in these zones is not warranted.
 - a. Please clarify whether you are referring to "zone of attenuation" under Part 811.

ANSWER: The location of the water-bearing unit may or may not be within the zone of attenuation. Many of these water-bearing units are isolated and not functional for obtaining water for potable uses due to the limited extent of the zone, low hydraulic conductivity of the deposit, or limited quantities available. Since these are not viable sources of potable water, there is no reason to apply the Class I or Class II standards. Adjustment of the Class standard should be allowed for such units.

b. If so, are you aware that groundwater within the "zone of attenuation" is classified as Class IV under Part 620 where Class I standards will not apply.

<u>ANSWER</u>: Yes, I am aware that Section 620.240(a) states that groundwater within a zone of attenuation is Class IV – but that's not how these regulations work in practice. The Class IV standards for organic constituents (as PFAS is) will default to Class II, except for a few not relevant here. As to PFAS constituents, as with many other organic constituents, the Class II standards are equivalent to the Class I standards – and that's what's being proposed here.

c. If not, clarify whether you are referring to Class I groundwater outside the zone of attenuation that is currently not being used as a drinking water source.

ANSWER: See above answer, but also when evaluating risk to public health and environment, actual risks associated with whether the water will be used as a potable water source should be considered – and to not do so is to not properly evaluate economic reasonableness and technical feasibility.

CERTIFICATE OF SERVICE

I, the undersigned, certify that on this 23rd day of November 2022, I electronically served the **PRE-FILED ANSWERS OF ERIC BALLENGER ON BEHALF OF NATIONAL WASTE & RECYCLING ASSOCIATION** upon the individuals on the attached service list. I further certify that my email address is cmanning@bhslaw.com.

Dated: November 23, 2022

By /s/ Claire A. Manning

BROWN, HAY & STEPHENS, LLP

Claire A. Manning 205 S. Fifth Street, Suite 1000 P.O. Box 2459 Springfield, IL 62705-2459 (217) 544-8491 cmanning@bhslaw.com

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November 7, 2022

Submitted electronically to: https://www.regulations.gov

Ms. Michelle Schutz Office of Superfund Remediation and Technology Innovation (5202T) U.S. Environmental Protection Agency 1200 Pennsylvania Avenue, N.W. Washington, DC 20460

Re: Docket ID No. EPA-HQ-OLEM-2019-0341; Designation of Perfluorooctanoic Acid (PFOA) and Perfluorooctanesulfonic Acid (PFOS) as CERCLA Hazardous Substances

Dear Ms. Schutz:

The undersigned organizations—representing "passive receivers" of perfluorooctanoic acid (PFOA) and perfluorooctanesulfonic acid (PFOS) that may be present in drinking water, wastewater, and solid waste facility influent—are concerned that the U.S. Environmental Protection Agency's (EPA's) proposal to designate these compounds as hazardous substances under the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA), without accompanying relief, could result in significant increased costs for essential public service providers and the communities they serve while undercutting the Administration's broader human health and environmental protection goals.

Drinking water treatment plants, municipal wastewater treatment facilities, and solid waste landfills and composting facilities neither manufacture nor use per- and polyfluoroalkyl substance (PFAS); instead, they are passive receivers of media containing PFAS—compounds that are ubiquitous in the stream of commerce and environment. Each of these public services is interdependent; landfills rely on wastewater treatment facilities for their leachate discharge while water and wastewater treatment facilities depend on landfills for biosolids management and disposal of spent water filtration systems. Designating PFOA and PFOS as CERCLA hazardous substances would disrupt this interdependence by driving each sector to revisit its acceptance of influent streams containing concentrations of PFOA and PFOS.

CERCLA designation thus would lead to significant cost increases on public service providers and the communities they serve while impeding EPA's commitments espoused in the agency's PFAS Strategic Roadmap:

- There currently are no cost-effective techniques available to treat or remove PFOA or PFOS for the sheer volume of drinking water, wastewater, and landfill leachate managed daily by passive receiver facilities, as advanced treatment techniques at this scale are very costly. Undertaking additional treatment for PFOA and PFOS would add significantly to the costs of facility operation.
- Drinking water and wastewater facilities must manage media containing concentrations of PFOA and PFOS generated from influent treatment. The management of biosolids via incineration or land application, for example, is under increasing scrutiny in many states, and any additional disruption to available disposal outlets could result in additional cost increases for wastewater treatment.
- Passive receivers could be held liable for the entire cost of cleanup of a contaminated site, both on a prospective basis and for lawful activities going back decades. Regardless of EPA's use of enforcement discretion in initiating remedial actions, CERCLA designation would result in third-party contribution and cost recovery claims, likely leading to substantial litigation costs for public service providers and the communities they serve.
- These foreseeable cost increases, combined with actions taken by passive receivers to curtail acceptance of influent with concentrations of PFOA or PFOS, could impact the ability of some public service providers to continue operating, frustrate EPA cleanup activities around military installations and other affected communities, and disproportionately impact low-income communities that rely on the affordability of passive receiver services.

ATTACHMENT A

The undersigned organizations recommend that EPA, the Interagency Policy Committee on PFAS, and the broader Administration acknowledge the full unintended consequences of the proposed rule, evaluate all relevant authorities that could provide relief to passive receivers and the communities they serve, and reinstate the "polluter pays" principle of the statute in lieu of a "community pays" approach in which public service providers would be subject to CERCLA liability. Thank you for your consideration of our input, and we look forward to continuing to partner with EPA on actions to address PFAS under the PFAS Strategic Roadmap.

Sincerely,

Scott D. Grayson, CAE Chief Executive Officer American Public Works Association

alle DChoe

Matthew D. Chase Chief Executive Officer & Executive Director National Association of County Officials

Clarence E. Anthony Chief Executive Officer & Executive Director National League of Cities

Farell Z. Smith

Darrell K. Smith President & Chief Executive Officer National Waste & Recycling Association

Janine Burke-Wells Executive Director North East Biosolids & Residuals Association

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David Biderman Executive Director & Chief Executive Officer Solid Waste Association of North America

Frank Franciosi Executive Director U.S. Composting Council

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Tom Cochran Chief Executive Officer & Executive Director U.S. Conference of Mayors

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Gerard J. Neuser Chair Wisconsin Counties Solid Waste Management Association

ATTACHMENT A





November 7, 2022

Submitted electronically to: https://www.regulations.gov

Ms. Michelle Schutz Office of Superfund Remediation and Technology Innovation (5202T) U.S. Environmental Protection Agency 1200 Pennsylvania Avenue, N.W. Washington, DC 20460

Re: Docket ID No. EPA-HQ-OLEM-2019-0341; Designation of Perfluorooctanoic Acid (PFOA) and Perfluorooctanesulfonic Acid (PFOS) as CERCLA Hazardous Substances

Dear Ms. Schutz:

The National Waste & Recycling Association (NWRA) and Solid Waste Association of North America (SWANA) are pleased to submit comments on the U.S. Environmental Protection Agency's (EPA's) proposal to designate perfluorooctanoic acid (PFOA) and perfluorooctanesulfonic acid (PFOS) as hazardous substances under the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA). NWRA and SWANA represent companies, municipalities, and professionals in the solid waste industry. NWRA is a not-for-profit trade association representing private solid waste and recycling collection, processing, and management companies that operate in all fifty states. SWANA is a not-for-profit professional association in the solid waste management field with more than 10,000 members from both the private and public sectors across North America. Members of both organizations strive to deliver collection, composting, recycling, and disposal services that are protective of the environment in a safe, science-based, and technologically advanced manner.

NWRA and SWANA members are pleased that EPA has committed to numerous actions under the agency's PFAS Strategic Roadmap to safeguard public health, protect the environment, and hold accountable manufacturers and heavy users of these compounds. Our sector also supports EPA's focus on broadening and accelerating the cleanup of per- and polyfluoroalkyl substance (PFAS) contamination; nevertheless, we are concerned that designating PFOA and PFOS as CERCLA hazardous substances, without providing accompanying relief in recognition of the unique role served by the solid waste industry, would impede cleanup efforts and lead to substantial environmental cleanup liability, impose significant additional costs on essential public services and their customers, and have broad repercussions throughout the economy, without any measurable environmental benefit. We therefore request that EPA consider these comments in ensuring that the rulemaking adheres to the "polluter pays" principle of CERCLA.

ATTACHMENT B

I. Modern Landfills are Effective Solutions to Manage Wastes Containing PFAS.

Modern landfills are essential public services¹ that are subject to extensive and evolving federal, state, and local environmental, health, and safety requirements, including the Resource Conservation and Recovery Act (RCRA), the Clean Air Act, and the Clean Water Act. Regulations established under Subtitle D of RCRA establish minimum federal criteria for the operation of municipal solid waste, industrial waste, and special waste landfills, including design criteria, location restrictions, financial assurance, strict environmental monitoring, corrective action protocols (if triggered), and closure and post-closure periods to ensure facilities will not be a threat to human health and the environment. Similarly, Subtitle C of RCRA and its accompanying regulations govern the permanent disposal of hazardous wastes, and these facilities employ even greater environmental controls, which can include double liner systems, waste immobilization techniques, advanced leachate collection systems, extensive groundwater monitoring systems, offsite discharge mitigation protocols, leak detection systems, and enclosed and controlled offload areas. Both Subtitle C and Subtitle D landfills are highly regulated by permit(s) at the state level, as they typically are subjected to additional monitoring obligations as well as construction and operational requirements that go beyond the federal framework.

As a result of the stringent environmental controls required by federal and state regulation, and in recognition of our role as stewards of the environment, our industry has made significant investments to ensure that landfills are designed, constructed, and operated to reduce their environmental impact. For these reasons, EPA recognized in its *Interim Guidance on the Destruction and Disposal of PFAS and Materials Containing PFAS* that disposal of PFAS-contaminated wastes at hazardous or solid waste landfills can be effective options for managing PFAS by sequestering these compounds and preventing society from being re-exposed.²

II. The Proposed Rule would Replace CERCLA's "Polluter Pays" Principle with a "Community Pays" Model, Imposing Significant Costs on Landfill Customers and Ratepayers.

It is important for EPA to recognize that landfills neither manufacture nor use PFAS; instead, they are <u>passive receivers</u> of materials containing PFAS—compounds that are ubiquitous in residential and commercial waste streams—that must be managed once discarded. Research has shown that landfills effectively sequester a high percentage of PFAS compounds, especially longer-chain compounds such as PFOA and PFOS.³ As rain percolates through landfills, the liquid will pick up some contaminants including a small amount of PFAS compounds not sequestered in the landfill environment. The resultant liquid is called leachate. Landfills are legally required to remove leachate from landfill collection systems and to properly manage this wastewater in order to protect groundwater resources. These management techniques can include onsite management, treatment prior to disposition or discharge, or collection and transport to wastewater treatment facilities. All of these activities are subject to regulatory permitting and oversight.

Despite the stringent management processes currently followed by our industry, a designation of PFOA and PFOS as CERCLA hazardous substances virtually guarantees that private parties—manufacturers of these compounds and other parties responsible for site contamination—will bring CERCLA claims for contribution

https://www.epa.gov/system/files/documents/2021-11/epa-hq-olem-2020-0527-0002_content.pdf.

¹ See Guidance on the Essential Critical Infrastructure Workforce: Ensuring Community and National Resilience in COVID-19 Response, V. 4.0, Cyber Security & INFRASTRUCTURE SECURITY AGENCY (Aug. 18, 2020).

² See Interim Guidance on the Destruction and Disposal of Perfluoroalkyl and Polyfluoroalkyl Substances and Materials Containing Perfluoroalkyl and Polyfluoroalkyl Substances, U.S. ENVT'L PROT. AGENCY (Dec. 18, 2020), at

³ See, e.g., PFAS Waste Source Testing Report, SANBORN, HEAD & ASSOCIATES, INC. (Oct. 2019), at

https://anrweb.vt.gov/PubDocs/DEC/SolidWaste/OL510/OL510%202019.10.15%20NEWSVT%20PFAS%20Source%20Testing %20Rpt%20-%20Final.pdf.

against landfills and other essential public service providers such as water and wastewater utilities that are also passive receivers of PFAS. Given that CERCLA imposes joint, several, and retroactive environmental cleanup liability to parties connected with the presence of a hazardous substance at a site, designating PFOA and PFOS as hazardous substances will, at a minimum, generate significant litigation costs for lawful PFAS-containing waste disposal and discharges going back decades.

This type of inequitable outcome has occurred in previous CERCLA matters. As an example, industrial parties determined to be responsible under CERCLA for the cleanup of the Passaic River in New Jersey brought contribution actions against 261 third-party defendants—including 70 municipalities and other public entities— contending that they bore site cleanup responsibility. This action resulted in litigation spanning eight years and culminating in a payment of \$35.4 million by these minor parties, many of whom were merely passive receivers of the contamination at issue.

Extensive litigation costs, as well as potential significant costs relating to PFAS remediation, would be passed along to communities, drinking water and wastewater treatment facilities, and the biosolids management sector—all of which rely on landfills for disposal of media containing PFAS. These cost increases, as well as similar cost increases passed through to drinking water and wastewater treatment ratepayers, likely would have a significant and disproportionate impact on low-income households that rely on the affordability of services that the waste sector and other passive receivers provide.

III. PFAS Treatment and Residuals Management Will Increase Costs to Communities but Will Not Reduce CERCLA Liability.

It has been suggested that the industry could simply treat leachate to eliminate any PFAS prior to discharging to wastewater treatment plants in order to reduce potential CERCLA liability presented by the proposed rulemaking. This premise is flawed for several reasons. Firstly, implementing treatment methods in the present day and into the future does not address potential liabilities for contribution actions that may be brought for cleanups stemming from prior POTW discharges.

Secondly, this premise does not recognize the current limitations of PFAS treatment technologies and their associated uncertainties and costs. Our industry is at the forefront of developing technologies for PFAS treatment and residuals management, however technologies for PFAS removal from leachate at scale are still developing and require a multi-step process that includes (1) pretreatment of leachate to address non-PFAS constituents, (2) subsequent PFAS treatment using one or more removal technologies (which creates PFAS-containing residuals), and (3) PFAS residuals treatment/management. Since most landfills rely on wastewater treatment plants for their leachate discharge, undertaking leachate pretreatment followed by PFAS treatment will add significantly to the costs of landfill operation.⁴ The estimated capital cost to implement leachate pretreatment and PFAS treatment at a moderate-sized landfill (i.e., biological treatment of 30,000-40,000 gallons per day of leachate) to the extent necessary to minimize PFAS in leachate ranges from \$2 million to \$12 million, or potentially far more.⁵ An additional layer of potential CERCLA liability could drive up these costs significantly and would ultimately be borne by the communities that rely on economical solid waste management services instead

⁴ These costs will be driven, in part, by potential future regulation under the Safe Drinking Water Act, Clean Water Act, and other federal and state authorities.

⁵ The standards that would govern a PFOA or PFOS cleanup action currently are unclear, complicated by a patchwork of state regulatory standards, unknown criteria that would be required for remedial actions, and EPA's interim drinking water health advisories for PFOA and PFOS. As such, the costs of PFAS treatment borne by landfills and their customers could far exceed these estimates.

of PFAS producers and manufacturers.

Moreover, since current technologies are unable to completely destroy PFAS, further management of residual PFAS waste streams—including biosolids and spent filters—is necessary to stabilize or otherwise limit their ability to reenter leachate. The costs and operational effectiveness for PFAS residuals management is less understood as most technologies have not been evaluated at full-scale. Based on general conversations with technology developers and estimates/extrapolations from small-scale studies, however, we anticipate that implementing new technologies for PFAS removal and subsequent residuals management could increase the costs of treating landfill leachate by approximately \$0.06 to \$0.39 (potentially even higher) per gallon of raw leachate processed (i.e., a cost increase of at least 400% to 800%). Increased costs associated with PFAS management thus could total approximately \$966 million to \$8.187 billion per year for municipal solid waste landfills alone. These costs typically cannot be absorbed by local governments with municipally operated landfills.

IV. The Mere Prospect of Designating PFOA and PFOS as CERCLA Hazardous Substances Already is Disrupting the Interdependence of Drinking Water and Wastewater Treatment Facilities, Biosolids Management, and Landfill Operations—and Could Have Much Broader Unintended Consequences on Administration Priorities.

Wastewater treatment facilities generate biosolids as a byproduct of their treatment activities. Similarly, drinking water treatment facilities generate spent filter materials from their operations. Expectedly, these biosolids and spent filter media may contain some amount of PFAS removed from the final treated wastewater and drinking water. Wastewater treatment facilities rely on landfills for biosolids management and drinking water treatment facilities for disposal of filter materials that may contain PFAS. At present, there are three viable options for management of biosolids: incineration, land application, and landfilling. At a time when incineration and land application are increasingly being prohibited, any further disruption to biosolids management could have a tremendous impact on municipal budgets and the environment.

Designating PFOA and PFOS as hazardous substances under CERCLA would impel landfill operators to revisit their waste acceptance criteria, likely choosing to limit inbound wastes with known elevated concentrations of PFAS—including filter materials, biosolids, and impacted soils—and/or increase disposal costs for certain media. Indeed, the mere prospect of a CERCLA designation has begun to disrupt the interdependence of the drinking water, wastewater, and solid waste sectors, as wastewater treatment facilities have begun to prohibit the acceptance of leachate while landfills are considering similar restrictions on the acceptance of biosolids and other PFAS-containing materials.

Regulation of PFOA and PFOS under CERCLA also could inadvertently undercut the Administration's broader environmental goals. The increased costs associated with disposal that are attributable to the rulemaking could incentivize bad actors to seek alternative means of disposal of PFAS-contaminated media and remediation wastes that are less protective of public health and the environment. Landfill operators choosing to limit specific inbound streams of waste containing elevated levels of PFAS also could curtail the ability of some wastewater treatment facilities to continue operating and <u>frustrate EPA and DOD cleanup activities around military installations</u> and other affected communities.

Moreover, EPA's action could lead to decreased composting services nationwide. Food waste compost may contain PFAS due to contact with PFAS-lined packaging materials. As a result, a CERCLA designation could result in communities diverting food waste from organics recycling programs, hindering federal, state, and local climate and waste reduction goals. Finally, and as mentioned above, the increased costs on ratepayers that are

ATTACHMENT B

attributable to the proposed rule likely will have disproportionate adverse impacts on low-income communities and frustrate the Administration's broader policies around environmental justice.

V. Recommendations

The solid waste sector and the communities we serve should not be held financially or legally liable under CERCLA for PFAS contamination, as landfills are only passive receivers of PFAS and are part of the long-term solution to manage these compounds. In its proposed designation, EPA announced that it "will use enforcement discretion and other approaches to ensure fairness for minor parties who may have been inadvertently impacted."⁶ We greatly appreciate EPA's apparent willingness to exercise its discretion to foster equitable outcomes in direct enforcement matters; however, our industry remains concerned that <u>this assurance would not sufficiently insulate landfills from third-party contribution litigation as discussed above</u>. Accordingly, we suggest that concrete liability protections should be implemented in conjunction with this proposed rulemaking and respectfully request that EPA and the Interagency Policy Committee on PFAS. *See, e.g.,* 42 U.S.C. §§ 9602(a) and 9615 (providing flexibility in the promulgation of regulations under CERCLA).

In the event EPA opines that it has limited authority to provide the solid waste sector with relief from third-party contribution litigation, the Administration should work with Congress to support a narrow legislative exemption from CERCLA liability in cases where a landfill discharges leachate in compliance with all applicable laws and regulations. Doing so would keep CERCLA liability on the industries that created and profited from these PFAS compounds — not on taxpayers.

Thank you for your consideration of our comments, and we look forward to continuing to partner with EPA to ensure the safe and effective management of waste streams containing PFAS. Should you have any questions about this letter, please contact Anne Germain, COO & SVP of Regulatory Affairs for NWRA, at agermain@wasterecycling.org. You may also contact Jesse Maxwell, Senior Manager, Advocacy & Safety for SWANA, at jmaxwell@swana.org.

Very truly yours,

Fanell Z. Smith

Darrell K. Smith President & CEO National Waste & Recycling Association

Deen Balen

David Biderman Executive Director & CEO Solid Waste Association of North America

ATTACHMENT B

⁶ EPA Proposes Designating Certain PFAS Chemicals as Hazardous Substances Under Superfund to Protect People's Health, U.S. ENVT'L PROT. AGENCY (Aug. 26, 2022), at https://www.epa.gov/newsreleases/epa-proposes-designating-certain-pfas-chemicals-hazardous-substances-under-superfund.

⁷ We request that the interagency committee broaden its scope when considering CERCLA liability concerns caused by the use of PFAS-containing firefighting foams at airports to include similar concerns from the waste sector. Just as certain airports are required by law to use firefighting foam containing PFAS, permitting authorities often require landfills to accept waste streams containing PFAS.

Michigan Waste & Recycling Association Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent

TECHNICAL REPORT

Completed in Collaboration with Michigan Department of Environmental Quality March 1, 2019 (Second Revision March 6, 2019)



1.0 INTRODUCTION & OVERVIEW

This report summarizes the results of a statewide study completed on behalf of the Michigan Waste & Recycling Association (MWRA) to determine levels of PFOA and PFOS in the leachate of those landfills participating in the study, and to estimate the leachate's relative contribution to the total amount found in wastewater influent at water resource recovery facilities (WRRFs) (aka POTWs or publicly owned treatment works, or sewage or wastewater treatment plants). The study involved testing leachate at 32 active municipal solid waste landfills (Type II landfills) located throughout the state. This report presents general background information on PFAS, summarizes testing results, and summarizes available PFAS information from WRRFs that receive leachate and those that do not.

PFOA and PFOS are two compounds in a class of compounds known as Per- and polyfluoroalkyl substances (PFAS). They have been used for over 50 years in household products such as non-stick coatings in cookware, in stain and water-resistant coatings and fabrics, and in industrial products such as firefighting foam. More recently, certain PFAS compounds were identified as having potentially adverse effects on human health and the environment. In general, PFAS compounds are resistant to natural degradation, and can therefore persist in the environment for a long time.

Each solid waste landfill in the study is licensed by the State of Michigan to accept household, commercial, and industrial solid waste generated by the communities they serve. Some of the wastes received for disposal contain PFAS. Leachate is the liquid that occurs in landfills when rainwater combines with moisture contained within the waste. Chemicals present in the waste may be present in the leachate. The leachate is effectively captured by utilizing engineered liner and active liquid collection systems. A common method of leachate management is through discharge to a local WRRF where it is handled with other household, commercial, and various industrial wastewaters. In this way, leachate is managed in a closed system where there is no direct exposure to the public.

Landfill leachate sent to a WRRF is typically directly discharged via pipeline or stored in onsite tanks prior to being transferred to tanker trucks and hauled to the treatment facility. WRRFs are engineered structures that apply various technologies to treat wastewater to meet certain regulatory criteria prior to discharge of these waters.

In 2018, the Michigan Department of Environmental Quality (MDEQ) and various WRRFs requested that landfills test for PFAS in leachate as part of a statewide effort to better understand the presence of PFAS in the environment and to work toward plans for PFAS reduction, where needed. The information was also useful to examine the interdependent cycle of waste disposal, leachate generation, wastewater treatment, and wastewater sludge disposal.

Rather than participating landfills sampling and reporting individually, the MWRA (with MDEQ concurrence) conducted a collective study involving 32 active municipal solid waste landfills (Type II landfills) located throughout the state. This effort represents one of the largest studies conducted on active landfill leachate to-date. The main objective of the study was to gather information on PFOA and PFOS concentration in leachate at individual landfills and to examine its potential significance to WRRF influent across the state.

NTH Consultants, Ltd, (NTH), a Michigan-based professional environmental and engineering consulting firm, conducted the MWRA study. NTH prepared this technical report that provides testing results for individual landfills, details of the sampling and analysis procedures, characteristic leachate discharge volumes, and available flow and PFAS testing information from the potentiallyaffected WRRFs.

Electronic Filing: Received, Clerk's Office 11/23/2022 2.0 REGULATORY STATUS AND GLOBAL LANDFILL LEACHATE CONCENTRATIONS

2.1 Status Of Regulatory Action In Michigan

Information on various adverse health effects associated with certain PFAS compounds has been evolving since the early 2000's. Two of the most widely-utilized PFAS compounds, PFOA and PFOS, have received early environmental regulatory focus. These and related compounds have been used in thousands of applications worldwide. Largely for these reasons, the manufacture of PFOA and PFOS has been voluntarily phased-out in the United States.

In response to concerns regarding the increasingly common detection of PFAS in the environment, the Michigan PFAS Action Response Team (MPART) was formed by an Executive Directive issued by then-Governor Snyder in November 2017. MPART, a multiagency group, is comprised of a team of local, state, and federal agencies that are working to understand the exposure risks and ways to mitigate PFAS impacts to the environment.

MPART emphasizes the need for cooperation and coordination among agencies at all levels of government charged with identifying PFAS contaminants, informing the public, and mitigating the potential effects.

The EPA established a drinking water health advisory (HA) for PFOA and PFOS of 70 ppt in 2016. Although the HA is not an enforceable drinking water standard, it was established as a protective guidance for the most sensitive subpopulations over a lifetime of exposure. In January of 2018, the MDEQ incorporated the information contained in the HA and established the same 70 ppt value as groundwater cleanup criteria under Part 201, Environmental Remediation, of the Natural Resources and Environmental Protection Act, 1994 P.A. 451, as amended (Act 451). Currently, this value is used by the Michigan Department of Health and Human Services (DHHS) as guidance when evaluating PFAS concentrations in public and private drinking water supplies.

The MDEQ also promulgated Water Quality Standards (WQS) for PFOA and PFOS in surface water in May 2011 and March 2014, respectively. These WQS values were developed for use by MDEQ when evaluating permits for discharge to surface water and were promulgated in

accordance with the Part 4 Rule 57 administrative rules (Rule 57) pursuant to Water Resources Protection (Part 31) of Act 451. Michigan's WQS values include chemical-specific values that represent the water quality values protective of aquatic life, human health, or wildlife; and acute chemical-specific values protective of aquatic life. The applicable most restrictive WQS values developed by the State are listed in below in Table 2-1, Rule 57 Values.

Table 2-1 – Rule 57 Values

Chemicals	HNV (non-drinking water*)	HNV (drinking water**)
PFOS	12 ppt	11 ppt
PFOA	12,000 ppt	420 ppt

HNV: Human Non-cancer Value

ppt: parts per trillion (laboratory reports in nanograms per liter (ng/L)

* "non-drinking water" means the surface water body receiving the discharge is not designated as a public drinking water source

** "drinking water" means the surface water body receiving the discharge is used as a public drinking water source

Other states have or are considering establishing regulatory limits for PFAS compounds. The variability in existing values between states is generally attributable to differences in the selection and interpretation of the choice of uncertainty factors, and the approach used for animal-to-human extrapolation mostly using the same key toxicity data. Differences in values between regulatory agencies may also be due to the choice of exposure assumptions, including the amount of water consumed, life stage used, and the relative source contributions (percentage exposure assumed to come from non-drinking water sources). All of this contributes to the overall uncertainty across the US in how to most appropriately establish risk-based criteria for these compounds and more consistency is needed in this important area.

2.2 Literature Summary Of PFOA & PFOS Concentrations In Landfill Leachate

To provide a basis for comparison of the results of the MRWA landfill leachate study, NTH completed a review of current literature regarding PFOA and PFOS concentrations in landfill leachate. Sources include professional journals, regulatory documents, and government agency websites. A summary of the information we reviewed is presented below.

2.2.1 Worldwide PFOA and PFOS

Literature review focused on documents published over the past 15 years. Two recent and comprehensive publications regarding PFAS concentrations in leachate includes a worldwide perspective by Hamid, et al (2018) and its associated multiple references, and the USfocused paper by Lang, et al (2017).

Unlike Hamid, et al (2018), Lang, et al (2017) focused on an evaluation of climatic effects on leachate PFAS concentrations and associated mass loading to municipal wastewater treatment plants located in the US. This study, which included 87 samples from 18 landfills, representing one of the largest databases of any similar investigation to date, demonstrates PFOA and PFOS concentrations in leachate generally have been decreasing over time, with greater rates of decline in humid regions (i.e., precipitation greater than 75 cm/year), which is where landfills that contain nearly half the annual volume of solid waste disposed in the US are located.

Hamid, et al (2018) compiled data from 11 selected literature sources, published between 2004 to 2017, that include PFAS leachate concentrations from landfills located in Australia, Canada, China, Denmark, Germany, Norway, Spain , Sweden, and the USA . Together, these sources comprise dozens of landfills with a total of more than 162 leachate samples. To summarize the PFOA and PFOS leachate results from these various studies, we prepared Table 2-2, Study of Literature Study derived from Hamid, et al.'s database (Supplemental Information Table 1) and information from the Lang (2017) et al. study. This information is graphically depicted on Figure 2-1, PFOA & PFOS Concentration in Landfill Leachate (Worldwide – Separate Studies).

Figure 2-2, PFOA & PFOS Concentrations in Landfill Leachate (By Region) summarizes the PFOA and PFOS ranges observed in each of the world regions. As shown, PFOA and PFOS concentrations in landfill leachate vary considerably in different regions of the world and likely reflect the nature of the consumer products and industrial materials used, produced, and disposed in each country. The age of waste materials, as well as climatic conditions to which landfills are subject, appear important factors that govern the rate of degradation of PFAS materials to PFOA and PFOS, both considered "terminal" products of precursor compounds.

In summary, the preceding information reveals a wide range of leachate PFOA and PFOS concentrations worldwide including the United States. China's values are much higher than elsewhere in the world, likely a result of their continued production of consumer goods (as well as industrial waste associated with related manufacturing processes) with PFAS compounds. These products are then distributed throughout the world for purchase, including in the US and eventually disposed.

Figure 2-1 **PFOA & PFOS Concentrations in Landfill Leachate** (Worldwide - Separate Studies)

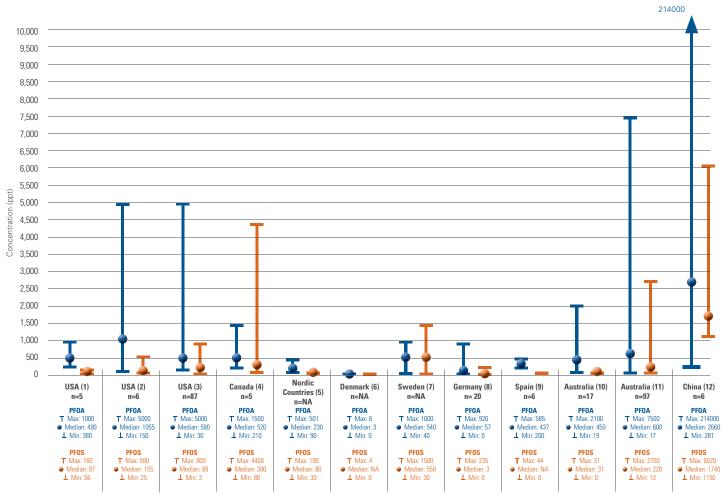
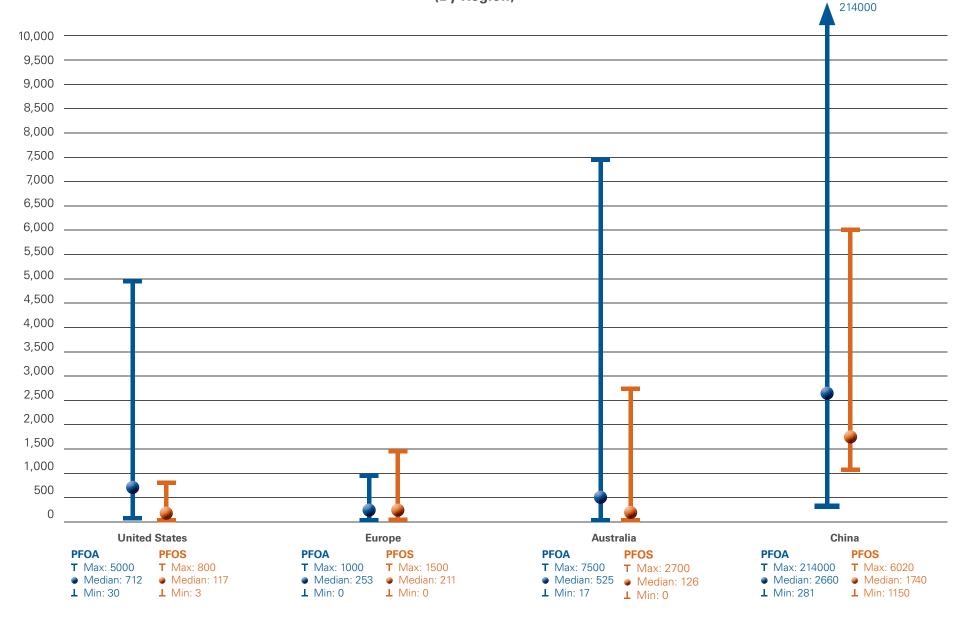


Table 2.2: Summary of Literature Study - PFOA & PFOS Concentrations in Landfill Leachate

Source Cited	Location/ Region	Sample Size		PFOA		PFOS		
		Detection Frequency %	Concentration Range (ng/l)	Median (ng/l)	Detection Frequency %	Concentration Range (ng/l)	Median (ng/l)	
1. Huset, et al (2011)	USA	5	100	380 - 1,000	490	100	56 -160	97
2. Allred, et al (2015)	USA	6	100	150 - 5,000	1,055	100	25 - 590	155
3. Lang, et al (2017)	USA	87	100	30 - 5,000	590	96	3-800	99
4. Benskin, et al (2012)	Canada	5	100	210 - 1,500	520	100	80 - 4,400	390
5. Kallenborn, et al (2004)	Nordic Countries	NA	NA	90-501	230	NA	30 - 190	80
6. Bossi, et al (2008)	Denmark	NA	NA	0 - 6	3	NA	0 - 4	NA
7. Woldegiorgis, et al (2008)	Sweden	NA	NA	40 - 1,000	540	NA	30 - 1,500	550
8. Busch, et al (2010)	Germany	20	95	0 - 926	57	100	0 - 235	3
9. Fuertes, et al (2017)	Spain	6	100	200 - 585	437	17	0 - 44	NA
10. Gullen, et al (2016)	Australia	17	100	19 - 2,100	450	89	0 - 100	31
11. Gullen, et al (2017)	Australia	97	64	17 - 7,500	600	65	13 - 2,700	220
12. Yan, et al (2015)	China	6	100	281 - 214,000	2,260	100	1,150 - 6,020	1,740

Figure 2-2 PFOA & PFOS Concentrations in Landfill Leachate (By Region)



Concentration (ppt)

ATTACHMENT C

3.0 LEACHATE SAMPLING PROGRAM

This section includes information regarding the statewide PFAS sampling program participants, along with sample collection methods and analytical techniques. The sampling program included 32 sites located in the Lower and Upper Peninsulas of Michigan, as shown on the attached Figure 3-1, Site Location Map. Each site is an active, Type II, municipal solid waste landfill. As explained later in this report, we included three additional landfills with leachate data available for comparison as part of our overall evaluation. The locations of these three disposal facilities (i.e., City of Riverview Landfill, South Kent County Landfill, and Smiths Creek Landfill) are also shown on Figure 3-1.

3.1 Field Methods

3.1.1 Site Sampling Planning & Coordination

NTH working with Test America Laboratories (TAL) sampled leachate at the 32 MWRA-member landfills over a period of 14 days, beginning on Monday, November 19, 2018, and concluding on Wednesday, December 12, 2018. NTH accompanied TA staff during the first 5 days of sampling to verify TAL followed MDEQ-recommended sampling methods and protocol in the guidance documents referenced below.

NTH contacted each of the 32 participating facilities and requested information including site contacts, leachate system discharge configuration, access limitations, specialized site requirements, pretreatment installations, leachate discharge volume, and receiving WRRF locations. The relevant information from the sites is summarized on Table 3-1, Landfill Leachate Discharge Information.

Additionally, NTH prepared and distributed a sampling schedule based on logistical groupings to maximize efficiency and coordinate acceptable sampling times at each site. NTH remained in contact with TAL to maintain the established schedule according to sitespecific approvals. NTH provided TAL the compiled site information for use as a guide during the sampling to help streamline and prepare for the field work.

3.1.2 Sampling Collection Overview

Experienced TAL field staff completed leachate sampling with oversight by Mr. Michael McNamara (NTH) during

the first 5 sampling days. Mr. McNamara previously completed PFAS sampling training conducted by the MDEQ in April 2018. The MDEQ training included fieldsampling of leachate and groundwater along with the collection equipment blanks using laboratory-supplied PFAS-free water (LSPFW). MDEQ has issued a number of draft guidance documents for PFAS sample collection, including:

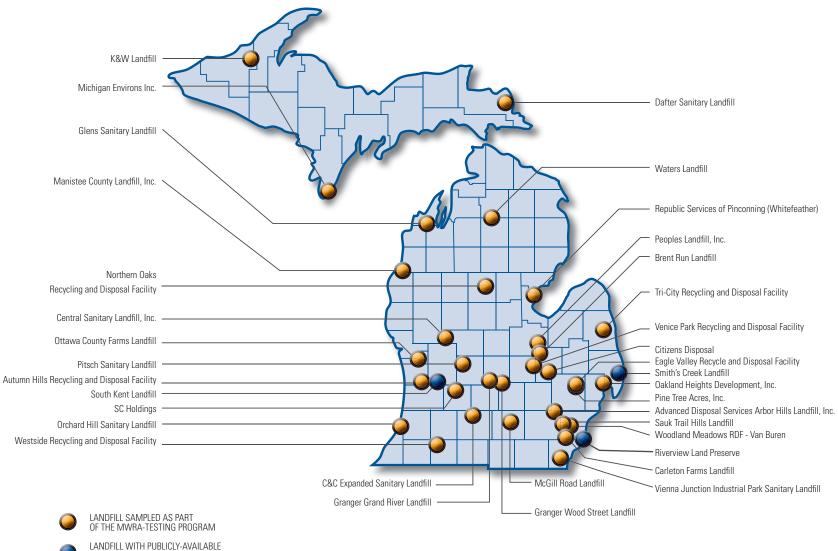
- "Standard Operating Procedure Collection of Landfill Leachate Samples for Analysis of Polyfluorinated Alkyl Substances (draft)," dated April 2018,
- "Wastewater PFAS Sampling Guidance," dated October 2018, and
- "General PFAS Sampling Checklist," dated October 2018.

Both NTH and TAL reviewed and followed these documents during sampling activities. To maintain consistency and uniformity with the program sampling, TAL dedicated two experienced representatives (Gary Schafer and Zachary Nelson) to this project, who remained involved for the duration of the entire 32-site program, as indicated in Table 3-1. During the first five days of sampling, which included 14 of the 32 sites, NTH accompanied the designated TAL sampling crew and verified that TAL followed the MDEQ PFAS-sampling protocols. A summary of the sampling procedures is included in Appendix A, Sampling and Testing Methods.

3.1.3 Sample Analysis

Consistent with MWRA's agreement with MDEQ, the sample analysis for this study included PFOA and PFOS using EPA Method 537 (modified). This was done to focus the study on the two compounds with Michigan Part 201 and Rule 57 standards. TA analyzed the samples at their Sacramento laboratory following their US EPA Method 537 (modified) standard operating procedures (SOPs).

Figure 3-1



LANDFILL WITH PUBLICLY-AVAILABLE PFOA AND PFOS DATA AVAILABLE

ATTACHMENT C

Electronic Filing: Received, Clerk's Office 11/23/2022 Table 3-1

Landfill Leachate Generation & Disposal Methods

		LEACHATE	DISCHARGE INFORMAT	ION
MWRA-Member Landfill Designation	Leachate Treatment Facility	Discharge Configuration	Pretreatment	Approximate Daily Dispos- al Volume at WRRF (Gallons)
	Discharge to Sanitary Sewer			
ADVANCED DISPOSAL SERVICES ARBOR HILLS LANDFILL INC	Ypsilanti Community Utilities Authority (YCUA) Pump and Haul to CWT eventually discharges to GLWA (~38,000 gpd)	Manhole to Sewer	N/A	60,400
BRENT RUN LANDFILL	Anthony Ragnone WWTP (Genesee County)	Manhole to Sewer	N/A	16,400
CITIZENS DISPOSAL	Anthony Ragnone WWTP (Genesee County)	Manhole to Sewer	N/A	32,900
EAGLE VALLEY RECYCLE & DISPOSAL FACILITY	Great Lakes Water Authroity WRRF (GLWA)	Forcemain to Sewer	N/A	32,900
GRANGER GRAND RIVER LANDFILL	Southern Clinton County Utilities Authority (SCCMUA)	Manhole to Sewer	N/A	64,400
GRANGER WOOD STREET LANDFILL	City of Lansing WWTP (Lansing)	Manhole to Sewer	N/A	19,200
OAKLAND HEIGHTS DEVELOPMENT INC	Clinton River Water Resource Recovery Facility in Pontiac (CRWRRF)	Manhole to Sewer	N/A	17,800
PINE TREE ACRES INC	Great Lakes Water Authroity WRRF (GLWA)	Manhole to Sewer	N/A	74,000
SAUK TRAIL HILLS LANDFILL	Ypsilanti Community Utilities Authority (YCUA)	Manhole to Sewer	N/A	20,500
SC HOLDINGS	City of Hastings WWTP (Hastings)	Direct Discharge	Ammonia Treatment	16,000
VENICE PARK RECYCLING & DISPOSAL FACILITY	Anthony Ragnone WWTP (Genesee County)	Two Manholes to Sewer	N/A	32,900
WESTSIDE RECYCLING & DISPOSAL FACILITY	City of Three Rivers WWTP (Three Rivers)	Direct Discharge	N/A	60,800
WOODLAND MEADOWS RDF-VAN BUREN	Great Lakes Water Authroity WRRF (GLWA)	Manhole to Sewer	N/A	54,800
	Pump and Haul to WRRF	1		1
AUTUMN HILLS RECYCLING AND DISPOSAL				
FACILITY	City of Grand Rapids WWTP (Grand Rapids)	Loadout Pad	N/A	54,800
DAFTER SANITARY LANDFILL	City of Sault Ste. Marie WWTP (Sault St. Marie)	Loadout Pad	N/A	16,500
GLENS SANITARY LANDFILL	Betsie Lake Utility Authority (BLUA)	Loadout Pad	Site Evaporator	3,800
K & W LANDFILL	Portage Lake Water and Sewage Authority's WWTF (Portage Lake) Iron-Gogebic Wastewater Authority's Treatment Facility (Ironwood)	Loadout Pad	N/A	17,500
	City of Ludington WWTP (Ludington) (approx 4,700 gpd)	Loadout Pad	N/A	
MANISTEE COUNTY LANDFILL INC	Packaging Corporation of America (PCA) - approx 30,000 gpd	Loadout Pad	N/A	4,700
MICHIGAN ENVIRONS INC	City of Menominee WWTF (Menominee)	Loadout Pad	N/A	13,100
PITSCH SANITARY LANDFILL	Belding WRRF (Belding), with Grand Rapids as a backup	Loadout Pad	N/A	15,000
TRI-CITY RECYCLING AND DISPOSAL FACILITY	City of Sandusky WWTP (Sandusky)	Loadout Pad	N/A	9,600
	Pump and Haul to Centralized Waste Treatment			
ADVANCED DISPOSAL SERVICES ARBOR HILLS LANDFILL INC	YCUA (60,400 gpd) Pump and Haul to CWT eventually discharges to GLWA	Loadout Pad	N/A	38,000
C & C EXPANDED SANITARY LANDFILL	Dart/Clean Earth in Detroit (DART) - GLWA	Loadout Pad	N/A	42,000
CARLETON FARMS LANDFILL	Dart/Clean Earth in Detroit (DART) - GLWA	Loadout Pad	N/A	123,300
CENTRAL SANITARY LANDFILL INC	SET Environmental Inc - Grand Rapids	Loadout Pad	N/A	30,100
MCGILL ROAD LANDFILL	Usher Oil (Detroit) (Usher) - GLWA	Loadout Pad	N/A	13,700
NORTHERN OAKS RECYCLING AND DISPOSAL FACILITY	Plummer's Environmental Services - Wyoming, MI (Plummer's)	Loadout Pad	Site Evaporator	12,300
ORCHARD HILL SANITARY LANDFILL	Third Party Pretreatment Facility in Holland, MI - Holland WRRF"	Loadout Pad	Reverse Osmosis	12,500
OTTAWA COUNTY FARMS LANDFILL	SET Environmental Inc - Grand Rapids	Loadout Pad	N/A	82,200
PEOPLES LANDFILL INC	Usher - GLWA	Loadout Pad	N/A	21,900
VIENNA JUNCTION INDUSTRIAL PARK SANITARY LANDFILL	Half to City of Toledo - Toledo (Out of state so not included in total) Half to Usher in Romulus, MI - GLWA	Loadout Pad	N/A	13,700
	Pump and Haul to Deep Injection Well for Disposal	1	19/75	10,700
WHITEFEATHER LANDFILL	Deep Injection Well In Pinconning -approx 12,600 gpd	Loadout Pad	N/A	Deep Well Disposal - No offsite leach- ate disposal
WATERS LANDFILL	Northeastern Exploration (Deep Well) in Johannesburg, MI-approx 8,200 gpd	Loadout Pad	Site Evaporator	Deep Well Disposal - No offsite leachate disposal

ATTACHMENT C

3.2 Leachate Disposal Methods, Daily Leachate Volume, & Receiving WRRFs

In this section, we present details regarding leachate disposal methods, annual leachate volumes, and the water resource recovery facilities (WRRFs) that treat leachate generated by the participating landfills, including relevant summary tables and graphics.

3.2.1 Disposal Methods

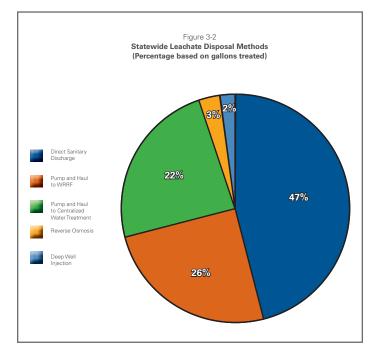
We obtained disposal information from a pre-sampling questionnaire completed by each facility owner representative. Based on the compiled data included in Table 3-1, the participating landfills manage leachate either by direct sanitary sewer discharge (DSD); pumpand-haul (PAH) for discharge; deep well injection (DWI); or a combination of these three methods. One site, Orchard Hill Landfill, primarily treats leachate for direct discharge to surface water using a reverse-osmosis (RO) system or whenever necessary, manages leachate by PAH. Figure 3-2, Statewide Leachate Disposal Methods illustrates the percentage by leachate volume of each disposal method utilized by the participating landfills.

3.2.2 Daily Leachate Volumes

Each site representative accessed their respective site Operating Records that include leachate flow measurements. The average daily leachate volumes by site, are included on Table 3-1. As indicated on Table 3-1 and graphed on Figure 3-3, Average Daily Leachate Volume Managed at Michigan WRRFs, the leachate volume discharged to WRRFs varies, ranging from approximately 3,800 gallons per day (gpd) at Glen's Sanitary Landfill to approximately 123,000 gpd at Carleton Farms Landfill. The daily flow from all 32 landfills is just over 1 million gallons. In general, the larger landfills produce more leachate than smaller ones, but other factors affect leachate generation including timing of cell closures, new cell development, leachate minimization practices, precipitation and recirculation.

3.2.3 Receiving WRRFs

As summarized on Table 3-1, with the exception of DWI, leachate from the original 32 MWRA-member landfills participating in this study are ultimately discharged to a WRRF, regardless of disposal/conveyance/pretreatment method employed. Statewide, the leachate from 18 facilities (more than half the participating sites) is managed at one of the five following, relatively large, regional WRRFs located in the southern half of Michigan's Lower Peninsula

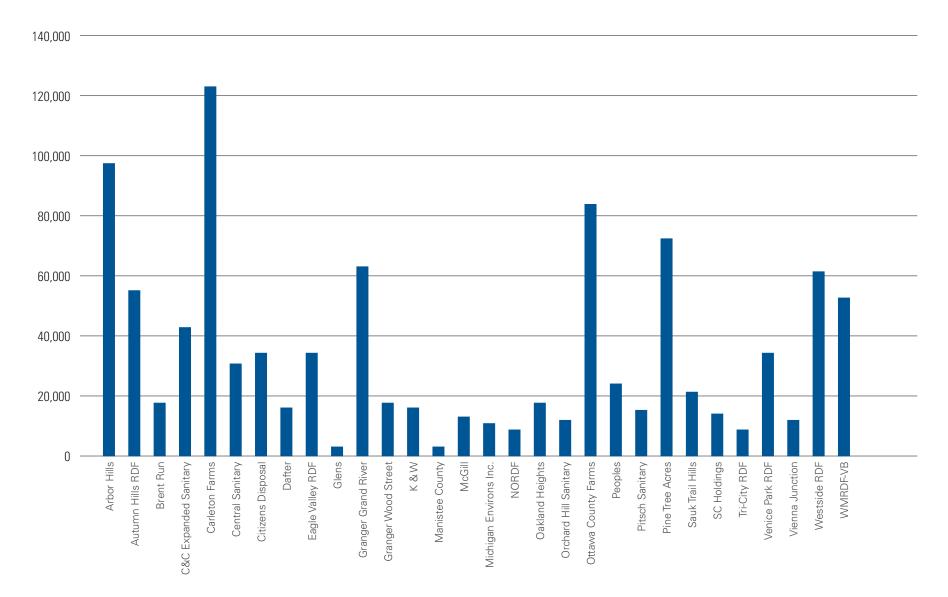


- 1. Great Lakes Water Authority in Detroit (GLWA), used by nine landfills,
- 2. Clinton River Water Resource Recovery in Pontiac (CRWRR), used by one landfill;
- 3. Grand Rapids Water Resource Recovery (GRWRR), used by four landfills
- 4. Anthony Ragnone Wastewater Treatment Plant near Flint (Ragnone), used by three landfills
- Ypsilanti Community Utilities Authority (YUCA), used by two landfills (one of these landfills also PAH to GLWA).

Leachate from the remaining 12 participating landfills is managed at individual, local and generally smaller-scale WRRFs, primarily located in less-densely populated regions of the state (e.g., Mid-Michigan, SW-Michigan, Northern-Michigan, and various locations in the Upper Peninsula), as indicated in Table 3-1.

Figure 3-3 Average Daily Leachate Volume Managed at Michigan WRRFs.





4.2 Statewide PFOA and PFOS Leachate Concentrations

Analytical data reports prepared by TAL, are contained in Appendix C, Analytical Data Reports. Table 4-2A, PFOA and PFOS Concentrations and Mass in Active Type II Landfills Leachate presents the concentrations of these PFAS compounds detected in 39 separate leachate samples collected from 35 active Type II landfills located in Michigan. We note three landfills included two or more leachate samples/locations (Venice Park, two samples; Riverview LF, three samples; and South Kent County LF, two samples).

As shown on Table 4-2A, PFOA concentrations for the MWRA participating landfills ranged from 240 ppt to 3,200 ppt. For all 35 Michigan active Type II landfills with data the PFOA concentration ranged from 16 ppt to 3,200 ppt with the lowest concentration in leachate detected in a Western-Michigan landfill and greatest concentration at a SE-Michigan landfill. The median PFOA leachate concentration was 1,000 ppt and the "average" concentration was approximately 1,187 ppt.

For PFOS, the leachate concentrations ranged from 100 to 710 ppt for the MWRA 32 participating landfills. For all 35 Michigan active Type II landfills with data the PFOS concentration ranged from 9 to 960 ppt, and the median value is 220 ppt. The lowest PFOS concentration was detected in leachate from a SE-Michigan landfill; the greatest from a Western-Michigan landfill. The average PFOS concentration was 287 ppt and the median concentration was 220 ppt.

4.3 MWRA Landfill Leachate PFOA & PFOS Concentrations Compared To Other Studies

Table 4-3, Michigan vs. Worldwide PFOA and PFOS Leachate Concentration Ranges compares ranges of PFOA and PFOS leachate concentrations observed as part of this study ("Michigan") to the ranges reported for other areas, based on the literature review discussed in Section 2.1. As shown, the worldwide leachate range for PFOA concentrations, is non-detect to 214,000 ppt and the corresponding PFOS range is non-detect to 6,020 ppt.

As indicated in Table 4-3, Michigan's PFOA and PFOS ranges are within those observed in the US based on available published literature. The Michigan PFOS concentration range is consistent with that reported in other Western world regions, but nearly an order-ofmagnitude lower than what is reported for China. The apparent reason China's concentrations are greater is their continued use of PFAS compounds in consumer-goods manufacturing.

 Table 4-3

 Michigan vs. Worldwide PFOA and PFOS Leachate

 Concentrations Ranges

Region	PFOA (ppt)	PFOS (ppt)
Michigan*	16 to 3,200	9 to 960
United States	30 to 5,000	3 to 800
Europe	ND to 1,000	ND to 1,500
Australia	17 to7,500	13 to 2,700
China	281 to 214,000	1,150 to 6,020
Worldwide Range	ND to 214,000	ND to 6,020

* Based on leachate analyses from 32 MWRA-member landfills participating in this statewide study and leachate data obtained on MiWaters.com.

4.4 Leachate PFOA And PFOS Concentrations vs. MDEQ Criteria

As indicated in Section 2.1, Michigan has established both groundwater clean-up criteria and surface water quality standards (WQS) for PFOA and PFOS. The Michigan Part 201 groundwater cleanup criteria for PFOA and PFOS is 70 ppt, either individually or as a combined limit. This is not an enforceable standard for public drinking water supplies but has been used in Michigan as a protective guideline during site investigations.

The Rule 57 PFOA WQS is 420 ppt for surface water that may be used as a drinking water (DW) source and 12,000 ppt for non-drinking water (NDW) sources. For PFOS, the WQS for drinking and non-drinking water sources are 11 ppt and 12 ppt, respectively.

It is not appropriate regulatory policy to compare the leachate results to surface water quality standards (WQS) because leachate is not being discharged to surface water. Nevertheless, the WQS are used as a means of putting the leachate results in some context.

Individually, as shown on Table 4-2A, the concentration of PFOA in leachate collected from two landfills during this study are below the 420 ppt DW WQS as are the concentrations from two samples from two separate landfills with data obtained from MiWaters. The other samples are above the 420 ppt value. The concentration of PFOA in the leachate from all sites was considerably lower than the 12,000 ppt NDW WQS. The concentration of PFOS at all locations exceeded the DW and NDW WQS.

Concentrations and Mass of PFOA AND PFOS Michigan Active Type II Landfills' Leachate

wichigan Active Type II Landhiis Leachate							
MWRA Participating Landfill Designation	Average Leachate Volume GPD	PFOA (ppt)	PFOS (ppt)	"PFOA Daily Mass (Ib/day)"	"PFOS Daily Mass (Ib/day)"		
Arbor Hills Landfill	98,400	3200	220	0.0026	0.00018		
Autumn Hills RDF	54,800	1300	380	0.0006	0.00017		
Brent Run Landfill	16,400	540	110	0.0001	0.00002		
C&C Expanded Sanitary Landfill	42,000	1300	450	0.0004	0.00015		
Carleton Farms Landfill	123,300	1800	250	0.0018	0.00026		
Central Sanitary Landfill	30,100	2500	470	0.0006	0.00012		
Citizen's Disposal Inc.	32,900	1100	180	0.0003	0.00005		
Dafter Sanitary Landfill	16,500	680	130	0.0001	0.00002		
Eagle Valley RDF	32,900	490	170	0.0001	0.00005		
Glens Sanitary Landfill	3,800	770	210	0.00002	0.00001		
Granger Grand River Landfill	64,400	240	160	0.0001	0.00009		
Granger Wood Street Landfill	19,200	470	110	0.0001	0.00002		
K&W Landfill	17,500	830	170	0.0001	0.00002		
Manistee County Landfill	4,700	420	220	0.000016	0.000009		
McGill Road Landfill	13,700	760	170	0.0001	0.00002		
Michigan Environs Inc. (Menominee)	13,100	1400	100	0.0002	0.00001		
Northern Oaks RDF	12,300	1000	220	0.0001	0.00002		
Oakland Heights Development	17,800	780	230	0.0001	0.00003		
Orchard Hill Sanitary Landfill	12,500	650	110	0.0001	0.00001		
Ottawa County Farms Landfill	82,200	1800	530	0.0012	0.0004		
People's Landfill	21,900	2500	710	0.0005	0.00013		
Pine Tree Acres RDF	74,000	1800	430	0.001	0.0003		
Pitsch Sanitary Landfill	15,000	1300	260	0.0002	0.00003		
Sauk Trail Hills Landfill	20,500	2800	610	0.0005	0.00010		
SC Holdings	16,000	960	410	0.0001	0.00005		
Tri-City RDF	9,600	1200	160	0.0001	0.00001		
Venice Park RDF MH#20*		910	190		İ		
Venice Park RDF MH#21*	32,900	1500	630	0.0007	0.0002		
Vienna Junction Industrial Park Sanitary Landfill	13,700	1300	130	0.0001	0.00001		
Waters Landfill	NONE	930	230	NONE	NONE		
Westside RDF	60,800	1300	160	0.0007	0.00008		
Whitefeather Landfill	NONE	1700	550	NONE	NONE		
Woodland Meadows RDF -Van Buren	54,800	2000	510	0.0009	0.00023		
Other Active Type II Landfill Leachate Data Obtained from MIWaters		PFOA (ppt)	PFOS (ppt)	PFOA Daily Mass (lb/day)	PFOS Daily Mass (Ib/day)		
Riverview 003* Riverview 004*		1900 860	270 140				
Riverview 004* Riverview 007*	37,400	38	8.5	0.0003	0.00004		
South Kent Outfall* South Kent Hauled*	48,000	725 16	960 130	0.0001	0.0002		
Smith's Creek Landfill*	32,900	510	120	0.0001	0.00003		
	minimum maximum median average n	16 3200 1000 1186 39	9 960 220 287 39	0.000016 0.003 0.0001 0.0004 33	0.000007 0.0004 0.00005 0.0001 33		

Notes:

1. There are a total 45 Active Type II Landfills in Michigan; 35 are represented in this table.

* - These facilities reported multiple laboratory results. In these cases, we calculated mass based on the averaged concentrations for PFOA and PFOS.

2. Riverview, South Kent, and Smith's Creek leachate are managed by the Downriver, Wyoming, and Port Huron WRRFs, respectively.

4.5: Statewide PFOA and PFOS WRRF Influent Concentrations

WRRFs serve all users within their respective service areas. Landfill leachate mixes with other wastewater from homes and workplaces, as well as public and private facilities (e.g., churches, restaurants and stores), that is delivered via municipal sanitary sewer networks. The WRRF treats the combined wastewater before adequately-treated water is discharged to a local surface water body or via infiltration beds.

Although very effective at removing bacteria, pathogens, and most undesirable chemicals present in wastewater, most WRRFs are not currently designed to significantly remove PFOA and PFOS.

Table 4-2B, WRRF Influent PFOA & PFOS Concentrations & Daily Mass, summarizes available data obtained from MiWaters organized by three groups. "Group A" includes the 14 (11 with available data) WRRFs that accept leachate from MWRA-member landfills; "Group B" nine (8 with data) that represent WRRF's that accept leachate from other active Type II landfills; and "Group C" 39 (20 with data) identify WRRFs that do not accept leachate from active Type II landfills.

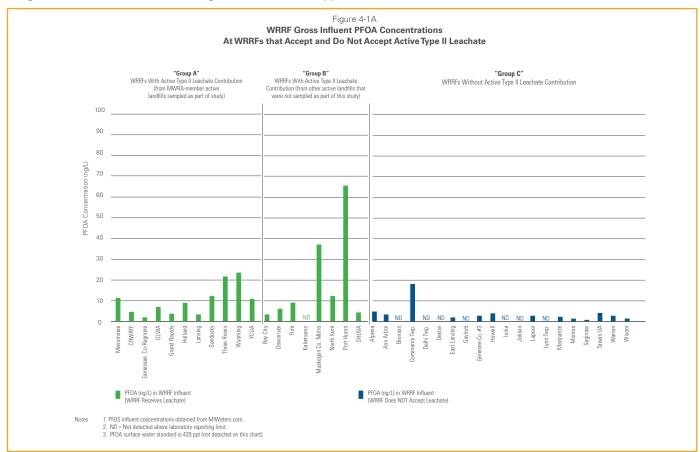
Reviewing all three groups, PFOA influent concentrations ranged from non-detect (ND) at eight WRRFs to 64.6 ppt.

The median PFOA influent concentration was 5.06 ppt and the average was 10.3 ppt, based on 31 sample with reported detections.

For PFOS in all groups, influent concentrations ranged from ND (at the same six WRRFs as before) to approximately 500 ppt. The median and average PFOS influent concentrations were 8.6 ppt and 34.5 ppt respectively, based on 29 samples with results above the method detection limit (MDL).

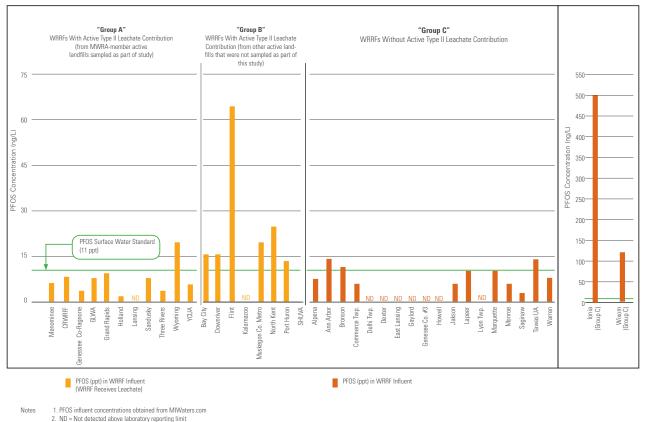
Figure 4-1A, WRRF Gross Influent PFOA Concentrations, graphically depicts available data for influent PFOA concentrations at WRRFs that accept leachate from active Type II landfills and those that do not, categorized by the groupings described above and on the graphic. Based on visual analyses of Figure 4-1A, we note that all influent values (Group A, Group B, and Group C) were below the most stringent 420 ppt PFOA WQS.

Figure 4-1B, WRRF Gross Influent PFOS Concentrations, depicts available data for influent PFOS concentrations at WRRFs that accept leachate from active Type II landfills and those that do not, categorized by the groupings described above and on the graphic. Based on visual analyses of Figure 4-1B, we note that more than half (12 of 19) of the WRRFs that accept landfill leachate (Group A and Group B) were below 11 ppt, the most stringent WQS for PFOS.



NTH Statewide Study on Landfill Leachate PFOA and PFOS Impact Technical Report
ATTACHMENT C

Figure 4-1B WRRF Gross Influent PFOS Concentrations At WRRFs that Accept and Do Not Accept Active Type II Leachate



4.6 PFOA & PFOS Leachate and WRRF Mass Comparison

In order to estimate the mass contribution of PFOA and PFOS in landfill leachate to the total WRRFs influent mass that were evaluated in the study, we again relied on information available from MWRA-member landfills (combined with data available for other landfills) and data provided via MiWaters (for influent and WRRF design flows). This information was used to calculate an estimated mass contribution of PFOA and PFOS from each landfill to their associated WRRF. We also estimated the total mass contribution of PFOA and PFOS from all study landfills and other wastewater sources that contribute to WRRF influent.

4.6.1: Influent Leachate PFOA and PFOS Mass

Table 4-2A, summarizes the calculated daily mass of PFOA in leachate from 33 landfills (2 landfills do not discharge to WRRFs) included in this study. The total daily PFOA estimated mass from all 33 landfills' leachate was 0.014 lb. Daily mass for PFOA was from a low of 0.000016 lb. (Northern-Michigan landfill) to a high of 0.0026 lb. (SE-Michigan landfill). The median daily PFOA mass was 0.0001 lb. and the average daily PFOA mass was 0.0004 lb. These small mass values illustrate that although some of the concentration results appear high when viewed in parts per trillion values, the mass contributions are actually quite low.

The calculated daily mass of PFOS in leachate from the 33 landfills is also include on Table 4-2A. The total daily PFOS estimated mass in leachate from all 33 landfills' leachate was 0.0031 lb. The daily mass ranged from a low of 0.000007 lb. (Northern-Michigan landfill) to a high of 0.0004 lb. (Western Michigan Landfill). The median daily PFOS mass was 0.00005 lb. and the average daily mass for PFOS was 0.0001 lb.

4.6.2: WRRF PFOA and PFOS Mass

Table 4-2B, provides a summary of all WRRFs used in our analyses. We note that the influent flow calculation is based on the WRRF design flow capacity provided in each WRRF's NPDES permit. This design flow was used since actual flow information is not known or published via MiWaters. Further, we note that most of the WRRF influent mass calculations rely on a single or very limited number of samples. Based on these considerations, the calculated masses are provided as estimates and actual mass may fluctuate over time, depending on a number of inter-related factors (e.g., precipitation, seasonality, etc.)

From Table 4-2B, based on 27 results, estimated daily WRRF influent PFOA mass ranged from non-detect

(at 10 facilities) to 0.03 lb., with a median of 0.0007 lb. and average of 0.003 lb. For PFOS, based on 25 results, estimated daily WRRF influent ranged from non-detect (at several locations) to 0.04 lb.; the associated median and average values were 0.0019 lb. and 0.005 lb., respectively.

Figure 4-2A, PFOA Mass: Influent Leachate vs. Overall WRRF Influent, depicts the total PFOA mass contribution from leachate versus overall estimated WRRF influent mass on a daily basis for the 13 facilities that receive leachate and have PFOA and/or PFOS data. Review of this graphic reveals the following:

- PFOA mass from leachate represents a relatively minor proportion of the individual WRRFs estimated influent mass at a majority of the WRRFs.
- GLWA's PFOA influent mass is at least twice that of any of the other 12 WRRFs, which is based on its permitted treatment capacity and large area served including many industrial facilities; and
- The influent PFOA mass for the other WRRFs that serve large, densely-populated metropolitan areas are

generally greater than observed at smaller WRRFs that serve less-populated areas.

Figure 4-2B, PFOS Mass: Influent Leachate vs. Overall WRRF Influent, depicts the total PFOS mass contribution from leachate versus overall estimated WRRF influent mass on a daily basis for the 13 facilities that receive leachate and have PFOA and or PFOS data. Visual evaluation of this stacked bar chart graph reveals the following:

- PFOS mass from leachate represents a relatively minor proportion of most the individual WRRFs and overall;
- GLWA'S PFOS influent mass is at least twice that of any of the other WRRFs, based on its large permitted treatment capacity and large area served including many industrial facilities; and
- Other than Lansing, which did not detect PFOS in their influent, the influent PFOS mass for the WRRFs that serve large, metropolitan areas are generally greater than smaller WRRFs that serve less populated areas.

		Influent Co	ncentration	Influent Mass	
Leachate Disposal/WRRF Facility	Permitted Capicity (MGD)*	PFOA (ppt)	PFOS (ppt)	PFOA (lb/ day	PFOS (lb/day)
		Min to Max	Min to Max		
Group A: WRRFs Utilized by MWRA-member	r Active, Type II Land	fills Participating	in this Study		
Belding	3.07	NA	NA	NA	NA
Menominee	3.2	12	5.6	0.0003	0.0001
Clinton River	30.6	4.94	7.68	0.0013	0.0019
Genesee Co-Ragnone	25.9	4	5.22	0.0009	0.0012
GLWA	650	6.02	7.54	0.0324	0.0406
Grand Rapids	61.1	5.06	12.7	0.0026	0.0066
Hastings	2	NA	NA	NA	NA
Holland	12	8.93	3.79	0.0009	0.0004
Lansing	35	4.98	ND	0.0014	ND
Ludington	4.5	NA	NA	NA	NA
Sandusky	2.55	12.2	7.98	0.0003	0.0002
Three Rivers	2.75	21.44	7.39	0.0005	0.0002
Wyoming	22	5.08 to 25	6.2 to 26.4	0.0046	0.0048
YCUA	51.2	12	4.8 to 7.51	0.0051	0.0032
Group B: WRRFs Utilized to Dispose I	Leachate from Other	Active, Type II La	ndfills		
Bay City	18	4.87	18.2	0.0007	0.0027
Downriver	125	7.2	22.2	0.0075	0.0230
Flint	50	10.3	62.4	0.0043	0.0258
Kalamazoo	53.5	ND	ND	ND	ND
KI Sawyer	0.65	NA	NA	NA	NA
Muskegon Co Metro	43	11.7 to 36.9	10.5 to 24.3	0.0131	0.0086
North Kent S A	8	11.2	31.1	0.0007	0.0021
Port Huron	20	64.6	19.5	0.0107	0.0032
S Huron Valley UA (SHUVA)	24	3.76	ND	0.0007	ND

Table 4-2B WRRF Influent PFOA and PFOS Concentrations (Page 1 of 2)

* WRRF permitted daily flow and PFOA and PFOS data provided by MIWaters.com.

Influent mass calculated using the single sample or the maximum value where multiple data are available.

NA: data not available

ND : Not detected. Detection limit unknown. Excluded from average and median calculations.

Table 4-	-2B		
WRRF Influent PFOA and PFOS	Concentr	ations (Page 2 of 2)	
	MODE	Influent Concentration	

	WRRF	Influent Co	Influent Concentration		Influent Mass	
Leachate Disposal/WRRF Facility	Permitted Capicity (MGD)*	PFOA (ppt)	PFOS (ppt)	PFOA (lb/ day	PFOS (lb/day)	
		Min to Max	Min to Max			
Group C: WRRFs that do not 1	Freat Active Type	II Leachate				
Adrian	7	NA	NA	NA	NA	
Alpena	5.5	5.94	5.44	0.0003	0.0002	
Ann Arbor	29.5	2.91 to 4.3	16.5 to 20	0.0011	0.0049	
AuGres	0.221	NA	NA	NA	NA	
Battle Creek	18	NA	NA	NA	NA	
Benton Harbor - St. Joseph	15.3	NA	NA	NA	NA	
Boyne City	0.9	NA	NA	NA	NA	
Bronson	0.5	ND	12	ND	0.0001	
Charlotte	1.8	NA	NA	NA	NA	
Commerce Twp	8.5	17.9	6.38	0.0013	0.0004	
Delhi Twp	4	ND	ND	ND	ND	
Dexter	0.58	ND	ND	ND	ND	
East Lansing	18.75	2.21	ND	0.0004	ND	
Gaylord	2.2	ND	ND	ND	ND	
Genesee Co #3	11	2.6	ND	0.0002	ND	
Gladwin	0.65	NA	NA	NA	NA	
Greenville	1.75	NA	NA	NA	NA	
Holly	1.35	NA	NA	NA	NA	
Howell	2.4	4.42	ND	0.0001	ND	
Ionia	4	ND	499.36	ND	0.0165	
Jackson	18	ND	5.98	ND	0.0009	
Lapeer	1.5	4.2	8.6	0.0001	0.0001	
Lyon Twp	1.095	ND	ND	ND	ND	
Marquette	3.85	3.27	10.3	0.0001	0.0003	
Marysville	2.4	NA	NA	NA	NA	
Milan WWTP	2.5	NA	NA	NA	NA	
Monroe	24	2.89	5.5	0.0006	0.0011	
Mt Clemens	6	NA	NA	NA	NA	
Petoskey	2.5	NA	NA	NA	NA	
Saginaw Twp	4.8	NA	NA	NA	NA	
Saginaw	32	2.56	4.19	0.0007	0.0011	
Saline	1.81	NA	NA	NA	NA	
South Lyon	2.5	NA	NA	NA	NA	
Sturgis	2.8	NA	NA	NA	NA	
Tawas Utility Authority	2.4	6.2	17	0.0001	0.0004	
Warren	36	4.61	7.31	0.0014	0.0022	
West Bay County Regional	10.28	NA	NA	NA	NA	
Wixom	2.8	3.07	128	0.0001	0.0029	
Zeeland	1.65	NA	NA	NA	NA	
	minimum maximum	ND 64.6	ND 499.36	ND 0.03	ND 0.04	
Summary Statistics - all Groups (A, B, C)	median	5.06	8.6	0.0007	0.0019	
	average	10.3	34.5	0.003	0.005	
	n	31	29	31	29	

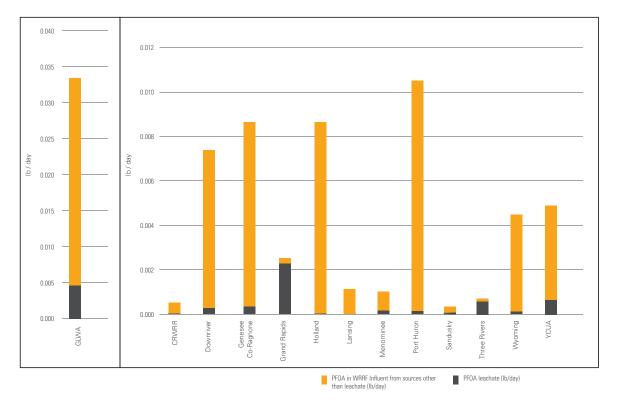
* WRRF permitted daily flow and PFOA and PFOS data provided by MIWaters.com

Influent mass calculated using the single sample or the maximum value where multiple data are available.

NA: data not available

ND · Not detected Detection limit unknown Excluded from average and median calculations





Note: Gray shading indicates active Type II landfill leachate loading to WWRF for PFOA mass. This graph includes a total of 13 WRRFs utilized by 26 landfills. Eleven of the WRRFs treat 24 active landfills [23 which were sampled as part of this study and South Kent landfill). Two of the WRRFs are utilized by two additional active landfills that were not sampled as part of this study. PFOA and PFOS influent concentrations were unavailable from the WRRFs that treat other active Type II landfills. The mass represents a calculated value on a single sample, permitted discharge volume, and average daily leachate discharge.

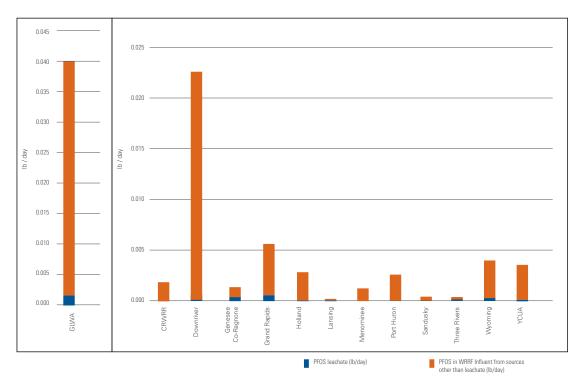


Figure 4-2B PFOS Mass: Influent Leachate vs. Overall WRRF Influent

Note : Blue shading represents active Type II landfill leachate loading for PFOS mass at each WRRF. This graph includes a total of 13 WRRFs utilized by 26 landfills. Eleven of the WRRFs treat 24 active landfills (23 which were sampled as part of this study and South Kent landfill). Two of the WRRFs are utilized by two additional active landfills that were not sampled as part of this study. PFOS influent concentrations were unavailable for the WRRFs that treat other active Type II landfills. The mass represents a calculated value on a single sample, permitted discharge volume, and average daily leachate discharge.

5.0: OTHER CONSIDERATIONS

In this section, we discuss other concerns related to the current understanding of PFOA and PFOS in the environment that need to be addressed to help guide future regulatory, toxicological, and best-management practices (BMPs).

5.1: WRRF Influent, Effluent, and Biosolids

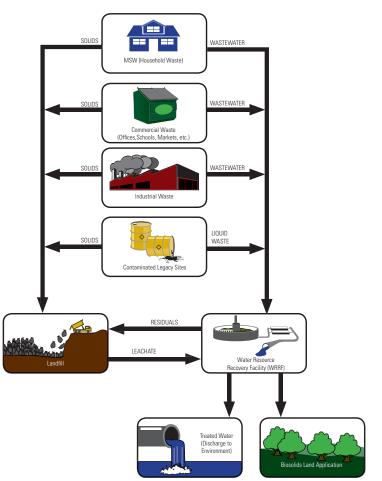
It is documented that WRRF biosolids typically contain PFAS (NEBRA, 2018). A recent comprehensive study was completed for the North East Biosolids and Residuals Association (NEBRA) that examined PFOA and PFOS concentrations in WRRF biosolids. Although the biosolids data are reported for solid/sludge samples and leachate samples are liquids, based on our review, the biosolids concentrations were typically two orders-of-magnitude greater than observed in active, Type II landfill leachate on a ppt basis.

Related specifically to PFOA and PFOS mass in leachate and WRRF biosolids, there are complexities between these two media that need evaluation to optimize future management of these two waste streams:

- the role of biochemical processes in WRRFs;
- fate and transport of PFOA/PFOS contained in biosolids
- temporal and spatial variation effects;
- waste age and state of decomposition in landfills;
- impact of equipment and infrastructure residual contamination; and
- appropriate and effective current BMPs.

While beyond the scope of this study to assess these factors, recent and ongoing research by others may provide direction. For example, work by Hamid (2018) and Lang (2017) indicate some PFAS compounds typically increase in WRRF effluent as compared to influent from biochemical degradation of related PFAS chemicals within the waste stream. Other factors could include residual PFAS from WRRF processing equipment.

For landfills, the existing literature (Lang, et al, and related references) indicates that PFOA+PFOA leachate mass decreases over time with more rapid declines observed in temperate, humid climates. This observation is significant with respect to long-term PFAS leachate management and reduction.



5.2: Proper PFAS Waste Management: Interdependence between Landfills, WRRFs, and General Public

Our study and previous investigations confirm PFAS presence in LF leachate – it comes from many sources that cannot be easily identified or eliminated including various consumer products disposed in landfills. As indicated throughout this report, PFAS have been used for over 50 years in household products. Managing PFAScontaining waste is a challenge that touches all sectors of the economy, including the solid waste industry, manufacturing and commercial sectors, and the general public. It is a societal concern that we need to work together to effectively address.

The leachate is effectively managed at landfills through active leachate collection via engineered liner systems. In Michigan, the most viable method for leachate management is its discharge to a local WRRF where it is handled with other household, commercial, and various industrial wastewaters. In this way, leachate is managed in a closed system where there is no direct exposure to the public. WRRFs treat wastewater to meet certain regulatory criteria prior to discharge of the treated water.



Considering data collected and evaluated during this study, the impact that PFOA and PFOS in landfill leachate has on WRRFs influent concentrations is presented on Figures 4-2A and 4-2B. These data indicate that:

- a. leachate provides a relatively minor contribution to the overall PFOA and PFOS concentration/mass in most WRRF influent because of the relatively low leachate discharge volumes;
- b. non-leachate sources of PFOA and PFOS significantly contribute to WRRF influent and at higher volumes. It is noteworthy that the WRRF influent that have no landfill leachate contribution show a similar concentration range for PFOA and PFOS as WRRF influent that has leachate contribution; and
- c. although reduction of landfill leachate concentrations of PFOA and PFOS to the WRRF influent could be beneficial to meeting WQS in the WRRF effluent, the impact may be minor in most cases since leachate typically contributes a relatively small volume to the overall WRRF influent.

As discussed above, WRRFs also produce biosolids (i.e., "sewage sludge") with elevated concentrations of PFAS. These biosolids are normally either land applied as fertilizer or incinerated (which potentially create separate environmental exposures), or are disposed at landfills (which likely contributes to higher PFAS concentrations in leachate at those landfills).

Each of these WRRF biosolids management methods have potential unintended adverse consequences. Incineration emissions may contribute to airborne PFAS, although this is largely un-studied. Similar cross-media impacts may be related to land application. Disposing of biosolids in landfills likely increases the concentrations of PFAS in leachate discharged to WRRFs. However, of the three disposal methods, landfilling in properly built and managed landfills appears to pose the least risk because landfills have engineering controls and environmental monitoring systems.

Accordingly, landfills and WRRFs have an important and mutually-beneficial relationship: landfills need to dispose of leachate and WRRFs need to safely manage society's biosolids. Together, these two critical environmental infrastructure components would benefit from enhanced cooperation to manage PFAS to serve the needs of both industries and protect the environment.

Electronic Filing: Received, Clerk's Office 11/23/2022 6.0: CONCLUSIONS

PFOA and PFOS were detected in all of the leachate samples taken in the study. The concentration ranges were similar to previous leachate studies conducted elsewhere in the US. The variability from landfill to landfill may reflect variations in waste-types, waste age, size of landfills in the study, and the relative state of decomposition. In summary:

- In leachate sampled from MWRA member landfills that participated in this study, PFOA ranged from 240 to 3,200 ppt and PFOS ranged from 100 to 710 ppt.
- In published studies of landfill leachate in the United States, PFOA ranged from 30 to 5,000 ppt and PFOS ranged from 3 to 800 ppt.
- Michigan leachate concentrations were substantially lower than some other countries, such as China, where published studies show PFOA ranged from 281 to 214,000 ppt and PFOS ranged from 1,150 to 6,020 ppt.

Comparing leachate volume and mass contribution from the 35 landfills examined to the total influent mass at the 39 WRRFs shows that the contribution of PFOA and PFOS is mostly from non-landfill sources.

 On a statewide basis, available data indicates that the 35 landfills contribute approximately one million gallons of leachate to WRRF influent, with approximately 0.01 lbs / day of PFOA and 0.003 lbs / day of PFOS. On a statewide basis, available data indicates that the 34 WRRFs that have influent data receive approximately 1.4 billion gallons of influent daily (based on design capacity), with approximately 0.09 lbs / day of PFOA and 0.15 lbs / day of PFOS.

The ranges of PFOA and PFOS concentrations in WRRF influent that do not accept leachate show overlap with those that do accept leachate.

- In WRRFs that do not accept landfill leachate, influent levels of PFOA range from non-detect to 17.9 ppt while PFOS ranges from non-detect to 499 ppt (next highest value is 128 ppt).
- In WRRFs that accept landfill leachate, influent levels of PFOA range from non-detect to 64.6 ppt while PFOS ranges from non-detect to 62.4 ppt.
- Available data show that PFOA levels in WRRF influent are well below Michigan's most conservative surface water criteria (420 ppt) at all WRRFs examined, and that PFOS levels in WRRF influent are below Michigan's most conservative surface water criteria (11 ppt) at approximately two-thirds of the WRRFs examined.
- The data collected during this study indicate that leachate provides a relatively minor contribution to the overall PFOA and PFOS concentration in most WRRF influent; non-leachate sources of PFOA and PFOS contribute greater mass to WRRF influent than leachate.

7.0: RECOMMENDATIONS

Based on the results of this study, we present the following recommendations:

- The solid waste industry in Michigan (and nationally) must continue working to understand the significance of the contribution of leachate to PFOA and PFOS received by WRRFs and work towards reduction solutions.
- The conclusions of this study are based mainly on a single leachate sample from each landfill and limited available data for WRRFs. Therefore, calculated mass values are estimates and more data and information are needed. This should include additional leachate data, WRRF influent data, and biosolids data.
- Facilities will need to present and discuss their individual results with the WRRF receiving their leachate to help evaluate any appropriate solutions on a local basis.

The information gathered during this study and other research can be used to develop, where needed, improved practices for management of waste that contains PFAS within and between landfills and WRRFs. Future collaboration should involve forming a workgroup consisting of MWRA members, MDEQ, MPART, and WRRFs. Discussions should take into consideration the unique aspects of landfills as a component of PFAS management and their interdependence with WRRFs in providing an important function to society. Further, the stakeholder parties need to work with toxicologists and other environmental scientists to better understand the potential impacts of PFOA and PFOS on human health in the context of landfill leachate and in general.

MWRA is committed to continue playing an active role in this process, as demonstrated by its funding of this statewide leachate report and ongoing participation with state and federal technical and scientific committees working toward solutions that follows sound scientific principles and implements best management practices where needed.

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North Carolina Collective Study Report

Collective Study of PFAS and 1,4-Dioxane in Landfill Leachate and Estimated Influence on Wastewater Treatment Plant Facility Influent

National Waste & Recycling Association - Carolinas Chapter

H&H Job No. NWA-001 March 10, 2020



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Table of Contents

Section Pag	e
1.0 Introduction	1
2.0 General Overview	3
2.1 Background Information	3
2.2 Waste Management System Considerations	4
2.3 Other Related Studies	5
2.4 Regulatory Status	7
3.0 Sampling Activities	9
3.1 Locations Sampled	9
3.2 Sampling Methodology	9
3.3 Laboratory Analyses	2
3.4 Discussion of Sampling Results and Comparison to Other Studies1	3
4.0 Influence on WWTP Influent1	5
4.1 Description of Receiving WWTPs1	5
4.2 WWTP Sampling Data Source1	5
4.3 Discussion of WWTP Influent Sampling Results and Comparison to Other Studies	6
4.4 Leachate Contribution to WWTP Daily Mass1	7
5.0 Conclusions and Recommendations1	8
6.0 References2	0



List of Tables

Table 1	Literature Summary of PFOS and PFOA in Landfill Leachate
Table 2	Landfill and WWTP Facility Information
Table 3	Leachate Analytical Data
Table 4	PFOS and PFOA Daily Leachate Mass Calculations
Table 5	1,4-Dioxane Daily Leachate Mass Calculations
Table 6	PFOS and PFOA Daily WWTP Mass Calculations
Table 7	1,4-Dioxane Daily WWTP Mass Calculations
Table 8	Percent of WWTP Daily Mass Contributed by Landfill Leachate

List of Figures

- Figure 1 PFOA & PFOS Concentrations in Landfill Leachate Based on Literature Summary
- Figure 2 Facility Location Map
- Figure 3 PFOS and PFOA Daily Leachate Mass Summary
- Figure 4 PFOS and PFOA Daily WWTP Mass Summary
- Figure 5 PFOS Landfill Leachate Contribution to WWTP Daily Mass
- Figure 6 PFOA Landfill Leachate Contribution to WWTP Daily Mass
- Figure 7 1,4-Dioxane Daily Leachate Mass Summary
- Figure 8 1,4-Dioxane Landfill Leachate Contribution to WWTP Daily Mass

List of Appendices

Appendix A Laboratory Analytical Reports



ii

North Carolina Collective Study Report National Waste & Recycling Association - Carolinas Chapter <u>H&H Job No. NWA-001</u>

1.0 Introduction

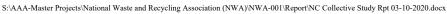
Hart & Hickman, PC (H&H) has prepared this North Carolina Collective Study Report on behalf of the Carolinas Chapter of the National Waste & Recycling Association (NWRA) and certain member companies. This report documents the results of a study of perfluoroalkyl and polyfluoroalkyl substances (PFAS) and 1,4-dioxane in municipal solid waste landfill (MSWLF) leachate and its possible influence on wastewater treatment plant (WWTP) facility influent.

In February 2019, the North Carolina Department of Environmental Quality (NCDEQ) met with representatives of the landfill industry to discuss the potential presence of PFAS and 1,4-dioxane in leachate as part of a statewide effort to better understand the presence of these emerging chemicals in the environment. During the meeting, NCDEQ inquired about sampling landfill leachate to begin to understand PFAS and 1,4-dioxane content and its influence on leachate treatment/disposal practices, including publicly owned WWTPs that receive leachate for treatment. Rather than participating landfills sampling and reporting individually, representatives of the landfill industry agreed to participate in a collective study involving active MSWLFs in North Carolina. From these discussions with NCDEQ, the Carolinas Chapter of the NWRA committed to collect leachate samples from nine privately-owned or operated MSWLFs, including four landfills that transport leachate to WWTPs located within the Cape Fear River Basin and five landfills that transport leachate to WWTPs located across the remainder of the State. This report documents the scope and results of the sampling program. Where available, the results of the sampling were evaluated in conjunction with WWTP influent volumes and published sampling data in order to estimate the relative contribution of landfill leachate to overall WWTP influent mass of PFAS and 1,4-dioxane. The goals and objectives of the sampling program were presented to NCDEQ in a Scoping Document, dated August 8, 2019. NCDEQ issued a letter, dated August 14, 2019, concurring with the plan outlined in the Scoping Document.



This North Carolina Collective Study Report is organized into sections to include the following:

- General overview of PFAS and 1,4-dioxane in landfill leachate, including background • information, waste management system considerations, a summary of previous studies, and North Carolina regulatory status;
- Description of sampling activities and results; and ٠
- Discussion of the WWTPs receiving the landfill leachate and calculations related to • estimating the contribution of landfill leachate to overall WWTP influent mass.





ATTACHMENT D

2.0 General Overview

2.1 Background Information

PFAS are a group of man-made chemicals that have been manufactured and used in a variety of industries worldwide since the 1940s. The most extensively produced and studied PFAS compounds are perfluorooctanoic acid (PFOA) and perfluorooctanesulfonate (PFOS). Another notable PFAS compound is 2,3,3,3-tetrafluoro-2-(1,1,2,2,3,3,3-heptafluoropropoxy)-propanoic acid (PFPrOPrA), which has the trade name GenX and is used in manufacturing nonstick coatings (United States Environmental Protection Agency [EPA], 2019a).

PFAS have been used to make a variety of consumer products that are resistant to water, grease, or stains. PFAS have also been used in firefighting foams and various industrial processes (Interstate Technology and Regulatory Council [ITRC], 2017). PFAS do not occur naturally, but are widespread in the environment and have been found in people, wildlife, and fish all over the world. Certain PFAS can accumulate in the human body for long periods of time and do not break down easily in the environment (Agency for Toxic Substances and Disease Registry [ATSDR], 2020).

PFOS and PFOA have been largely phased out by industry in the United States, with this phase-out beginning in the early 2000s. However, PFOS and PFOA are still being produced internationally and imported into the United States in consumer goods. Landfills receive a large variety of residential and industrial waste containing PFAS compounds (see inset) (ITRC, 2017).

Products/Wastes with Potential PFAS

Consumer products Paper and packaging Clothing and carpets Outdoor textiles and sporting equipment Ski and snowboard waxes Non-stick cookware Cleaning agents and fabric softeners Polishes and waxes Pesticides and herbicides Hydraulic fluids Windshield wipers Paints, varnishes, dyes, and inks Adhesives Medical products Personal care products (for example, shampoo, hair conditioners, sunscreen, cosmetics, toothpaste, dental floss) Sewage sludge Industrial wastes Auto shredder residue Debris from fire cleanup **Discarded AFFF** Other sources



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PFAS are considered to be contaminants of emerging concern (CECs). CECs are chemicals that present known or potential human health effects or environmental risks, but either do not have regulatory cleanup standards or regulatory standards are evolving due to new science, detection capabilities or pathways, or both (ITRC, 2017). PFAS were the primary focus of the North Carolina Collective Study; however, at the request of the NCDEQ, another CEC, 1,4-dioxane, was also included in the sampling and analytical program. 1,4-Dioxane has been used as a solvent in the manufacture of other chemicals, as a stabilizer for chlorinated solvents, and as a laboratory reagent. It can also be found as a by-product in many consumer and industrial products (EPA, 2017a, ATSDR, 2011, and ATSDR, 2012) (see inset). Disposal of these products in landfills can result in 1,4-dioxane in landfill leachate (Maine Department of Environmental Protection [MDEP], 2020).

2.2 Waste Management System Considerations

Landfills and WWTPs play an important role in managing wastes for our communities. It is important to note that landfills and WWTPs are receivers of PFAS and 1,4-dioxane and are not the producers or original sources. Rather, consumer products and other wastes disposed of in these facilities represent the source. Modern landfills are well-engineered and managed facilities designed to protect the environment from contaminants that may be present in the waste stream. MSWLFs must meet stringent regulatory requirements (see inset) (EPA, 2017b). North Carolina Administrative Code (NCAC) Title 15A Subchapter 13B requires that MSWLF liner systems include either 1) a geomembrane liner installed above and in direct and uniform contact with a compacted clay liner with a minimum thickness of 24 inches and a permeability

Products/Wastes with Potential 1,4-Dioxane

Consumer products Household cleaners Detergents Shampoos Deodorants Cosmetics Food supplements Paint Paint strippers Dyes Greases Antifreeze Aircraft deicing fluids Adhesives Pesticides Industrial wastes Laboratory wastes

MSWLF Regulatory Requirements

Location restrictions Composite liner requirements Leachate collection and removal systems Operating practices Federal, state, and local environmental monitoring requirements (groundwater, surface water, stormwater, air, leachate) Closure and post-closure care requirements Corrective action provisions Financial assurance Others



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of no more than $1.0 \ge 10^{-7}$ cm/sec or 2) a geomembrane liner installed above and in direct and uniform contact with a geosynthetic clay liner (GCL) overlying a compacted clay liner with a minimum thickness of 18 inches and a permeability of no more than $1.0 \ge 10^{-5}$ cm/sec. Landfill leachate is generated from rainfall travelling through landfill waste or liquids within the waste itself. The leachate is effectively captured through liner and leachate collection systems. A common method of leachate disposal is discharge to a local publicly-owned WWTP where it is handled with other household, commercial, and various industrial wastewaters. Management of leachate in this way provides for a closed system where there is no direct exposure to the public (NTH Consultants, Ltd. [NTH], 2019).

Because PFAS and 1,4-dioxane are so ubiquitous, publicly-owned WWTPs receive wastewater from multiple sources that may contain PFAS and 1,4-dioxane. In addition to landfill leachate, other potential sources containing PFAS and/or 1,4-dioxane include wastewater from industrial, commercial, and agricultural operations and domestic sewage generated from homes, workplaces, and other public and private facilities. Biosolids (sewage sludge) from WWTPs may contain PFAS compounds (EPA, 2018; MDEP, 2020a; Michigan Department of Environment, Great Lakes, and Energy, 2020). Biosolids are commonly disposed of via land application, incineration, or landfilling. Because MSWLFs are strictly regulated and include liners and leachate collection systems engineered to prevent releases of pollutants to the environment, disposal of biosolids in MSWLFs may represent the preferred management option.

2.3 Other Related Studies

NTH, on behalf of the Michigan Waste & Recycling Association (MWRA), recently performed a statewide study of landfill leachate PFAS impacts on WWTP influent in the State of Michigan (herein referred to as the Michigan Study). This effort represented one of the largest studies conducted on active landfill leachate to date. The results of the study were documented in a Technical Report dated March 1, 2019 (NTH, 2019). Testing performed as part of the Michigan Study included collection of leachate samples from 32 active MSWLFs located in the State of Michigan and analysis of the samples for PFOS and PFOA. Data related to leachate disposal methods and volumes were gathered for each of the MSWLFs tested. The results were evaluated



5

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with respect to publicly available sampling data for WWTPs located across the State of Michigan. The North Carolina Collective Study presented in this report was performed using an approach similar to the Michigan Study. The results of the Michigan Study are discussed in conjunction with the results of the North Carolina Collective Study in Sections 3.4 and 4.0 of this report.

The Michigan Study also included a review of literature related to PFAS in landfill leachate. The literature review identified two key publications: National Estimate of Per- and Polyfluoroalkyl Substances (PFAS) Release to U.S. Municipal Landfill Leachate (Lang et al, 2017) and Review of the Fate and Transformation of Per- and Polyfluoroalkyl Substances (PFASs) in Landfills (Hamid et al, 2018). Lang et al (2017) evaluated the concentrations of PFAS compounds in 95 samples of leachate from landfills of varying climates and waste ages in the United States. According to the summary presented in the Michigan Study report, Lang et al demonstrated that PFOA and PFOS concentrations in leachate generally have been decreasing over time, with greater rates of decline in humid regions. Hamid et al (2018) compiled data from 11 literature sources that document PFAS leachate concentrations from dozens of landfills and more than 162 leachate samples from across the globe. The data show that PFOS and PFOA concentrations vary widely in different regions of the world, and are likely reflective of the consumer products and industrial materials used, produced, and disposed in each country. Reported concentrations for landfills in China were notably higher than elsewhere, which is likely due to the continued production of consumer goods containing PFAS and associated industrial waste from the manufacturing processes. Note that PFAS-containing products manufactured in China and other countries are often imported into the United States for purchase and eventually disposed of in United States landfills. PFOS and PFOA concentration data based on the literature review performed during the Michigan Study are summarized in Table 1 and depicted in Figure 1.

Additional studies of PFAS in landfill leachate are underway since the date of the Michigan Study. Locally, the North Carolina Policy Collaboratory (NC Collaboratory) has funded research being performed by the NC PFAS Testing (PFAST) Network. The NC Collaboratory was established by the North Carolina General Assembly in 2016 to facilitate and fund research and make recommendations to the General Assembly. The PFAST Network consists of investigators from



various NC universities performing multiple studies related to PFAS. One of these studies is being led by Dr. Morton Barlaz at North Carolina State University and focuses on PFAS in landfill leachate. The purpose of the study is to assess the relative importance of MSWLFs and domestic wastewater as contributors of PFAS to WWTPs and potentially to surface water (PFAST Network, 2019). The results of the PFAST Network study have not yet been published and therefore could not be incorporated into the North Carolina Collective Study documented in this report.

No comprehensive studies have been identified regarding 1,4-dioxane concentrations in landfill leachate. More data are available regarding 1,4-dioxane concentrations in public water systems (PWS). Monitoring of 1,4-dioxane in PWS was required by the EPA Third Unregulated Contaminant Monitoring Rule (UCMR 3). Adamson et al (2017) documents an evaluation of 1,4-dioxane concentrations in PWS located across the United States based on data collected under the UCMR 3. The results of the study identified detectable concentrations of 1,4-dioxane in 21% of 4,864 PWS. The study concluded that the data indicated a decreasing trend in concentrations and detection frequency over time. The study also concluded that detections of 1,4-dioxane were highly associated with detections of other chlorinated solvent compounds, which is attributed to the use of 1,4-dioxane as a solvent stabilizer.

2.4 Regulatory Status

The regulatory status of PFAS and 1,4-dioxane are currently evolving as additional studies are completed regarding human health risks and ecological effects. No regulatory standards or screening levels have been developed by EPA or the State of North Carolina that are applicable to landfill leachate. Levels that have been established for drinking water are summarized below, but it should be noted that these levels do not apply to landfill leachate.

PFAS

EPA has not adopted Federal regulatory standards or Maximum Contaminant Levels (MCLs) for PFAS compounds to date. EPA has established a Health Advisory Level for combined or individual PFOS and PFOA of 70 nanograms per liter (ng/L, equivalent to parts per trillion). EPA's



health advisories are non-enforceable and non-regulatory but provide technical information to state agencies and other public health officials on health effects, analytical methodologies, and treatment technologies associated with drinking water contamination (EPA, 2019b).

North Carolina also has not adopted regulatory standards for PFAS compounds to date. North Carolina has established a Drinking Water Health Goal for PFPrOPrA (GenX) of 140 ng/L. According to the North Carolina Department of Health and Human Services (NCDHHS), the PFPrOPrA Drinking Water Health Goal is not a regulatory level and is not a boundary line between a "safe" or "dangerous" level, but can be used to provide information to affected communities and residents about potential risks from exposure to GenX through drinking water (NCDHHS, 2020).

1,4-Dioxane

EPA has not adopted Federal regulatory standards or MCLs for 1,4-dioxane to date. EPA has established a Drinking Water Health Advisory Level of 35 micrograms per liter (μ g/L, equivalent to parts per billion). As referenced above, EPA's health advisories are non-enforceable and non-regulatory but provide technical information to state agencies and other public health officials (EPA, 2019b).

North Carolina has established a 2L Groundwater Standard under Title 15A NCAC 2L .0202 of 3 μ g/L for 1,4-dioxane. The 2L Standards are the maximum allowable concentrations resulting from any discharge of contaminants that may be tolerated without creating a threat to human health or would otherwise render the groundwater unsuitable for it intended best usage. Although a 2L Groundwater Standard has been established, NCDEQ has relied on the EPA Drinking Water Health Advisory Level of 35 μ g/L when evaluating the potential for impacts to public water supplies (NCDEQ, 2020).

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3.0 Sampling Activities

3.1 Locations Sampled

In accordance with the August 2019 Scoping Document, leachate samples were collected from the following nine active MSWLF facilities located across the State of North Carolina:

- 1. Great Oak Landfill (7607-MSWLF-2015)
- 2. Sampson County Disposal, LLC (8202-MSWLF-2000)
- 3. South Wake MSW Landfill (9222-MSWLF-2008)
- 4. Upper Piedmont Regional Landfill (7304-MSWLF-1997)
- 5. BFI-Charlotte Motor Speedway Landfill V (1304-MSWLF-1992)
- 6. Uwharrie Environmental Regional Landfill (6204-MSWLF-1995)
- 7. East Carolina Regional Landfill (0803-MSWLF-1993)
- 8. Chambers Development MSW Landfill (0403-MSWLF-2010)
- 9. Foothills Environmental Landfill (1403-MSWLF-1998)

Prior to sampling, H&H contacted each landfill and requested information regarding site contacts, leachate collection and disposal systems, access limitations, typical leachate sampling locations, leachate volumes, and leachate disposal methods. This information is summarized in **Table 2**. The landfill locations are shown on **Figure 2**.

3.2 Sampling Methodology

Sampling was performed by H&H staff with experience sampling for PFAS and other constituents of concern. Sampling procedures were in accordance with the guidance document "PFC Sampling Procedures, January 2019" issued by the NCDEQ Division of Waste Management (DWM) Solid Waste Section (herein referred to as NC DWM Sampling Guidance). Prior to sampling, a Health & Safety Plan was prepared to cover safety concerns associated with the proposed field activities. Sampling bottles, bottle coolers, and PFAS-free water for blanks and decontamination were



obtained from the laboratory, GEL Laboratories, LLC (GEL) located in Charleston, South Carolina.

Because PFAS are present in many commonly used materials, the PFCs Sampling Checklist form included with the NC DWM Sampling Guidance was followed by field personnel to reduce the potential for cross-contamination of samples with PFAS from external sources. Each sampler washed their hands before sampling and utilized a minimum of three layers of nitrile gloves at each sampling location to maintain a "clean hands" approach after encountering various surfaces. Sampling supplies were placed on new high-density polyethylene (HDPE) sheeting in close proximity to the sampling location.

Sampling was performed September 16 through 19, 2019. Leachate collection/management systems vary by landfill facility; therefore, samples were collected under three general scenarios as described below. The sampling scenario for each facility is indicated on **Table 2**.

Valve at Bottom of Holding Tank/Discharge Line

At locations where a sample port was located at the bottom of the holding tank and/or the discharge line (all locations except BFI-Charlotte Motor Speedway Landfill V and Great Oak Landfill), the valve was opened to clear any potential sediment and to adjust the flow to an appropriate rate for sample collection. Using fresh nitrile gloves, the sampler then removed the lid of the sample container and collected the sample keeping the sample container lid in the opposite hand. Upon completion of sampling, bottles were capped, placed in Zip-lock bags, and placed into laboratory-supplied coolers filled with ice. Because samples were collected directly into laboratory-supplied sampling containers and no separate sampling apparatus was used, no equipment blanks were collected for these locations.

Direct From Lagoon

• At the BFI-Charlotte Motor Speedway Landfill V, the sampling team mobilized to the leachate lagoon and set up a sampling station on the edge of the lagoon utilizing new HDPE sheeting. Samples were collected by submerging a new unpreserved laboratory-supplied

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ATTACHMENT D

sample container approximately 1-foot below the surface of the lagoon, then transferring the contents into the laboratory-supplied sample containers to be submitted for analysis. Upon completion of sampling, bottles were capped, placed in Zip-lock bags, and placed into laboratory-supplied coolers filled with ice. Because samples were collected using laboratory-supplied sampling containers and no separate sampling apparatus was used, no equipment blanks were collected for this location.

Direct From Holding Tank

At the Great Oak Landfill, the level of leachate in the holding tank was insufficient to collect a sample from the discharge port; therefore, samples were collected directly from the manhole hatch located at the top of the leachate holding tank. On September 17, 2019, samples were collected using a new properly decontaminated HDPE bucket and cotton string for analysis of both PFAS and 1,4-dioxane. H&H returned to the site on September 30, 2019, to resample for 1,4-dioxane due to issues with damage to sample containers during transport to the laboratory. During the sampling on September 30, 2019, samples were collected using a new HDPE bailer and cotton string for analysis of 1,4-dioxane. To complete the sampling, leachate was extracted from the holding tank using the bucket or bailer and transferred into the sample containers. The sampling station at the platform on top of the holding tank was covered with new HDPE sheeting. In addition, the "windmill" technique was utilized while bailing to prevent the bailer or string from contacting potential PFAS containing surfaces. Upon completion of sampling, bottles were capped, placed in Zip-lock bags, and placed into laboratory-supplied coolers filled with ice. For quality assurance/quality control (QA/QC), an equipment blank was collected during each sampling event from the bucket or bailer using PFAS-free water provided by the laboratory.

Each sample was assigned a unique identification number beginning with the first four digits of the NCDEQ permit number. Samples collected for analysis of PFAS were placed in coolers separate from samples collected for analysis of 1,4-dioxane. The sample coolers were shipped to GEL under chain-of-custody protocol for analysis as described in Section 3.3.



3.3 Laboratory Analyses

The samples from each facility were analyzed for PFAS by modified EPA Method 537.1 using Method PFAS by LCMSMS Compliant with Table B-15 of Department of Defense Quality Systems Manual (QSM) Version 5.3 and 1,4-dioxane by EPA Method 8270 Selective Ion Monitoring. The list of PFAS compounds included in the analyses was based on prior discussions between NWRA member companies and NCDEQ staff. At the request of NCDEQ, samples from Sampson County Disposal, LLC were also analyzed for PFPrOPrA by modified EPA Method 537.1.

Three items were identified during review of the laboratory QA/QC data which are discussed below:

- For sample 0403-1 (Chambers Development MSWLF), the surrogate recovery for the 1,4dioxane sample analysis was below acceptable limits. The analytical results indicated 60% surrogate recovery with an estimated sample concentration of 9.22 µg/L. If this concentration is adjusted upward based on 100% recovery instead of 60%, the estimated 1,4-dioxane concentration in the sample would be 15.4 µg/L ([9.22 µg/L x 100%] / 60% = 15.4 µg/L). Following the initial analysis, GEL re-analyzed a second portion of the sample. However, the re-analysis was performed outside the method-recommended holding time. The results of the second analysis indicated a concentration of 14.8 µg/L. Based on the adjusted initial sample analysis result and the re-analysis result, H&H concludes that there is sufficient data to conclude the concentration in the sample is reasonably on the order of approximately 15 µg/L.
- For sample 1304-1 (BFI-Charlotte Motor Speedway Landfill V), GEL inadvertently did not analyze the 1,4-dioxane sample collected on September 16, 2019. A second sample (ID 1,1A,2,2A) was collected by landfill facility personnel on December 4, 2019 and analyzed for 1,4-dioxane.



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• The equipment blank sample collected from Great Oak Landfill (sample 7607-EB) contained perfluorobutyric acid (PFBA) at a laboratory estimated concentration of 1.12 ng/L. The concentration detected was J-flagged, which means the concentration is estimated above the laboratory method detection limit but below the quantification/reporting limit. PFBA was also detected in the primary leachate sample collected from Great Oak Landfill (sample 7607). Based on these data, there is less confidence in PFBA concentrations reported for the Great Oak Landfill.

Laboratory analytical reports are included in Appendix A.

3.4 Discussion of Sampling Results and Comparison to Other Studies

The results of the laboratory analyses indicated detectable concentrations of PFOS, PFOA, and other PFAS compounds in each of the collected samples. 1,4-Dioxane was also detected in each of the samples. A summary of laboratory analytical data for the full set of constituents of concern is provided in **Table 3**.

Concentrations of PFOS and PFOA detected in the samples were compared to concentrations detected in leachate samples collected during the Michigan Study. The comparison data are summarized in **Table 4**. The results of the comparison indicated mean concentrations detected during the North Carolina Collective Study were generally similar to those detected during the Michigan Study (see inset). Variations in minimum and maximum

PFOS and PFOA Concentrations in Leachate							
Parameter Min Max Mean							
PFOS	NC	82	402	199			
(ng/L)	MI	9	960	222			
PFOA	NC	108	3,690	1,005			
(ng/L)	MI	16	3,200	881			

concentrations between the North Carolina and Michigan studies are likely a result of differing sample sizes. Comparison to published literature references (as referenced in Section 2.2) indicates that concentrations detected during the North Carolina Collective Study are also within

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ATTACHMENT D

S:\AAA-Master Projects\National Waste and Recycling Association (NWA)\NWA-001\Report\NC Collective Study Rpt 03-10-2020.docx

the range of values reported during other studies in the United States and other Western world regions, but more than an order of magnitude lower than maximum values reported for China.

Similar to the procedure followed during the Michigan Study, the analytical data and estimated

annual leachate volumes provided by each MSWLF facility were used to calculate the daily mass of PFOS and PFOA contained within landfill leachate for each facility. The calculations based on the North

	•	5						
PFOS and PFOA Daily Mass in Leachate								
Parameter Min Max Mean								
PFOS Daily	NC	0.00001	0.00014	0.00004				
Mass (lbs/day)	MI	0.00001	0.00040	0.00005				
PFOA Daily Mass (lbs/day)	NC	0.00001	0.00098	0.00013				
	MI	0.00002	0.00260	0.00022				

Carolina Collective Study data indicate a mean daily mass of less than 0.001 lbs/day of PFOS or PFOA (see inset). Comparison of estimated daily mass values for the North Carolina Collective Study to those calculated during the Michigan Study indicate generally similar values. Daily mass calculations for PFOS and PFOA are summarized in **Table 4** and depicted on **Figure 3**.

Daily mass calculations were also performed for 1,4-dioxane based on data collected during the North Carolina Collective Study. The results of the calculations indicated a mean daily mass of less than 0.1 lbs/day of 1,4-dioxane (see inset). The Michigan Study did not include analysis for 1,4-dioxane, nor were comprehensive published references identified for typical 1,4-dioxane

concentrations in landfill leachate. As such, no additional data are available for comparison. However, based on the general similarity in PFAS concentrations

1,4-Dioxane Concentration and Daily Mass in Leachate							
Parameter Min Max Mean							
1,4-Dioxane Concentration (µg/L)	14.8	469	120				
1,4-Dioxane Daily Mass (lbs/day)	0.0022	0.0944	0.0255				

reported in the North Carolina Collective Study, Michigan Study, and United States published literature, the 1,4-dioxane concentrations detected during the North Carolina Collective Study are expected to be similar to those for other MSWLFs across the United States. Daily mass calculations for 1,4-dioxane are summarized in **Table 5** and depicted on **Figure 7**.



4.0 Influence on WWTP Influent

4.1 Description of Receiving WWTPs

The MSWLFs covered under the North Carolina Collective Study each dispose of leachate via one or more publicly-owned WWTPs. H&H compiled locations for the receiving WWTPs based on information provided by each landfill. A summary of the receiving WWTP names, addresses, and National Pollutant Discharge Elimination System (NPDES) permit numbers is provided in **Table 2**. H&H determined the permitted flow for each WWTP based on information obtained from permit applications on the NCDEQ on-line Laserfiche document repository. Permitted flows are used rather than actual flows to be consistent with the approach used by NCDEQ during evaluation of the WWTP sampling data referenced below.

4.2 WWTP Sampling Data Source

In 2019, the NCDEQ DWR issued letters to publicly owned utilities with pretreatment programs and industrial dischargers in the Cape Fear River Basin requiring influent sampling for 1,4-dioxane and PFAS for three consecutive months beginning in July 2019. The sampling was performed in July, August, and September 2019. H&H retrieved the results of the sampling from the NCDEQ website (NCDEQ, 2020). Discussions in this report are based on average concentrations detected during the three monthly sampling events between July and September 2019.

The NCDEQ website contains PFAS and 1,4-dioxane data for the following WWTPs which receive leachate from landfills in the North Carolina Collective Study, including:

- City of Asheboro WWTP
- East Burlington WWTP
- Utley Creek Water Reclamation Facility
- Harnett County Lillington Plant

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ATTACHMENT D

4.3 Discussion of WWTP Influent Sampling Results and Comparison to Other Studies

The WWTP sampling data are summarized on **Table 6**. For the WWTPs that receive leachate from facilities in the North Carolina Collective Study, the concentrations of PFOS and PFOA in the influent range from 8.86 to 49.5 ng/L (based on the average of the samples collected at each WWTP). Based on documentation provided on the NCDEQ website, NCDEQ concluded that the PFOS and PFOA concentrations for these facilities would not cause levels at downstream PWS intakes that exceed the EPA Drinking Water Health Advisory Level of 70 ng/L.

For 1,4-dioxane, the average concentrations of WWTP influent range from 5.95 to 18.5 μ g/L, with the exception of one outlier which indicated a significantly higher average concentration of 163 μ g/L. Based on documentation provided on the NCDEQ website, the elevated outlier concentration is primarily attributed to an industrial discharger rather than a landfill leachate source. Overall, for the WWTPs that receive leachate from facilities in the North Carolina Collective Study, NCDEQ concluded that the 1,4-dioxane concentrations for these WWTPs are not anticipated to cause levels that exceed the EPA Drinking Water Health Advisory Level of 35 μ g/L at downstream PWS intakes.

The WWTP sampling and flow data were used to calculate the estimated daily mass of PFOS, PFOA, and 1,4-dioxane for each facility with available data. For PFOS and PFOA, the calculated daily mass values were then compared to WWTP daily mass values calculated during the Michigan Study. The results of this comparison indicated that the daily PFOS and PFOA mass for the North Carolina WWTPs are generally similar to or lower than the corresponding daily mass for the Michigan WWTPs. Daily WWTP mass calculations summarized in **Tables 6** and **7**, and depicted on **Figures 4** and **8**.

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ATTACHMENT D

4.4 Leachate Contribution to WWTP Daily Mass

In order to evaluate the relative contribution of landfill leachate to WWTP daily mass, the daily mass values calculated for leachate were compared to the daily mass values calculated for WWTP influent. The results of these calculations for the North Carolina Collective Study facilities are summarized in **Table 8**. The PFOS and PFOA data are depicted along with similar data from the

Percent Contribution to WWTP Influent Daily Mass						
Constituent	Leachate Sources	Non- Leachate Sources				
PFOS	0.7 to 2.9%	97.1 to 99.3%				
PFOA	0.6 to 10.2%	89.8 to 99.4%				
1,4-Dioxane	0.3 to 3.6%	96.4 to 99.7%				

Michigan Study on **Figures 5** and **6**, respectively. The 1,4-dioxane data are depicted on **Figure 8**. **Review of the graphical depiction demonstrates that the mass of PFOS, PFOA, and 1,4-dioxane from landfill leachate represents a minor contribution to overall WWTP influent mass**. The estimated percent contribution of landfill leachate to overall WWTP mass for the sites in the North Carolina Collective Study ranges from only 0.3 to 10.2% for PFOS, PFOA, and 1,4-dioxane (see

inset), with an average of 3.3%. The PFOS and PFOA results are corroborated by the larger data set included in the Michigan Study, which also confirms that landfill leachate represents a minor contribution to overall WWTP influent mass and

Review of the graphical depictions on Figures 5, 6, and 8 demonstrates that the mass of PFOS, PFOA, and 1,4-dioxane from landfill leachate represents a minor contribution to overall WWTP influent mass.

non-leachate sources represent a much larger contribution.



5.0 Conclusions and Recommendations

The North Carolina Collective Study included collection of leachate samples from nine MSWLF facilities located across the State of North Carolina for analysis of PFAS constituents and 1,4-dioxane. Where available, the results of the sampling were evaluated in conjunction with WWTP influent volumes and published sampling data in order to estimate the relative contribution of landfill leachate to overall WWTP influent mass of PFAS and 1,4-dioxane. The data were also evaluated with respect to the results of a larger study performed in Michigan using similar methodology.

The results of the North Carolina Collective Study clearly show that **landfill leachate represents** a minor contribution of PFOS, PFOA, and 1,4-dioxane mass to overall WWTP influent mass for these compounds. Non-leachate sources contribute significantly more mass to WWTP influent than leachate. These conclusions are supported by both the North Carolina Collective Study and the Michigan Study. Importantly, NCDEQ concluded that WWTP influent sampling data for facilities in the Cape Fear River Basin that receive leachate from landfills in the Collective Study indicate that PFOS, PFOA, and 1,4-dioxane concentrations do not pose a threat to downstream PWS intakes.

MSWLFs and WWTPs generally have an interdependent relationship for waste management (WWTPs accept leachate from MSWLFs and MSWLFs accept biosolids from WWTPs). Landfills and WWTPs are not producers of the original sources of PFAS and 1,4-dioxane. Rather, they both receive and manage PFAS contaminated waste and wastewater from households, business, and industry. MSWLFs and WWTPs are designed to manage waste in ways that are protective of human health and the environment. If long term reductions of CECs in the environment are to be achieved, then manufacturing and product utilization in society need to be addressed. The evidence provided by this report that landfill leachate represents only a small percentage of total influent mass of PFAS and 1,4-dioxane into WWTPs indicates the ubiquitous nature of these compounds in society. In spite of this ubiquitous nature, it is encouraging to note

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that in the Cape Fear River basin, NCDEQ concluded that WWTP discharges do not represent a threat to drinking water supplies in most cases.

Based on the findings of both the North Carolina Collective Study and the Michigan Study, continued work towards PFAS and 1,4-dioxane source reduction solutions, such as the United States' phase-out of PFOS and PFOA in manufacturing, is recommended. We also recommend collaboration between the solid waste and WWTP industries, NCDEQ, and the scientific community in order to identify best management practices and other solutions for safe management of wastes generated by our communities.

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ATTACHMENT D

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20

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TABLES



Table 1Literature Summary of PFOS and PFOA in Landfill LeachateNorth Carolina Collective StudyH&H Job No. NWA-001

				PFOA ¹			PFOS ²	
Source Cited	Location/ Region	Sample Size	Detection Frequency (%)	Concentration Range (ng/l) ³	Median (ng/l)	Detection Frequency (%)	Concentration Range (ng/l)	Median (ng/l)
Huset, et al (2011)	USA	5	100	380 - 1,000	490	100	56 -160	97
Allred, et al (2015)	USA	6	100	150 - 5,000	1,055	100	25 - 590	155
Lang, et al (2017)	USA	87	100	30 - 5,000	590	96	3 - 800	99
Benskin, et al (2012)	Canada	5	100	210 - 1,500	520	100	80 - 4,400	390
Kallenborn, et al (2004)	Nordic Countries	NA	NA	90 - 501	230	NA	30 - 190	80
Bossi, et al (2008)	Denmark	NA	NA	0 - 6	3	NA	0 - 4	NA
Woldegiorgis, et al (2008)	Sweden	NA	NA	40 - 1,000	540	NA	30 - 1,500	550
Busch, et al (2010)	Germany	20	95	0 - 926	57	100	0 - 235	3
Fuertes, et al (2017)	Spain	6	100	200 - 585	437	17	0 - 44	NA
Gullen, et al (2016)	Australia	17	100	19 - 2,100	450	89	0 - 100	31
Gullen, et al (2017)	Australia	97	64	17 - 7,500	600	65	13 - 2,700	220
Yan, et al (2015)	China	6	100	281 - 214,000	2,260	100	1,150 - 6,020	1,740

Notes:

1. PFOA = Perfluorooctanoic acid

2. PFOS = Perfluorooctanesulfonate

3. ng/L = nanograms per liter

Data Source: Michigan Waste & Recycling Association Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent (March 2019).

Table 2					
Landfill and WWTP Facility Information					
North Carolina Collective Study					
H&H Job No. NWA-001					

Landfill Name	NCDEQ Permit Number	Landfill Address	Estimated Annual Leachate Volume (gallons/day)	Description of Sampling Location	Receiving WWTP ¹ Name	WWTP NPDES ² Permit Number	WWTP Permitted Flow Limit (gallons/day)*	Receiving WWTP Address	Receiving WWTP River Basin
Foothills Environmental Landfill	1403-MSWLF-1998	2800 Cheraw Road Lenoir, NC 28645	24,364	Valve at Bottom of Holding Tank	Henry Fork WWTP	NC0040797	9,000,000	4014 River Road Hickory, NC	Catawba
BFI-Charlotte Motor Speedway Landfill V	1304-MSWLF-1992	5105 Morehead Road Concord, NC 28027	40,027	Direct from Lagoon	Rocky River Regional WWTP	NC0036269	26,500,000	6400 Breezy Lane Concord, NC	Yadkin Pee Dee
Chambers Development MSWLF	0403-MSWLF-2010	375 Dozer Drive Polkton, NC 28135	17,452	Valve at Bottom of Holding Tank	Anson County WWTP	NC0041408	3,500,000	1306 Hollywood Road Wadesboro, NC	Yadkin Pee Dee
Uwharrie Environmental Regional Landfill	6204-MSWLF-1995	500 Landfill Road Mt Gilead, NC 27306	31,649	Valve at Bottom of Holding Tank	Town of Troy WWTP	NC0028916	1,200,000	Troy, NC	Yadkin Pee Dee
Great Oak Landfill	7607-MSWLF-2015	3597 Old Cedar Falls Road Randleman, NC 27317	9,589	Direct from Holding Tank	City of Asheboro WWTP	NC0026123	9,000,000	1032 Bonkemeyer Dr Asheboro, NC	Cape Fear
Upper Piedmont Regional Landfill	7304-MSWLF-1997	9650 Oxford Road Rougemont, NC 27572	31,830	Valve at Bottom of Holding Tank	East Burlington WWTP	NC0023868	12,000,000	225 Stone Quarry Road Haw River, NC	Cape Fear
Wake County South Wake	9222-MSWLF-2008	6124 Old Smithfield Road	5,260	- Valve on Discharge Line	Utley Creek Water Reclamation Facility	NC0063096	6,000,000**	150 Treatment Plant Road Holly Springs, NC	Cape Fear
MSWLF	3222-IVISVVLF-2000	Apex, NC 27502	3,890	0	City of Lumberton WWTP	NC0024571	20,000,000	700 Lafayette Street Lumberton, NC	Lumber
			8,658		Harnett County Lillington Plant	NC0021636	7,500,000	175 Bain Street Lillington, NC	Cape Fear
Sampson County Disposal,	8202-MSWLF-2000	7434 Roseboro Highway	16,219		Harnett County South Plant	NC0088366	15,000,000	3224 Shady Grove Road Spring Lake, NC	Cape Fear
LLC	0202-1000001 -2000	Roseboro, NC 28382	20,411	 Valve on Discharge Line 	City of Lumberton WWTP	NC0024571	20,000,000	700 Lafayette Street Lumberton, NC	Lumber
			22,137		Not applicable - Evaporation	Not applicable	Not applicable	Not applicable	Not applicable
East Carolina Regional Landfill	0803-MSWLF-1993	1922 Republican Road Aulander, NC 27805	41,044	Valve at Bottom of Holding Tank	Tar River Regional WWTP	NC0030317	21,000,000	3031 Treatment Plant Road Rocky Mount, NC	Tar-Pamlico

Notes:

1. WWTP = wastewater treatment plant

2. NPDES = National Pollutant Discharge Elimination System

* = Permitted flow obtained from Section A.6 of latest NPDES permit application retrieved from North Carolina Department of Environmental Quality on-line Laserfiche document repository in December 2019. ** = After receiving an Authorization to Construct, the treatment capacity will increase to 8 millions of gallons per day.

Table 3Leachate Analytical DataNorth Carolina Collective StudyH&H Job No. NWA-001

	Sample ID		9222-1	1403-1	1304-1	0403-1	6204-1	7607-1	0803-1	7304-1	8202-1
	Sampling Da	te	09/18/19	09/16/19	09/16/19*	09/16/19	09/17/19	09/17/19**	09/19/19	09/17/19	09/18/19
Parameter	Landfill Name		Wake County South Wake MSWLF ¹	South Wake Environmental	BFI-Charlotte Motor Speedway Landfill V	Chambers Development MSWLF	Uwharrie Environmental Bogiopol Londfill	Great Oak Landfill	East Carolina Regional Landfill	Upper Piedmont Regional Landfill	Sampson County Disposal, LLC
	Laboratory Method	Laboratory Method Units ²		Landfill		MSWLF	Regional Landfill		_	_	-
Fluorotelomer sulfonate 4:2 (4:2 FTS)	EPA 537.1 Mod	ng/L	ND ³	ND	ND	ND	ND	ND	ND	ND	ND
Fluorotelomer sulfonate 6:2 (6:2 FTS)	EPA 537.1 Mod	ng/L	ND	ND	ND	180J ⁴	ND	ND	ND	ND	ND
Fluorotelomer sulfonate 8:2 (8:2 FTS)	EPA 537.1 Mod	ng/L	ND	ND	39.7	ND	35.8J	ND	ND	ND	ND
N-ethylperfluoro-1-octanesulfonamidoacetic acid (NEtFOSAA)	EPA 537.1 Mod	ng/L	ND	101	87.2	14.9J	68.0	15.6J	237	48.7	43.8
N-methylperfluoro-1-octanesulfonamidoacetic acid (NMeFOSAA)	EPA 537.1 Mod	ng/L	35.8J	257	258	50.5	180	42.4	230	106	104
Perfluorobutyric acid (PFBA)	EPA 537.1 Mod	ng/L	600	744	1920	831	2400	303EB ⁵	650	743	4770
Perfluorobutanesulfonate (PFBS)	EPA 537.1 Mod	ng/L	1420	4400	5260	6290	2870	72.2	3850	1420	7530
Perfluorotetradecanoic acid (PFTeDA)	EPA 537.1 Mod	ng/L	ND	ND	ND	ND	ND	ND	ND	ND	ND
Perfluorotridecanoic acid (PFTrDA)	EPA 537.1 Mod	ng/L	ND	ND	ND	ND	ND	ND	ND	ND	ND
Perfluorodecanesulfonate (PFDS)	EPA 537.1 Mod	ng/L	ND	ND	6.87J	ND	ND	7.10J	ND	14.9J	ND
Perfluorodecanoic acid (PFDA)	EPA 537.1 Mod	ng/L	17.3J	82.6	590	23.6	632	18.5J	90.8	48.0	90.9
Perfluorododecanoic acid (PFDoA)	EPA 537.1 Mod	ng/L	7.40J	ND	63.3	ND	184	ND	ND	ND	9.17J
Perfluoroheptanesulfonate (PFHpS)	EPA 537.1 Mod	ng/L	ND	6.82J	8.17J	ND	9.40J	ND	9.39J	ND	ND
Perfluoroheptanoic acid (PFHpA)	EPA 537.1 Mod	ng/L	241	571	983	249	1560	68.4	689	344	5520
Perfluorohexanesulfonate (PFHxS)	EPA 537.1 Mod	ng/L	237	794	925	218	640	59.1	536	190	424
Perfluorohexanoic acid (PFHxA)	EPA 537.1 Mod	ng/L	2940	3920	3470	2200	5540	449	3610	2350	6730
Perfluorononanesulfonate (PFNS)	EPA 537.1 Mod	ng/L	20.7	ND	ND	ND	ND	ND	ND	13.4J	ND
Perfluorononanoic acid (PFNA)	EPA 537.1 Mod	ng/L	28.8	71.4	269	15.5J	326	32.8	89.0	44.1	128
Perfluorooctanesulfonamide (PFOSA)	EPA 537.1 Mod	ng/L	ND	7.08J	11.5J	ND	ND	8.75J	17.3J	ND	ND
Perfluorooctanesulfonate (PFOS)	EPA 537.1 Mod	ng/L	82.3	296	356	84.2	356	83.9	402	254	222
Perfluorooctanoic acid (PFOA)	EPA 537.1 Mod	ng/L	803	1650	2210	345	3690	108	1640	884	1790
Perfluoropentanesulfonate (PFPeS)	EPA 537.1 Mod	ng/L	32.3	50.6	73.2	19.6	41.4	10.3J	54.7	28.1	61.0
Perfluoropentanoic acid (PFPeA)	EPA 537.1 Mod	ng/L	577	1070	2160	780	2150	159	1220	621	86400
Perfluoroundecanoic acid (PFUdA)	EPA 537.1 Mod	ng/L	ND	7.04J	30.8	ND	33.0	7.44J	ND	ND	10.2J
2,3,3,3-Tetrafluoro-2-(1,1,2,2,3,3,3-heptafluoropropoxy)-propanoic acid (PFPrOPrA) ⁶	EPA 537.1 Mod	ng/L	NA ⁷	NA	NA	NA	NA	NA	NA	NA	10800
1,4-Dioxane	EPA 8270 SIM	µg/L	30.0	99.7	214	14.8Q ⁸	357	469	157	177	184

Notes:

1. MSWLF = municipal solid waste landfill

2. ng/L = nanograms per liter; μ g/L = micrograms per liter

3. ND = Not detected above laboratory method detection limt

4. J = Estimated concentration between method detection limit and reporting limit

5. EB = Constituent was also detected in associated equipment blank sample

6. PFPrOPrA also known by trade name GenX

7. NA = Not analyzed

8. Q = Value indicates results of reanalysis outside laboratory holding time

* = BFI-Charlotte Motor Speedway Landfill V was resampled for 1,4-dioxane (new sample ID 1,1A,2,2A) on 12/4/19

** = Great Oak Landfill (sample ID 7607-1) was resampled for 1,4-dioxane analysis on 9/30/19

Table 4 PFOS and PFOA Daily Leachate Mass Calculations North Carolina Collective Study H&H Job No. NWA-001

Sampling Reference	Average Leachate Volume (gallons/day)	PFOS ¹ (ng/L) ³	PFOA ² (ng/L)	PFOS Daily Mass (Ibs/day) ⁴	PFOA Daily Mass (Ibs/day)
	North Carolina Colle	ective Study			
Wake County South Wake MSWLF ⁵	9,151	82.3	803	0.00001	0.00001
Foothills Environmental Landfill	24,364	296	1,650	0.00006	0.00006
BFI-Charlotte Motor Speedway Landfill V	40,027	356	2,210	0.00012	0.00074
Chambers Development MSWLF	17,452	84	345	0.00001	0.00005
Uwharrie Environmental Regional Landfill	31,649	356	3,690	0.00009	0.00098
Great Oak Landfill	9,589	84	108	0.00001	0.00001
East Carolina Regional Landfill	41,044	402	1,640	0.00014	0.00056
Upper Piedmont Regional Landfill	31,830	254	884	0.00007	0.00024
Sampson County Disposal, LLC*	45,288	222	1,790	0.00008	0.00068
Minimum	9,151	82	108	0.00001	0.00001
Maximum	45,288	402	3,690	0.00014	0.00098
Geometric Mean	24,152	199	1,005	0.00004	0.00013
	Michigan Stu		•	r	•
Arbor Hills Landfill	98,400	220	3,200	0.00018	0.0026
Autumn Hills RDF ⁷	54,800	380	1,300	0.00017	0.0006
Brent Run Landfill	16,400	110	540	0.00002	0.0001
C&C Expanded Sanitary Landfill	42,000	450	1,300	0.00015	0.0004
Carleton Farms Landfill	123,300	250	1,800	0.00026	0.0018
Central Sanitary Landfill	30,100	470	2,500	0.00012	0.0006
Citizen's Disposal Inc.	32,900	180	1,100	0.00005	0.0003
Dafter Sanitary Landfill	16,500	130	680	0.00002	0.0001
Eagle Valley RDF	32,900	170	490	0.00005	0.0001
Glens Sanitary Landfill	3,800	210	770	0.00001	0.00002
Granger Grand River Landfill	64,400	160	240	0.00009	0.0001
Granger Wood Street Landfill	19,200	110	470	0.00002	0.0001
K&W Landfill	17,500	170	830	0.00002	0.0001
Manistee County Landfill	4,700	220	420	0.000009	0.000016
McGill Road Landfill	13,700	170	760	0.00002	0.0001
Michigan Environs Inc. (Menominee)	13,100	100	1,400	0.00001	0.0002
Northern Oaks RDF	12,300	220	1,000	0.00002	0.0001
Oakland Heights Development	17,800	230	780	0.00003	0.0001
Orchard Hill Sanitary Landfill	12,500	110	650	0.00001	0.0001
Ottawa County Farms Landfill	82,200	530	1,800	0.0004	0.0012
People's Landfill	21,900	710	2,500	0.00013	0.0005
Pine Tree Acres RDF	74,000	430	1,800	0.0003	0.001
Pitsch Sanitary Landfill	15,000	260	1,300	0.00003	0.0002
Sauk Trail Hills Landfill	20,500	610	2,800	0.00010	0.0005
SC Holdings	16,000	410	960	0.00005	0.0001
Tri-City RDF	9,600	160	1,200	0.00001	0.0001
Venice Park RDF MH#20/Venice Park RDF MH#21**	32,900 —	<u>190</u> 630	910 1,500	0.0002	0.0007
Vienna Junction Industrial Park Sanitary Landfill	13,700	130	1,300	0.00001	0.0001
Waters Landfill	NONE	230	930	NONE	NONE
Westside RDF	60,800	160	1,300	0.00008	0.0007
Whitefeather Landfill	NONE	550	1,700	NONE	NONE
Woodland Meadows RDF -Van Buren	54,800	510	2,000	0.00023	0.0009
		270	1,900		
Riverview 003/Riverview 004/Riverview 007**	37,400	140	860	0.00004	0.0003
		8.5	38		
South Kent Outfall/South Kent Hauled**	48,000	960 130	725 16	0.0002	0.0001
Smith's Creek Landfill**	32,900	120	510	0.00003	0.0001
Minimum	3,800	9	16	0.00001	0.00002
Maximum	123,300	960	3,200	0.00040	0.00260
Geometric Mean	25,501	222	881	0.00005	0.00022

Notes:

1. PFOS = Perfluorooctanesulfonate

2. PFOA = Perfluorooctanoic acid

3. ng/L = nanograms per liter

4. lbs/day = pounds per day

5. MSWLF = municipal solid waste landfill

6. Michigan Study = Michigan Waste & Recycling Association Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent (March 2019)

7. RDF = recycling and disposal facility

* = Leachate volume does not include volume disposed of via evaporation.

** = Multiple laboratory results reported, average used for daily mass calculations.

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Table 5 1,4-Dioxane Daily Leachate Mass Calculations North Carolina Collective Study H&H Job No. NWA-001

Sampling Reference	Average Leachate Volume (gallons/day)	1,4-Dioxane (μg/L) ¹	1,4-Dioxane Daily Mass (Ibs/day) ²
North Carolina Co	ollective Study		
Wake County South Wake MSWLF ³	9,151	30.0	0.0023
Foothills Environmental Landfill	24,364	99.7	0.0203
BFI-Charlotte Motor Speedway Landfill V	40,027	214	0.0716
Chambers Development MSWLF	17,452	14.8Q ⁴	0.0022
Uwharrie Environmental Regional Landfill	31,649	357	0.0944
Great Oak Landfill	9,589	469	0.0376
East Carolina Regional Landfill	41,044	157	0.0538
Upper Piedmont Regional Landfill	31,830	177	0.0471
Sampson County Disposal, LLC*	45,288	184	0.0696
Minimum	9,151	14.8	0.0022
Maximum	45,288	469	0.0944
Geometric Mean	24,152	120	0.0255

Notes:

1. $\mu g/L = micrograms per liter$

2. lbs/day = pounds per day

3. MSWLF = municipal solid waste landfill

4. Q = value indicates results of reanalysis outside laboratory holding time

* = Leachate volume is representative of volume disposed at WWTPs.

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Table 6 PFOS and PFOA Daily WWTP Mass Calculations North Carolina Collective Study H&H Job No. NWA-001

Facility	WWTP ¹ Permitted Flow Limit (gallons/day)*	PFOS ² Concentration (ng/l) ⁴	PFOA ³ Concentration (ng/l)	PFOS Daily Mass (Ibs/day) ⁵	PFOA Daily Mass (Ibs/day)
	eive leachate from land		Collective Study		
City of Asheboro WWTP	9,000,000	10.6	19.3	0.0008	0.0014
East Burlington WWTP	12,000,000	49.5	39.6	0.0050	0.0040
Utley Creek Water Reclamation Facility	6,000,000**	10	9.8	0.0005	0.0005
Harnett County Lillington Plant	7,500,000	8.86	20.2	0.0006	0.0013
Michigan Study ⁶	WWTPs that receive lease	achate from landfills	included in Study		
Menominee	3,200,000	5.6	12	0.0001	0.0003
Clinton River	30,600,000	7.68	4.94	0.0019	0.0013
Genesee Co-Ragnone	25,900,000	5.22	4	0.0012	0.0009
GLWA	650,000,000	7.54	6.02	0.0406	0.0324
Grand Rapids	61,100,000	12.7	5.06	0.0066	0.0026
Holland	12,000,000	3.79	8.93	0.0004	0.0009
Lansing	35,000,000	ND ⁷	4.98	ND	0.0014
Sandusky	2,550,000	7.98	12.2	0.0002	0.0003
Three Rivers	2,750,000	7.39	21.44	0.0002	0.0005
Wyoming	22,000,000	6.2 to 26.4	5.08 to 25	0.0048	0.0046
ÝCUA	51,200,000	4.8 to 7.51	12	0.0032	0.0051
Michigan Study W	WTPs that receive leac	hate from landfills no	t included in Study		
Bay City	18,000,000	18.2	4.87	0.0027	0.0007
Downriver	125,000,000	22.2	7.2	0.0230	0.0075
Flint	50,000,000	62.4	10.3	0.0258	0.0043
Kalamazoo	53,500,000	ND	ND	ND	ND
Muskegon Co Metro	43,000,000	10.5 to 24.3	11.7 to 36.9	0.0086	0.0131
North Kent S A	8,000,000	31.1	11.2	0.0021	0.0007
Port Huron	20,000,000	19.5	64.6	0.0032	0.0107
S Huron Valley UA (SHUVA)	24,000,000	ND	3.76	ND	0.0007

Notes:

1. WWTP = wastewater treatment plant

2. PFOS = Perfluorooctanesulfonate

3. PFOA = Perfluorooctanoic acid

4. ng/L = nanograms per liter

5. lbs/day = pounds per day

6. Michigan Study = Michigan Waste & Recycling Association Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent (March 2019)

7. ND = not detected

* = Permitted flow obtained from Section A.6 of latest National Pollutant Discharge Elimination System permit application retrieved from North Carolina Department of Environmental Quality on-line Laserfiche document repository in December 2019.

** = After receiving an Authorization to Construct, the treatment capacity will increase to 8 millions of gallons per day. The

lower value of 6 millions of gallons per day was conservatively used for concentration calculations.

Table only shows facilities for which sampling data are available.

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For Michigan sites, daily mass calculations performed using maximum value where multiple data are available. For North Carolina sites, concentrations shown and associated daily mass calculations are based on average values for three sampling events performed between July and September 2019.

Table 7 1,4-Dioxane Daily WWTP Mass Calculations North Carolina Collective Study H&H Job No. NWA-001

Facility	WWTP ¹ Permitted Flow Limit (gallons/day)*	1,4-Dioxane Concentration (μg/l) ²	1,4-Dioxane Daily Mass (Ibs/day) ³
WWTPs that receive leachate from lar	dfills in North Carolina	Collective Study	
City of Asheboro WWTP	9,000,000	163	12.2927
East Burlington WWTP	12,000,000	18.5	1.8583
Utley Creek Water Reclamation Facility	6,000,000**	7.3	0.3635
Harnett County Lillington Plant	7,500,000	5.95	0.3729

Notes:

1. WWTP = wastewater treatment plant

2. µg/L = micrograms per liter

3. lbs/day = pounds per day

* = Permitted flow obtained from Section A.6 of latest National Pollutant Discharge Elimination System permit application retrieved from North Carolina Department of Environmental Quality on-line Laserfiche document repository in December 2019. ** = After receiving an Authorization to Construct, the treatment capacity will increase to 8 millions of gallons per day. The lower value of 6 millions of gallons per day was conservatively used for concentration calculations.

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Concentrations shown and associated daily mass calculations are based on average values for three sampling events performed between July and September 2019.

Table 8 Percent of WWTP Daily Mass Contributed by Landfill Leachate North Carolina Collective Study H&H Job No. NWA-001

	Average	Beechdag	WWTP Permitted			Concentration Data	1	Daily Ma	iss Data	Percentage of WWTP Influent	
Landfill Name	Leachate Volume (gallons/day)	Receiving WWTP ¹ Name	Flow Limit (gallons per day)*	Constituent	Concentration Units ²	Landfill Leachate Concentration	WWTP Influent Concentration	Landfill Leachate Daily Mass (Ibs/day) ³	WWTP Influent Daily Mass (Ibs/day)	Daily Mass Associated with Landfill Leachate***	
		Utley Creek Water		PFOS ⁵	ng/L	82.3	10	0.00000	0.0005	0.7%	
	5,260	Reclamation	6,000,000**	PFOA ⁶	ng/L	803	9.8	0.00004	0.0005	7.2%	
Wake County	0,200	Facility	0,000,000	PFOS+PFOA	ng/L	885	20	0.00004	0.0010	3.9%	
South Wake				1,4-Dioxane	µg/L	30	7.3	0.00132	0.3635	0.4%	
MSWLF ⁴				PFOS	ng/L	82.3	NS'	0.00000	NS	NS	
WOWLI	3,890	City of Lumberton	20,000,000	PFOA	ng/L	803	NS	0.00003	NS	NS	
	3,690	WWTP	20,000,000	PFOS+PFOA	ng/L	885	NS	0.00003	NS	NS	
				1,4-Dioxane	µg/L	30	NS	0.00098	NS	NS	
				PFOS	ng/L	296	NS	0.00006	NS	NS	
Foothills		Henry Fork		PFOA	ng/L	1650	NS	0.00034	NS	NS	
Environmental	24,364	WWTP	9,000,000	PFOS+PFOA	ng/L	1946	NS	0.00040	NS	NS	
Landfill				1,4-Dioxane	µg/L	99.7	NS	0.02030	NS	NS	
BFI-Charlotte		Rocky River		PFOS PFOA	ng/L	356 2210	NS NS	0.00012 0.00074	NS NS	NS NS	
Motor Speedway	40,027	Regional WWTP	26,500,000	PFOA PFOS+PFOA	ng/L ng/L	2210 2566	NS	0.00074	NS NS	NS NS	
Landfill V		Regional WWIT		1,4-Dioxane	μg/L	2300	NS	0.07157	NS	NS	
Chambers				PFOS	ng/L	84.2	NS	0.00001	NS	NS	
Development	17,452	Anson County	3,500,000	PFOA	ng/L	345	NS	0.00005	NS	NS	
MSWLF		WWTP	- ,	PFOS+PFOA	ng/L	429	NS	0.00006	NS	NS	
				1,4-Dioxane	µg/L	14.8Q ⁸	NS	0.00216	NS	NS	
Uwharrie				PFOS	ng/L	356	NS	0.00009	NS	NS	
Environmental	31.649	Town of Troy	1.200.000	PFOA	ng/L	3690	NS	0.00098	NS	NS	
Regional Landfill	31,049	WWTP	1,200,000	PFOS+PFOA	ng/L	4046	NS	0.00107	NS	NS	
rtegional Eanann				1,4-Dioxane	µg/L	357	NS	0.09441	NS	NS	
				PFOS	ng/L	83.9	10.6	0.00001	0.0008	0.8%	
		City of Asheboro		PFOA	ng/L	108	19.3	0.00001	0.0014	0.6%	
Great Oak Landfill	9,589	WWTP	9,000,000	PFOS+PFOA	ng/L	192	29.9	0.00002	0.0022	0.7%	
				1,4-Dioxane	µg/L	469	163	0.03758	12.2927	0.3%	
				PFOS	24 M	402	NS	0.00014	NS	NS	
East Carolina		Tar River Regional		PFOA	ng/L ng/L	1640	NS	0.00056	NS	NS	
Regional Landfill	41,044	WWTP	21,000,000	PFOS+PFOA	ng/L	2042	NS	0.00070	NS	NS	
rtogional zanami				1.4-Dioxane	µg/L	157	NS	0.05384	NS	NS	
Harris Blades at		Fact Deally stars		PFOS	ng/L	254	49.5	0.00007	0.0050	1.4%	
Upper Piedmont Regional Landfill	31,830	East Burlington WWTP	12,000,000	PFOA PFOS+PFOA	ng/L	884 1138	39.6 89.0	0.00024 0.00030	0.0040	5.9% 3.4%	
Regional Landilli		VVVIP		1,4-Dioxane	ng/L µg/L	177	18.5	0.00030	1.8583	2.5%	
				1,4-Dioxane	µg/L	177	10.5	0.04707	1.0303	2.378	
				PFOS	ng/L	222	8.86	0.00002	0.0006	2.9%	
	0.050	Harnett County		PFOA	ng/L	1790	20.2	0.00013	0.0013	10.2%	
	8,658	Lillington Plant	7,500,000	PFOS+PFOA	ng/L	2012	29.0	0.00015	0.0018	8.0%	
		-		1,4-Dioxane	µg/L	184 10800	5.95 NS	0.01331	0.3729	3.6%	
				PFPrOPrA ⁹	ng/L	10600	6VI	0.00078	NS	NS	
				PFOS	ng/L	222	NS	0.00003	NS	NS	
		Harnett County		PFOA	ng/L	1790	NS	0.00024	NS	NS	
	16,219	South Plant	15,000,000	PFOS+PFOA	ng/L	2012	NS	0.00027	NS	NS	
				1,4-Dioxane	µg/L	184	NS	0.02494	NS	NS	
Sampson County				PFPrOPrA	ng/L	10800	NS	0.00146	NS	NS	
Disposal, LLC				PFOS	ng/L	222	NS	0.00004	NS	NS	
		City of Lumberton		PFOA	ng/L	1790	NS	0.00031	NS	NS	
	20,411	WWTP	20,000,000	PFOS+PFOA	ng/L	2012	NS	0.00034	NS	NS	
				1,4-Dioxane	µg/L	184	NS	0.03138	NS	NS	
				PFPrOPrA	ng/L	10800	NS	0.00184	NS	NS	
				PFOS	ng/L	222	NA ¹⁰	NA	NA	NA	
				PFOA	ng/L	1790	NA	NA	NA	NA	
1	22,137	Evaporation	Not applicable	PFOS+PFOA	ng/L	2012	NA	NA	NA	NA	
				1,4-Dioxane	μg/L	184	NA	NA	NA	NA	
				PFPrOPrA	ng/L	10800	NA	NA	NA	NA	

Notes

1. WWTP = wastewater treatment plant WWTF = wastewater treatment plant
 ng/L = morgarans per liter; µg/L = micrograms per liter
 lbs/day = pounds per day
 MSWLF = municipal solid waste landfill
 FPOS = perfluorooctanesulfonate

6. PFOA = perfluorooctanoic acid7. NS = no sampling data available

8. Q = value indicates results of reanalysis outside laboratory holding time

9. PFPrOPrA = 2,3,3,3-tetrafluoro-2-(1,1,2,2,3,3,3-heptafluoropropoxy)-propanoic acid (trade name GenX)

10. NA = not applicable

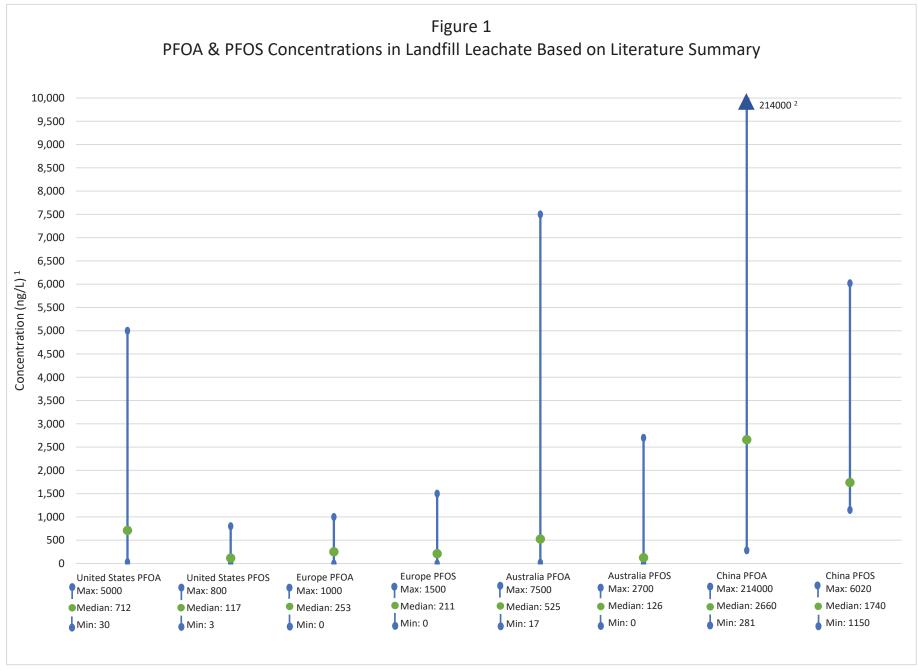
Permitted flow obtained from Section A.6 of latest National Pollutant Discharge Elimination System permit application retrieved from North Carolina Department of Environmental Quality on-line Laserfiche

document repository in December 2019. ** = After receiving an Authorization to Construct, the treatment capacity will increase to 8 millions of gallons per day. The lower value of 6 millions of gallons per day was conservatively used for concentration

calculations.
*** = WWTP mass attributed to landfill leachate only includes contributions from landfills covered under the North Carolina Collective Study.

FIGURES





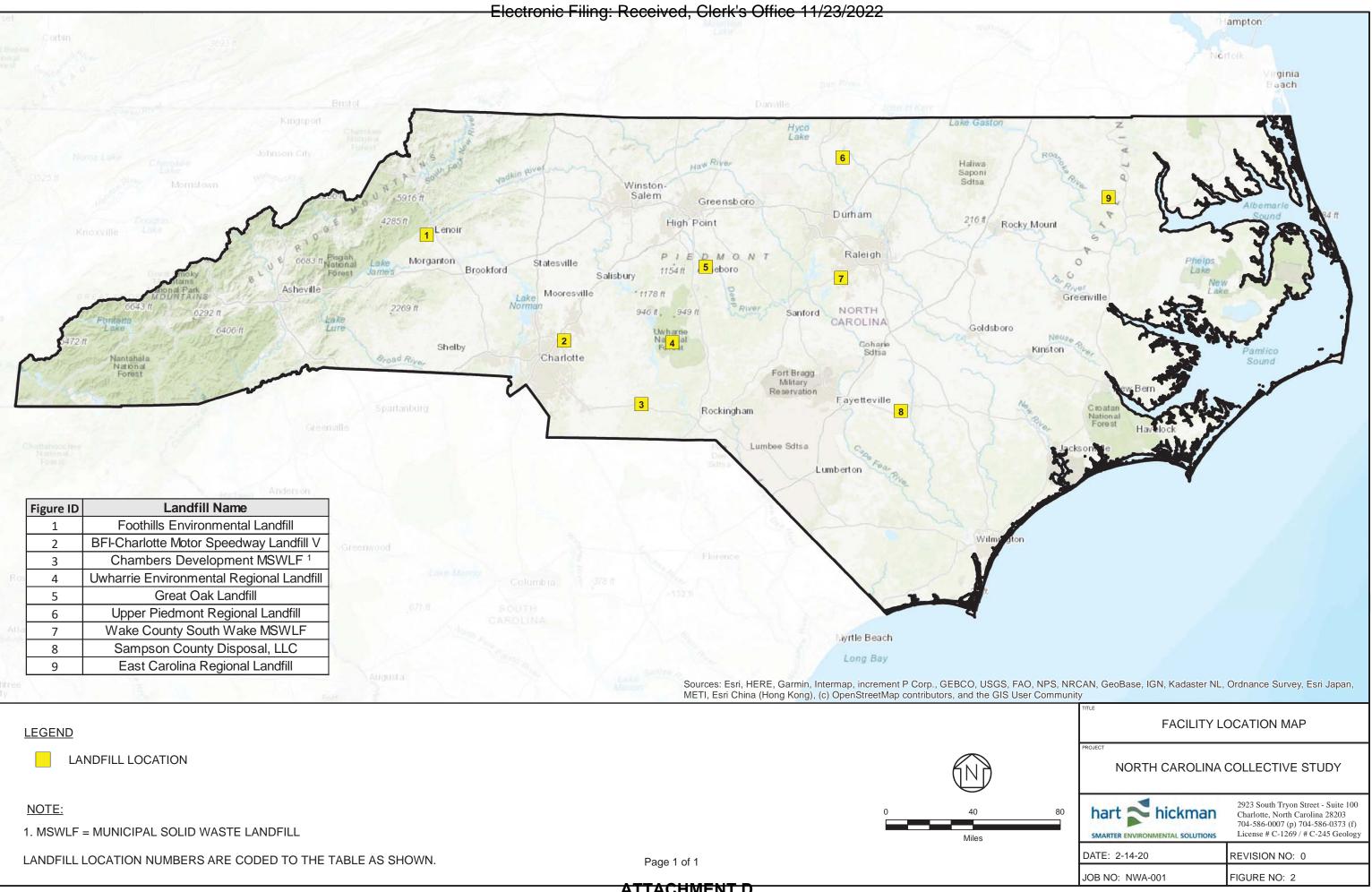
Notes:

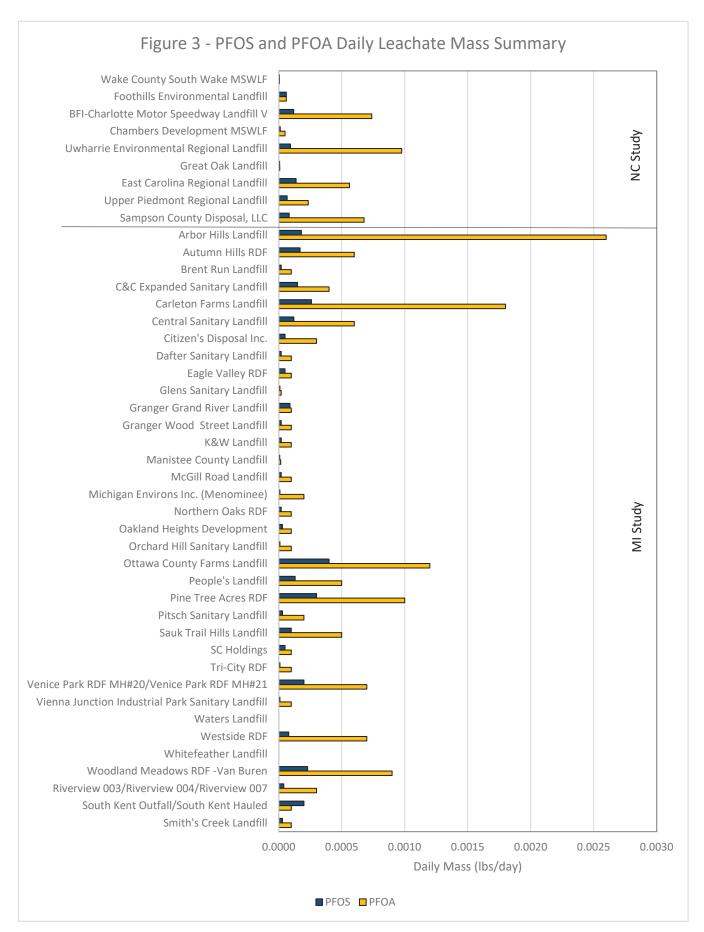
ng/L = nanograms per liter
 Concentration is beyond the scale of the graph (>20 times scale of graph)

Source: Michigan Waste & Recycling Association Statewide Study on

Landfill Leachage PFP And PFP MENT Voter Resource Recovery Facility Influent (March 2019)

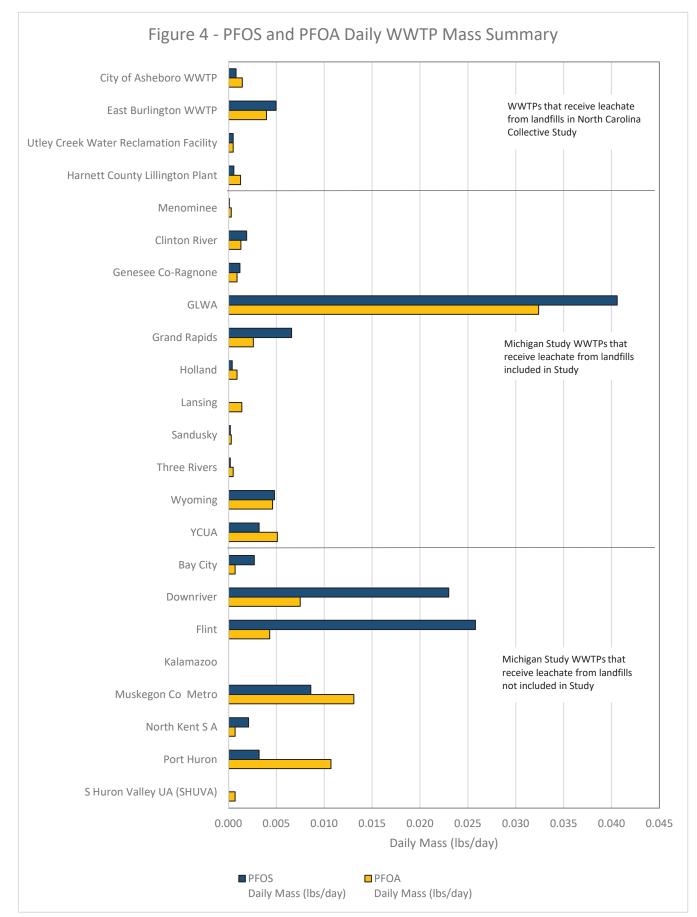
Page 1 of 1





Notes:

1. MI Study = Michigan Waste & Recycling Association Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent (March 2019) 2. lbs/day = pounds per day



ATTACHMENT D

Notes:

MI Study = Michigan Waste & Recycling Association
 Statewide Study on Landfill Leachate PFOA and PFOS Impact on
 Water Resource Recovery Facility Influent (March 2019)
 Ibs/day = pounds per day

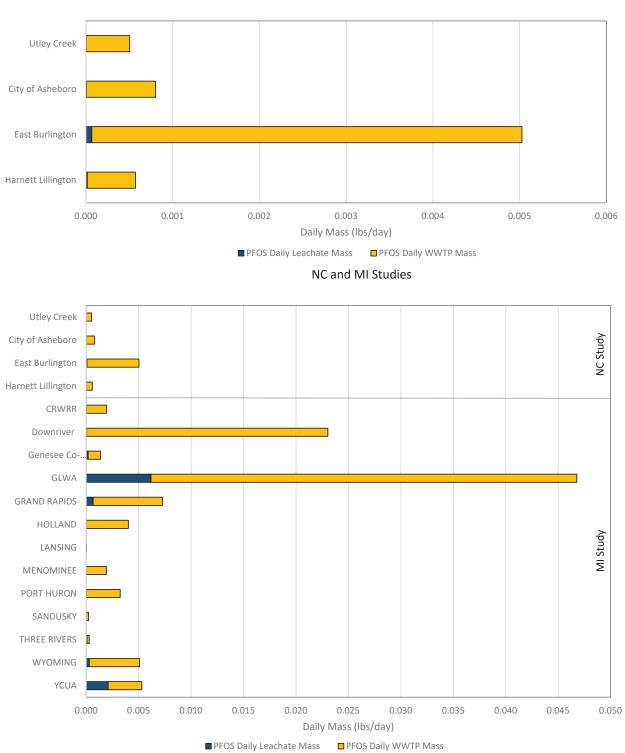
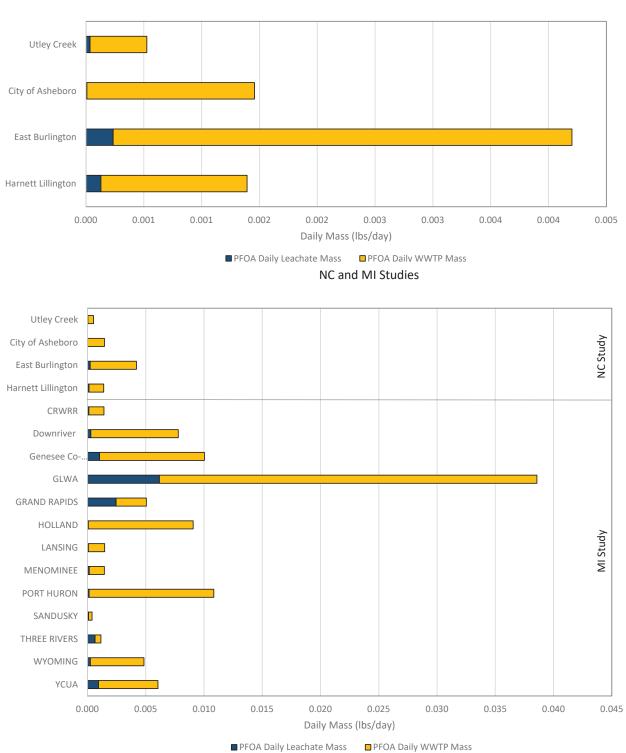


Figure 5 - PFOS Landfill Leachate Contribution to WWTP Daily Mass

NC Study

Notes: 1. MI Study = Michigan Waste & Recycling Association Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent (March 2019) 2. Ibs/day = pounds per day



ATTACHMENT D

Figure 6 - PFOA Landfill Leachate Contribution to WWTP Daily Mass

NC Study

 MI Study = Michigan Waste & Recycling Association
 Statewide Study on Landfill Leachate PFOA and PFOS Impact on Water Resource Recovery Facility Influent (March 2019)
 Ibs/day = pounds per day

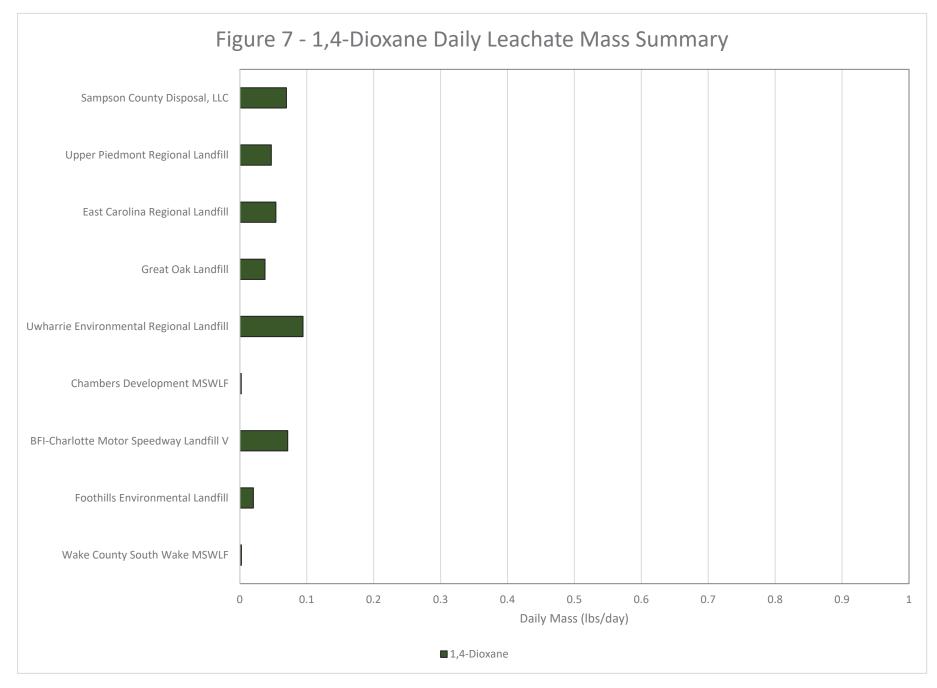
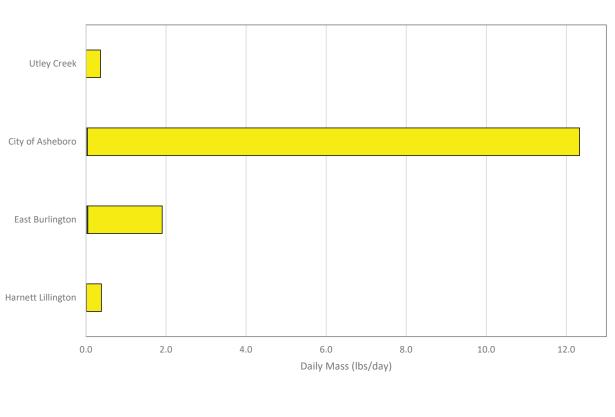


Figure 8 - 1,4-Dioxane Landfill Leachate Contribution to WWTP Daily Mass



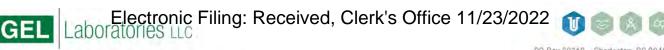
NC Study

■ 1,4-Dioxane Daily Leachate Mass □ 1,4-Dioxane Daily WWTP Mass

APPENDIX A

LABORATORY ANALYTICAL REPORTS





a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for South Wake MSWLF Work Order: 490673

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

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Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490673 GEL Work Order: 490673

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Electroni

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forSouth Wake MSWLF		
Client Sample ID:	9222-1	Project:	NWRA00119
Sample ID:	490673001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	18-SEP-19 10:00		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Anal	lyst Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs by LO	C-MS/MS	"As Received"										
N-ethylperfluoro-1-	U	ND	13.2	40.0	ng/L	0.200	1	JLS	10/04/19	1109	1921240	1
octanesulfonamidoacetic acid (N	1-				U							
EtFOSAA)	_				_							
N-methylperfluoro-1-	J	35.8	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (N MeFOSAA)	N-											
Perfluorobutanesulfonic acid (PI	FBS)	1420	6.60	17.8	ng/L	0.200	1					
Perfluorodecanesulfonic acid	U	ND	6.60	19.4	ng/L	0.200						
(PFDS)	C				8							
Perfluorodecanoic acid (PFDA)	J	17.3	7.80	20.0	ng/L	0.200	1					
Perfluorododecanoic acid (PFDc	DA) J	7.40	6.60	20.0	ng/L	0.200	1					
Perfluoroheptanesulfonic acid	U	ND	6.60	19.0	ng/L	0.200	1					
(PFHpS)					-							
Perfluoroheptanoic acid (PFHpA	A)	241	6.60	20.0	ng/L	0.200						
Perfluorohexanesulfonic acid (PFHxS)		237	6.60	18.2	ng/L	0.200	1					
(PFHXS) Perfluorohexanoic acid (PFHxA)	2940	6.60	20.0	ng/L	0.200	1					
Perfluorononanesulfonic acid)	20.7	7.00	19.2	ng/L	0.200						
(PFNS)		20.7	7.00	19.2	iig/L	0.200	1					
Perfluorononanoic acid (PFNA)		28.8	6.60	20.0	ng/L	0.200	1					
Perfluorooctanesulfonamide	U	ND	6.60	18.6	ng/L	0.200	1					
(PFOSA)												
Perfluorooctanesulfonic acid (PI	FOS)	82.3	8.00	20.0	ng/L	0.200						
Perfluorooctanoic acid (PFOA)		803	7.00	20.0	ng/L	0.200						
Perfluoropentanesulfonic acid		32.3	6.60	18.8	ng/L	0.200	1					
(PFPeS) Perfluoropentanoic acid (PFPeA	`	577	6.60	20.0	ng/L	0.200	1					
Perfluoroundecanoic acid (PFUc	·	ND	6.60	20.0	ng/L	0.200						
Fluorotelomer sulfonate 8:2 (8:2	,	ND	132	20.0 384	ng/L	0.200		пс	10/02/19	0622	1921240	2
FTS)	. 0	ND	152	564	ng/L	0.200	10	JLO	10/02/19	0022	1921240	2
Perfluorobutyric acid (PFBA)		600	66.0	200	ng/L	0.200	10					
Perfluorotetradecanoic acid	U	ND	66.0	200	ng/L	0.200						
(PFTeDA)					U							
Perfluorotridecanoic acid (PFTrl	,	ND	66.0	200	ng/L	0.200						
Fluorotelomer sulfonate 4:2 (4:2	U	ND	1320	3760	ng/L	0.200	100	JLS	10/02/19	1016	1921240	3
FTS)		ND	1226	2000	/*	0.000	100					
Fluorotelomer sulfonate 6:2 (6:2 FTS)	U	ND	1320	3800	ng/L	0.200	100					
Semi-Volatile-GC/MS												

Semi-Volatile-GC/MS

Electronic Filing Elecared RAFORIES

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
Contact:	Arlington, Virginia 22202 Mr. Jim Riley		
Project:	Analytical forSouth Wake MSWLF		
Tiojeci.	Allarytical forsouth wake wis will		
Client Sample ID:	9222-1	Project:	NWRA00119
Sample ID:	490673001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Analyst Date	Time Batch	Method	
Semi-Volatile-GC/MS											
SW846 8270 SIM 1,4-D	oioxane in Li	quid "As Received"									
1,4-Dioxane		30.0	10.0	20.0	ug/L	0.200	10	JMB3 09/24/19	1314 1919444	4	
The following Prep Met	hods were pe	erformed:									
Method	Description	n		Analyst	Date		Time	Prep Batch			
EPA 537.1 Mod, PFAS, Comp	ol PFCs Extract	ion in Liquid		LM1	09/27/19)	0830	1921239			
SW846 3535A	SW8270E SI	M Prep 1,4-Dioxane		SJW1	09/23/19)	1200	1919441			
The following Analytical Methods were performed:											
Method	Description	l				Analys	t Cor	nments			
1	EPA 537.1 M	od, PFAS, Compliant with QSM T	able B-1	5							
2	EPA 537.1 M	od, PFAS, Compliant with QSM T	able B-1	5							
3	EPA 537.1 M	od, PFAS, Compliant with QSM 7	able B-1	5							
4	SW846 3535A	A/8270E SIM									
Surrogate/Tracer Recover	ery Test				Result	Nomin	al	Recovery%	Acceptable L	imits	
1,4-Dioxane-d8	SW846 Receive	8270 SIM 1,4-Dioxane in Liquid	"As		26.2 ug/L	40).0	66*	(70%-130%))	
Notes:											

Column headers are defined as follows:DF: Dilution FactorLc/LC: Critical LevelDL: Detection LimitPF: Prep FactorMDA: Minimum Detectable ActivityRL: Reporting LimitMDC: Minimum Detectable ConcentrationSQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 7

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia

Contact: Mr. Jim Riley Workorder: 490673

Parmname NOM Sample Qual QC Units RPD/D% **REC%** Range Anlst Date Time Perfluorinated Compounds 1921240 Batch OC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 18.2 15.7 JLS 10/02/19 06:05 ng/L 86 (60%-145%) FTS) Fluorotelomer sulfonate 6:2 (6:2 18.5 20.4ng/L 110 (56% - 143%)FTS) Fluorotelomer sulfonate 8:2 (8:2 18.7 17.5 ng/L 94 (57% - 138%)FTS) N-ethylperfluoro-1-19.5 19.3 ng/L 99 (63%-131%) octanesulfonamidoacetic acid (N-EtFOSAA) N-methylperfluoro-1-19.5 21.5 ng/L 111 (62%-133%) octanesulfonamidoacetic acid (N-MeFOSAA) Perfluorobutanesulfonic acid 17.2 16.6 ng/L 96 (68%-136%) (PFBS) Perfluorobutyric acid (PFBA) 19.5 19.7 ng/L 101 (70% - 133%)Perfluorodecanesulfonic acid 18.8 16.8 ng/L 89 (53%-142%) (PFDS) Perfluorodecanoic acid (PFDA) 19.5 18.0 ng/L 93 (62% - 135%)19.5 19.5 100 Perfluorododecanoic acid (PFDoA) ng/L (66%-131%) Perfluoroheptanesulfonic acid 18.5 18.1 ng/L 98 (66%-138%) (PFHpS) Perfluoroheptanoic acid (PFHpA) 19.5 17.9 92 ng/L (67%-135%) Perfluorohexanesulfonic acid 17.7 14.5 ng/L 82 (64%-137%) (PFHxS) 19.5 18.9 Perfluorohexanoic acid (PFHxA) ng/L 97 (67%-133%)

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QC Summary

Workorder: 490673		<u>x</u>	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	<u> </u>					
Parmname	NOM	Sample Q	ual QC	Units	RPD/D%	REC%	Range	Anlst	Page 2 of 7 Date Time
Perfluorinated Compounds Batch 1921240	110111	Sample Q		Cints	<u> </u>	<u>KEC 70</u>	Kange	Allist	Date Thit
Perfluorononanesulfonic acid (PFNS)	18.7		17.5	ng/L		93	(66%-130%)) JLS	10/02/19 06:05
Perfluorononanoic acid (PFNA)	19.5		21.1	ng/L		108	(66%-134%))	
Perfluorooctanesulfonamide (PFOSA)	19.5		21.5	ng/L		111	(68%-137%))	
Perfluorooctanesulfonic acid (PFOS)	19.5		19.8	ng/L		102	(61%-131%))	
Perfluorooctanoic acid (PFOA)	19.5		18.8	ng/L		97	(63%-145%))	
Perfluoropentanesulfonic acid (PFPeS)	18.3		16.5	ng/L		90	(62%-139%))	
Perfluoropentanoic acid (PFPeA)	19.5		19.3	ng/L		99	(69%-132%))	
Perfluorotetradecanoic acid (PFTeDA)	19.5		22.5	ng/L		115	(65%-143%))	
Perfluorotridecanoic acid (PFTrDA)	19.5		19.9	ng/L		102	(57%-149%))	
Perfluoroundecanoic acid (PFUdA)	19.5		19.1	ng/L		98	(65%-134%))	
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6		20.5	ng/L	26	116	(0%-35%))	10/02/19 06:14
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9		17.6	ng/L	14	98	(0%-36%))	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1		19.9	ng/L	13	110	(0%-39%))	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8		20.1	ng/L	4	107	(0%-25%))	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8		21.9	ng/L	2	116	(0%-26%))	

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QC Summary

		<u> </u>	IIIIIai	<u> </u>				
Workorder: 490673				_				Page 3 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	. 4	103	(0%-30%) JLS	10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	. 10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	. 4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	. 1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	. 4	98	(0%-29%)	

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QC Summary

Workorder: 490673		-							Page 4 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240									
Perfluoropentanoic acid (PFPeA)	18.8			20.0	ng/L	3	106	(0%-30%) JLS	3 10/02/19 06:14
Perfluorotetradecanoic acid (PFTeDA)	18.8			20.6	ng/L	9	109	(0%-30%)	
Perfluorotridecanoic acid (PFTrDA)	18.8			17.7	ng/L	11	94	(0%-35%)	
Perfluoroundecanoic acid (PFUdA)	18.8			21.2	ng/L	10	112	(0%-28%)	
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)			U	ND	ng/L				10/02/19 05:56
Fluorotelomer sulfonate 6:2 (6:2 FTS)			U	ND	ng/L				
Fluorotelomer sulfonate 8:2 (8:2 FTS)			U	ND	ng/L				
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)			U	ND	ng/L				
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)			U	ND	ng/L				
Perfluorobutanesulfonic acid (PFBS)			U	ND	ng/L				
Perfluorobutyric acid (PFBA)			U	ND	ng/L				
Perfluorodecanesulfonic acid (PFDS)			U	ND	ng/L				
Perfluorodecanoic acid (PFDA)			U	ND	ng/L				
Perfluorododecanoic acid (PFDoA)			U	ND	ng/L				
Perfluoroheptanesulfonic acid (PFHpS)			U	ND	ng/L				

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QC Summary

Workorder: 490673								Page	e 5 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D% REC	C% Range	Anlst	Date	
Perfluorinated CompoundsBatch1921240									
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L			JLS	10/02/1	9 05:56
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L					
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L					
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L					
Perfluorononanoic acid (PFNA)		U	ND	ng/L					
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L					
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L					
Perfluorooctanoic acid (PFOA)		U	ND	ng/L					
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L					
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L					
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L					
Perfluorotridecanoic acid (PFTrDA)		U	ND	ng/L					
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L					
Semi-Volatile-GC/MS Batch 1919444									
QC1204387349 LCS **1,4-Dioxane-d8	4.00		3.55	ug/L	89	9 (70%-130%	6) JMB3	09/24/1	9 12:24

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QC Summary

Workorder: 49	0673										Page	e 6 of 7
Parmname		NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Semi-Volatile-GC/MS Batch 1919												
QC1204387350 **1,4-Dioxane-d8	LCSD	4.00			3.18	ug/L		79	(70%-130%)	JMB3	09/24/1	9 12:49
QC1204387348 1,4-Dioxane	MB			U	ND	ug/L					09/24/1	9 11:59
**1,4-Dioxane-d8		4.00			3.05	ug/L		76	(70%-130%)			

Notes:

The Qualifiers in this report are defined as follows:

** Analyte is a surrogate compound

- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N/A RPD or %Recovery limits do not apply.

N1 See case narrative

- ND Analyte concentration is not detected above the detection limit
- NJ Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier
- P Organics--The concentrations between the primary and confirmation columns/detectors is >40% different. For HPLC, the difference is >70%.
- Q One or more quality control criteria have not been met. Refer to the applicable narrative or DER.

R Sample results are rejected

- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.
- UJ Compound cannot be extracted

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QC Summary

Workord	der: 490673										Pag	ge 7 of 7
Parmnan	ne	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Х	Consult Case Narrative, I	Data Summary package	, or Project N	Manager	concerning t	his qualifi	ier					

Y QC Samples were not spiked with this compound

^ RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.

h Preparation or preservation holding time was exceeded

N/A indicates that spike recovery limits do not apply when sample concentration exceeds spike conc. by a factor of 4 or more or %RPD not applicable. ^ The Relative Percent Difference (RPD) obtained from the sample duplicate (DUP) is evaluated against the acceptance criteria when the sample is greater than five times (5X) the contract required detection limit (RL). In cases where either the sample or duplicate value is less than 5X the RL, a control limit of +/- the RL is used to evaluate the DUP result.

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490673

GC/MS Semivolatile

<u>Product:</u> Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1919444

Preparation Method: SW846 3535A **Preparation Procedure:** GL-OA-E-073 REV# 2 **Preparation Batch:** 1919441

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490673001	9222-1
1204387348	Method Blank (MB)
1204387349	Laboratory Control Sample (LCS)
1204387350	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Surrogate Recoveries

Sample (See Below) did not meet surrogate recovery acceptance criteria. The sample was analyzed at a dilution. As a result, one or more surrogates were diluted out of the acceptance limits.

Sample	Analyte	Value
490673001 (9222-1)	1, 4-Dioxane-d8	66* (70%-130%)

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

Technical Information

Sample Dilutions

Sample 490673001 (9222-1) was diluted due to the presence of non-target analytes. The data from the dilution are reported.

Electronic Filing: Received, Clerk's Office 11/23/2022 <u>LCMSMS-Misc</u>

<u>Product:</u> The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 <u>Analytical Procedure:</u> GL-OA-E-076 REV# 7 <u>Analytical Batches:</u> 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490673001	9222-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490673001 (9222-1).

Amelanta	490673
Analyte	001
Fluorotelomer sulfonate 4:2 (4:2 FTS)	100X
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Fluorotelomer sulfonate 8:2 (8:2 FTS)	10X
Perfluorobutyric acid (PFBA)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

<u> </u>	·	1	T				=lect	ipi	ųς	ЦŊ	y	Ŗε	ęce	IV.	eμ,	er	<u>Ƙ S</u>		Щ	:е _г	11	/23	/20	JZZ							
GEL Laboratories, LLC 2040 Savage Road Charleston, SC 29407	Phone: (843) 556-8171 Fax: (843) 766-1178	(Fill in the number of containers for each test)	Preservative Type (6)				required for sample specific QC									V: (Subject to Surcharge)		[] Level 2 [] Level 3 [] Level 4		[] No Cooler Temp:C	[,] Others - 15 Stables Advector Advector			al		Please provide any additional details	below regarding handling and/or disposal commerce of a - Origin of completed managed	of site collected from, odd matrices, etc.)			
ity Analytics		Analysis Requested (5)		PEAS 21 cmnd list hv EPA 537 mod				×								TAT Requested: Normal: X Rush: Specify:	[] Yes [X] No	Select Deliverable: [] C of A [] QC Summary [] level 1	emarks:	ing Use Only: Custody Seal 1	enter Pacifice Central Mountain	ab, C = Composite		 Matrix Lodes: DW=Druking Water, GW=Groundwater, SW=Surface Water, WW=Water, W=Water, W=Wa	6.) Preservative Type: Ha = Hydrochloric Acid, NI = Nitric Acid, SH = Sodium Hydroxide, SA = Solium Hydroxide, AA = Ascorbic Acid, HX = Hixtane, ST = Sodium Thiosultiste, if no preservative is added = leave field hand.		estos, herviltium, irritants, other				
Laboratories LLC Chemistry I Radiochemistry Radiobioassay Speciaty Analytics	Manager:	Samp	Should this	sample be	COM 45	0 10 0 0	étionoibe Raoiope In organist An organist									TAT Re	Fax Results: [] Yes	D Select Delive	Additional Remarks.	For Lab Rec	ownipie Concentral Little Colles X] Eastern	ike Duplicate Sample, \mathbf{G} = \mathbf{Gn}		diment, SL∞Sludge, SS≃Sofid	, out us // = / us - 1). . ST = Sodium Thiosultate, If 1	$\frac{\text{Other}}{\text{OT} = \text{Other} / 11 \text{element}}$	(i.e.: High/lov	misc. health hazards, etc.)	Description;		
Laborato	GEL Project	919-847-4241	704-586-0007		lolson@harthickman		QC Field Sample Code (2) Filtered (3) Matrix (4)	N N N									Date Time	580 61161		Sec. 1. C. W. W.		pike Sample, MSD = Matrix Spi	was not field filtered.	"=Mise Liquid, SO=Soil, SD=Se.	 Ascorbic Acid, HX = Hexane, 	Listed Waste LW= Listed Waste	(F,K,P and U-listed wastes.)	vde(s):			
190073 COM LADOLATORIAS LLC 1906 Males Streen Chemistry 1 Radiobloases 1 Specie	GEL Work Order Number:	Phone # 0	Fax# 7		Send Results To:Genna Olson go	and a state of the	*Date Collected Collected (Military) (hhum)	09-18-19 1000									Received by (signed) I	5 220		Raviau farm (CDD)	A Second minformation	quipment Blank. MS = Matrix S	dd filtered or - N - for sample wa	W=Waste Water, W=Water, MI. MA) and mumber of non-nineer a	roxide. SA = Sulfaric Acid, AA =	Hazards Listed	3	Waste code(s):	p	rinated	S
10101	GEL Work O	Hickman, PC	SWLF		Send Results			0						-		ain of Custod		1630	5 .	S Sample Receipt &		c, FD = Field Duplicate, EB = Ec	i - Y - for yes the sample was the	dwater, SW#Surface Water, WY concered (i.e. 82608-60108/747	 Nitrie Acid, SH = Sodium Hydi 	S Characteristic Hazards FL = Flammable/lenitable	CO = Corrosive	RE = Reactive	TSCA Regulated	PCB = Polychlor	Diphenyls
1 of 1 NWA-001 NWRA Quote	NA	NWRA c/o Hart & Hickman, PC	1	Apex, NC	Patrick Stevens		Sample ID For composites - indicate start and stop date/time	9222-1						-			1	. Junt 13-10-13		> For sumple shinning and delivery details, see Samulo Receipt & Review forme (SDD)	1.) Chain of Custody Number = Client Determined	2) QC Codes: N = Neurol Sample, TB = Trip Black, FD = Field Duplicate, EB = Equipment Blank. MS = Matrix Spike Sample, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite	3.) Field Filtered: For fiquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample	 Watrix Codes: DW=Drnking Water, GW=Groundwater, SW=Surface Water, WW=Water, W=Water, W=Water, ML=Mise Liquid, SO=Soil, SD=Sediment, SL=Sladge, Sumple Analysis Requested: Analytical method requested (i.e. 8760R. 6010R73470A) and workers of memory sciences. A second science of iteration for the second science of the science of the second science of the sc	x HA ≏ Hydrochloric Acid, NI ∞	7.) Are there any known or possible hazards associated with these samples?		Hg= Mercury	Se= Selenium	Ag= Sitver MD= Miscallynamic	Shoalipitonetiki _VIN
Page: Project # GEL Quote #: COC Number ⁽¹	PO Number:	Client Name:	Project/Site Name:	Address:	Collected By:		* For co									0.1.1.2.2.0	keimquisned By (Signed)	- Man H. Area	7	> For sumple sh	1.) Chain of Custody	2.) QC Codes: $N = h$	3.) Field Filtered: Fo	 4.) Mutrix Codes: D' 5.) Sample Analysis I 	6.) Preservative Type	7.) Are there any associated win		RCKA Metals As = Arsenic	Ba = Barium	Ca = Cadmium Cr = Chromium	

Page 14 of 16 SDG: 490673 Rev1

ATTACHMENT D

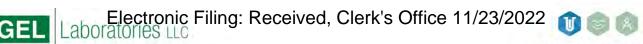
11/00/0000

Laboratories um	Th	SAMPLE DECE
Client: NWBA		SAMPLE RECEIPT & REVIEW FORM
Received By: ATA		Date Received: 9/19/19
Carrier and Tracking Number		Endex Express FedEx Ground UPS Field Services Courier Other 7762 7563 2308 -1°, 7762 7563 3418-1° 7762 7563 2764 -1°, 7762 7563 3290 -1° (Net Counter > 100-
Suspected Hazard Information	In No Co	"Net Counts > 1000mm an
A)Shipped as a DOT Hazardous?	Ha	If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation neard Class Shipped: UN#: UN2910, Is the Radioactive Shipment Survey Compliant? Yes No
B) Did the client designate the samples are to be received as radioactive?	Лсо	C notation or radioactive stickers on containers equal client designation.
C) Did the RSO classify the samples as radioactive?	/ Ma:	ximum Net Counts Observed* (Observed Counts - Area Background Counts):CPM1 / mR/Hr
D) Did the client designate samples are hazardous?		C notation or hazard labels on containers equal client documents
E) Did the RSO identify possible hazards? Sample Receipt Criteria	РСВ	of E is ves, select Hazarda L. L
1 Shipping containers received intact and sealed?		Comments/Qualifiers (Required for Non-Conforming Items) Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
 2 Chain of custody documents included with shipment? 3 Samples requiring cold preservation within (0 < 6 day over 10 day) 		ircle Applicable: Client contacted and provided COC COC created upon receiver
J Daily check performed and provide the		reservation Method: Aver to lice Packs Dry ice None Other:
temperature gun? 5 Sample containers intact and sealed?		emperature Device Serial #: 7. B.4 – 16 condary Temperature Device Serial # (If Applicable): refe Applicable: Seale Junior Device Serial # (If Applicable):
6 Samples requiring chemical preservation at proper pH?	1 6	cle Applicable: Scalsbroken Damaged container Leaking cograiner Other (describe) 2044-1 (1boff)e) cap received Oracked
Do any samples require Volatile Analysis?	Do	reservation added. Lot# es, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) liquid VOA vials free of headspace? Yes No NA
Samples received within holding time?		and tests affected:
Sample ID's on COC match ID's on bottles?		nud containers affected:
Number of containers received	Circle	Applicable: No dates on containers No times on containers COC missing info Other (describe)
Are sample containers identifiable as		Applicable: No container count on COC Other (describe)
COC form is properly signed in relinquished/received sections?	Circle	Applicable: (Not relinquished Other (describe)
(in the interview of the interview);		
PM (or PMA) review	: Initials	Dav Dav Page of

Page 15 of 16 SDG: 490673 Rev1

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water Louisiana NELAP	LA024
	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019



a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for Foothills Environmental Landfill Work Order: 490860

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490860 GEL Work Order: 490860

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Electroni

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Certificate of Analysis

Report Date: November 8, 2019

Company : Address :	NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804		
Address .	1550 Crystar Dilve, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forFoothills Environmental Landfill		
Client Sample ID:	1403-1	Project:	NWRA00119
Sample ID:	490860001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	16-SEP-19 09:20		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Analy	yst Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs by LO	C-MS/MS	"As Received"										
Fluorotelomer sulfonate 4:2 (4:2		ND	13.2	37.6	ng/L	0.200	1	JLS	10/02/19	0849	1921240	1
FTS)					U							
N-ethylperfluoro-1-		101	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (N	V -											
EtFOSAA) N-methylperfluoro-1-		257	13.2	40.0	ng/I	0.200	1					
octanesulfonamidoacetic acid (N	J_	237	15.2	40.0	ng/L	0.200	1					
MeFOSAA)	•											
Perfluorodecanesulfonic acid	U	ND	6.60	19.4	ng/L	0.200	1					
(PFDS)												
Perfluorodecanoic acid (PFDA)		82.6	7.80	20.0	ng/L	0.200						
Perfluorododecanoic acid (PFDe	,	ND	6.60	20.0	ng/L	0.200						
Perfluoroheptanesulfonic acid	J	6.82	6.60	19.0	ng/L	0.200	1					
(PFHpS)	• •	571	((0	20.0		0.200	1					
Perfluoroheptanoic acid (PFHpA	4)		6.60		ng/L	0.200						
Perfluorohexanesulfonic acid (PFHxS)		794	6.60	18.2	ng/L	0.200	1					
Perfluorononanesulfonic acid	U	ND	7.00	19.2	ng/L	0.200	1					
(PFNS)	C	112	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	1712	iig, 2	0.200						
Perfluorononanoic acid (PFNA)		71.4	6.60	20.0	ng/L	0.200	1					
Perfluorooctanesulfonamide	J	7.08	6.60	18.6	ng/L	0.200	1					
(PFOSA)												
Perfluorooctanesulfonic acid (Pl	FOS)	296	8.00	20.0	ng/L	0.200						
Perfluoropentanesulfonic acid		50.6	6.60	18.8	ng/L	0.200	1					
(PFPeS) Perfluoropentanoic acid (PFPeA	`	1070	6.60	20.0	no/I	0.200	1					
Perfluoroundecanoic acid (PFPeA	·	7.04	6.60 6.60	20.0	ng/L	0.200						
Fluorotelomer sulfonate 8:2 (8:2	,	7.04 ND	132	20.0 384	ng/L	0.200		пс	10/02/10	0657	1021240	2
FTS)	2 0	ND	152	364	ng/L	0.200	10	JLS	10/02/19	0037	1921240	2
Perfluorobutanesulfonic acid (Pl	FBS)	4400	66.0	178	ng/L	0.200	10					
Perfluorobutyric acid (PFBA))	744	66.0	200	ng/L	0.200						
Perfluorohexanoic acid (PFHxA	3	3920	66.0	200	ng/L	0.200						
Perfluorooctanoic acid (PFOA)	-	1650	70.0	200	ng/L	0.200						
Perfluorotetradecanoic acid	U	ND	66.0	200	ng/L	0.200						
(PFTeDA)	C				8							
Perfluorotridecanoic acid (PFTr	DA) U	ND	66.0	200	ng/L	0.200	10					
Fluorotelomer sulfonate 6:2 (6:2	2 U	ND	1320	3800	ng/L	0.200	100	JLS	10/02/19	1059	1921240	3
FTS)												
Semi-Volatile-GC/MS												

Semi-Volatile-GC/MS

Electronic Filing Elecared RAFORIES ice C1/23/2022

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	A discours Missicia 22202		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forFoothills Environmental Landfill		
Client Sample ID:	1403-1	Project:	NWRA00119
Sample ID:	490860001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF D	F Analyst Date	Time Batch	Method
Semi-Volatile-GC/MS									
SW846 8270 SIM 1,4-D	ioxane in Lic	uid "As Received"							
1,4-Dioxane		99.7	2.00	4.00	ug/L	0.200 2	JMB3 09/24/19	1829 1919444	4
The following Prep Met	hods were pe	rformed:							
Method	Description	l		Analyst	Date	Tin	ne Prep Batch	l	
EPA 537.1 Mod, PFAS, Comp	ol PFCs Extraction	on in Liquid		LM1	09/27/19	083	0 1921239		
SW846 3535A	SW8270E SIN	/I Prep 1,4-Dioxane		SJW1	09/23/19	120	0 1919441		
The following Analytic	The following Analytical Methods were performed:								
Method	Description				1	Analyst C	omments		
1	EPA 537.1 Mo	d, PFAS, Compliant with QSM T	able B-1	5		-			
2	EPA 537.1 Mo	d, PFAS, Compliant with QSM T	able B-1	5					
3	EPA 537.1 Mo	d, PFAS, Compliant with QSM T	able B-1	5					
4	SW846 3535A	/8270E SIM							
Surrogate/Tracer Recov	ery Test				Result	Nominal	Recovery%	Acceptable Li	mits
1,4-Dioxane-d8	SW846 Received	8270 SIM 1,4-Dioxane in Liquid ' d"	'As		30.0 ug/L	40.0	75	(70%-130%)	
Notes:									

Column headers are defined as follows:	
DF: Dilution Factor	Lc/LC: Critical Level
DL: Detection Limit	PF: Prep Factor
MDA: Minimum Detectable Activity	RL: Reporting Limit
MDC: Minimum Detectable Concentration	SQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 7

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia

Contact: Mr. Jim Riley

Workorder: 490860

Parmname	NOM	Sample Qual	QC	Units RPD/D	% REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240							
QC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2		15.7	ng/L	86	(60%-145%) JL	S 10/02/19 06:05
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5		20.4	ng/L	110	(56%-143%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7		17.5	ng/L	94	(57%-138%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5		19.3	ng/L	99	(63%-131%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5		21.5	ng/L	111	(62%-133%)	
Perfluorobutanesulfonic acid (PFBS)	17.2		16.6	ng/L	96	(68%-136%)	
Perfluorobutyric acid (PFBA)	19.5		19.7	ng/L	101	(70%-133%)	
Perfluorodecanesulfonic acid (PFDS)	18.8		16.8	ng/L	89	(53%-142%)	
Perfluorodecanoic acid (PFDA)	19.5		18.0	ng/L	93	(62%-135%)	
Perfluorododecanoic acid (PFDoA)	19.5		19.5	ng/L	100	(66%-131%)	
Perfluoroheptanesulfonic acid (PFHpS)	18.5		18.1	ng/L	98	(66%-138%)	
Perfluoroheptanoic acid (PFHpA)	19.5		17.9	ng/L	92	(67%-135%)	
Perfluorohexanesulfonic acid (PFHxS)	17.7		14.5	ng/L	82	(64%-137%)	
Perfluorohexanoic acid (PFHxA)	19.5		18.9	ng/L	97	(67%-133%)	

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QC Summary

Workorder: 490860										
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Page 2 of 7 Date Time	
Perfluorinated Compounds Batch 1921240			<u></u>		11 D/D / 0	111070	Tunge		Dute Thire	
Perfluorononanesulfonic acid (PFNS)	18.7		17.5	ng/L		93	(66%-130%)) JLS	10/02/19 06:05	
Perfluorononanoic acid (PFNA)	19.5		21.1	ng/L		108	(66%-134%))		
Perfluorooctanesulfonamide (PFOSA)	19.5		21.5	ng/L		111	(68%-137%))		
Perfluorooctanesulfonic acid (PFOS)	19.5		19.8	ng/L		102	(61%-131%))		
Perfluorooctanoic acid (PFOA)	19.5		18.8	ng/L		97	(63%-145%))		
Perfluoropentanesulfonic acid (PFPeS)	18.3		16.5	ng/L		90	(62%-139%))		
Perfluoropentanoic acid (PFPeA)	19.5		19.3	ng/L		99	(69%-132%))		
Perfluorotetradecanoic acid (PFTeDA)	19.5		22.5	ng/L		115	(65%-143%))		
Perfluorotridecanoic acid (PFTrDA)	19.5		19.9	ng/L		102	(57%-149%))		
Perfluoroundecanoic acid (PFUdA)	19.5		19.1	ng/L		98	(65%-134%))		
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6		20.5	ng/L	26	116	(0%-35%))	10/02/19 06:14	
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9		17.6	ng/L	14	98	(0%-36%))		
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1		19.9	ng/L	13	110	(0%-39%))		
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8		20.1	ng/L	4	107	(0%-25%))		
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8		21.9	ng/L	2	116	(0%-26%))		

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QC Summary

Workorder: 490860				<u> </u>				Page 3 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	. 4	103	(0%-30%) JLS	5 10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	. 2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	. 3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	. 16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	. 3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	. 2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	. 9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	. 15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	. 10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	. 4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	. 12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	<i>.</i> 6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	. 1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	<i>.</i> 0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	. 4	98	(0%-29%)	

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QC Summary

Workorder: 490860									Page 4 of 7			
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time		
Perfluorinated CompoundsBatch1921240												
Perfluoropentanoic acid (PFPeA)	18.8			20.0	ng/L	3	106	(0%-30%)	JLS	10/02/19 06:14		
Perfluorotetradecanoic acid (PFTeDA)	18.8			20.6	ng/L	9	109	(0%-30%)				
Perfluorotridecanoic acid (PFTrDA)	18.8			17.7	ng/L	11	94	(0%-35%)				
Perfluoroundecanoic acid (PFUdA)	18.8			21.2	ng/L	10	112	(0%-28%)				
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)			U	ND	ng/L					10/02/19 05:56		
Fluorotelomer sulfonate 6:2 (6:2 FTS)			U	ND	ng/L							
Fluorotelomer sulfonate 8:2 (8:2 FTS)			U	ND	ng/L							
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)			U	ND	ng/L							
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)			U	ND	ng/L							
Perfluorobutanesulfonic acid (PFBS)			U	ND	ng/L							
Perfluorobutyric acid (PFBA)			U	ND	ng/L							
Perfluorodecanesulfonic acid (PFDS)			U	ND	ng/L							
Perfluorodecanoic acid (PFDA)			U	ND	ng/L							
Perfluorododecanoic acid (PFDoA)			U	ND	ng/L							
Perfluoroheptanesulfonic acid (PFHpS)			U	ND	ng/L							

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QC Summary

Workorder: 490860								Pag	e 5 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D% REC	C% Range	Anlst		Time
Perfluorinated CompoundsBatch1921240									
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L			JLS	10/02/1	19 05:56
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L					
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L					
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L					
Perfluorononanoic acid (PFNA)		U	ND	ng/L					
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L					
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L					
Perfluorooctanoic acid (PFOA)		U	ND	ng/L					
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L					
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L					
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L					
Perfluorotridecanoic acid (PFTrDA)		U	ND	ng/L					
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L					
Semi-Volatile-GC/MS Batch 1919444									
QC1204387349 LCS **1,4-Dioxane-d8	4.00		3.55	ug/L	8	39 (70%-130%	6) JMB3	09/24 /1	19 12:24

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QC Summary

Workorder: 49	0860										Page	e 6 of 7
Parmname		NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Semi-Volatile-GC/MS Batch 1919												
QC1204387350 **1,4-Dioxane-d8	LCSD	4.00			3.18	ug/L		79	(70%-130%)	JMB3	09/24/1	19 12:49
QC1204387348 1,4-Dioxane	MB			U	ND	ug/L					09/24/1	19 11:59
**1,4-Dioxane-d8		4.00			3.05	ug/L		76	(70%-130%)			

Notes:

The Qualifiers in this report are defined as follows:

- ** Analyte is a surrogate compound
- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N/A RPD or %Recovery limits do not apply.

N1 See case narrative

- ND Analyte concentration is not detected above the detection limit
- NJ Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier
- P Organics--The concentrations between the primary and confirmation columns/detectors is >40% different. For HPLC, the difference is >70%.
- Q One or more quality control criteria have not been met. Refer to the applicable narrative or DER.
- R Sample results are rejected
- U \qquad Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.
- UJ Compound cannot be extracted

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QC Summary

Workor	der: 490860										Page	e 7 of 7
Parmnar	ne	NOM	Sample Q	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Х	Consult Case Narrative, I	Data Summary package	, or Project Ma	nager o	concerning t	his qualifi	er					

Y QC Samples were not spiked with this compound

^ RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.

h Preparation or preservation holding time was exceeded

N/A indicates that spike recovery limits do not apply when sample concentration exceeds spike conc. by a factor of 4 or more or %RPD not applicable. ^ The Relative Percent Difference (RPD) obtained from the sample duplicate (DUP) is evaluated against the acceptance criteria when the sample is greater than five times (5X) the contract required detection limit (RL). In cases where either the sample or duplicate value is less than 5X the RL, a control limit of +/- the RL is used to evaluate the DUP result.

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490860

GC/MS Semivolatile

<u>Product:</u> Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1919444

Preparation Method: SW846 3535A **Preparation Procedure:** GL-OA-E-073 REV# 2 **Preparation Batch:** 1919441

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490860001	1403-1
1204387348	Method Blank (MB)
1204387349	Laboratory Control Sample (LCS)
1204387350	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

Technical Information

Sample Dilutions

Sample 490860001 (1403-1) was diluted due to the presence of one or more over-range target analytes.

LCMSMS-Misc

<u>Product:</u> The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 <u>Analytical Procedure:</u> GL-OA-E-076 REV# 7 <u>Analytical Batches:</u> 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490860001	1403-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490860001 (1403-1).

Amelanta	490860
Analyte	001
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Fluorotelomer sulfonate 8:2 (8:2 FTS)	10X
Perfluorobutanesulfonate (PFBS)	10X
Perfluorobutyric acid (PFBA)	10X
Perfluorohexanoic acid (PFHxA)	10X
Perfluorooctanoic acid (PFOA)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

Page: 1 of 1			l F		-	-				GEL Lab	GEL Laboratories, LLC		
1					100f	d lol	LADOIAIONESLLC			2040 Sav	2040 Savage Road		
GEL Quote #: NWRA Quote COC Number ⁽¹⁾ . NA	191	-19 DX100	orice Cha	ain of Ct	nistry i Rad	iochemistr and An:	getcom ¹ Chemistry I Radiochemistry I Radiobioassay I Specialty Analytics Chain of Custody and Analytical Reguest	y i Specialty An JUEST	alytics	Charlesto	Charleston, SC 29407 Phone: (843) 556-8171		
	GEL Work	GEL Work Order Number:			GEL Project Manager:	iect Mai	iager:			Fax: (842	Fax: (843) 766-1178		
Client Name: NWRA c/o Hart & Hickman, PC	Hickman, PC		Phone # 9	919-847-4241	-4241		5	ample Ana	Sample Analysis Requested ⁽⁵⁾	^b (Fill in the number of containers for each test)	er of container	s for each test)	i Nice Maria
Project/Site Name: Foothills Environmental Landfill	mental Landfill		Fax# 7	704-586-0007	-0007		Should this	st:				< Preservative Type (6)	e (6)
Address: Lenoir, NC			والمراجع المراجع				sample be considered:) niejn [PFAS 21 0	PFAS 21 cmpd list by EPA 537 mod	A 537 mod		
Collected By: Patrick Stevens	Send Resu	Send Results To:Genna Olson golson@harthickman	Olson go	lson@h	arthick		spin E S S	1. 01 60	r1,4-Dioxa	1,4-Dioxane by EPA 8270SIM	WIS0.	Note: extra sample is	le is
Sample ID * For composites - indicate start and stor date time	stor date/time	*Date Collected (mm-dd-vv)	*Time Collected (Military) (hhum)	QC Code ⁽³⁾ F	Field Sample Filtered ⁽³⁾ Matrix ⁽⁴⁾	Sample Matrix (4) Matrix (4)	ezey əlqissod 10 u.nouy (L) Ofuj əldətəsi Alddas əsvələ	rodmun letoT			<u>}</u>	required for sample specific QC	
1403-1		09-16-19	0320	z	z	ML		4 X	×				
					_								<u>.</u>
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													1977-4 1977-4
Dalinouishad Br./Girmad) Date		Cuain of Custory Signatures	nod)	Date	Time			macanhaw IVI		ado	oherið-		
1 Value II. And 09	1630	A A	5	19/10	08	Sa	Fax R(Select	Fax Results: [] Yes Select Deliverable: [s [X] No] C of A [] QC	Fax Results: []Yes [X]No Select Deliverable: []C of A []]QC Summary []] level 1	1 [] Level 2	[] Level 3 [] Level 4	<u>s</u> 7
2 / aller 11' 20000		2^{I}					Additi	Additional Remarks.					
e N		3					For La	For Lab Receiving Use Only:	Use Only: Custody	Custody Seal Intact? [] Yes	[] No	Couler Temp:OC	
> For sample shipping and delivery details, see Sample Receipt & Review form (SRR.)	ls, see Sample Recei	nt & Review form	(SRR.)		ample Co	llection T	Sample Collection Time Zone: [X] Eastern	1222	[] Pacific [] Central	itral [] Mountain	t [] Other:		
 Chain of Custody Number = Client Determined QC Codes: N = Nonnel Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sample, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite 	t, FD = Field Duplicate, El	3 = Equipment Blank.	MS = Matrix 1	Spike Sampio	e, MSD = M	atrix Spíke I	Juplicate Sample	G = Grub, C = (omposite				1/23
3.) Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered.	 Y • for yes the sample w 	as field filtered or - N	- for sample w	was not field filtered.	litered.	ch	and the P3 can		owne muchim prive				/20
 Niatrix Codes: UW=Urnking water, UW=Crountwater, SW=Surrise water, W=Water, W=Water, MI=Nise Liquid, SU=SOR, G0108/7470A + 1). Sample Analysis Requested: Analytical method requested (i.e. 8260B, 6010B/7470A) and number of containers provided for each (i.e. 8260B + 3, 6010B/7470A + 1). 	uwater, 3W = Surface Wate	a, w w waster water B/7470A) and number	of containers	provided for	na, 50-50n each (i.e. 82	60B - 3, 60	0 <i>B/7470</i> .4 - 1).	0-000 W 4900	0-04, F-TBRI, F-WI	00, U≕UIDBC, F≂FCC31, M≕	Nasar		
(6.) Preservative Type: Ha = Hydrochloric Acid, NI = Nitric Acid, SH = Sodium Hydroxide. SA = Sulfritic Acid, AA = Ascorbic Acid, HX = Hexane, ST = Sodium Thiosulfate. If no preservative is added = leave field blank	= Nitric Acid, SH = Sodiun	n Hydroxide. SA = Sul	furie Acid, AA	= Ascorbic	Acid. HX = 1	Hexane, ST	= Sodium Thiost	lfate. If no presei	vative is added = leave	field blank	a na si n		a di seconda
(.) Are there any known or possible hazaras associated with these samples?		Characteristic Hazards FL = Flammable/Ignitable CO = Corrosive	$\frac{ \text{LW} }{ \text{LW} }$	Listed Waste LW= Listed Waste (F.K.P and U-listed wastes.)	ste ted wastes	<u>،</u> ا	$\frac{0 \text{ Iner}}{0 t = 0}$	Other OT= Other / Unknown (i.e.: High/law pH. asb	Uttter OT= Other / Unknown G.e.: Hiehlow pH. asbestos, bervllium, irritants, other	irritants, other	Please provi below regari concerns (i	Please provide any additional details below regarding handling and/or disposal concerns. (1.e Oriein of sample(s), type	ils lisposal frine
RCRA Metals As = Arsenic Hg= Mercury	$\mathbf{RE} = \mathbf{Reactive}$	itve	Waste	Waste code(s):			misc. health Description:	mise. health hazards, etc.) Descrintion:	; etc.)		of site collec	of stie collected from, odd matrices, etc.)	etc.)
	TSCA Regulated	ulated											
Cd = Cadmum Ag = Silver Cr = Chromium MR = Miscellaneous	PCB = Poly	Polychlorinated hinhenvls											
	3												

Page 14 of 16 SDG: 490860 Rev1

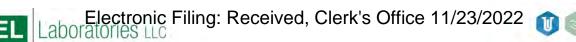
Laboratories Laboratories	
Client: NWBA Jh	SAMPLE RECEIPT & REVIEW FORM
Received By: ATA	SDG/AR/COC/Work Order: 40108 000
Carrier and Tracking Number	Birth Received: 9/1/1/19 Circle Applicable: FedEx Ground UPS Field Services Courier Other 7762 7563 2308 -1°, 7762 7563 3418-1° 7762 7563 2764-1°, 7762 7563 3290-1°
Suspected Hazard Information $\frac{3}{2}$ $\frac{2}{2}$	*If Net Counts > 100cmu on complex
A)Shipped as a DOT Hazardous?	*If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation. -lazard Class Shipped: UN#: f UN2910, Is the Radioactive Shipment Survey Compliant? Yes No
B) Did the client designate the samples are to be received as radioactive?	OC notation or radioactive stickers on containers equal client designation.
radioactive?	laximum Net Counts Observed' (Observed Counts - Area Back
b) bid the client designate samples are	lassified as: Rad L Rad 2 Rad 3
E) Did the RSO identify possible hazards?	D or E is yes, select Hazards below. B's Flammable Foreign Soil BCD and the second s
Sample Receipt Criteria	Comment (O manual Comment)
1 Shipping containers received intact and sealed? 2 2	Comments/Qualifiers (Required for Non-Conforming Items) Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
2 Chain of custody documents included with shipment?	Circle Applicable: Client contacted and provided COC COC created upon receipt
3 Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?*	Preservation Method: Available to part
4 Daily check performed and passed on IR	Temperature Device Section 4. 2017
5 Sample containers intact and sealed?	Secondary Temperature Device Serial # (If Applicable): Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
6 at proper pH?	Sample ID's and Containers Affected:
7 Do any samples require Volatile Analysis?	Image: Preservation added_Lot# If Yes, are Encores or Soil Kits present for solids? YesNoNA(If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? YesNoNA(If unknown, select No) re liquid VOA vials free of headspace? YesNoNA(If unknown, select No) umple ID's and containers affected:
	D's and tests affected:
y Sample ID's on COC match ID's on bottles?	's and containers affected:
10 Date & time on COC match date & time Cin	rele Applicable: No dates on containers No times on containers COC missing info Other (describe)
number indicated on COC?	cle Applicable: No container count on COC Other (describe)
12 Are sample containers identifiable as	
I3 COC form is properly signed in relinquished/received sections? Y Comments (Use Continuation Form if needed): Circle	cle Applicable: Not relinquished Other (describe)
PM (or PMA) review: Initials	Date 125 1 Page GL-CHL-SR-001 Rev 6

ATTACHMENT D

4.....

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water Louisiana NELAP	LA024
	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019





PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for BFI-Charlotte motor Speedway Landfill V Work Order: 490866

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

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Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490866 GEL Work Order: 490866

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecared RAFORIES ice C1/23/2022

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forBFI-Charlotte motor Speedway Landfill V		
Client Sample ID:	1304-1	Project:	NWRA00119
Sample ID:	490866001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	16-SEP-19 12:55		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter Quali	fier	Result	DL	RL	Units	PF	DF	Anal	yst Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs by LC-MS	/MS '	'As Received"										
Fluorotelomer sulfonate 8:2 (8:2		39.7	13.2	38.4	ng/L	0.200	1	JLS	10/02/19	0907	1921240	1
FTS)					, in the second s							
N-ethylperfluoro-1-		87.2	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (N- EtFOSAA)												
N-methylperfluoro-1-		258	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (N-		200	1012		8		-					
MeFOSAA)												
Perfluorodecanesulfonic acid	J	6.87	6.60	19.4	ng/L	0.200	1					
(PFDS) Perfluorodecanoic acid (PFDA)		590	7.80	20.0	no/I	0.200	1					
Perfluorododecanoic acid (PFDA)		63.3	6.60	20.0	ng/L ng/L	0.200						
Perfluoroheptanesulfonic acid	J	8.17	6.60	20.0 19.0	ng/L	0.200						
(PFHpS)	J	0.17	0.00	19.0	ng/L	0.200	1					
Perfluoroheptanoic acid (PFHpA)		983	6.60	20.0	ng/L	0.200	1					
Perfluorohexanesulfonic acid		925	6.60	18.2	ng/L	0.200	1					
(PFHxS)					-							
Perfluorononanesulfonic acid (PFNS)	U	ND	7.00	19.2	ng/L	0.200	1					
Perfluorononanoic acid (PFNA)		269	6.60	20.0	ng/L	0.200	1					
Perfluorooctanesulfonamide (PFOSA)	J	11.5	6.60	18.6	ng/L	0.200	1					
Perfluorooctanesulfonic acid (PFOS)		356	8.00	20.0	ng/L	0.200	1					
Perfluoropentanesulfonic acid (PFPeS)		73.2	6.60	18.8	ng/L	0.200	1					
Perfluoroundecanoic acid (PFUdA)		30.8	6.60	20.0	ng/L	0.200	1					
Perfluorobutanesulfonic acid (PFBS)		5260	66.0	178	ng/L	0.200	10	JLS	10/02/19	0706	1921240	2
Perfluorobutyric acid (PFBA)		1920	66.0	200	ng/L	0.200	10					
Perfluorohexanoic acid (PFHxA)		3470	66.0	200	ng/L	0.200	10					
Perfluorooctanoic acid (PFOA)		2210	70.0	200	ng/L	0.200						
Perfluoropentanoic acid (PFPeA)		2160	66.0	200	ng/L	0.200						
Perfluorotetradecanoic acid (PFTeDA)	U	ND	66.0	200	ng/L	0.200	10					
Perfluorotridecanoic acid (PFTrDA)	U	ND	66.0	200	ng/L	0.200						
Fluorotelomer sulfonate 4:2 (4:2 FTS)	U	ND	1320	3760	ng/L	0.200	100	JLS	10/02/19	1117	1921240	3
Fluorotelomer sulfonate 6:2 (6:2 FTS)	U	ND	1320	3800	ng/L	0.200	100					
The following Prep Methods w	ere p	erformed:										

Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Electroni

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forBFI-Charlotte motor Speedway Landfill V		
Client Sample ID:	1304-1	Project:	NWRA00119
Sample ID:	490866001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF DF	Analyst Date	Time Batch	Method
The following Prep	Methods were pe	erformed:							
Method	Description	n		Analyst	Date	Time	Prep Batch		
EPA 537.1 Mod, PFAS, C	Compl PFCs Extracti	ion in Liquid		LM1	09/27/19	0830	1921239		
The following Anal	ytical Methods v	were performed:							
Method	Description	l			А	nalyst Con	nments		
1	EPA 537.1 Mo	od, PFAS, Compliant wi	ith QSM Table B-15	5		-			
2	EPA 537.1 Mo	od, PFAS, Compliant wi	ith QSM Table B-15	5					
3	EPA 537.1 Mo	od, PFAS, Compliant wi	ith QSM Table B-15	5					

Notes:

Column headers are defined as follows:	
DF: Dilution Factor	Lc/LC: Critical Level
DL: Detection Limit	PF: Prep Factor
MDA: Minimum Detectable Activity	RL: Reporting Limit
MDC: Minimum Detectable Concentration	SQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 6

NWRA - Carolinas Chapter
1550 Crystal Drive, Suite 804
Arlington, Virginia

Contact: Mr. Jim Riley

Workorder: 490866

Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date 7	Гіте
Perfluorinated CompoundsBatch1921240										
QC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2		15.7	ng/L		86	(60%-145%)	JLS	10/02/19	06:05
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5		20.4	ng/L		110	(56%-143%)			
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7		17.5	ng/L		94	(57%-138%)			
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5		19.3	ng/L		99	(63%-131%)			
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5		21.5	ng/L		111	(62%-133%)			
Perfluorobutanesulfonic acid (PFBS)	17.2		16.6	ng/L		96	(68%-136%)			
Perfluorobutyric acid (PFBA)	19.5		19.7	ng/L		101	(70%-133%)			
Perfluorodecanesulfonic acid (PFDS)	18.8		16.8	ng/L		89	(53%-142%)			
Perfluorodecanoic acid (PFDA)	19.5		18.0	ng/L		93	(62%-135%)			
Perfluorododecanoic acid (PFDoA)	19.5		19.5	ng/L		100	(66%-131%)			
Perfluoroheptanesulfonic acid (PFHpS)	18.5		18.1	ng/L		98	(66%-138%)			
Perfluoroheptanoic acid (PFHpA)	19.5		17.9	ng/L		92	(67%-135%)			
Perfluorohexanesulfonic acid (PFHxS)	17.7		14.5	ng/L		82	(64%-137%)			
Perfluorohexanoic acid (PFHxA)	19.5		18.9	ng/L		97	(67%-133%)			

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QC Summary

Workorder: 490866			Page 2 of 6							
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated CompoundsBatch1921240										
Perfluorononanesulfonic acid (PFNS)	18.7			17.5	ng/L		93	(66%-130%) JLS	10/02/19 06:05
Perfluorononanoic acid (PFNA)	19.5			21.1	ng/L		108	(66%-134%)	
Perfluorooctanesulfonamide (PFOSA)	19.5			21.5	ng/L		111	(68%-137%)	
Perfluorooctanesulfonic acid (PFOS)	19.5			19.8	ng/L		102	(61%-131%)	
Perfluorooctanoic acid (PFOA)	19.5			18.8	ng/L		97	(63%-145%)	
Perfluoropentanesulfonic acid (PFPeS)	18.3			16.5	ng/L		90	(62%-139%)	
Perfluoropentanoic acid (PFPeA)	19.5			19.3	ng/L		99	(69%-132%)	
Perfluorotetradecanoic acid (PFTeDA)	19.5			22.5	ng/L		115	(65%-143%)	
Perfluorotridecanoic acid (PFTrDA)	19.5			19.9	ng/L		102	(57%-149%)	
Perfluoroundecanoic acid (PFUdA)	19.5			19.1	ng/L		98	(65%-134%)	
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6			20.5	ng/L	26	116	(0%-35%)	10/02/19 06:14
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9			17.6	ng/L	14	98	(0%-36%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1			19.9	ng/L	13	110	(0%-39%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8			20.1	ng/L	4	107	(0%-25%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8			21.9	ng/L	2	116	(0%-26%)	

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QC Summary

Workorder: 490866				Page 3 of 6				
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	4	103	(0%-30%) JLS	10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	4	98	(0%-29%)	

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QC Summary

Workorder: 490866		_			<u> </u>					Deve Arf (
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Page 4 of 6 Date Time
Perfluorinated CompoundsBatch1921240		·								
Perfluoropentanoic acid (PFPeA)	18.8			20.0	ng/L	3	106	(0%-30%)) JLS	10/02/19 06:14
Perfluorotetradecanoic acid (PFTeDA)	18.8			20.6	ng/L	9	109	(0%-30%))	
Perfluorotridecanoic acid (PFTrDA)	18.8			17.7	ng/L	11	94	(0%-35%))	
Perfluoroundecanoic acid (PFUdA)	18.8			21.2	ng/L	10	112	(0%-28%))	
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)			U	ND	ng/L					10/02/19 05:56
Fluorotelomer sulfonate 6:2 (6:2 FTS)			U	ND	ng/L					
Fluorotelomer sulfonate 8:2 (8:2 FTS)			U	ND	ng/L					
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)			U	ND	ng/L					
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)			U	ND	ng/L					
Perfluorobutanesulfonic acid (PFBS)			U	ND	ng/L					
Perfluorobutyric acid (PFBA)			U	ND	ng/L					
Perfluorodecanesulfonic acid (PFDS)			U	ND	ng/L					
Perfluorodecanoic acid (PFDA)			U	ND	ng/L					
Perfluorododecanoic acid (PFDoA)			U	ND	ng/L					
Perfluoroheptanesulfonic acid (PFHpS)			U	ND	ng/L					

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QC Summary

		$\underline{\mathbf{v}}$	IIIIIai	y					
Workorder: 490866									Page 5 of
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated Compounds Batch 1921240									
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L				JLS	10/02/19 05:5
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L					
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L					
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L					
Perfluorononanoic acid (PFNA)		U	ND	ng/L					
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L					
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L					
Perfluorooctanoic acid (PFOA)		U	ND	ng/L					
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L					
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L					
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L					
Perfluorotridecanoic acid (PFTrDA)		U	ND	ng/L					
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L					

Notes:

The Qualifiers in this report are defined as follows:

- ** Analyte is a surrogate compound
- < Result is less than value reported
- > Result is greater than value reported

Page 9 of 15 SDG: 490866 Rev1

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QC Summary

armnan	ne	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
А	The TIC is a suspected	aldol-condensation prod	uct									
В	The target analyte was	detected in the associate	d blank.									
С	Analyte has been confin	rmed by GC/MS analysi	S									
D	Results are reported fro	om a diluted aliquot of th	e sample									
Е	Concentration of the tar	rget analyte exceeds the	instrument ca	alibration	range							
Н	Analytical holding time	e was exceeded										
J	See case narrative for a	n explanation										
J	Value is estimated											
JNX	Non Calibrated Compo	und										
N	OrganicsPresumptive on nearest internal stand Presumptive evidence b internal standard respor RPD or %Recovery lim	dard response factor based on mass spectral li nse factor	-	-				-				
	See case narrative											
ND	Analyte concentration i	s not detected above the	detection lin	nit								
NJ	Consult Case Narrative	, Data Summary packag	e, or Project	Manager	concerning	this qualif	ier					
Р	OrganicsThe concentr	rations between the prim	ary and conf	irmation	columns/de	tectors is >	>40% differen	t. For HPL	C, the diffe	rence is >'	70%.	
Q	One or more quality con	ntrol criteria have not be	en met. Refe	r to the ap	oplicable na	arrative or 1	DER.					
R	Sample results are reject	cted										
U	Analyte was analyzed f	or, but not detected abo	ve the MDL,	MDA, M	DC or LOI).						
UJ	Compound cannot be ex	xtracted										
Х	Consult Case Narrative	, Data Summary packag	e, or Project	Manager	concerning	this qualif	ïer					
Y	QC Samples were not s	piked with this compou	nd									
٨	RPD of sample and dup	plicate evaluated using +	/-RL. Conce	entrations	are <5X th	e RL. Qua	lifier Not Ap	olicable for	Radiochem	istry.		
h	Preparation or preserva	tion holding time was ex	ceeded									
The Re	icates that spike recover- elative Percent Difference es (5X) the contract requ ed to evaluate the DUP 1	ce (RPD) obtained from ired detection limit (RL	the sample d	uplicate ((DUP) is ev	aluated ag	ainst the acce	ptance criter	ria when th	e sample i	s greater	

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Electronic Filing: Received, Clerk's Office 11/23/2022 LCMSMS-Misc Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490866

Product: The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 <u>Analytical Procedure:</u> GL-OA-E-076 REV# 7 <u>Analytical Batches:</u> 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490866001	1304-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490866001 (1304-1).

Amelanta	490866
Analyte	001
Fluorotelomer sulfonate 4:2 (4:2 FTS)	100X
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Perfluorobutanesulfonate (PFBS)	10X
Perfluorobutyric acid (PFBA)	10X
Perfluorohexanoic acid (PFHxA)	10X
Perfluorooctanoic acid (PFOA)	10X
Perfluoropentanoic acid (PFPeA)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

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GEL Laboratorics, LLC 2040 Savage Road Charleston, SC 29407 Phone: (843) 556-8171	Fax: (843) 766-1178	(Fill in the number of containers for each test)	C Preservative Type (6)	v 537 mod	Comments SIM Notes aver comments											fy: (Subject to Surcharge)		[] Level 2 [] Level 3 [] Level 4		[] No Cooler Temp:C	 [] Others and Shares are a strated with several 			tsał		Please nravide any additional details	below regarding handling and/or disposal	concerns. (i.e.: Origin of sample(s), type	of sue conected from, odd matrices, etc.)			
aity Analytics		Sample Analysis Requested ⁽⁵⁾ (Fill in the numbe		PFAS 21 cmpd list by EPA 537 mod												TAT Requested: Normal: X Rush: Specify:	[] Ycs [X] No	Select Deliverable: [] C of A [] QC Summary [] level 1	emarks:	ing Use Only: (em []] Pacific []] Central []] Mountain	tab. C = Composite		4.) Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=Serface Water, WW=Waste Water, WEWater, ML=Mise Liquid, SO=Soil, SD=Sediment, SL=Sludge, SS=Solid Waste. 0=Oil, F=Filter, P=Wipe, U=Urine, F=Feeal, N=Nata	 Sample Analysis Requested: Analytical method requested (i.e. 82608, 60108/7470A) and number of containers provided for each (i.e. 82608 - 3, 60108/7470A - 1). Preservative Tone H a Educated MI = Minie Acid SH = Sodium Redevoide SA = Sodium Acid A = According Acid HY = Provided For Economic Acid HY = Provided Size (Size (no preservative is added = leave held blank	Unknown	(i.e.: High/low pH, asbestos, beryllium, irritants, other mise health harards are)				
Laboratorias LLC chemistry I Radiochemistry I Radiobioassay I Specialty Analytics Custody and Analytical Request	Manager:	Samp		sample be	com	Radionetive Please supply (7) Known or (7) K										TATR	Fax Results: [] Yes	Select Delive	Additional Remarks.	For Lab Rec	Sample Collection Time Zone: [X] Eastern	nike Duplicate Sample, G = Gr		ediment, SL=Sludge, SS=Soli,	. 6010B/7470.4 - 1). sr = s-4:	Stress of Other	OT = Other / Unknown	(i.e.: High/lo misc health l	Description:			
ain of	GEL Project Manager:	919-847-4241	704-586-0007		olson@harthickman.	1 QC Field Sample Code (2) Filtered (2) Matrix (4)	z z										Date Time	19/19 2850			Sample Collectic	x Spike Sample, MSD = Matrix Sp	was not field filtered.	ML=Mise Liquid, SO=Soil, SD=S	s provided for each (i.e. 8260B - 3 A = Association Aniel ETV - Harmers	Listed Waste	Listed Waste	(F.K.P and U-listed wastes.) Waste code(s):				
	ber:	Phone #	Landfill V Fax #		Send Results To:Genna Olson g	*Time *Date Collected (Millary) (nm-dd-yy) (hhmu)										Chain of Custody Signatures	Received by (signed)	1- AA	2	3	r & Keview Jorm (SRR.)	 C Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sumple, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite 	3.) Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered.	r, WW≃Waste Water, W≂Water, 1	 Sample Analysis Requested: Analytical method requested (i.e. 8260B, 6010B/7470A) and number of containers provided for each (i.e. 8260B - 3, 6010B/7470A - 1). Preservative Tone HA = Hortnohlmic Anid NI = Ninie Acid SH = Sodium Hortnoide SA = Solition Axid Axid Axid Divertient Contention (i.e. 8260B - 2, 6010B/7470A - 1). 	Characteristic Hazards Liste	e/Ignitable			tlated chlorinated	biphenyls	
1	GEL Wor	NWRA c/o Hart & Hickman, PC	BFI-Charlotte Motor Speedway Landfill V	O		D 1 and stop date time												-19 1630			> F or sumple supping and delivery details, see Sample Receipt & Review form (SRR). 1) Chain of Castody Number = Client Determined	o Blank, FD = Field Duplicate, EB	ϵ with a - Y - for yes the sample w	Groundwater, SW=Surface Water	thod requested (i.e. 82608, 60101 d NI = Nitrie Acid SH = Sodium	azards Characteris	FL = Flamn	CO = Corrosive RE = Reactive		TSCA Regulated PCR = Polychlorinated		
ŽŽ	r: NA			Concord, NC	By: Patrick Stevens	Sample ID * For composites - indicate start and stop date time	1304-1										1	and H. Junt - 13-18-19			For sample stupping and delivery detail. [1] Chain of Custody Number at Client Determined	N = Normal Sample, TB = Trip	ed: For liquid matrices, indicate	les: DW=Drinking Water, GW=	alysis Requested: Analytical met a Tyne: HA = Hydrochlaric Acid	7.) Are there any known or possible hazards	associated with these samples?	als		u Se= Selenium ium Ar= Silver		RCRA metals
Page: 1 Project # GEL Quote #: COC Number ⁽¹⁾ ;	PO Number:	Client Name:	Project/Site Name:	Address:	Collected By:	• F											Keinquish	- /attu	2	3 2.16/02/17/25	> r or samp 1.) Chain of Ci	2.) QC Codes:	3.) Field Filten	4.) Matrix Cod	 Sample An. Preservative 	7.) Are there	associate	RCRA Metals	As = Arsenic	Ba = Barium Cd = Cadmium	Cr = Chromium	Pb = Lead

Page 13 of 15 SDG: 490866 Rev1

ATTACHMENT D

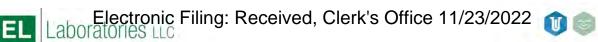
Electropic Eiling: Received Clark's Office 11/22/2022

Laboratories Laboratories	T R	SAMPLE DROP
Client: NWBA	Jh	SAMPLE RECEIPT & REVIEW FORM
Received By: ATA		Date Received: 9/19/19
Carrier and Tracking Number		Edex Express Fedex Ground UPS Field Services Courier Other 7762 7563 2308 -1°, 7762 7563 3418-1° 7762 7563 2764 -1°, 7762 7563 3290 -1°
Suspected Hazard Information	Y _{cs} No	*If Net Counts > 100cpm on samples not made in
A)Shipped as a DOT Hazardous?	/	•If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation. Hazard Class Shipped: UN#: If UN2910, Is the Radioactive Shipment Survey Compliant? Yes No
B) Did the client designate the samples are to be received as radioactive?	Λ	COC notation or radioactive stickers on containers equal client designation.
C) Did the RSO classify the samples as radioactive?		dasimum Net Country Of
D) Did the clieft designate samples are hazardous?		Classified as: Rad 1 Rad 2 Rad 3 COC notation or hazard labels on containers equal client designation.
E) Did the RSO identify possible Inzards?		D or E is yes, select Hazards below. CB's Flammable Foreign Suit and
Sample Receipt Criteria		Asbestos Beryllium Other:
I Shipping containers received intact and sealed?		Comments/Qualifiers (Required for Non-Conforming Items) Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
2 Chain of custody documents included with shipment?	/	Circle Applicable: Client contacted and provided COC COC created upon receipt
3 Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?*		Preservation Method: Aver the Land to the
A Daily check performed and passed on IR temperature gun?		Temperature Davies Sect 4 TO TEMP: 10
5 Sample containers intact and sealed?	7	Secondary Temperature Device Serial #: <u>7.84–18</u> Circle Applicable: Seals,broken Damaged container Leaking container Other (describe)
6 Samples requiring chemical preservation at proper pH?		Sample ID's and Containers Affected:
7 Do any samples require Volatile Analysis?		If Preservation added. Lot# If Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) Are liquid VOA vials free of headspace? Yes No NA ample ID's and containers affected:
8 Samples received within holding time?		D's and tests affected:
9 Sample ID's on COC match ID's on bottles?		D's and containers affected:
10 Date & time on COC match date & time on bottles?	Ci	ircle Applicable: No dates on containers No times on containers COC missing info Other (describe)
11 Number of containers received match number indicated on COC? 12 Are sample containers identifiable as	Ci	rele Applicable: No container count on COC Other (describe)
COC form is properly signed in		
relinquished/received sections?		cle Applicable: Not relinquished Other (describe)
		Var
PM (or PMA) review	: Initials	Caller [N.D.] Page or]
		GL-CHL-SR-001 Rev 6

ATTACHMENT D

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012 SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water	LA024
Louisiana NELAP	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235–19–15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019



a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for Chambers Development MSWLF Work Order: 490872

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

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Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490872 GEL Work Order: 490872

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- Q One or more quality control criteria have not been met. Refer to the applicable narrative or DER.
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.
- h Preparation or preservation holding time was exceeded

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Electroni

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical for Chambers Development MSWLF		
Client Sample ID:	0403-1	Project:	NWRA00119
Sample ID:	490872001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	16-SEP-19 15:30		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Anal	yst Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs by	y LC-MS/MS	"As Received"										
Fluorotelomer sulfonate 8:2		ND	13.2	38.4	ng/L	0.200	1	JLS	10/02/19	0915	1921240	1
FTS)					-							
N-ethylperfluoro-1-	J	14.9	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic ac EtFOSAA)	1d (N-											
N-methylperfluoro-1-		50.5	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic ac	id (N-	50.5	15.2	10.0	iig/L	0.200						
MeFOSAA)												
Perfluorodecanesulfonic aci	id U	ND	6.60	19.4	ng/L	0.200	1					
(PFDS)		22.6	7.00	20.0	æ	0.000	1					
Perfluorodecanoic acid (PF		23.6	7.80	20.0	ng/L	0.200						
Perfluorododecanoic acid (I	,	ND	6.60	20.0	ng/L	0.200						
Perfluoroheptanesulfonic ac (PFHpS)	cid U	ND	6.60	19.0	ng/L	0.200	1					
Perfluoroheptanoic acid (PF	FHpA)	249	6.60	20.0	ng/L	0.200	1					
Perfluorohexanesulfonic aci	1 /	218	6.60	18.2	ng/L	0.200						
(PFHxS)					6							
Perfluorononanesulfonic ac	id U	ND	7.00	19.2	ng/L	0.200	1					
(PFNS)			<i>c.co</i>	•••	~							
Perfluorononanoic acid (PF	,	15.5	6.60	20.0	ng/L	0.200						
Perfluorooctanesulfonamide (PFOSA)	e U	ND	6.60	18.6	ng/L	0.200	1					
(PFOSA) Perfluorooctanesulfonic aci	d (PFOS)	84.2	8.00	20.0	ng/L	0.200	1					
Perfluorooctanoic acid (PFC	· · ·	345	7.00	20.0	ng/L	0.200						
Perfluoropentanesulfonic ac	· ·	19.6	6.60	18.8	ng/L	0.200						
(PFPeS)		19.0	0.00	1010	iig z	0.200	-					
Perfluoropentanoic acid (PF	FPeA)	780	6.60	20.0	ng/L	0.200	1					
Perfluoroundecanoic acid (H	PFUdA) U	ND	6.60	20.0	ng/L	0.200	1					
Fluorotelomer sulfonate 6:2	(6:2 J	180	132	380	ng/L	0.200	10	JLS	10/02/19	0714	1921240	2
FTS)					-							
Perfluorobutanesulfonic aci	· · ·	6290	66.0	178	ng/L	0.200						
Perfluorobutyric acid (PFB)	,	831	66.0	200	ng/L	0.200						
Perfluorohexanoic acid (PF	,	2200	66.0	200	ng/L	0.200						
Perfluorotetradecanoic acid (PFTeDA)	U	ND	66.0	200	ng/L	0.200	10					
(PFTeDA) Perfluorotridecanoic acid (F	PFTrDA) U	ND	66.0	200	ng/L	0.200	10					
Fluorotelomer sulfonate 4:2	,	ND	1320	3760	ng/L	0.200		ILS	10/02/19	1125	1921240	3
FTS)	0	1,12	1520	2700	11-67-12	0.200	100	10	10,02,19	1120		5
Semi-Volatile-GC/MS	1											

Semi-Volatile-GC/MS

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Certificate of Analysis

Report Date: November 8, 2019

Compan	y :	NWRA - Carolinas Chapter		
Address	:	1550 Crystal Drive, Suite 804		
		Arlington, Virginia 22202		
Contact:		Mr. Jim Riley		
Project:		Analytical for Chambers Development MSWLF		
Client Sa	mple ID:	0403-1	Project:	NWRA00119
Sample I	D:	490872001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF D	F Analyst Date	e Time Batch	Method	
Semi-Volatile-GC/MS										
SW846 8270 SIM 1,4-I	Dioxane in Li	quid "As Received"								
1,4-Dioxane	Q	9.22	1.00	2.00	ug/L	0.200	1 JMB3 09/24/1	9 1854 1919444	4	
1,4-Dioxane	h	14.8	1.00	2.00	ug/L	0.200	1 JMB3 10/02/1	9 1652 1922216	5	
The following Prep Me	thods were p	erformed:								
Method	Description	n		Analyst	Date	Ti	me Prep Bate	ch		
EPA 537.1 Mod, PFAS, Con	npl PFCs Extract	ion in Liquid]	LM1	09/27/19	08	30 1921239			
SW846 3535A	SW8270E SI	M Prep 1,4-Dioxane	:	SJ	10/02/19	10	00 1922215			
SW846 3535A	SW8270E SI	M Prep 1,4-Dioxane	:	SJW1	09/23/19	12	00 1919441			
The following Analytical Methods were performed:										
Method	Description Analyst Comments									
1	EPA 537.1 M	od, PFAS, Compliant with QS	M Table B-15							
2	EPA 537.1 M	od, PFAS, Compliant with QS	M Table B-15							
3	EPA 537.1 M	od, PFAS, Compliant with QS	M Table B-15							
4	SW846 3535A	A/8270E SIM								
5	SW846 35354	A/8270E SIM								
Surrogate/Tracer Recov	very Test			F	Result	Nominal	Recovery%	Acceptable L	imits	
1,4-Dioxane-d8	SW846 Receive	8270 SIM 1,4-Dioxane in Liq	uid "As	24	4.2 ug/L	40.0	60*	(70%-130%))	
1,4-Dioxane-d8		8270 SIM 1,4-Dioxane in Liq	uid "As	37	7.7 ug/L	40.0	94	(70%-130%))	
Notes:										
Column headers are de	fined as follo	ws:								

DF: Dilution Factor	Lc/LC: Critical Level
DL: Detection Limit	PF: Prep Factor
MDA: Minimum Detectable Activity	RL: Reporting Limit
MDC: Minimum Detectable Concentration	on SQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 7

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia

Mr. Jim Riley

Workorder: 490872

Contact:

Parmname	NOM	Sample Qual	QC	Units RPD	/D% REC%	Range Anls	t Date Time
Perfluorinated Compounds Batch 1921240							
QC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2		15.7	ng/L	86	(60%-145%) J.	LS 10/02/19 06:05
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5		20.4	ng/L	110	(56%-143%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7		17.5	ng/L	94	(57%-138%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5		19.3	ng/L	99	(63%-131%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5		21.5	ng/L	111	(62%-133%)	
Perfluorobutanesulfonic acid (PFBS)	17.2		16.6	ng/L	96	(68%-136%)	
Perfluorobutyric acid (PFBA)	19.5		19.7	ng/L	101	(70%-133%)	
Perfluorodecanesulfonic acid (PFDS)	18.8		16.8	ng/L	89	(53%-142%)	
Perfluorodecanoic acid (PFDA)	19.5		18.0	ng/L	93	(62%-135%)	
Perfluorododecanoic acid (PFDoA)	19.5		19.5	ng/L	100	(66%-131%)	
Perfluoroheptanesulfonic acid (PFHpS)	18.5		18.1	ng/L	98	(66%-138%)	
Perfluoroheptanoic acid (PFHpA)	19.5		17.9	ng/L	92	(67%-135%)	
Perfluorohexanesulfonic acid (PFHxS)	17.7		14.5	ng/L	82	(64%-137%)	
Perfluorohexanoic acid (PFHxA)	19.5		18.9	ng/L	97	(67%-133%)	

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QC Summary

Workorder: 490872		•								Page 2 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated Compounds Batch 1921240										
Perfluorononanesulfonic acid (PFNS)	18.7			17.5	ng/L		93	(66%-130%) JLS	10/02/19 06:05
Perfluorononanoic acid (PFNA)	19.5			21.1	ng/L		108	(66%-134%)	
Perfluorooctanesulfonamide (PFOSA)	19.5			21.5	ng/L		111	(68%-137%)	
Perfluorooctanesulfonic acid (PFOS)	19.5			19.8	ng/L		102	(61%-131%)	
Perfluorooctanoic acid (PFOA)	19.5			18.8	ng/L		97	(63%-145%)	
Perfluoropentanesulfonic acid (PFPeS)	18.3			16.5	ng/L		90	(62%-139%)	
Perfluoropentanoic acid (PFPeA)	19.5			19.3	ng/L		99	(69%-132%)	
Perfluorotetradecanoic acid (PFTeDA)	19.5			22.5	ng/L		115	(65%-143%)	
Perfluorotridecanoic acid (PFTrDA)	19.5			19.9	ng/L		102	(57%-149%)	
Perfluoroundecanoic acid (PFUdA)	19.5			19.1	ng/L		98	(65%-134%)	
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6			20.5	ng/L	26	116	(0%-35%)	10/02/19 06:14
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9			17.6	ng/L	14	98	(0%-36%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1			19.9	ng/L	13	110	(0%-39%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8			20.1	ng/L	4	107	(0%-25%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8			21.9	ng/L	2	116	(0%-26%)	

Page 6 of 17 SDG: 490872 Rev1

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QC Summary

Workorder: 490872								Page 3 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	. 4	103	(0%-30%) JLS	5 10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	. 2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	. 16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	. 3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	. 2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	, 9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	. 15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	. 10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	. 4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	. 12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	. 6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	. 1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	. 0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	. 4	98	(0%-29%)	

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QC Summary

Workorder: 490872								
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Page 4 of 7 Date Time
Perfluorinated Compounds Batch 1921240		Bumpie Qua	<u><u>v</u>v</u>	Cints	R D D / 0	REC /U	Kunge Amst	
Perfluoropentanoic acid (PFPeA)	18.8		20.0	ng/L	3	106	(0%-30%) JLS	10/02/19 06:14
Perfluorotetradecanoic acid (PFTeDA)	18.8		20.6	ng/L	9	109	(0%-30%)	
Perfluorotridecanoic acid (PFTrDA)	18.8		17.7	ng/L	11	94	(0%-35%)	
Perfluoroundecanoic acid (PFUdA)	18.8		21.2	ng/L	10	112	(0%-28%)	
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)		U	ND	ng/L				10/02/19 05:56
Fluorotelomer sulfonate 6:2 (6:2 FTS)		U	ND	ng/L				
Fluorotelomer sulfonate 8:2 (8:2 FTS)		U	ND	ng/L				
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)		U	ND	ng/L				
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)		U	ND	ng/L				
Perfluorobutanesulfonic acid (PFBS)		U	ND	ng/L				
Perfluorobutyric acid (PFBA)		U	ND	ng/L				
Perfluorodecanesulfonic acid (PFDS)		U	ND	ng/L				
Perfluorodecanoic acid (PFDA)		U	ND	ng/L				
Perfluorododecanoic acid (PFDoA)		U	ND	ng/L				
Perfluoroheptanesulfonic acid (PFHpS)		U	ND	ng/L				

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QC Summary

Workorder: 490872								Page 5 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D% REC%	% Range	Anlst	Date Time
PerfluorinatedCompoundsBatch1921240								
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L			JLS	10/02/19 05:56
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L				
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L				
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L				
Perfluorononanoic acid (PFNA)		U	ND	ng/L				
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L				
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L				
Perfluorooctanoic acid (PFOA)		U	ND	ng/L				
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L				
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L				
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L				
Perfluorotridecanoic acid (PFTrDA)		U	ND	ng/L				
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L				
Semi-Volatile-GC/MS Batch 1919444								
QC1204387349 LCS *1,4-Dioxane-d8	4.00		3.55	ug/L	89	(70%-130%) JMB3	09/24/19 12:24

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QC Summary

Workorder: 49()872				_					Page	6 of 7
Parmname		NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Semi-Volatile-GC/MS Batch 19194	144										
QC1204387350 **1,4-Dioxane-d8	LCSD	4.00		3.18	ug/L		79	(70%-130%)	JMB3	09/24/1	9 12:49
QC1204387348 1,4-Dioxane	MB		U	ND	ug/L					09/24/1	9 11:59
**1,4-Dioxane-d8		4.00		3.05	ug/L		76	(70%-130%)			
Batch 19222											
QC1204393997 **1,4-Dioxane-d8	LCS	4.00		4.08	ug/L		102	(70%-130%)	JMB3	10/02/1	9 15:34
QC1204393998 **1,4-Dioxane-d8	LCSD	4.00		3.76	ug/L		94	(70%-130%)		10/02/1	9 16:02
QC1204393996 1,4-Dioxane	MB		U	ND	ug/L					10/02/1	9 15:07
**1,4-Dioxane-d8		4.00		3.87	ug/L		97	(70%-130%)			

Notes:

The Qualifiers in this report are defined as follows:

- ** Analyte is a surrogate compound
- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded

J See case narrative for an explanation

- J Value is estimated
- JNX Non Calibrated Compound

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QC Summary

	•		-	X U D	ummu	<u>J</u>						
Workon	rder: 490872										Pag	e 7 of '
Parmna	me	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Ν	OrganicsPresumptive evide on nearest internal standard r	response factor	•	-								
N N/A	Presumptive evidence based internal standard response fa RPD or %Recovery limits do	ctor	brary search	to make a	tentative id	entificatio	on of the analy	te (TIC). Ç	uantitation	is based of	on neares	t
N1	See case narrative											
ND	Analyte concentration is not	detected above the	detection lin	nit								
NJ	Consult Case Narrative, Data	a Summary package	e, or Project I	Manager	concerning	his qualif	ier					
Р	OrganicsThe concentrations between the primary and confirmation columns/detectors is >40% different. For HPLC, the difference is >70%.											
Q	One or more quality control criteria have not been met. Refer to the applicable narrative or DER.											
R	Sample results are rejected											
U	Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.											
UJ	Compound cannot be extracted											
Х	Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier											
Y	QC Samples were not spiked with this compound											
٨	RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.											
h	Preparation or preservation h	olding time was ex	ceeded									
^ The R five tim	licates that spike recovery limi telative Percent Difference (RF les (5X) the contract required of sed to evaluate the DUP result	PD) obtained from the letection limit (RL)	the sample du	uplicate (DUP) is eva	luated ag	ainst the accep	otance criter	ia when the	e sample i	s greater	

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490872

GC/MS Semivolatile

<u>Product:</u> Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1919444

Preparation Method: SW846 3535A **Preparation Procedure:** GL-OA-E-073 REV# 2 **Preparation Batch:** 1919441

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490872001	0403-1
1204387348	Method Blank (MB)
1204387349	Laboratory Control Sample (LCS)
1204387350	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Surrogate Recoveries

Sample (See Below) did not meet surrogate recovery acceptance criteria. The sample was re-extracted out of holding and met acceptance criteria for all surrogates. Both sets of data results have been reported.

Sample	Analyte	Value		
490872001 (0403-1)	1, 4-Dioxane-d8	60* (70%-130%)		

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

<u>Product:</u> Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2

Page 12 of 17 SDG: 490872 Rev1

Analytical Batch: 1922216

<u>Preparation Method:</u> SW846 3535A <u>Preparation Procedure:</u> GL-OA-E-073 REV# 2 <u>Preparation Batch:</u> 1922215

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	<u>Client Sample Identification</u>
490872001	0403-1
1204393996	Method Blank (MB)
1204393997	Laboratory Control Sample (LCS)
1204393998	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

Technical Information

Holding Time Specifications

Sample (See Below) was re-extracted out of holding due to QC failure. The failure did not confirm, so both sets of results are reported and have been qualified accordingly.

Sample	Value
490872001 (0403-1)	Received 19-SEP-19, within holding, prepped 02-OCT-19, out of holding 23-SEP-19

Miscellaneous Information

Manual Integrations

Sample (See Below) required manual integration in order to properly identify one or more peaks and/or to correctly position the baseline as set in the calibration standard injections.

Sample	Analyte	Value		
490872001 (0403-1)	Tetrahydrofuran-d8	Result 100ug/L		

LCMSMS-Misc

<u>Product:</u> The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15

Page 13 of 17 SDG: 490872 Rev1

Analytical Procedure: GL-OA-E-076 REV# 7 Analytical Batches: 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490872001	0403-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490872001 (0403-1).

A 1.	490872
Analyte	001
Fluorotelomer sulfonate 4:2 (4:2 FTS)	100X
Fluorotelomer sulfonate 6:2 (6:2 FTS)	10X
Perfluorobutanesulfonate (PFBS)	10X
Perfluorobutyric acid (PFBA)	10X
Perfluorohexanoic acid (PFHxA)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

Page: 1	of 1												GEL Labor	GEL Laboratories, LLC			<u> </u>
1#	NWA-001	0077.9	22		<u>м</u>	Laboratories LLC	tories	SLLC					2040 Savage Road	ze Road			
GEL Quote #:	NWRA Quote		1	gel.com	n Chem	Chemistry I Radiochemistry I Radiobioassay Specialty Analytics	temistry I Ra-	diobioassay	Specialty /	Analytics			Charleston, SC 29407	SC 29407			
COC Number ⁽¹⁾ :					n of Cu	Chain of Custody and Analytical Request	d Analyti	ical Req.	uest				Phone: (84)	Phone: (843) 556-8171			
PO Number:	NA	GEL Work (GEL Work Order Number:			GEL Project Manager:	et Manage	r: 					Fax: (843) 766-1178	766-1178			Π
Client Name: 1	NWRA c/o Hart & Hickman, PC	nan, PC		Phone # 9.	919-847-4241	4241		Sa	mple An	alysis R	Sample Analysis Requested ⁽⁵⁾		he number	of container	(Fill in the number of containers for each test)		
Project/Site Name	Project/Site Name: Chambers Development MSWLF	tent MSWLF		Fax # 7(704-586-0007	0007	Shot	Should this	51						< Preserval	< Preservative Type (6)	
Address:	Polkton, NC						san Cons	sample be considered:	onieon	†@	FAS 21	cmpd list	by EPA	PFAS 21 cmpd list by EPA 537 mod			
Collected By: 1	Patrick Stevens	Send Results	Send Results To:Genna Olson g	lson gol	son@hɛ	olson@harthickman	8	t.q2	t of co	μ <u>–</u>	1,4-Dioxane by	ne by El	EPA 8270SIM	SIM	Note: extra	Comments Note: extra sample is	
* For com	Sample ID * For composites - indicate start and stop dateritme		*Date Collected (mm-dd-vv)	*Time Collected (Military) (hhmn)	QC Code ⁽²⁾ Fil	Field Sample Filtered (3) Matrix (4)	Norobic noro Algores subly Bradionetive Bradionetive	possible haza (7) Known or Marad	Total number	\rightarrow]	required f specif	required for sample specific QC	Elect
	0403-1		09-16-19		z	N ML			4 ×	×							
																	
																	μιι
																	y.
																	Re
																	, u
	С	Chain of Custody Signatures	ly Signatures					TA	TAT Requested: Normal:	ted: N		X Rush:	Specify:	y:	(Subject to Surcharge)	surcharge)	Ч.
Relinquished By (Signed)	Date	Time	Received by (signed)		Date	Time		Fax Rest	Fax Results: [] Yes	cs [X] No	No						κs
1- Farrier H. Auros	. Junt 09-18-19 1630	30	2	6	10	906	R	Select D.	eliverable.	[]Cot	Select Deliverable: [] C of A [] QC Summary	Summary	[] level []	[] Level 2	[] Level 3	[] Level 4	<u> </u>
2		5	>		-			Addition	Additional Remarks.	:X:							
9		[]						For Lab	For Lab Receiving Use Only:	g Use On	dy: Custod	Custody Seal Intact? [1? [] Yes	[] No Co	Cooler Temp:	°C	
> For sample sh	> For sumple shipping and delivery details, see Sample Receipt & Review form (SRR.)	Sample Receipt	& Review form .	(SRR.)	S	Sample Collection Time Zone: [X] Eastern	ction Time.	Zone: [X]	Eastern	[] Pacific	ic [] Central	1 . .	[] Mountain	[] Other:			<u>'</u>
 Chain of Custody QC Codes: N = N 	 Chain of Custody Number = Client Determined Partice Sumple, NS = Matrix Spike Sumple, NS = Matrix Spike Sumple, MS = Matrix Spike Sumple, MS = Matrix Spike Duplicate Sumple, G = Grab, C = Composite 	Field Duplicate, EB =	Equípment Blank, 1	MS = Matrix Si	oike Sample,	MSD = Matrix	: Spike Duplic	ate Sample, G	; = Grab, C -	- Composíu	5						1/23
3.) Field Filtered: Fo	 Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered 	or yes the sample was	field filtered or - N -	for sample wa.	s not field fill	tered.											"_
4.) Matrix Codes: DV	4.) Matrix Codes: DW=Drinking Water, GW=Greundwater, SW=Surface Water, WW=Water, WL=Mise Liquid, SO=Soil, SD=Sediment, SL=Sludge, SS=Soid Waste, O=Oil, F=Filter, P=Wipe, U=Head, N=Nast	SW'=Surface Water,	WW=Waste Water,	W=Water, ML	=Mise Liqui	d, SO=Soil, SD	⊨Sediment, SI	L=Sludge, SS	=Solid Wast	s. 0≔0ìl, F	'=Filter, P=Wi	pe. U≃Urine, F	²=Fecal, N=Nau	sal			
5.) Sample Analysis I	 Sample Analysis Requested: Analytical method requested (i.e. 82608, 601087/470A) and number of containers provided for each (i.e. 82608 - 3, 601087/4704 - 1). 	l (i.e. 8260B, 6010B/.	7470A) and number-	of containers p.	rovided for e.	ach (i.e. <i>82608</i>		170.4 - 1). 	5	-	-						
7.) Are there any i	7.) Are there any known or possible hazards	Characteristic Hazards	c Hazards	Listed V	Waste		and 21 - 200	Other	ite, ti no pre	ci vauve 15	aducu Icave			Please provi	Please provide any additional details	tal details	182
associated wh	associated with these samples?	FL = Flammable/Ignitable	ble/Ignitable	LW=L	LW= Listed Waste	e]	01= OI	OT= Other / Unknown	umo				below regard	below regarding handling and/or disposal	ind/or disposi	2
RCRA Metals		CO = Corrosive RE = Reactive	ve	(F,K,P and U-I Waste code(s):	and U-liste ode(s):	P and U-listed wastes.) le code(s):		(i.e.: Hig misc. hee	(i.e.: High/low pH, asbeste mise_health harards_etc)	asbesto.	s, beryllium	(i.e.: High/low pH, asbestos, beryllium, irritants, other misc health hazards_etc)	other	concerns. (i.	concerns. (i.e.: Origin of sample(s), type of eits collocied from odd wattriens atc.)	umple(s), type atricas atc.)	
As = Arsenic	Hg= Mercury				- /			Description:	ion:	(mm (en				ס) אוג באוגרי	teu from, ouu n	ומותוכא, כורי)	1904) 1904)
Ba = Barium Cd = Cadmium	Se= Selenium Ag= Silver	TSCA Regulated PCB = Polychlorinated	ated														
Cr = Chromium	,,	biphenyls	nvls														जन
Pb = Lead		ių.															

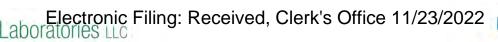
Page 15 of 17 SDG: 490872 Rev1

Client: NIARA		Jh	SAMPLE RECEIPT & REVIEW FORM
Received By: Art			SDG/AR/COC/Work Order: 400612
<u> </u>			Date Received: 9/19/19
Carrier and Tracking Number			Circle Applicable: FedEx Express FedEx Ground UPS Field Services Courier Other FF62 F563 230B -1, FF62 F563 3418-1° F762 F563 2764 -1°, F762 F563 3290 -1°
Suspected Hazard Information	Ves	No	*If Net Counts > 100cpm on samples not marked #
A)Shipped as a DOT Hazardous?		Л	 If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigatio Hazard Class Shipped: UN#: If UN2910, Is the Radioactive Shipment Survey Compliant? Yes No.
B) Did the client designate the samples are to be received as radioactive?	ΙĪ,	Λ	COC notation or radioactive stickers on containers equal client designation.
C) Did the RSO classify the samples as radioactive?		1	Maximum Net Counts Observed (Observed Counts Are Date
D) Did the client designate samples are hazardous?	Ţ,	Λ	Constituent of the country of the background County):CPM / mR/Hr
E) Did the RSO identify possible hazards?	V	Λ^{μ}	D or E is yes, select Hazards below. CB's Flammable Foreign Soil P.C.P.A.
Sample Receipt Criteria	Ves NA	2	RCKA Asbestos Beryllium Other:
I Shipping containers received intact and sealed?			Comments/Qualifiers (Required for Non-Conforming Items) Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
2 Chain of custody documents included with shipment?	1		Circle Applicable: Client contacted and provided COC COC created upon receipt
3 Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?* Daily check performed and passed on IR	1		Preservation Method; Wet the Ice Packs Dry ice None Other: 10
temperature gun?			Temperature Device Serial #: 7.84 - 16* TEMP:
Sample containers intact and sealed? Samples requiring chemical preservation			Circle Applicable: Seals broken Damaged contained Leaking container Other (describe)
at proper pH?	И		Sample (D's and Confainers Affected: If Preservation added, Lord:
Do any samples require Volatile Analysis?		,	If Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) Are liquid VOA vials free of headspace? Yes No NA Sample ID's and containers affected:
Samples received within holding time?			D's and tests affected:
Sample ID's on COC match 1D's on bottles?		-	D's and containers affected:
Date & time on COC match date & time		c	ircle Applicable: No dates on containers No times on containers COC missing info Other (describe)
Number of containers received match number indicated on COC?		Ci	rele Applicable: No container count on COC Other (describe)
Are sample containers identifiable as <u>TEL provided?</u> OC form is properly signed in		+-	
elinquished/received sectione?	₿/	Ci	rcle Applicable: Not relinquished Other (describe)
ents (Use Continuation Form if needed):	<u>v</u>	1	
PM (or PMA) revis	w: Init	ials	Date 123 Page of

Page 16 of 17 SDG: 490872 Rev1

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water Louisiana NELAP	LA024
	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019





a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for Uwharrie Environmental Regional Landfill Work Order: 490875

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490875 GEL Work Order: 490875

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Electroni

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	•		
riojeci.	Analytical forUwharrie Environmental Regional Landfill		
Client Sample ID:	6204-1	Project:	NWRA00119
Sample ID:	490875001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	17-SEP-19 08:55		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Analys	t Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs by L	C-MS/MS	"As Received"										
Fluorotelomer sulfonate 8:2 (8:		35.8	13.2	38.4	ng/L	0.200	1	JLS	10/02/19	0924	1921240	1
FTS)					e							
N-ethylperfluoro-1-		68.0	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (N-											
EtFOSAA) N-methylperfluoro-1-		180	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (N-	180	13.2	40.0	iig/L	0.200	1					
MeFOSAA)												
Perfluorodecanesulfonic acid	U	ND	6.60	19.4	ng/L	0.200	1					
(PFDS)												
Perfluorodecanoic acid (PFDA	,	632	7.80	20.0	ng/L	0.200						
Perfluorododecanoic acid (PFD	,	184	6.60	20.0	ng/L	0.200						
Perfluoroheptanesulfonic acid (PFHpS)	J	9.40	6.60	19.0	ng/L	0.200	1					
Perfluoroheptanoic acid (PFHp	A)	1560	6.60	20.0	ng/L	0.200	1					
Perfluorohexanesulfonic acid		640	6.60	18.2	ng/L	0.200						
(PFHxS)					8		-					
Perfluorononanesulfonic acid (PFNS)	U	ND	7.00	19.2	ng/L	0.200	1					
Perfluorononanoic acid (PFNA	.)	326	6.60	20.0	ng/L	0.200	1					
Perfluorooctanesulfonamide	U	ND	6.60	18.6	ng/L	0.200	1					
(PFOSA)					_							
Perfluorooctanesulfonic acid (I	PFOS)	356	8.00	20.0	ng/L	0.200						
Perfluoropentanesulfonic acid (PFPeS)		41.4	6.60	18.8	ng/L	0.200	1					
Perfluoroundecanoic acid (PFL	JdA)	33.0	6.60	20.0	ng/L	0.200	1					
Perfluorobutanesulfonic acid (I	PFBS)	2870	66.0	178	ng/L	0.200	10	JLS	10/02/19	0731	1921240	2
Perfluorobutyric acid (PFBA)		2400	66.0	200	ng/L	0.200	10					
Perfluorohexanoic acid (PFHx)	A)	5540	66.0	200	ng/L	0.200	10					
Perfluorooctanoic acid (PFOA))	3690	70.0	200	ng/L	0.200	10					
Perfluoropentanoic acid (PFPe	A)	2150	66.0	200	ng/L	0.200	10					
Perfluorotetradecanoic acid (PFTeDA)	U	ND	66.0	200	ng/L	0.200	10					
Perfluorotridecanoic acid (PFT	rDA) U	ND	66.0	200	ng/L	0.200	10					
Fluorotelomer sulfonate 4:2 (4: FTS)	2 U	ND	1320	3760	ng/L	0.200	100	JLS	10/02/19	1134	1921240	3
Fluorotelomer sulfonate 6:2 (6: FTS)	2 U	ND	1320	3800	ng/L	0.200	100					
Sami Valatila CC/MS												

Semi-Volatile-GC/MS

Page 3 of 16 SDG: 490875 Rev1

Electronic Filing Elecared RAFORIES

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forUwharrie Environmental Regional Landfill		
Client Sample ID:	6204-1	Project:	NWRA00119
Sample ID:	490875001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Analyst Date	Time Ba	atch	Method
Semi-Volatile-GC/MS											
SW846 8270 SIM 1,4-D	Dioxane in Li	quid "As Received"									
1,4-Dioxane		357	10.0	20.0) ug/L	0.200	10	JMB3 09/24/19	1528 191	9444	4
The following Prep Met	hods were pe	erformed:									
Method	Description	1		Analyst	Date	Т	ìme	Prep Batch			
EPA 537.1 Mod, PFAS, Comp	pl PFCs Extracti	on in Liquid		LM1	09/27/19	0	830	1921239			
SW846 3535A	SW8270E SI	M Prep 1,4-Dioxane		SJW1	09/23/19	1	200	1919441			
The following Analytic	al Methods v	vere performed:									
Method	Description					Analyst	Con	nments			
1	EPA 537.1 Mo	od, PFAS, Compliant with QSM T	able B-1	5							
2	EPA 537.1 Mo	od, PFAS, Compliant with QSM T	able B-1	5							
3	EPA 537.1 Mo	od, PFAS, Compliant with QSM T	able B-1	5							
4	SW846 3535A	/8270E SIM									
Surrogate/Tracer Recov	ery Test				Result	Nomina	1	Recovery%	Acceptab	le Li	mits
1,4-Dioxane-d8	SW846 Receive	8270 SIM 1,4-Dioxane in Liquid ' d"	As		40.4 ug/L	40.0	0	101	(70%-1	30%)	
Notes:											

Column headers are defined as follows:DF: Dilution FactorLc/LC: Critical LevelDL: Detection LimitPF: Prep FactorMDA: Minimum Detectable ActivityRL: Reporting LimitMDC: Minimum Detectable ConcentrationSQL: Sample Quantitation Limit

2040 Savage Road Charleston, SC 29407 - (843) 556-8171 - www.gel.com

QC Summary

Report Date: November 8, 2019

Page 1 of 7

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia Mr. Jim Riley

Workorder: 490875

Contact:

Parmname	NOM	Sample Qual	QC	Units RPD/D	% REC%	Range Anl	st Date Time
Perfluorinated CompoundsBatch1921240							
QC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2		15.7	ng/L	86	(60%-145%)	ILS 10/02/19 06:05
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5		20.4	ng/L	110	(56%-143%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7		17.5	ng/L	94	(57%-138%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5		19.3	ng/L	99	(63%-131%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5		21.5	ng/L	111	(62%-133%)	
Perfluorobutanesulfonic acid (PFBS)	17.2		16.6	ng/L	96	(68%-136%)	
Perfluorobutyric acid (PFBA)	19.5		19.7	ng/L	101	(70%-133%)	
Perfluorodecanesulfonic acid (PFDS)	18.8		16.8	ng/L	89	(53%-142%)	
Perfluorodecanoic acid (PFDA)	19.5		18.0	ng/L	93	(62%-135%)	
Perfluorododecanoic acid (PFDoA)	19.5		19.5	ng/L	100	(66%-131%)	
Perfluoroheptanesulfonic acid (PFHpS)	18.5		18.1	ng/L	98	(66%-138%)	
Perfluoroheptanoic acid (PFHpA)	19.5		17.9	ng/L	92	(67%-135%)	
Perfluorohexanesulfonic acid (PFHxS)	17.7		14.5	ng/L	82	(64%-137%)	
Perfluorohexanoic acid (PFHxA)	19.5		18.9	ng/L	97	(67%-133%)	

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QC Summary

Workorder: 490875		•								Page 2 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated Compounds Batch 1921240										
Perfluorononanesulfonic acid (PFNS)	18.7			17.5	ng/L		93	(66%-130%) JLS	10/02/19 06:05
Perfluorononanoic acid (PFNA)	19.5			21.1	ng/L		108	(66%-134%)	
Perfluorooctanesulfonamide (PFOSA)	19.5			21.5	ng/L		111	(68%-137%)	
Perfluorooctanesulfonic acid (PFOS)	19.5			19.8	ng/L		102	(61%-131%)	
Perfluorooctanoic acid (PFOA)	19.5			18.8	ng/L		97	(63%-145%)	
Perfluoropentanesulfonic acid (PFPeS)	18.3			16.5	ng/L		90	(62%-139%)	
Perfluoropentanoic acid (PFPeA)	19.5			19.3	ng/L		99	(69%-132%)	
Perfluorotetradecanoic acid (PFTeDA)	19.5			22.5	ng/L		115	(65%-143%)	
Perfluorotridecanoic acid (PFTrDA)	19.5			19.9	ng/L		102	(57%-149%)	
Perfluoroundecanoic acid (PFUdA)	19.5			19.1	ng/L		98	(65%-134%)	
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6			20.5	ng/L	26	116	(0%-35%)	10/02/19 06:14
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9			17.6	ng/L	14	98	(0%-36%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1			19.9	ng/L	13	110	(0%-39%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8			20.1	ng/L	4	107	(0%-25%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8			21.9	ng/L	2	116	(0%-26%)	

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QC Summary

Workorder: 490875								Page 3 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	. 4	103	(0%-30%) JLS	S 10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	. 2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	. 3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	. 16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	. 3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	. 2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	. 9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	. 15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	. 10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	. 4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	. 12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	<i>.</i> 6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	. 1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	. 0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	. 4	98	(0%-29%)	

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QC Summary

Workorder: 490875		-			<u></u>					Dess 4 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Page 4 of 7 Date Time
Perfluorinated Compounds Batch 1921240		·	•					.,		
Perfluoropentanoic acid (PFPeA)	18.8			20.0	ng/L	3	106	(0%-30%)	JLS	10/02/19 06:14
Perfluorotetradecanoic acid (PFTeDA)	18.8			20.6	ng/L	9	109	(0%-30%)		
Perfluorotridecanoic acid (PFTrDA)	18.8			17.7	ng/L	11	94	(0%-35%)		
Perfluoroundecanoic acid (PFUdA)	18.8			21.2	ng/L	10	112	(0%-28%)		
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)			U	ND	ng/L					10/02/19 05:56
Fluorotelomer sulfonate 6:2 (6:2 FTS)			U	ND	ng/L					
Fluorotelomer sulfonate 8:2 (8:2 FTS)			U	ND	ng/L					
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)			U	ND	ng/L					
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)			U	ND	ng/L					
Perfluorobutanesulfonic acid (PFBS)			U	ND	ng/L					
Perfluorobutyric acid (PFBA)			U	ND	ng/L					
Perfluorodecanesulfonic acid (PFDS)			U	ND	ng/L					
Perfluorodecanoic acid (PFDA)			U	ND	ng/L					
Perfluorododecanoic acid (PFDoA)			U	ND	ng/L					
Perfluoroheptanesulfonic acid (PFHpS)			U	ND	ng/L					

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QC Summary

Workorder: 490875								Page	e 5 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D% RE	C% Range	Anlst		Time
Perfluorinated CompoundsBatch1921240									
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L			JLS	10/02/1	9 05:56
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L					
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L					
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L					
Perfluorononanoic acid (PFNA)		U	ND	ng/L					
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L					
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L					
Perfluorooctanoic acid (PFOA)		U	ND	ng/L					
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L					
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L					
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L					
Perfluorotridecanoic acid (PFTrDA)		U	ND	ng/L					
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L					
Semi-Volatile-GC/MS Batch 1919444									
QC1204387349 LCS **1,4-Dioxane-d8	4.00		3.55	ug/L	8	89 (70%-130%	6) JMB3	09/24/1	19 12:24

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QC Summary

Workorder: 49	0875										Page	e 6 of 7
Parmname		NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Semi-Volatile-GC/MS Batch 1919												
QC1204387350 **1,4-Dioxane-d8	LCSD	4.00			3.18	ug/L		79	(70%-130%)	JMB3	09/24/1	9 12:49
QC1204387348 1,4-Dioxane	MB			U	ND	ug/L					09/24/1	9 11:59
**1,4-Dioxane-d8		4.00			3.05	ug/L		76	(70%-130%)			

Notes:

The Qualifiers in this report are defined as follows:

** Analyte is a surrogate compound

- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N/A RPD or %Recovery limits do not apply.

N1 See case narrative

- ND Analyte concentration is not detected above the detection limit
- NJ Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier
- P Organics--The concentrations between the primary and confirmation columns/detectors is >40% different. For HPLC, the difference is >70%.
- Q One or more quality control criteria have not been met. Refer to the applicable narrative or DER.
- R Sample results are rejected
- U \qquad Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.
- UJ Compound cannot be extracted

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QC Summary

Workordo	er: 490875									Page	e 7 of 7
Parmnam	e	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
X (Consult Case Narrative, Data	Summary package	e, or Project Manag	er concerning	this qualif	ïer					

Y QC Samples were not spiked with this compound

^ RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.

h Preparation or preservation holding time was exceeded

N/A indicates that spike recovery limits do not apply when sample concentration exceeds spike conc. by a factor of 4 or more or %RPD not applicable. ^ The Relative Percent Difference (RPD) obtained from the sample duplicate (DUP) is evaluated against the acceptance criteria when the sample is greater than five times (5X) the contract required detection limit (RL). In cases where either the sample or duplicate value is less than 5X the RL, a control limit of +/- the RL is used to evaluate the DUP result.

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490875

GC/MS Semivolatile

Product: Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1919444

Preparation Method: SW846 3535A **Preparation Procedure:** GL-OA-E-073 REV# 2 **Preparation Batch:** 1919441

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490875001	6204-1
1204387348	Method Blank (MB)
1204387349	Laboratory Control Sample (LCS)
1204387350	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

Technical Information

Sample Dilutions

Sample 490875001 (6204-1) was diluted due to the presence of one or more over-range target analytes.

LCMSMS-Misc

<u>Product:</u> The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 <u>Analytical Procedure:</u> GL-OA-E-076 REV# 7 <u>Analytical Batches:</u> 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490875001	6204-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490875001 (6204-1).

Amelanta	490875
Analyte	001
Fluorotelomer sulfonate 4:2 (4:2 FTS)	100X
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Perfluorobutanesulfonate (PFBS)	10X
Perfluorobutyric acid (PFBA)	10X
Perfluorohexanoic acid (PFHxA)	10X
Perfluorooctanoic acid (PFOA)	10X
Perfluoropentanoic acid (PFPeA)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

[T	T	T		Flect	φn	μC.	ĘΠ	μŋ	1.1	≺e	çe	ļve	α,	بال	SLF	<u>('S</u>	\mathbf{O}	ШÇ	ce,	1	1/23/2	UZ								
GEL Laboratories, LLC 2040 Savage Road Charleston, SC 29407 Discos. (212) SC6 6171	766-1178	(Fill in the number of containers for each test)	C Preservative Type (6)		SIM Note extra sample is												y: (Subject to Surcharge)		[] Level 2 [] Level 3 [] Level 4		[] No Cooler Temp:C	[] Other.		22		Please provide any additional details	below regarding handling and/or disposal	concerns. (i.e., Origin of sample(s), type of site collected from, odd matrices, etc.)				
atty Analytics		Sample Analysis Requested ⁽⁵⁾ (Fill in the number		PFAS 21 cmpd list by EPA 537 mod	1	lotal number											TAT Requested: Normal: X Rush: Specify:	Fax Results: [] Yes [X] No	Select Deliverable: [] C of A [] QC Summary [] level 1	Additional Remarks:	For Lab Receiving Use Only: Custody Seal Intact? [] Yes	[] Pacific [] Central [] Mountain	$\mathbf{G} = \mathbf{G}_{\mathbf{T}}\mathbf{a}\mathbf{b}, \ \mathbf{C} = \mathbf{C}_{\mathbf{D}}\mathbf{n}\mathbf{p}\mathbf{e}\mathbf{s}\mathbf{i}\mathbf{t}\mathbf{c}$	4.) Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=Surface Water, WW=Waste Water, WL=Mise Liquid, SO=Soil, SD=Sediment, SL=Studge, SS=Solid Waste. O=Oil. F=Filter, P=Wipe, U=Urine, F=Feed, N=Nused	ltâte. If no preservative is added ≃ leave field blank	Other Other / Linkson	octor hamiltinu imitanta aitan					
Laboratories LLC chemistry - Radiobioassay 1 Speciatly Analytics Custody and Analytical Request	GEL Project Manager:		Should this	sample be considered:	com	Marry S Marry S Marry S Marry S Madioactive Madioactive Madioactive Marry S Marry S Ma	in S Bill										F	Fax Ro	SO Select	Additio	For La	Sample Collection Time Zone: [X] Eastern	atrix Spike Duplicate Sample.	. SD=Sediment, SL=Sludge, S	60B - 3, 6010B/7470A - 1). fexane, ST = Sodium Thiosul	Other			Description:			
Chain of	GEL Pro	Phone # 919-847-4241	Fax # 704-586-0007		Olson golson@harthickman.	*Time Collected (Military) (fihumi) Code ⁽²⁾ Fileld Scode ⁽²⁾ Filered ⁽²⁾ M	z z											d) Date Time	allered us		8		S = Matrix Spike Sample, MSD = Ma r sample was not field filtered.	"=Water, ML=Mise Liquid, SO=Soil,	containers provided for each (i.e. 826 ie Acid, AA = Ascorbic Acid, HX = H	Listed Waste	(F.K.P. and U-listed wastes)	Waste code(s):				
4910876	GEL Work Order Number:				Send Results To:Genna Ol	*Date Collected *Date Collected ne (nm-dd-yy)											Chain of Custody Signatures	Received by (signed)		2 * 2	3	ıple Receipt & Review form (S	Duplicate, EB = Equipment Blank, M the sample was field fillered or - N - fo	Surface Water, WW≠Waste Water, W	82608, 60108/7470.A) and number of SH = Sodium Hydroxide. SA = Sulfuri	Characteristic Hazards FL = Flammable/Ionitable	CO = Corrosive	RE = Reactive	SCA Regulated	PCB = Polychlorinated	biphenyls	
Page: 1 of 1 Project # NWA-001 0	NA	Client Name: NWRA c/o Hart & Hickman, PC	Project/Site Name: Uwharrie Environmental Regional Landfill	Address: Mt Gilead, NC	Collected By: Patrick Stevens S	Sample ID * For composites - indicate start and stop date time	6204-1											Date	- / and H. June 19-19-19 1030			> For sample shipping and delivery details, see Sample Receipt & Review form (SRR.)	 Chain of Custody Number = Client Determined Chain of Custody Number = Client Determined QC Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sample, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite QC Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sample, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite Pfield Filtered: For fiquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered. 	 Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=6 	 Sample Analysis Requested: Analytical method requested (i.e. 82608, 60108/7470A) and number of containers provided for each (i.e. 82608 - 3, 60/08/7470A - 1). Preservative Type: Ha = Hydrochloric Acid, NI = Nitric Acid, SH = Sodium Hydroxide. Sa = Sulfuric Acid, AA = Ascorbis Acid, HX = Hexane, ST = Sodium Thiosulfate, If no preservative is added = keave field blank 	7.) Are there any known or possible hazards C		ls university of the second se		Ag= Silver	nium MR= Miscellaneous	Pb = Lead RCRA metals

Page 14 of 16 SDG: 490875 Rev1

ATTACHMENT D

1/00/0000

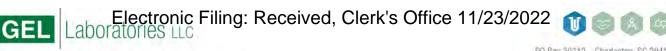
Laboratories Lo	
Client: NWBA Jh	SAMPLE RECEIPT & REVIEW FORM
Received By: Art	SUGAROCOC/Work Order: 400 8
y.c	Date Received: 9/19/19
Carrier and Tracking Number	Circle Applicable: FedEx Express FedEx Ground UPS Field Services Courier Other FF62 F563 230B -1, FF62 F563 3418-1° FF62 F563 2F64 -1°, FF62 F563 3290 -1°
Suspected Hazard Information	"If Net Counts > 100 nm an and 100 nm an
report us a DOT mazardous?	*If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation. Hazard Class Shipped: UN#: f UN2910, Is the Radioactive Shipment Survey Compliant? Yes No
B) Did the client designate the samples are to be ceeived as radioactive?	COC notation or radioactive stickers on containers equal client designation.
radioactive?	faximum Net Counte Ohanna ta car
by bid the chent designate samples are	Inssified as: Rad I Rad 2 Rad 3 OC notation or hazard labels on containers equal elient designation.
E) Did the RSO identify possible hazards?	D'or E is yes, select Hazards below. B's Flammable Foreign Soil RCD A state
Sample Receipt Criteria	Asbestos Beryllium Other:
I Shipping containers received intact and sealed?	Comments/Qualifiers (Required for Non-Conforming Items) Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
2 Chain of custody documents included with shipment?	Circle Applicable: Client control (
3 Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?*	Preservation Method: Vet lo Ice Packs Dry ice None Other:
4 Daily check performed and passed on IR temperature gun?	Temperature Device Serial #: 7.84-16 Secondary Temperature Device Serial # (If Applicable):
• Sumple containers intact and sealed?	Crete Applicable: Seals, broken Damaged container Leaking container Other (describe)
6 at proper pH?	Sample ID's and Containers Affected:
7 Do any samples require Volatile Analysis?	Preservation added, Lott [#] TYes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) To liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) re liquid VOA vials free of headspace? Yes No NA mple ID's and containers affected:
	y's and tests affected:
V REAL	's and containers affected:
10 Date & time or COC match date & time Cin	cle Applicable: No dates on containers No times on containers COC missing info Other (describe)
11 number indicated on COC?	ele Applicable: No container count on COC Other (describe)
COC form in	
Control of the state of th	ele Applicable: Not relinquished Diher (describe)
PM (or PMA) review: Initials	Vor Date 125 Page for

ATTACHMENT D

4

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water Louisiana NELAP	LA024
	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019



a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for Great Oak Landfill Work Order: 490876

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490876 GEL Work Order: 490876

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecared RAFORIES ice C1/23/2022

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forGreat Oak Landfill		
Client Sample ID:	7607-1	Project:	NWRA00119
Sample ID:	490876001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	17-SEP-19 13:10		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Anal	yst Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs b	y LC-MS/MS	"As Received"										
N-ethylperfluoro-1-	J	15.6	13.2	40.0	ng/L	0.200	1	JLS	10/04/19	1052	1921240	1
octanesulfonamidoacetic ad	cid (N-				-							
EtFOSAA)		42.4	12.2	40.0		0.200	1					
N-methylperfluoro-1- octanesulfonamidoacetic ad	rid (N-	42.4	13.2	40.0	ng/L	0.200	1					
MeFOSAA)												
Perfluorobutanesulfonic ac	id (PFBS)	72.2	6.60	17.8	ng/L	0.200	1					
Perfluorobutyric acid (PFB	A)	303	6.60	20.0	ng/L	0.200	1					
Perfluorodecanesulfonic ac	id J	7.10	6.60	19.4	ng/L	0.200	1					
(PFDS)		10 5	- 00	20.0		0.000						
Perfluorodecanoic acid (PF	,	18.5	7.80	20.0	ng/L	0.200						
Perfluorododecanoic acid (, -	ND	6.60	20.0	ng/L	0.200						
Perfluoroheptanesulfonic a (PFHpS)		ND	6.60	19.0	ng/L	0.200						
Perfluoroheptanoic acid (P		68.4	6.60	20.0	ng/L	0.200						
Perfluorohexanesulfonic ac	cid	59.1	6.60	18.2	ng/L	0.200	1					
(PFHxS) Perfluorohexanoic acid (PF	ZT I A \	440	((0	20.0		0.200	1					
Perfluorononanesulfonic ac	· ·	449 ND	6.60 7.00	20.0 19.2	ng/L ng/L	0.200						
(PFNS)					-							
Perfluorononanoic acid (PF		32.8	6.60	20.0	ng/L	0.200						
Perfluorooctanesulfonamid (PFOSA)	e J	8.75	6.60	18.6	ng/L	0.200	1					
Perfluorooctanesulfonic ac	id (PFOS)	83.9	8.00	20.0	ng/L	0.200	1					
Perfluorooctanoic acid (PF	OA)	108	7.00	20.0	ng/L	0.200						
Perfluoropentanesulfonic a (PFPeS)	cid J	10.3	6.60	18.8	ng/L	0.200	1					
Perfluoropentanoic acid (P	FPeA)	159	6.60	20.0	ng/L	0.200	1					
Perfluoroundecanoic acid (PFUdA) J	7.44	6.60	20.0	ng/L	0.200	1					
Fluorotelomer sulfonate 8:2 FTS)	2 (8:2 U	ND	132	384	ng/L	0.200	10	JLS	10/02/19	0740	1921240	2
Perfluorotetradecanoic acid (PFTeDA)	U U	ND	66.0	200	ng/L	0.200	10					
Perfluorotridecanoic acid (PFTrDA) U	ND	66.0	200	ng/L	0.200	10					
Fluorotelomer sulfonate 4:	,	ND	1320	3760	ng/L	0.200		JLS	10/02/19	1143	1921240	3
FTS)					6							
Fluorotelomer sulfonate 6:2 FTS)	2 (6:2 U	ND	1320	3800	ng/L	0.200	100					
The following Prep M	lethods were p	erformed:										
	1											

Electronic Filing Elecared RAFORIES ice C1/23/2022

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forGreat Oak Landfill		
Client Sample ID:	7607-1	Project:	NWRA00119
Sample ID:	490876001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF DF	Analyst Date	Time Batch	Method			
The following Prep Methods were performed:												
Method	Description		1	Analyst	Date	Time	Prep Batch					
EPA 537.1 Mod, PFAS, G	Compl PFCs Extraction	on in Liquid	I	LM1	09/27/19	0830	1921239					
The following Anal	ytical Methods w	ere performed:										
Method	Description				А	.nalyst Con	nments					
1	EPA 537.1 Mod	d, PFAS, Compliant with	QSM Table B-15									
2	EPA 537.1 Mod	d, PFAS, Compliant with	QSM Table B-15									
3	EPA 537.1 Mod	d, PFAS, Compliant with	QSM Table B-15									

Notes:

Column headers are defined as follows:	
DF: Dilution Factor	Lc/LC: Critical Level
DL: Detection Limit	PF: Prep Factor
MDA: Minimum Detectable Activity	RL: Reporting Limit
MDC: Minimum Detectable Concentration	SQL: Sample Quantitation Limit

Electronic Filing Elecared RAFORIES ice C1/23/2022

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Certificate of Analysis

Report Date: November 8, 2019

Company : Address :	NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804		
Traditests .			
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forGreat Oak Landfill		
Client Sample ID:	7607-EB	Project:	NWRA00119
Sample ID:	490876002	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	17-SEP-19 12:40		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter Q	ualifier	Result	DL	RL	Units	PF	DF	Analyst Date	Time	e Batch	Method
LCMSMS PFCs											
EPA 537Mod PFCs by LC-	MS/MS	"As Received"									
Fluorotelomer sulfonate 4:2 (4:2	U	ND	1.15	3.29	ng/L	0.0175	1	JLS 10/02/19	9 0941	1921240	1
FTS)											
Fluorotelomer sulfonate 6:2 (6:2	U	ND	1.15	3.32	ng/L	0.0175	1				
FTS) Fluorotelomer sulfonate 8:2 (8:2	U	ND	1.15	3.36	ng/L	0.0175	1				
FTS)	U	ND	1.15	5.50	ng/L	0.0175	1				
N-ethylperfluoro-1-	U	ND	1.15	3.50	ng/L	0.0175	1				
octanesulfonamidoacetic acid (N-											
EtFOSAA)		ND	1.15	2.50	/T	0.0175	1				
N-methylperfluoro-1- octanesulfonamidoacetic acid (N-	U	ND	1.15	3.50	ng/L	0.0175	1				
MeFOSAA)											
Perfluorobutanesulfonic acid (PFB	S) U	ND	0.577	1.56	ng/L	0.0175	1				
Perfluorobutyric acid (PFBA)	J	1.12	0.577	1.75	ng/L	0.0175	1				
Perfluorodecanesulfonic acid	U	ND	0.577	1.70	ng/L	0.0175	1				
(PFDS)					-						
Perfluorodecanoic acid (PFDA)	U	ND	0.682	1.75	ng/L	0.0175	1				
Perfluorododecanoic acid (PFDoA	/	ND	0.577	1.75	ng/L	0.0175	1				
Perfluoroheptanesulfonic acid (PFHpS)	U	ND	0.577	1.66	ng/L	0.0175	1				
Perfluoroheptanoic acid (PFHpA)	U	ND	0.577	1.75	ng/L	0.0175	1				
Perfluorohexanesulfonic acid (PFHxS)	U	ND	0.577	1.59	ng/L	0.0175	1				
Perfluorohexanoic acid (PFHxA)	U	ND	0.577	1.75	ng/L	0.0175	1				
Perfluorononanesulfonic acid	U	ND	0.612	1.68	ng/L	0.0175	1				
(PFNS)											
Perfluorononanoic acid (PFNA)	U	ND	0.577	1.75	ng/L	0.0175	1				
Perfluorooctanesulfonamide	U	ND	0.577	1.63	ng/L	0.0175	1				
(PFOSA) Perfluorooctanesulfonic acid (PFO	DS) U	ND	0.699	1.75	ng/L	0.0175	1				
Perfluorooctanoic acid (PFOA)	U (5)	ND	0.612	1.75	ng/L ng/L	0.0175	1				
Perfluoropentanesulfonic acid	U	ND	0.577	1.64	ng/L ng/L	0.0175	1				
(PFPeS)	U	ND	0.577	1.04	ng/L	0.0175	1				
Perfluoropentanoic acid (PFPeA)	U	ND	0.577	1.75	ng/L	0.0175	1				
Perfluorotetradecanoic acid	U	ND	0.577	1.75	ng/L	0.0175	1				
(PFTeDA)											
Perfluorotridecanoic acid (PFTrDA	· ·	ND	0.577	1.75	ng/L	0.0175	1				
Perfluoroundecanoic acid (PFUdA	.) U	ND	0.577	1.75	ng/L	0.0175	1				
The following Prep Method	ls were po	erformed:									

Electronic Filing Elecaised RAI # OKIES ice C1/23/2022

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forGreat Oak Landfill		
Client Sample ID:	7607-EB	Project:	NWRA00119
Sample ID:	490876002	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Analyst Date	Time Batch	Method
The following Pr	rep Methods were pe	erformed:								
Method	Description	n		Analyst	Date		Time	Prep Batch		
EPA 537.1 Mod, PFA	AS, Compl PFCs Extract	ion in Liquid		LM1	09/27/19		0830	1921239		
The following A	analytical Methods v	vere performed:								
Method	Description	l			A	Analys	st Con	nments		
1 EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15										

Notes:

Column headers are defined as follows:DF: Dilution FactorLc/LC: Critical LevelDL: Detection LimitPF: Prep FactorMDA: Minimum Detectable ActivityRL: Reporting LimitMDC: Minimum Detectable ConcentrationSQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 6

NWRA - Carolinas Chapter
1550 Crystal Drive, Suite 804
Arlington, Virginia

Contact: Mr. Jim Riley

Workorder: 490876

Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Perfluorinated CompoundsBatch1921240											
QC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2			15.7	ng/L		86	(60%-145%)	JLS	10/02/1	19 06:05
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5			20.4	ng/L		110	(56%-143%))		
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7			17.5	ng/L		94	(57%-138%))		
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5			19.3	ng/L		99	(63%-131%))		
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5			21.5	ng/L		111	(62%-133%))		
Perfluorobutanesulfonic acid (PFBS)	17.2			16.6	ng/L		96	(68%-136%))		
Perfluorobutyric acid (PFBA)	19.5			19.7	ng/L		101	(70%-133%))		
Perfluorodecanesulfonic acid (PFDS)	18.8			16.8	ng/L		89	(53%-142%))		
Perfluorodecanoic acid (PFDA)	19.5			18.0	ng/L		93	(62%-135%))		
Perfluorododecanoic acid (PFDoA)	19.5			19.5	ng/L		100	(66%-131%))		
Perfluoroheptanesulfonic acid (PFHpS)	18.5			18.1	ng/L		98	(66%-138%))		
Perfluoroheptanoic acid (PFHpA)	19.5			17.9	ng/L		92	(67%-135%))		
Perfluorohexanesulfonic acid (PFHxS)	17.7			14.5	ng/L		82	(64%-137%))		
Perfluorohexanoic acid (PFHxA)	19.5			18.9	ng/L		97	(67%-133%))		

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QC Summary

Workorder: 490876		<u>x</u> -		<u> </u>					
Parmname	NOM	Sample Qu	ial QC	Units	RPD/D%	REC%	Range	Anlst	Page 2 of 6 Date Time
Perfluorinated Compounds Batch 1921240					<u> </u>	KEC 70	Kange	Amst	
Perfluorononanesulfonic acid (PFNS)	18.7		17.5	ng/L		93	(66%-130%)) JLS	10/02/19 06:05
Perfluorononanoic acid (PFNA)	19.5		21.1	ng/L		108	(66%-134%))	
Perfluorooctanesulfonamide (PFOSA)	19.5		21.5	ng/L		111	(68%-137%))	
Perfluorooctanesulfonic acid (PFOS)	19.5		19.8	ng/L		102	(61%-131%))	
Perfluorooctanoic acid (PFOA)	19.5		18.8	ng/L		97	(63%-145%))	
Perfluoropentanesulfonic acid (PFPeS)	18.3		16.5	ng/L		90	(62%-139%))	
Perfluoropentanoic acid (PFPeA)	19.5		19.3	ng/L		99	(69%-132%))	
Perfluorotetradecanoic acid (PFTeDA)	19.5		22.5	ng/L		115	(65%-143%))	
Perfluorotridecanoic acid (PFTrDA)	19.5		19.9	ng/L		102	(57%-149%))	
Perfluoroundecanoic acid (PFUdA)	19.5		19.1	ng/L		98	(65%-134%))	
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6		20.5	ng/L	26	116	(0%-35%))	10/02/19 06:14
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9		17.6	ng/L	14	98	(0%-36%))	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1		19.9	ng/L	13	110	(0%-39%))	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8		20.1	ng/L	4	107	(0%-25%))	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8		21.9	ng/L	2	116	(0%-26%))	

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QC Summary

		<u>QC Su</u>	IIIIIai	y				
Workorder: 490876								Page 3 of 6
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	4	103	(0%-30%) JLS	10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	4	98	(0%-29%)	

Page 9 of 17 SDG: 490876 Rev1

2040 Savage Road Charleston, SC 29407 - (843) 556-8171 - www.gel.com

QC Summary

Workorder: 490876		د								
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Page 4 of 6 Date Time
Perfluorinated Compounds Batch 1921240		•						8		
Perfluoropentanoic acid (PFPeA)	18.8			20.0	ng/L	3	106	(0%-30%)	JLS	10/02/19 06:14
Perfluorotetradecanoic acid (PFTeDA)	18.8			20.6	ng/L	9	109	(0%-30%)		
Perfluorotridecanoic acid (PFTrDA)	18.8			17.7	ng/L	11	94	(0%-35%)		
Perfluoroundecanoic acid (PFUdA)	18.8			21.2	ng/L	10	112	(0%-28%)		
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)			U	ND	ng/L					10/02/19 05:56
Fluorotelomer sulfonate 6:2 (6:2 FTS)			U	ND	ng/L					
Fluorotelomer sulfonate 8:2 (8:2 FTS)			U	ND	ng/L					
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)			U	ND	ng/L					
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)			U	ND	ng/L					
Perfluorobutanesulfonic acid (PFBS)			U	ND	ng/L					
Perfluorobutyric acid (PFBA)			U	ND	ng/L					
Perfluorodecanesulfonic acid (PFDS)			U	ND	ng/L					
Perfluorodecanoic acid (PFDA)			U	ND	ng/L					
Perfluorododecanoic acid (PFDoA)			U	ND	ng/L					
Perfluoroheptanesulfonic acid (PFHpS)			U	ND	ng/L					

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QC Summary

		$\underline{\mathbf{v}}$	mmai	y						ŀ
Workorder: 490876				-					Page	5 of 6
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date T	lime
Perfluorinated CompoundsBatch1921240										
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L				JLS	10/02/19	05:56
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L						
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L						
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L						
Perfluorononanoic acid (PFNA)		U	ND	ng/L						
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L						
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L						
Perfluorooctanoic acid (PFOA)		U	ND	ng/L						
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L						
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L						
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L						
Perfluorotridecanoic acid (PFTrDA)		U	ND	ng/L						
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L						

Notes:

The Qualifiers in this report are defined as follows:

- ** Analyte is a surrogate compound
- < Result is less than value reported
- > Result is greater than value reported

Page 11 of 17 SDG: 490876 Rev1

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QC Summary

Parmnan	ne	NO	OM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
А	The TIC is a st	uspected aldol-condensa	tion product										
В	The target ana	lyte was detected in the	associated bl	ank.									
С	Analyte has been confirmed by GC/MS analysis												
D	Results are reported from a diluted aliquot of the sample												
Е	Concentration of the target analyte exceeds the instrument calibration range												
Н	Analytical hole	ding time was exceeded											
J	See case narra	tive for an explanation											
J	Value is estim	ated											
JNX	Non Calibrated	d Compound											
Ν	OrganicsPresumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor RPD or %Recovery limits do not apply.												
N1	See case narra												
ND	Analyte conce	ntration is not detected a	bove the det	ection lin	nit								
	•	Narrative, Data Summar				concerning	this qualif	ier					
Р	OrganicsThe	concentrations between	the primary	and confi	rmation of	columns/de	tectors is >	40% differen	t. For HPL	C, the diffe	rence is >7	70%.	
Q	One or more q	uality control criteria ha	ve not been i	net. Refe	r to the ap	plicable na	rrative or	DER.					
R	Sample results	are rejected											
U	Analyte was a	nalyzed for, but not dete	cted above th	ne MDL,	MDA, M	DC or LOE).						
UJ	Compound car	nnot be extracted											
Х	Consult Case I	Narrative, Data Summar	y package, oi	Project l	Manager	concerning	this qualif	ier					
Y	QC Samples w	vere not spiked with this	compound										
^	RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.												
h	Preparation or	preservation holding tin	ne was excee	ded									
^ The Re five time	elative Percent	e recovery limits do not a Difference (RPD) obtain tract required detection 1	ied from the	sample di	iplicate (DUP) is ev	aluated ag	ainst the acce	ptance criter	ria when th	e sample i	s greater	

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Electronic Filing: Received, Clerk's Office 11/23/2022 LCMSMS-Misc Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490876

Product: The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS Analytical Method: EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 **Analytical Procedure:** GL-OA-E-076 REV# 7 **Analytical Batches:** 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490876001	7607-1
490876002	7607-EB
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490876001 (7607-1).

Amelante	490876
Analyte	001
Fluorotelomer sulfonate 4:2 (4:2 FTS)	100X
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Fluorotelomer sulfonate 8:2 (8:2 FTS)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002(7607-EB) (7607-EB) due to elevated concentrations of target analytes. PFBA was detected in the following samples above the MDL but less than LOQ. The sample is identified as Field Reagent Blanks (FRB). All samples associated with these blanks contained PFBA concentrations greater than 10 times that found in the blank. 490876002 (7607-EB).

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

r			T	.	T		Flect	t O	μС	μn	g:	Ķе	ЧСE	ųν	ęa,	iei	KS	j L	лП	CE	<u>1</u>	1/2	5/Z	UΖ	ـــــ							
GEL Laboratories, LLC 2040 Savage Road	Cuarteston, SC 29407 Phone: (843) 556-8171	766-1178	(Fill in the number of containers for each test)	 Antipe (6) 		SIM Note extra sample is										v: (Subject to Surcharge)		[] Level 2 [] Level 3 [] Level 4		[] No Cooler Temp:C	[] Others			le		Please provide any additional details	below regarding handling and/or disposal	concerns. (i.e.: Origin of sample(s), type	oj sue conectea from, oda marrices, etc.)			
		Fax: (843) 766-1178	Sample Analysis Requested ⁽³⁾ (Fill in the number		PFAS 21 cmpd list by EPA 537 mod	H1.4-Dioxane by EPA 8270SIM			1							TAT Requested: Normal: X Rush: Specify:	ícs [X] No	Select Deliverable: [] C of A [] QC Summary [] level I	ks:	g Use Only: Custody Seal Intact? [] Yev	[] Pacific [] Central [] Mountain	= Composite		4.) Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=Surface Water, WW=Waste Water, WE=Water, ML=Mise Liquid, SO=Soil, SD=Sediment, SL=Studge, SS=Soild Waste, O=Oil, F=Filter, P=Wipe, U=Urine, F=Focal, N=Nasal		PUT	IOWN	(i.e.: High/low pH, asbestos, beryllium, irritants, other mise, health harards, ptr.)				
	Becom Customer reductions and additional specially reducing the control of Custody and Analytical Request	••	Sample An			Lds	ro nwon (7) Rishi oldiseoo Potal number		4 X							TAT Reques	Fax Results: [] Yes	Select Deliverable:	Additional Remarks.	For Lab Receiving Use Only:	Sample Collection Time Zone: [X] Eastern	e Sample. G = Grab, C =		sSludge, SS=Solid Waste	0.4 - 1). m Thiosulfara If no reas	Other	OT= Other / Unknown	(i.e.: High/low pH, asbest misc. health harards_etc.)	Description:			
ries	nalytic	anager		Shoul	sample be considered:	com	ofut ətdətəs Addas əsvəfd əxtrəvoipey														Time Zo	e Duplicate		ment, SL∞	0108/7470 T = Sodium						,	'
Laboratories LLC	<u>v</u> and <u>A</u>	GEL Project Manager:					Sample Matrix ⁽⁴⁾	٦	ML									20			ollection	Matríx Spik		iil, SD=Sed	82608 - 3, (= Havaoa S			es.)				
abo	Sustod	GEL I	919-847-4241	704-586-0007		olson@harthickman	Field Filtered ⁽³⁾	z	z								Tíme	108			Sample (le. MSD =	filtered.	µid, SO=Sı	r each (i.e. a Arid HV :		iste	(F,K,P and U-listed wastes.) Waste code(s):		: " 		
	ain of C		919-84	04-58		lson@	Code (1)	z	EB								Date	910				Spike Samp	as not field	L∞Misc Liq	provided for = Ascorbic	Listed Waste	isted Wa	(F.K.P and U-li. Waste code(s):				
			Phone #	Fax#		lson go	*Time Collected (Military) (hhmm)	1310	1240									7			SRR.)	15 = Matrix	for sample w	V=Water, MI	f containers _l rir Acid AA	Listed	LW=L	(F.K.P Waste				
	UNLO	GEL Work Order Number:				Send Results To:Genna Olson g	*Date Collected (mm-dif-vv)	09-17-19	09-17-19							 ody Signatures	Received by (signed)	XX	2	3	t & Review form (= Equipment Blank, A	s field filtered or - N -	WW=Waste Water, N	/7470A) and number o Hydroxide. SA = Sulfu	tic Hazards	FL = Flammable/Ignitable	ive ve	1 - 1 N ¹	lated Morinated	enyls	
	1 RUJUT	GEL Work	kman, PC			Send Resul	lateitime									Chain of Custody Signatures	Time	1630			e Sample Receipt	Field Duplicate, EB	for yes the sample wa	, SW=Surface Water,	zd (i.e. 8260B, 6010B : Acid. SH = Sodium	Characteristic Hazards	FL = Flamm	CO = Corrosive RE = Reactive		TSCA Regulated PCB = Polychlorinated	biphenyls	
of 1 NWA-001 NWRA Ounte	NA STATE	NA	NWRA c/o Hart & Hickman, PC	me: Great Oak Landfill	Randleman, NC	Patrick Stevens	Sample ID * For composites - indicate start and stop date time	7607-1	7607-EB								Date	and H. June 09-18-19 16			> For sample shipping and delivery details, see Sample Receipt & Review form (SRR.)	 Chain of Custody Number = Client Determined OC Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sample, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite 	3.) Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered.	OW=Drinking Water, GW=Groundwater	 Sample Analysis Requested: Analysical method requested (i.e. 82608, 60108/7470A) and number of containers provided for each (i.e. 82608 - 3, 60108/74704 - 1). Preservative Type: HA = Hydrochloric Acid. NI = Ninie Acid. SH = Sudfamily Acid. A at a According Acid. HX = Horanov CT = Section Thiomatica. In according to Acid. Acid. A at a According Acid. HX = Horanov CT = Section Thiomatica. In according to Acid. Acid. SH = Sudfamily Acid. SH = Sudfamily Acid. Acid. At a According Acid. HX = Horanov CT = Section Thiomatica. If Acid. Acid. Acid. Acid. Sci. Acid. Acid. Sci. Acid. Sci. Acid. Acid. Sci. Acid. Sci. Acid. Sci. Acid. Acid. Acid. Acid. Acid. Acid. Acid. Acid. Acid. Sci. Acid. Acid. Acid. Sci. Acid. /li>	7.) Are there any known or possible hazards	associated with these samples?		Hg= Mercury	Se= Selenium Ae= Silver		RCRA metals
Page: 1 Project # GF1 Outre #-	COC Number ⁽¹⁾ .	PO Number:	Client Name:	Project/Site Name:	Address:	Collected By:	* For ce										Relinquished By (Signed)	- Pater +	2	3	> For sample s	 Chain of Custos QC Codes: N = 	3.) Field Filtered: F	 Matrix Codes: L 	 Sample Analysis Preservative Tvr 	7.) Are there any	associated w	RCRA Metals	As = Arsenic	Ba = Barium Cd = Cadmium	Cr = Chromium	Pb = Lead

Page 15 of 17 SDG: 490876 Rev1

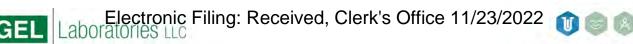
ATTACHMENT D

11/00/0000

Laboratories (1.4)								
Client: NWBA Jh	SAMPLE RECEIPT & REVIEW FORM							
Received By: Art	SUGAROCOC/Work Order: 40 08 0							
y.c.	Date Received: 9/19/19							
Carrier and Tracking Number	FF62 7563 2308 -1°, 7762 7563 3418-1° 7762 7563 1764 -1°, 7762 7563 3418-1°							
Suspected Hazard Information	If Net Counts > 100mm and 1							
A)Shipped as a DOT Hazardous?	If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation. Inzard Class Shipped: UN#: UN2910, Is the Radioactive Shipment Survey Compliant? Yes No							
B) Did the clicht designate the samples are to be Conceived as radioactive?	OC notation or radioactive stickers on containers equal client designation.							
C) Did the RSO classify the samples as Minatoria Classify the samples as Classify the samples as Classify Class	aximum Net Counts Observed* (Observed Counts - Area Background Counts):CPM / mR/Hr							
	DC notation or hazard labels on containers equal client designation.							
E) Did the RSO identify possible hazards?	or E is yes, select Hazards below. B's Flammable Foreign Soil non t							
Sample Receipt Criteria $\frac{3}{2} \neq \frac{3}{2}$	Allocatos Beryllium Other:							
	Comments/Qualifiers (Required for Non-Conforming Items) Circle Applicable: Seals broken Damaged container Leaking container Other (describe)							
2 Chain of custody documents included with shipment?	Circle Applicable: Client and Container Other (describe)							
3 Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?*	Preservation Method: Wet to Ice Packs Dry ice None Other:							
temparature and passed on IR	Temperature Device Sector To To Transformed Temperature							
sample containers intact and seated?	Secondary Temperature Device Serial # (If Applicable): 'irele Applicable: Seals,broken Damyed container Leaking container Other (describe)							
6 at proper pH?	ample ID's and Containers Affected:							
7 Do any samples require Volatile Analysis?	Preservation added, Lot# Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) o liquid VOA vials contain acid preservation? Yes No NA (If yes, take to VOA Freezer) e liquid VOA vials free of headspace? Yes No NA (If unknown, select No) nple ID's and containers affected: No NA (If unknown, select No)							
	s and tests affected:							
V 10323	s and containers affected:							
10 Date & time on COC match date & time Circ	ele Applicable: No dates on containers No times on containers COC missing info Other (describe)							
11 Number of containers received match number indicated on COC? Circ 12 Are sample containers identifiable as	le Applicable: No container count on COC Other (describe)							
Land provided?								
13 COC form is properly signed in relinquished/received sections?	e Applicable: Not relinquished Other (describe)							
Comments (Use Continuation Form if needed):								
PM (or PMA) review: Initials	VIV Ato 2 to 6							
	Date Date Date Of GL-CHL-SB-001 Box 6							

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water Louisiana NELAP	LA024
	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019



a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for East Carolina Reginal Landfill Work Order: 490877

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 20, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490877 GEL Work Order: 490877

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Elecarbor Alectric Electronic Filing Electroni

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forEast Carolina Reginal Landfill		
Client Sample ID:	0803-1	Project:	NWRA00119
Sample ID:	490877001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	19-SEP-19 10:35		
Receive Date:	20-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Analys	t Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs by	LC-MS/MS	"As Received"										
Fluorotelomer sulfonate 4:2		ND	13.2	37.6	ng/L	0.200	1	JLS	10/02/19	0950	1921240	1
FTS)												
N-ethylperfluoro-1-	1 (N	237	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic aci EtFOSAA)	a (N-											
N-methylperfluoro-1-		230	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic aci	d (N-				U							
MeFOSAA)												
Perfluorodecanesulfonic acid	d U	ND	6.60	19.4	ng/L	0.200	1					
(PFDS) Perfluorodecanoic acid (PFD		90.8	7.80	20.0	ng/I	0.200	1					
Perfluorododecanoic acid (P	,	90.8 ND	6.60	20.0	ng/L ng/L	0.200						
Perfluoroheptanesulfonic aci	,	9.39	6.60	20.0 19.0	ng/L	0.200						
(PFHpS)	iu J	9.39	0.00	19.0	ng/L	0.200	1					
Perfluoroheptanoic acid (PF	HpA)	689	6.60	20.0	ng/L	0.200	1					
Perfluorohexanesulfonic acid	d	536	6.60	18.2	ng/L	0.200	1					
(PFHxS)					-							
Perfluorononanesulfonic aci (PFNS)	d U	ND	7.00	19.2	ng/L	0.200	1					
Perfluorononanoic acid (PFN	NA)	89.0	6.60	20.0	ng/L	0.200	1					
Perfluorooctanesulfonamide	· ·	17.3	6.60	18.6	ng/L	0.200						
(PFOSA)					6							
Perfluorooctanesulfonic acid	l (PFOS)	402	8.00	20.0	ng/L	0.200						
Perfluorooctanoic acid (PFO	,	1640	7.00	20.0	ng/L	0.200						
Perfluoropentanesulfonic aci	id	54.7	6.60	18.8	ng/L	0.200	1					
(PFPeS)	D- 4)	1000	(())	20.0		0.200	1					
Perfluoropentanoic acid (PF Perfluoroundecanoic acid (P		1220 ND	6.60 6.60	20.0 20.0	ng/L	0.200						
Perfluorobutanesulfonic acid	,	3850	66.0	20.0 178	ng/L ng/L	0.200		JLS	10/02/19	0740	1921240	2
Perfluorobutyric acid (PFBA	· /	650	66.0	200	ng/L	0.200		JLS	10/02/19	0749	1921240	2
Perfluorohexanoic acid (PFF	· · · · · · · · · · · · · · · · · · ·	3610	66.0	200	ng/L	0.200						
Perfluorotetradecanoic acid	U	ND	66.0	200	ng/L	0.200						
(PFTeDA)	0	ND	00.0	200	ng/L	0.200	10					
Perfluorotridecanoic acid (P	FTrDA) U	ND	66.0	200	ng/L	0.200	10					
Fluorotelomer sulfonate 6:2	(6:2 U	ND	1320	3800	ng/L	0.200	100	JLS	10/02/19	1151	1921240	3
FTS)					-							
Fluorotelomer sulfonate 8:2 FTS)	(8:2 U	ND	1320	3840	ng/L	0.200	100					
Semi-Volatile-GC/MS												

Semi-Volatile-GC/MS

Electronic Filing Elecared RAF draie 51/23/2022

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington Minsiein 22202		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forEast Carolina Reginal Landfill		
Client Sample ID:	0803-1	Project:	NWRA00119
Sample ID:	490877001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF DF	Analyst Date	Time Batch	Method			
Semi-Volatile-GC/MS												
SW846 8270 SIM 1,4-D	SW846 8270 SIM 1,4-Dioxane in Liquid "As Received"											
1,4-Dioxane		157	4.00	8.00	ug/L	0.200 4	JMB3 09/24/19	1919 1919444	4			
The following Prep Methods were performed:												
Method	Description	n		Analyst	Date	Tim	e Prep Batch					
EPA 537.1 Mod, PFAS, Comp	pl PFCs Extract	ion in Liquid		LM1	09/27/19	0830	1921239					
SW846 3535A	SW8270E SI	M Prep 1,4-Dioxane		SJW1	09/23/19	1200	1919441					
The following Analytical Methods were performed:												
Method	Description	1			I	Analyst Co	mments					
1	EPA 537.1 M	od, PFAS, Compliant with QSM	Table B-1	5		•						
2	EPA 537.1 M	od, PFAS, Compliant with QSM	Table B-1	5								
3	EPA 537.1 M	od, PFAS, Compliant with QSM	Table B-1	5								
4	SW846 3535A	A/8270E SIM										
Surrogate/Tracer Recov	ery Test				Result 1	Nominal	Recovery%	Acceptable L	imits			
1,4-Dioxane-d8	SW846 Receive	8270 SIM 1,4-Dioxane in Liquid	"As		27.3 ug/L	40.0	68*	(70%-130%))			
Notes:												

Column headers are defined as follows:DF: Dilution FactorLc/LC: Critical LevelDL: Detection LimitPF: Prep FactorMDA: Minimum Detectable ActivityRL: Reporting LimitMDC: Minimum Detectable ConcentrationSQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 7

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia Mr. Jim Riley

Workorder: 490877

Contact:

Parmname	NOM	Sample Qual	QC	Units RPD/	D% REC%	Range Anls	t Date Time
Perfluorinated Compounds Batch 1921240							
QC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2		15.7	ng/L	86	(60%-145%) J	LS 10/02/19 06:05
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5		20.4	ng/L	110	(56%-143%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7		17.5	ng/L	94	(57%-138%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5		19.3	ng/L	99	(63%-131%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5		21.5	ng/L	111	(62%-133%)	
Perfluorobutanesulfonic acid (PFBS)	17.2		16.6	ng/L	96	(68%-136%)	
Perfluorobutyric acid (PFBA)	19.5		19.7	ng/L	101	(70%-133%)	
Perfluorodecanesulfonic acid (PFDS)	18.8		16.8	ng/L	89	(53%-142%)	
Perfluorodecanoic acid (PFDA)	19.5		18.0	ng/L	93	(62%-135%)	
Perfluorododecanoic acid (PFDoA)	19.5		19.5	ng/L	100	(66%-131%)	
Perfluoroheptanesulfonic acid (PFHpS)	18.5		18.1	ng/L	98	(66%-138%)	
Perfluoroheptanoic acid (PFHpA)	19.5		17.9	ng/L	92	(67%-135%)	
Perfluorohexanesulfonic acid (PFHxS)	17.7		14.5	ng/L	82	(64%-137%)	
Perfluorohexanoic acid (PFHxA)	19.5		18.9	ng/L	97	(67%-133%)	

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QC Summary

Workorder: 490877		•								Page 2 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated CompoundsBatch1921240										
Perfluorononanesulfonic acid (PFNS)	18.7			17.5	ng/L		93	(66%-130%) JLS	10/02/19 06:05
Perfluorononanoic acid (PFNA)	19.5			21.1	ng/L		108	(66%-134%)	
Perfluorooctanesulfonamide (PFOSA)	19.5			21.5	ng/L		111	(68%-137%)	
Perfluorooctanesulfonic acid (PFOS)	19.5			19.8	ng/L		102	(61%-131%)	
Perfluorooctanoic acid (PFOA)	19.5			18.8	ng/L		97	(63%-145%)	
Perfluoropentanesulfonic acid (PFPeS)	18.3			16.5	ng/L		90	(62%-139%)	
Perfluoropentanoic acid (PFPeA)	19.5			19.3	ng/L		99	(69%-132%)	
Perfluorotetradecanoic acid (PFTeDA)	19.5			22.5	ng/L		115	(65%-143%)	
Perfluorotridecanoic acid (PFTrDA)	19.5			19.9	ng/L		102	(57%-149%)	
Perfluoroundecanoic acid (PFUdA)	19.5			19.1	ng/L		98	(65%-134%)	
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6			20.5	ng/L	26	116	(0%-35%)	10/02/19 06:14
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9			17.6	ng/L	14	98	(0%-36%)	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1			19.9	ng/L	13	110	(0%-39%)	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8			20.1	ng/L	4	107	(0%-25%)	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8			21.9	ng/L	2	116	(0%-26%)	

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QC Summary

		<u>QC SI</u>	IIIIIai	<u>. y</u>				
Workorder: 490877				_				Page 3 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	. 4	103	(0%-30%) JLS	10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	. 2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	. 3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	. 16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	. 3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	. 2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	, 9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	. 15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	. 10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	. 4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	. 12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	. 6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	. 1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	<i>.</i> 0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	. 4	98	(0%-29%)	

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QC Summary

Workorder: 490877		-			<u></u>					Dess 4 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Page 4 of 7 Date Time
Perfluorinated Compounds Batch 1921240		-	•							
Perfluoropentanoic acid (PFPeA)	18.8			20.0	ng/L	3	106	(0%-30%)	JLS	10/02/19 06:14
Perfluorotetradecanoic acid (PFTeDA)	18.8			20.6	ng/L	9	109	(0%-30%)		
Perfluorotridecanoic acid (PFTrDA)	18.8			17.7	ng/L	11	94	(0%-35%)		
Perfluoroundecanoic acid (PFUdA)	18.8			21.2	ng/L	10	112	(0%-28%)		
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)			U	ND	ng/L					10/02/19 05:56
Fluorotelomer sulfonate 6:2 (6:2 FTS)			U	ND	ng/L					
Fluorotelomer sulfonate 8:2 (8:2 FTS)			U	ND	ng/L					
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)			U	ND	ng/L					
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)			U	ND	ng/L					
Perfluorobutanesulfonic acid (PFBS)			U	ND	ng/L					
Perfluorobutyric acid (PFBA)			U	ND	ng/L					
Perfluorodecanesulfonic acid (PFDS)			U	ND	ng/L					
Perfluorodecanoic acid (PFDA)			U	ND	ng/L					
Perfluorododecanoic acid (PFDoA)			U	ND	ng/L					
Perfluoroheptanesulfonic acid (PFHpS)			U	ND	ng/L					

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QC Summary

Workorder: 490877								Pag	e 5 of 7
Parmname	NOM	Sample Qua	l QC	Units	RPD/D% RE	C% Range	Anlst		Time
Perfluorinated CompoundsBatch1921240									
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L			JLS	10/02/1	19 05:56
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L					
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L					
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L					
Perfluorononanoic acid (PFNA)		U	ND	ng/L					
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L					
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L					
Perfluorooctanoic acid (PFOA)		U	ND	ng/L					
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L					
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L					
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L					
Perfluorotridecanoic acid (PFTrDA)		U	ND	ng/L					
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L					
Semi-Volatile-GC/MS Batch 1919444									
QC1204387349 LCS **1,4-Dioxane-d8	4.00		3.55	ug/L	8	89 (70%-130%	6) JMB3	09/24/1	19 12:24

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QC Summary

Workorder: 49	90877										Pag	e 6 of 7
Parmname		NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Semi-Volatile-GC/M Batch 1919	S 9444											
QC1204387350 **1,4-Dioxane-d8	LCSD	4.00			3.18	ug/L		79	(70%-130%)	JMB3	09/24/1	19 12:49
QC1204387348 1,4-Dioxane	MB			U	ND	ug/L					09/24/1	19 11:59
**1,4-Dioxane-d8		4.00			3.05	ug/L		76	(70%-130%)			

Notes:

The Qualifiers in this report are defined as follows:

** Analyte is a surrogate compound

.....

- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N/A RPD or %Recovery limits do not apply.

N1 See case narrative

- ND Analyte concentration is not detected above the detection limit
- NJ Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier
- P Organics--The concentrations between the primary and confirmation columns/detectors is >40% different. For HPLC, the difference is >70%.
- Q One or more quality control criteria have not been met. Refer to the applicable narrative or DER.
- R Sample results are rejected
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.
- UJ Compound cannot be extracted

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QC Summary

Workord	der: 490877										Pag	e 7 of 7
Parmnan	ne	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
X Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier												

Y QC Samples were not spiked with this compound

^ RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.

h Preparation or preservation holding time was exceeded

N/A indicates that spike recovery limits do not apply when sample concentration exceeds spike conc. by a factor of 4 or more or %RPD not applicable. ^ The Relative Percent Difference (RPD) obtained from the sample duplicate (DUP) is evaluated against the acceptance criteria when the sample is greater than five times (5X) the contract required detection limit (RL). In cases where either the sample or duplicate value is less than 5X the RL, a control limit of +/- the RL is used to evaluate the DUP result.

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490877

GC/MS Semivolatile

<u>Product:</u> Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1919444

Preparation Method: SW846 3535A **Preparation Procedure:** GL-OA-E-073 REV# 2 **Preparation Batch:** 1919441

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490877001	0803-1
1204387348	Method Blank (MB)
1204387349	Laboratory Control Sample (LCS)
1204387350	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Surrogate Recoveries

Sample (See Below) did not meet surrogate recovery acceptance criteria. The sample was analyzed at a dilution. As a result, one or more surrogates were diluted out of the acceptance limits.

Sample	Analyte	Value
490877001 (0803-1)	1, 4-Dioxane-d8	68* (70%-130%)

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

Technical Information

Sample Dilutions Sample 490877001 (0803-1) was diluted due to the presence of one or more over-range target analytes.

Electronic Filing: Received, Clerk's Office 11/23/2022 <u>LCMSMS-Misc</u>

<u>Product:</u> The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 <u>Analytical Procedure:</u> GL-OA-E-076 REV# 7 <u>Analytical Batches:</u> 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	<u>Client Sample Identification</u>
490877001	0803-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490877001 (0803-1).

A	490877
Analyte	001
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Fluorotelomer sulfonate 8:2 (8:2 FTS)	100X
Perfluorobutanesulfonate (PFBS)	10X
Perfluorobutyric acid (PFBA)	10X
Perfluorohexanoic acid (PFHxA)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

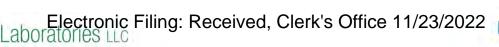
Proge: 1 of 1									GEL Labor	GEL Laboratories. LLC		
CULA			aboratories LLC	niesu	0 V				2040 Savage Road	ge Road		
GEL Quote #: <u>Nurza Queze</u>		tel.com Chemist	themistry I Rediochemistry I Radiobioassay I Specialty Analytics	stry I Radiobic	assay I Specia	alty Analytics			Charleston, SC 29407	SC 29407		
		Chain of Cust	Custody and Analytical Request	Nnalytical	Request	100			Phone: (843) 556-8171	3) 556-8171		
NWENT OF W	UEL Work Urder Number:		UEL Project Manager:	lanager:	14/10	(iso)	(12)		Fax: (843) 766-1178	766-1178		
clo that & Hickman	S.C.	Phone # QIQ - PMP	1424 - FHS		Sample	Sample Analysis Requested	kequested ⁽	(5) (Fill in	the number	(Fill in the number of containers for each test	for each test)	
Protect/Site Name: East (amlina Desirval	1 Bul			Should fhis		2					< Preservative Type (6)	(D) (0)
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	Send Results To Renner Olen	30 being hornicknem.c	Nickman	ł	r of co rtds	Two I					Comments Note: extra samul	nts amole is
Sample ID	*Time *Date Collected Collected (Military)	8	Field Sample	ofaj sidojo ijdans svoj avljavojpu	to nwonă (azad oldizec odmun lete	s ofe					required for sam specific QC	Elect
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Date Time	Received by (signed)		Time	Fax	Fax Results: [] Yes		MN0					s C
7-19-19 PW	-1- 7- 2- 2-	9-19-19	<u>do</u>	Sel	Select Deliverable: [] C of A	ble: [] C of		[] QC Summary	[] level 1	[] Level 2	[] Level 3 []	Level #1
ra na bana na dina ana ana ang na yang na pangang sang nang nang nang nang nang nan	2 a. allenter	- (120)	19 8:	SS Ad	Additional Remarks:	arks:					1	ce
	3 - 1		:	Fo	For Lab Receiving Use Only: Custody Seal Intact? [ing Use On	ly: Custody	Seal Intact] Yes		Cooler Temp: 1	11 ວ
 Chain of Custody Number = Client Determined Chain of Custody Number = Client Determined QC Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate. 	 Chain of Custody Number = Client Determined Chain of Custody Number = Client Determined Chain of Custody Number = Client Determined Control Cost of N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, Matrix Solice Sample, MSD = Matrix Solice Sample, TB = Gram G = Gram Science S	ix Spike Sample, M	ozanjav vo meodulu 1000 zonec [] Edistenn de MSD = Matrix Snike Dunlicate Samale G = Grab [e Dunlicate Sar	[] L'ASICHA mile G = Grab	[] racino Comosite	le Central		- Alountain	1 Officer		/23/2
te with a - Y - for yes the sampl	 Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered. 	e was not field filter	ed.				3					02
=Groundwater, SW=Surface W	4.) Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=Surface Water, WW=Water, W=Water, ML=Misc Liquid, SO=Soil, SD=Sediment, SL=Sludge, SS=Solid Wase, O=Oil, F=Filter, P=Wipe, U=Urine, F=Fecal, N=Nasal	ML=Mise Liquid, (SO=Soil, SD=Sed	liment, SL=Slud	ige, SS=Solid W	'aste, O=Oil, F	=Filter, P=Wip	e, U=Urine, ₽	=Fccal, N=Nasa			2
ethod requested (i.e. 8260B , 66 id, NI = Nitric Acid, SH = Sod	mber of co = Sulfuric <i>i</i>	ers provided for each AA = Ascorbic Acid	1 (i.e. <i>8260B</i> - 3, 4 I, HX = Hexane, S	6010B/7470A - ST = Sodium Th	1). iosulfate, If no J	preservative is	added = leave f	ield blank				
7.) Are there any known or possible hazards Characte associated with these samples? FL = Fla	Characteristic Hazards Liste FL = Flammable/Ignitable LW=	Listed Waste LW= Listed Waste		Other OT=(Other OT= Other / Unknown	known			1	Please provide Velow recordin	Please provide any additional details below recording handling and/or dismosal	letails or disnosal
CO = Corrosive DT = D		(F,K,P and U-listed	listed wastes.)	(i.e.	(i.e.: Highlow pH, asbestos, beryllium, irritants, other	oH, asbestos	i, beryllium,	irritants, oi		oncerns. (i.e.	concerns. (i.e.: Origin of sample(s), type	e(s), type
E I R		Waste code(s):		nis Des	misc. health hazards, etc.) Description:	ards, etc.)			<u>0</u>	of site collected	of site collected from, odd matrices, etc.)	ces, etc.)
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Ag- Juvei MR= Miscellaneous	onychtormateu binhenvis											
A DESCRIPTION OF A DESC	No beneficiaries de la compactación de la contexte d	A STREET AND A STREET	A CONTRACT OF A	A REPORT OF A R	And a subscription of the second s		A CONTRACTOR OF					STREET, STREET

GEL Laboratories, LLC 2040 Savage Road Charleston, SC 29407 Phone: (843) 556-8171	Fax: (843) 766-1178	(Fill in the number of containers for each test)	c Preservative Type (6)	Cammente	Note: extra sample is	required for sample			-ilir	ng:	R	ec	eiv				Specify: (Subject to Surcharge	\$ C	[] level 1 [] Level 2 [] Level 3 [] Level] Yes [] No Cooler Temp:C	[] Other:	3/2	02			Please provide any additional details		er concerns. (i.e.: Origin of sample(s), type of site collected from. odd matrices. etc.)	✓			
alty Analytics	Robin Son	Sample Analysis Requested ⁽⁵⁾ (Fill in th			sbrug ards	o nwonż stana b stana b	t (τ) szoq szoT C	×									TAT Requested: Normal: K Rush:	Fax Results: [] Yes [] Yuo] C of A [] QC Summary		For Lab Receiving Use Only: Custody Seal Intact? [one: [] Eastern [] Pacific [] Central [] Mountain	te Samule. $G = Grab. C = Connosite$		Liquid, SO=Soil, SD=Sediment, SL=Sludge, SS=Solid Waste, O=Oil, F=Filter, P=Wipe, U=Urine, F=Fecal, N=Nasal	0A - 1). un Thiosulfster If no mesercative is added = Jeave field blank	Other	OT= Other / Unknown	(i.e.: High/low pH, asbestos, beryllium, irritants, other misc. health hazards, etc.)	Description:			
GEL Laboratories LLC get.com Chemistry I Radiochemistry I Radiochemistry Analytics Chain of Custody and Analytical Request	GEL Work Order Number: GEL Project Manager:	14 ch-tho- an- and - ant- 42 HI	. h. 1. Fax #	NU 37805	Io: Chenne Olson 7 sishe horthideman in	Field	(hhorm) Code (2) Fultered (2) Matrix (4) 22 20	14-19-19 25 01 N-19-19									dy Signatures	Received by (signed) Date Time	Feelex 9-19-19 1200	2 0. (Uman 920129 8:55	3	& Review form (SRR,) Sample Collection Time Zone: [] Eastern	= Equipment Blank, MS = Matrix Spike Sample, MSD = Matrix Spike Duplica	s field filtered or - N - for sample was not field filtered.	WW=Waste Water, W=Water, ML=Mise Liquid, SO=Soil, SD=Sediment, SL	7470A) and number of containers provided for each (i.e. $8260B - 3$, $6010B/74$; 4vdroxide, $SA = Sulfarite Acid, AA = Ascorbic Acid, HX = Hexane, ST = Sudiu$	ic Hazards Listed Waste	o/Ignitable	ive (t',K,P and U-listed wastes.) • Waste code(s):	[ated	nyls	
Preme: 1 of 4 Premeet # 190772 NUV2A COMA GEL Quote #: NUV2A Que te COC Number (1): NA	BUIL to NURA	CHATT Name: NWERA UP 14 and Michanan	" Fost Carolina	Adress 1922 Republican Road, Aulaneter,	2.Stevens		nd stop date/time	_			A	\ T	ГА	Cł	HM	EN	L Chain of Custody Signatures	Ran quished By (Signed) Date Time	Fallert An PARA 1700			- For sample shipping and delivery details, see Sample Receipt & Review form (SRR,)	 Chain of Custody Number = Client Determined Concorders: N = Nortral Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sample, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite 	3.) Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered.	 Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=Surface Water, WW=Waster, W=Water, ML=Mise 	 Sample Analysis Requested: Analytical method requested (i.e. 82608, 60108/7470A) and number of containers provided for each (i.e. 8260B - 3, 6010B/7470A - 1). Preservative Type: HA = Hydrochloric Acid, NI = Nitric Acid, SH = Sodium Hydroxide, SA = Sulfaric Acid, HA = Accordic Acid, NI = Hydrochloric Acid, NI = Nitric Acid, SI = Sodium Hydroxide, SA = Sulfaric Acid, AA = Accordic Acid, HX = Hexane, ST = Sodium Thiosalfate. If no meservative is added = preventioned for the second seco	7.) Are there any known or possible hazards Characteristic Hazards	associated with these samples? FL = Flamm	RE Reactive RCRA Metals	As = Arsenic Hg= Mercury Re = Bestime co- Colonium (TSCA Domu)	E	nium MR= Miscellaneous	Pb = Lead RCRA metals

Laboratories con			
Client: MARA	JA		SAMPLE RECEIPT & REVIEW FORM SDG/AR/COC/Work Orger: 490877
Received By: ATA			Date Received: 1/20119
- Jul			Circle Applicable:
Construct To the Mark			FedEx Express FedEx Ground UPS Field Services Courier Other
Carrier and Tracking Number			7762 8638 8788
			7767 8638 8034
Suspected Hazard Information	Yes	°N N	*If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation.
A)Shipped as a DOT Hazardous?		/	Hazard Class Shipped: UN#: If UN2910, Is the Radioactive Shipment Survey Compliant? Yes No
B) Did the client designate the samples are to be received as radioactive?		\square	COC notation or radioactive stickers on containers equal client designation.
C) Did the RSO classify the samples as radioactive?		\square	Maximum Net Counts Observed " (Observed Counts - Area Background Counts):CPM / niR/Hr Classified as: Rad 1 Rad 2 Rad 3
D) Did the client designate samples are hazardous?		<u> </u>	COC notation or hazard labels on containers equal client designation.
E) Did the RSO identify possible hazards?		/	lf D or E is yes, select Hazards below. ?CB's Flammable Foreign Soil RCRA Asbestos Beryllium Other:
Sample Receipt Criteria	2	ź	
Shipping containers received intact and sealed?	7		Z Comments/Qualifiers (Required for Non-Conforming Items) Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
2 Chain of custody documents included with shipment?	7		Circle Applicable: Client contacted and provided COC COC created upon receipt
3 Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?*	Λ		Preservation Method Wet ice) Ice Packs Dry ice None Other: *all temperatures are recorded in Celsius TEMP:
4 Daily check performed and passed on IR temperature gun?	\bigwedge		Temperature Device Serial #: <u>J.B.4 – LB</u> Secondary Temperature Device Serial # (If Applicable):
5 Sample containers intact and sealed?	\bigwedge		Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
6 Samples requiring chemical preservation at proper pH?		Л	Sample ID's and Containers Affected:
7 Do any samples require Volatile			If Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer)
7 Analysis?			Are liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) Are liquid VOA vials free of headspace? Yes No NA Sample ID's and containers affected:
8 Samples received within holding time?	7		ID's and tests affected:
9 Sample ID's on COC match ID's on bottles?	/		ID's and containers affected:
Date & time on COC match date & time on bottles?	Λ		Circle Applicable: No dates on containers No times on containers COC missing info Other (describe)
1 Number of containers received match number indicated on COC?			Circle Applicable: No container count on COC Other (describe)
2 Are sample containers identifiable as GEL provided?	\mathbf{A}		
, COC form is properly signed in	\mathbf{A}		Circle Applicable: Not relinquished Other (describe)
relinquished/received sections? Art. 9/10 omments (Use Continuation Form if meded);		M V	
PM (or PMA	.) r.w.		1122 10 1 1
e se (of these	., ievi	unt. I	uitials 9 0 Date Page of GL-CHL-SR-001 Rev 6

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012 SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water	LA024
Louisiana NELAP	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235–19–15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019





a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for Upper Piedmont Regional Landfill Work Order: 490879

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019 and September 20, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

rilie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

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Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490879 GEL Work Order: 490879

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecared RAFORIES ice C1/23/2022

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical for Upper Piedmont Regional Landfill		
Client Sample ID:	7304-1	Project:	NWRA00119
Sample ID:	490879001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	17-SEP-19 15:25		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Anal	yst Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs	by LC-MS/MS	"As Received"										
N-ethylperfluoro-1-	5	48.7	13.2	40.0	ng/L	0.200	1	JLS	10/04/19	1100	1921240	1
octanesulfonamidoacetic	acid (N-				U							
EtFOSAA)		10.4	12.2	10.0	~							
N-methylperfluoro-1- octanesulfonamidoacetic	acid (N	106	13.2	40.0	ng/L	0.200	1					
MeFOSAA)												
Perfluorobutanesulfonic a	cid (PFBS)	1420	6.60	17.8	ng/L	0.200	1					
Perfluorodecanesulfonic a		14.9	6.60	19.4	ng/L	0.200	1					
(PFDS)												
Perfluorodecanoic acid (P	,	48.0	7.80	20.0	ng/L	0.200						
Perfluorododecanoic acid	-	ND	6.60	20.0	ng/L	0.200						
Perfluoroheptanesulfonic	acid U	ND	6.60	19.0	ng/L	0.200	1					
(PFHpS) Perfluoroheptanoic acid (2	DEU n (A)	344	6.60	20.0	ng/I	0.200	1					
Perfluorohexanesulfonic a	1 /	344 190	6.60	20.0 18.2	ng/L ng/L	0.200						
(PFHxS)	aciu	190	0.00	10.2	ng/L	0.200	1					
Perfluorononanesulfonic a	acid J	13.4	7.00	19.2	ng/L	0.200	1					
(PFNS)					-							
Perfluorononanoic acid (H	· ·	44.1	6.60	20.0	ng/L	0.200						
Perfluorooctanesulfonami	ide U	ND	6.60	18.6	ng/L	0.200	1					
(PFOSA) Perfluorooctanesulfonic a	aid (DEOS)	254	8.00	20.0	ng/L	0.200	1					
Perfluorooctanoic acid (P	· /	884	7.00	20.0	ng/L ng/L	0.200						
Perfluoropentanesulfonic	,	28.1	6.60	18.8	ng/L ng/L	0.200						
(PFPeS)	aciu	20.1	0.00	10.0	lig/L	0.200	1					
Perfluoropentanoic acid (PFPeA)	621	6.60	20.0	ng/L	0.200	1					
Perfluoroundecanoic acid		ND	6.60	20.0	ng/L	0.200	1					
Fluorotelomer sulfonate 8	:2 (8:2 U	ND	132	384	ng/L	0.200	10	JLS	10/02/19	0757	1921240	2
FTS)												
Perfluorobutyric acid (PF		743	66.0	200	ng/L	0.200						
Perfluorohexanoic acid (F	,	2350	66.0	200	ng/L	0.200						
Perfluorotetradecanoic ac	id U	ND	66.0	200	ng/L	0.200	10					
(PFTeDA) Perfluorotridecanoic acid	(PFTrDA) U	ND	66.0	200	ng/L	0.200	10					
Fluorotelomer sulfonate 4	-	ND	1320	3760	ng/L	0.200		пс	10/02/19	1200	1921240	3
FTS)	.2 (7.2 U		1520	5700	ng/L	0.200	100	100	10/02/19	1200	1721240	0
Fluorotelomer sulfonate 6	5:2 (6:2 U	ND	1320	3800	ng/L	0.200	100					
FTS)					-							
The following Prep	Methods were r	performed:										
	1	·										

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical for Upper Piedmont Regional Landfill		
Client Sample ID:	7304-1	Project:	NWRA00119
Sample ID:	490879001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF DF	Analyst Date	Time Batch	Method
The following Prep	Methods were pe	erformed:							
Method	Descriptior	1		Analyst	Date	Time	Prep Batch		
EPA 537.1 Mod, PFAS,	Compl PFCs Extracti	on in Liquid		LM1	09/27/19	0830	1921239		
The following Ana	lytical Methods w	vere performed:							
Method	Description				А	nalyst Corr	nments		
1	EPA 537.1 Mo	od, PFAS, Compliant wi	th QSM Table B-15						
2	EPA 537.1 Mo	od, PFAS, Compliant wi	th QSM Table B-15						
3	EPA 537.1 Mo	od, PFAS, Compliant wi	th QSM Table B-15						

Notes:

Column headers are defined as follows:	
DF: Dilution Factor	Lc/LC: Critical Level
DL: Detection Limit	PF: Prep Factor
MDA: Minimum Detectable Activity	RL: Reporting Limit
MDC: Minimum Detectable Concentration	SQL: Sample Quantitation Limit

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Certificate of Analysis

Report Date: November 8, 2019

	Company : Address :			inas Chapter rive, Suite 804										
	Contact: Project:	Mr.	Jim Riley	inia 22202 Ipper Piedmont I	Region	al Landfil	1							
	Client Sample ID:	7304	4-1					Pro	ject:		NWRA	00119		
	Sample ID:	4908	879002					Cli	ent ID	:	NWRA	001		
	Matrix:	Mis	c Liquid											
	Collect Date:	17-5	SEP-19 15:2	25										
	Receive Date:	20-5	SEP-19											
	Collector:	Clie	nt											
Parameter	Quali	fier	Result		DL	RL	Uni	its	PF	DF	Analys	t Date	Time Batch	Method
Semi-Volat											5			
	0 SIM 1,4-Dioxane	in Lie	nuid "As Re	eceived"										
1,4-Dioxane	o blin i, i bionaile		177		5.00	10.0) u	g/L	0.200	5	JMB3	09/24/19	1945 1919444	1
The followi	ing Prep Methods w	ere pe	erformed:											
Method	Desci	-				Analyst	Dat	e	,	Time	Pre	p Batch	l	
SW846 3535A	SW82	70E SIN	A Prep 1,4-Di	oxane		SJW1	09/2	3/19		1200	191	9441		
The follow	ing Analytical Meth	ods w	vere perform	ned:										
Method	Descr		1					A	Analys	t Con	nments			
1	SW846	3535A	/8270E SIM											
Surrogate/T	racer Recovery	Test					Result	ľ	Nomin	al	Recove	ery%	Acceptable L	imits
1,4-Dioxane-d		W846 Receive		Dioxane in Liquid "	As		24.2 ug/L		40	.0		61*	(70%-130%))
Notes:														
	aders are defined as	follo												
DF: Dilutio				Lc/LC: Critical										
DL: Detect	ion Limit			PF: Prep Factor										

MDA: Minimum Detectable Activity MDC: Minimum Detectable Concentration

RL: Reporting Limit SQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 7

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia

Mr. Jim Riley

Workorder: 490879

Contact:

Parmname	NOM	Sample Qua	I QC	Units RPD	/D% REC%	Range A	Anlst	Date Time
Perfluorinated Compounds Batch 1921240								
QC1204391614 LCS Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2		15.7	ng/L	86	(60%-145%)	JLS	10/02/19 06:05
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5		20.4	ng/L	110	(56%-143%)		
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7		17.5	ng/L	94	(57%-138%)		
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5		19.3	ng/L	99	(63%-131%)		
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5		21.5	ng/L	111	(62%-133%)		
Perfluorobutanesulfonic acid (PFBS)	17.2		16.6	ng/L	96	(68%-136%)		
Perfluorobutyric acid (PFBA)	19.5		19.7	ng/L	101	(70%-133%)		
Perfluorodecanesulfonic acid (PFDS)	18.8		16.8	ng/L	89	(53%-142%)		
Perfluorodecanoic acid (PFDA)	19.5		18.0	ng/L	93	(62%-135%)		
Perfluorododecanoic acid (PFDoA)	19.5		19.5	ng/L	100	(66%-131%)		
Perfluoroheptanesulfonic acid (PFHpS)	18.5		18.1	ng/L	98	(66%-138%)		
Perfluoroheptanoic acid (PFHpA)	19.5		17.9	ng/L	92	(67%-135%)		
Perfluorohexanesulfonic acid (PFHxS)	17.7		14.5	ng/L	82	(64%-137%)		
Perfluorohexanoic acid (PFHxA)	19.5		18.9	ng/L	97	(67%-133%)		

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QC Summary

Workorder: 490879		•			<u> </u>					Page 2 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated CompoundsBatch1921240										
Perfluorononanesulfonic acid (PFNS)	18.7			17.5	ng/L		93	(66%-130%)) JLS	10/02/19 06:05
Perfluorononanoic acid (PFNA)	19.5			21.1	ng/L		108	(66%-134%))	
Perfluorooctanesulfonamide (PFOSA)	19.5			21.5	ng/L		111	(68%-137%))	
Perfluorooctanesulfonic acid (PFOS)	19.5			19.8	ng/L		102	(61%-131%)	
Perfluorooctanoic acid (PFOA)	19.5			18.8	ng/L		97	(63%-145%)	
Perfluoropentanesulfonic acid (PFPeS)	18.3			16.5	ng/L		90	(62%-139%))	
Perfluoropentanoic acid (PFPeA)	19.5			19.3	ng/L		99	(69%-132%))	
Perfluorotetradecanoic acid (PFTeDA)	19.5			22.5	ng/L		115	(65%-143%))	
Perfluorotridecanoic acid (PFTrDA)	19.5			19.9	ng/L		102	(57%-149%))	
Perfluoroundecanoic acid (PFUdA)	19.5			19.1	ng/L		98	(65%-134%))	
QC1204391615 LCSD Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6			20.5	ng/L	26	116	(0%-35%))	10/02/19 06:14
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9			17.6	ng/L	14	98	(0%-36%))	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1			19.9	ng/L	13	110	(0%-39%))	
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8			20.1	ng/L	4	107	(0%-25%))	
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8			21.9	ng/L	2	116	(0%-26%))	

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QC Summary

		$\underline{\mathbf{v}}$	IIIIIai	<u>y</u>				
Workorder: 490879								Page 3 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	4	103	(0%-30%) JLS	10/02/19 06:14
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	1	106	(0%-27%)	
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	0	100	(0%-30%)	
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	4	98	(0%-29%)	

Page 8 of 19 SDG: 490879 Rev1

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QC Summary

Workorder: 490879		-			<u></u>					Dess 4 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Page 4 of 7 Date Time
Perfluorinated CompoundsBatch1921240				-						
Perfluoropentanoic acid (PFPeA)	18.8			20.0	ng/L	3	106	(0%-30%)	JLS	10/02/19 06:14
Perfluorotetradecanoic acid (PFTeDA)	18.8			20.6	ng/L	9	109	(0%-30%)		
Perfluorotridecanoic acid (PFTrDA)	18.8			17.7	ng/L	11	94	(0%-35%)		
Perfluoroundecanoic acid (PFUdA)	18.8			21.2	ng/L	10	112	(0%-28%)		
QC1204391613 MB Fluorotelomer sulfonate 4:2 (4:2 FTS)			U	ND	ng/L					10/02/19 05:56
Fluorotelomer sulfonate 6:2 (6:2 FTS)			U	ND	ng/L					
Fluorotelomer sulfonate 8:2 (8:2 FTS)			U	ND	ng/L					
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)			U	ND	ng/L					
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)			U	ND	ng/L					
Perfluorobutanesulfonic acid (PFBS)			U	ND	ng/L					
Perfluorobutyric acid (PFBA)			U	ND	ng/L					
Perfluorodecanesulfonic acid (PFDS)			U	ND	ng/L					
Perfluorodecanoic acid (PFDA)			U	ND	ng/L					
Perfluorododecanoic acid (PFDoA)			U	ND	ng/L					
Perfluoroheptanesulfonic acid (PFHpS)			U	ND	ng/L					

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QC Summary

Workorder: 490879		-	2							Page	e 5 of 7
Parmname	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst		Time
PerfluorinatedCompoundsBatch1921240											
Perfluoroheptanoic acid (PFHpA)			U	ND	ng/L				JLS	10/02/1	9 05:56
Perfluorohexanesulfonic acid (PFHxS)			U	ND	ng/L						
Perfluorohexanoic acid (PFHxA)			U	ND	ng/L						
Perfluorononanesulfonic acid (PFNS)			U	ND	ng/L						
Perfluorononanoic acid (PFNA)			U	ND	ng/L						
Perfluorooctanesulfonamide (PFOSA)			U	ND	ng/L						
Perfluorooctanesulfonic acid (PFOS)			U	ND	ng/L						
Perfluorooctanoic acid (PFOA)			U	ND	ng/L						
Perfluoropentanesulfonic acid (PFPeS)			U	ND	ng/L						
Perfluoropentanoic acid (PFPeA)			U	ND	ng/L						
Perfluorotetradecanoic acid (PFTeDA)			U	ND	ng/L						
Perfluorotridecanoic acid (PFTrDA)			U	ND	ng/L						
Perfluoroundecanoic acid (PFUdA)			U	ND	ng/L						
Semi-Volatile-GC/MS Batch 1919444											
QC1204387349 LCS **1,4-Dioxane-d8	4.00			3.55	ug/L		89	(70%-130%) JMB3	09/24/1	9 12:24

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QC Summary

Workorder: 49	0879										Page	e 6 of 7
Parmname		NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Semi-Volatile-GC/MS Batch 1919												
QC1204387350 **1,4-Dioxane-d8	LCSD	4.00			3.18	ug/L		79	(70%-130%)	JMB3	09/24/1	9 12:49
QC1204387348 1,4-Dioxane	МВ			U	ND	ug/L					09/24/1	9 11:59
**1,4-Dioxane-d8		4.00			3.05	ug/L		76	(70%-130%)			

Notes:

The Qualifiers in this report are defined as follows:

** Analyte is a surrogate compound

.....

- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N/A RPD or %Recovery limits do not apply.

N1 See case narrative

- ND Analyte concentration is not detected above the detection limit
- NJ Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier
- P Organics--The concentrations between the primary and confirmation columns/detectors is >40% different. For HPLC, the difference is >70%.
- Q One or more quality control criteria have not been met. Refer to the applicable narrative or DER.
- R Sample results are rejected
- U \qquad Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.
- UJ Compound cannot be extracted

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QC Summary

Workor	der: 490879										Page	e 7 of 7
Parmnan	ne	NOM	Sample Q	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Х	Consult Case Narrative, I	Data Summary package	e, or Project Ma	anager o	concerning t	his qualifi	er					

Y QC Samples were not spiked with this compound

^ RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.

h Preparation or preservation holding time was exceeded

N/A indicates that spike recovery limits do not apply when sample concentration exceeds spike conc. by a factor of 4 or more or %RPD not applicable. ^ The Relative Percent Difference (RPD) obtained from the sample duplicate (DUP) is evaluated against the acceptance criteria when the sample is greater than five times (5X) the contract required detection limit (RL). In cases where either the sample or duplicate value is less than 5X the RL, a control limit of +/- the RL is used to evaluate the DUP result.

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490879

GC/MS Semivolatile

<u>Product:</u> Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1919444

Preparation Method: SW846 3535A **Preparation Procedure:** GL-OA-E-073 REV# 2 **Preparation Batch:** 1919441

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490879002	7304-1
1204387348	Method Blank (MB)
1204387349	Laboratory Control Sample (LCS)
1204387350	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Surrogate Recoveries

Sample (See Below) did not meet surrogate recovery acceptance criteria. The sample was analyzed at a dilution. As a result, one or more surrogates were diluted out of the acceptance limits.

Sample	Analyte	Value
490879002 (7304-1)	1, 4-Dioxane-d8	61* (70%-130%)

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

Technical Information

Sample Dilutions Sample 490879002 (7304-1) was diluted due to the presence of one or more over-range target analytes.

Electronic Filing: Received, Clerk's Office 11/23/2022 <u>LCMSMS-Misc</u>

<u>Product:</u> The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 <u>Analytical Procedure:</u> GL-OA-E-076 REV# 7 <u>Analytical Batches:</u> 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490879001	7304-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490879001 (7304-1).

A mellente	490879
Analyte	001
Fluorotelomer sulfonate 4:2 (4:2 FTS)	100X
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Fluorotelomer sulfonate 8:2 (8:2 FTS)	10X
Perfluorobutyric acid (PFBA)	10X
Perfluorohexanoic acid (PFHxA)	10X
Perfluorotetradecanoic acid (PFTeDA)	10X
Perfluorotridecanoic acid (PFTrDA)	10X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

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GEL Laboratories, LLC 2040 Savage Road Charleston, SC 29407 Phome, (843) 556.8171	Fax: (843) 766-1178	(Fill in the number of containers for each test)	<pre>< Preservative Type (6)</pre>	537 mod	Note: extra sample is	required for sample specific QC							Note that you will receive a separate cooler	501 LI WITH DOTTIES TOF 1,4-010XANE ANALYSIS- for this site. If is OV to include these in the	rol uns sue. Tuis ON to include those in the - same lab report as these samples.	fy: (Subject to Surcharge)		I [] Level 2 [] Level 3 [] Level 4		[] No Cooler Temp:C	[.] Other:			asal		Please provide any additional details	below regarding handling and/or disposal	concerns. (i.e.: Origin of sample(s), type	of sue concerca from, oad maintees, etc.)			
	Fax: (843)	Sample Analysis Requested ⁽⁵⁾ (Fill in the numbe		PFAS 21 cmpd list by EPA									Note that you will	for this site it is f	same lab report as these samples	ed: Normal: X Rush: Specify:	cs [X] No	Select Deliverable: [] C of A [] QC Summary [] level 1	s:	For Lab Receiving Use Only: Custody Seal Intact? [] Yes] Pacific [] Central [] Mountain	Composite		4.) Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=Surface Water, WW=Waste Water, W=Waste, Liquid, SO=Soil, SD=Sediment, SL=Sludge, SS=Solid Waste, O=Oil, F=Filter, P=Wipe, U=Urine, F=Fecal, N=Naste		srvative is added = leave held blank	UMC	(i.e.: High/low pH, asbestos, beryllium, irritants, other mise health larands etc)	9, 510.)			
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getcom Chair		Phone # 91	Fax# 70		son gols	*Time Collected (Military)												0			RR.)	= Matrix Spi	r sample was i	Water, ML=	ontainers pro	Listed W	LW= Lis	(F,K,P and U-I Waste code(s):				
3 6LROUP	GEL Work Order Number:	Ph			Send Results To:Genna Olson g	*Date Collected (-									 Chain of Custody Signatures	Received by (signed)	A A	2	3	nt & Review form (S.	3 = Equipment Blank. MS	as field filtered or - N - fo	r, WW=Waste Water, W=	B/7470A) and number of o	d, 5H = 5001um Hydroxide, 5A = 5011um Characteristic Hazards	FL = Flammable/Ignitable	sive ive		ulated	hinhenvls	
,oloh	GEL Worl	an, PC	nal Land		Send Resu											ain of Cust					mple Recei	d Duplicate. El	es the sample w	^≖Surface Wate	e. 8260B, 6010	d, SH = Sodiun Characteri	FL, = Flamr	CO = Corrosive RE = Reactive		TSCA Regulated		
of 1 VWA-001 VWRA Quote NA	NA	NWRA c/o Hart & Hickman, PC	Project/Site Name: Upper Piedmont Regional Landfill	Rougemont, NC	Patrick Stevens	Sample ID	7304-1									Cha	Date	Munt 09-18-19 1630			> For sample shipping and delivery details, see Sample Receipt & Review form (SRR.)	 Clain of Custody Number = Client Determined Clain of Custody Number = Client Determined QC Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sumple, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite 	3.) Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered.	V=Drinking Water, GW=Groundwater, SW	 Sample Analysis Requested: Analytical method requested (i.e. 82608, 60108/7470A) and number of containers provided for each (i.e. 8260B - 3, 60108/7470A - 1). 	0.) Preservaive 19pe: HA = hytrochrone Acid, N = Nime Acid, SH = Solutin hytrochroe. SA = Ascortine Acid, HA = fexante, SI = Solutin I hocultate, II no preservative is added = leave held blank 7.) Are there any known or possible hazards Characteristic Hazards Listed Waste Other			Hg= Mercury	Se= Selenium	ellaneous	RCRA metals
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Page 15 of 19 SDG: 490879 Rev1

ATTACHMENT D

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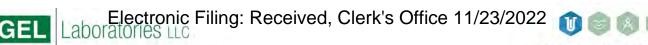
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 Treservauve 19pe, n.A. – пулочного Activity A. – имеет Activity (7.) Are there any known or possible hazards 	Characteristic Hazards	Listed Waste		Other				F	Vease provide	Please provide any additional details	details
associated with these samples?	FL = Flammable/Ignitable	LW= Listed Waste	te	0T=Oth	OT= Other / Unknown	. 1			elow regardi	below regarding handling and/or disposal	for disposal
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Ba = Barium Se = Selentum Cd = Cadmium Ag = Silver P	PCB = Polychlorinated										
MR= Miscellaneous PCP A metals	biphenyls										

	Laboratories Laboratories				
С	lient: NIABA		Th		SAMPLE RECEIPT & REVIEW FORM
R	eccived By: ATA				DG/AR/COC/Work Order: HALDS GARY HOLDS 79
F	<u> </u>		·		ate Received: 4/1/4/14
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A):	Shipped as a DOT Hazardous?		J	/Ha	izard Class Shipped: UN#: UN2910, Is the Radioactive Shipment Survey Compliant? Yes No
B) rec	Did the client designate the samples are to be eived as radioactive?		$\overline{}$	7	DC notation or radioactive stickers on containers equal elient designation.
C) rad	Did the RSO classify the samples as ioactive?		V	Ma Cli	aximum Net Counts Observed* (Observed Counts - Area Background Counts):CPM / mR/Hr
	Did the client designate samples are ardous?	1	\checkmark	co	C notation or hazard labels on containers equal elient designation.
E) [Did the RSO identify possible hazards?		\checkmark	IF D PC	D or E is yes, select Hazards below. B's Flammable Foreign Soil RCRA Asbestos Beryllium Other:
	Sample Receipt Criteria	Yes	VN	ŝ	Comments/Qualifiers (Required for Non-Conforming Items)
1	Shipping containers received intact and sealed?	\square			Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
2	Chain of custody documents included with shipment?	\square			Circle Applicable: Client contacted and provided COC COC created upon receipt
3	Samples requiring cold preservation within $(0 \le 6 \deg C)$?*	\square	ANGS		Preservation Method; Wet Ro lee Packs Dry ice None Other: *all temperatures are recorded in Celsius TEMP:
4	Daily check performed and passed on IR temperature gun?	\square			Temperature Device Serial #: <u>7.84 - 16</u> Secondary Temperature Device Serial # (If Applicable):
	Sample containers intact and sealed?			\checkmark	Circle Applicable: Seals broken Damaged container Leaking container Other (describe) <u>6204-1 (1botHe) Cap received Cracked</u> Sample ID's and Containers Affected;
6	Samples requiring chemical preservation at proper pH?				IC Preservation added. Lotte
7	Do any samples require Volatile Analysis?			V	If Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) Are liquid VOA vials free of headspace? Yes No NA Sample ID's and containers affected:
8	Samples received within holding time?	$\overline{\Lambda}$			ID's and tests affected:
,	Sample ID's on COC match ID's on bottles?	$\overline{/}$	の行う		ID's and containers affected:
	Date & time on COC match date & time on bottles?				Circle Applicable: No dates on containers No times on containers COC missing info Other (describe)
<u> </u>	Number of containers received match number indicated on COC?				Circle Applicable: No container count on COC Other (describe)
-	Are sample containers identifiable as GEL provided? COC form is properly signed in	Δ			Circle Applicable: Mot relinquished a Other (describe)
1	relinquished/received sections? ments (Use Continuation Form if needed):		國		Oner (describe)
0	itens (Ose Continuation Porth If Accord):				
<u></u>	PM (or PM.	A) rev	iew:	Initi	and a second sec
					GL-CHL-SR-001 Rev 6

WhA JA d Tracking Number d Information DT Hazardous? lesignate the samples are to be etive? assify the samples as lesignate samples are entify possible hazards? le Receipt Criteria Itainers received intact and tody documents included nt?		NA No	D: *If Ha If U Co Lif D PCI	SAMPLE RECEIPT & REVIEW FORM DG/AR/COC/Work Order: 400819 OR AR/COC/Work Order: 400819 ate Received: 1000000000000000000000000000000000000
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tody documents included nt?	1	plan, er	2 [°]	Comments/Qualifiers (Required for Non-Conforming Items)
nt?				Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
				Circle Applicable: Client contacted and provided COC COC created upon receipt
uiring cold preservation deg. C)?*				Preservation Method Wet Ice Packs Dry ice None Other: *all temperatures are recorded in Celsius TEMP:
performed and passed on II gun?				Temperature Device Serial #: <u>T.R.4 ~ L6</u> Secondary Temperature Device Serial # (IT Applicable):
niners intact and sealed?				Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
tiring chemical preservation	1			Sample ID's and Containers Affected: If Preservation added, Lot#:
amples require Volatile Analysis?			1	If Yes, are Encores or Soil Kits present for solids? YesNoNA(If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? YesNoNA(If unknown, select No) Are liquid VOA vials free of headspace? YesNoNA Sample ID's and containers affected:
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on COC match date & time	Л	050962		Circle Applicable: No dates on containers No times on containers COC missing info Other (describe)
ontainers received match ated on COC?	/			Circle Applicable: No container count on COC Other (describe)
1?	Д		_/	Circle Applicable: Not relinquished Other (describe)
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	amples require Volatile Analysis? ved within holding time? n COC match ID's on m COC match date & time ntainers received match tted on COC? mtainers identifiable as	amples require Volatile Analysis? ved within holding time? n COC match ID's on on COC match date & time mainers received match ted on COC? matainers identifiable as	amples require Volatile Analysis? ved within holding time? n COC match ID's on m COC match date & time ntainers received match ted on COC? matainers identifiable as	amples require Volatile Analysis? ved within holding time? n COC match ID's on m COC match date & time mtainers received match ted on COC? mtainers identifiable as ?

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012 SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water	LA024
Louisiana NELAP	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235–19–15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019



a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

November 08, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for Sampson County Disposal, LLC Work Order: 490881

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on September 19, 2019 and September 20, 2019. This revised data report has been prepared and reviewed in accordance with GEL's standard operating procedures. This package was revised to include PFPeA and PFOA.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

rilie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 490881 GEL Work Order: 490881

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- J See case narrative for an explanation
- J Value is estimated
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecared RAFORIES ice C1/23/2022

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forSampson County Disposal, LLC		
Client Sample ID:	8202-1	Project:	NWRA00119
Sample ID:	490881001	Client ID:	NWRA001
Matrix:	Misc Liquid		
Collect Date:	18-SEP-19 12:20		
Receive Date:	19-SEP-19		
Collector:	Client		

Parameter	Qualifier	Result	DL	RL	Units	PF	DF	Anal	yst Date	Time	Batch	Method
LCMSMS PFCs												
EPA 537Mod PFCs by L	C-MS/MS	"As Received"										
Fluorotelomer sulfonate 4:2 (4:2		ND	13.2	37.6	ng/L	0.200	1	JLS	10/02/19	1007	1921240	1
FTS)					-							
N-ethylperfluoro-1-	NT.	43.8	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (1 EtFOSAA)	N-											
N-methylperfluoro-1-		104	13.2	40.0	ng/L	0.200	1					
octanesulfonamidoacetic acid (I	N-				6							
MeFOSAA)												
Perfluorodecanesulfonic acid	U	ND	6.60	19.4	ng/L	0.200	1					
(PFDS) Perfluorodecanoic acid (PFDA)		90.9	7.80	20.0	ng/L	0.200	1					
Perfluorododecanoic acid (PFDA)		90.9 9.17	6.60	20.0	ng/L ng/L	0.200						
Perfluoroheptanesulfonic acid	U U	ND	6.60	20.0 19.0	ng/L ng/L	0.200						
(PFHpS)	U	ND	0.00	19.0	lig/L	0.200	1					
Perfluorohexanesulfonic acid		424	6.60	18.2	ng/L	0.200	1					
(PFHxS)					U							
Perfluorononanesulfonic acid	U	ND	7.00	19.2	ng/L	0.200	1					
(PFNS)		120	6.60	20.0		0.000	1					
Perfluorononanoic acid (PFNA)		128	6.60	20.0	ng/L	0.200						
Perfluorooctanesulfonamide (PFOSA)	U	ND	6.60	18.6	ng/L	0.200	1					
Perfluorooctanesulfonic acid (P	FOS)	222	8.00	20.0	ng/L	0.200	1					
Perfluorooctanoic acid (PFOA)	,	1790	7.00	20.0	ng/L	0.200						
Perfluoropentanesulfonic acid		61.0	6.60	18.8	ng/L	0.200						
(PFPeS)					8							
Perfluoroundecanoic acid (PFU	dA) J	10.2	6.60	20.0	ng/L	0.200	1					
2,3,3,3-Tetrafluoro-2-		10800	330	1000	ng/L	0.200	50	JLS	10/02/19	0806	1921240	2
(1,1,2,2,3,3,3-heptafluoropropo	xy)-											
propanoic acid (PFPrOPrA) Fluorotelomer sulfonate 8:2 (8:2	2 U	ND	660	1920	no/I	0.200	50					
FTS)	2 0	ND	000	1920	ng/L	0.200	30					
Perfluorobutanesulfonic acid (P	(FBS)	7530	330	890	ng/L	0.200	50					
Perfluorobutyric acid (PFBA)	,	4770	330	1000	ng/L	0.200	50					
Perfluoroheptanoic acid (PFHp	A)	5520	330	1000	ng/L	0.200	50					
Perfluorohexanoic acid (PFHxA	A)	6730	330	1000	ng/L	0.200	50					
Perfluorotetradecanoic acid	U	ND	330	1000	ng/L	0.200	50					
(PFTeDA)					-							
Perfluorotridecanoic acid (PFT)	-	ND	330	1000	ng/L	0.200						
Fluorotelomer sulfonate 6:2 (6:	2 U	ND	1320	3800	ng/L	0.200	100	JLS	10/02/19	1209	1921240	3
FTS)												

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Certificate of Analysis

Report Date: November 8, 2019

Company :	NWRA - Carolinas Chapter		
Address :	1550 Crystal Drive, Suite 804		
	Arlington, Virginia 22202		
Contact:	Mr. Jim Riley		
Project:	Analytical forSampson County Disposal, LLC		
Client Sample ID:	8202-1	Project:	NWRA00119
Sample ID:	490881001	Client ID:	NWRA001

Parameter	Qualifier	Result	DL	RL	Units	PF 1	DF	Analyst Date	Time Batch	h Method
LCMSMS PFCs										
EPA 537Mod PFCs by	LC-MS/MS "	As Received"								
Perfluoropentanoic acid (PFPeA) 86400			660	2000	ng/L	0.200	100			
The following Prep Met	erformed:									
Method Description				Analyst	Date	Т	ïme	Prep Batch		
EPA 537.1 Mod, PFAS, Com	EPA 537.1 Mod, PFAS, Compl PFCs Extraction in Liquid				09/27/19	0	830	1921239		
The following Analytic	al Methods w	vere performed:								
Method	Description				A	Analyst	Con	nments		
1	EPA 537.1 Mo	od, PFAS, Compliant with QSM Ta	ble B-15							
2	EPA 537.1 Mo	od, PFAS, Compliant with QSM Ta	ble B-15							
3	EPA 537.1 Mo	od, PFAS, Compliant with QSM Ta	ble B-15							
Notes:										

Column headers are defined as follows:DF: Dilution FactorLc/LC: Critical LevelDL: Detection LimitPF: Prep FactorMDA: Minimum Detectable ActivityRL: Reporting LimitMDC: Minimum Detectable ConcentrationSQL: Sample Quantitation Limit

Electronic Filing Elecarbor Alectronic Filing Elecarbor Al

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: November 8, 2019

	Company : Address :			inas Chapter rive, Suite 804											
	Contact: Project:	Mr.	Jim Riley	ginia 22202 ampson County	Dispos	sal, LLC									
	Client Sample ID:	8202	2-1					Pro	ject:		NWRA	400119			
	Sample ID:	490	881002					Cli	ent ID	:	NWRA001				
	Matrix:	Mis	c Liquid												
	Collect Date:	18-5	SEP-19 12:	20											
	Receive Date:	20-5	SEP-19												
	Collector:	Clie	nt												
Parameter	Quali	fier	Result		DL	RL	Uni	its	PF	DF	Analys	st Date	Time Batch	Meth	nod
Semi-Volat	ile-GC/MS														
	0 SIM 1,4-Dioxane	in Li	uid "As R	eceived"											
1,4-Dioxane	- · · · · · ·		184		5.00	10.0) u	g/L	0.200	5	JMB3	09/24/19	2011 1919444		1
The followi	ng Prep Methods w	ere pe	erformed:												
Method	Desci	ription	1			Analyst	Dat	e		Time	Pre	p Batch			
SW846 3535A	SW82	70E SIN	M Prep 1,4-Di	oxane		SJW1	09/2	3/19		1200	191	9441			
The follow	ing Analytical Meth	nods v	vere perfori	med:											
Method	Descr	iption	-					A	Analys	t Cor	nments				
1	SW846	3535A	/8270E SIM												
Surrogate/T	racer Recovery	Test					Result	ľ	Nomin	al	Recov	ery%	Acceptable L	imits	
1,4-Dioxane-d	-	W846 Receive		-Dioxane in Liquid '	'As		27.7 ug/L		40	0.0	69*		(70%-130%))	
Notes:															
Column he	aders are defined as	follo	ws:												
DF: Dilutio				Lc/LC: Critical											
DL: Detect	ion Limit			PF: Prep Factor											

MDA: Minimum Detectable Activity MDC: Minimum Detectable Concentration

RL: Reporting Limit SQL: Sample Quantitation Limit

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QC Summary

Report Date: November 8, 2019

Page 1 of 7

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia

Contact: Mr. Jim Riley

Workorder: 490881

Parmname	NOM	Sample Qual	QC	Units RPD)/D% REC%	Range A	nlst Da	te Time
Perfluorinated Compounds Batch 1921240								
QC1204391614 LCS 2,3,3,3-Tetrafluoro-2- (1,1,2,2,3,3,3- heptafluoropropoxy)-propanoic acid (PFPrOPrA)	19.5		17.1	ng/L	88	(70%-137%)	JLS 10/0	2/19 06:05
Fluorotelomer sulfonate 4:2 (4:2 FTS)	18.2		15.7	ng/L	86	(60%-145%)		
Fluorotelomer sulfonate 6:2 (6:2 FTS)	18.5		20.4	ng/L	110	(56%-143%)		
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.7		17.5	ng/L	94	(57%-138%)		
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	19.5		19.3	ng/L	99	(63%-131%)		
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	19.5		21.5	ng/L	111	(62%-133%)		
Perfluorobutanesulfonic acid (PFBS)	17.2		16.6	ng/L	96	(68%-136%)		
Perfluorobutyric acid (PFBA)	19.5		19.7	ng/L	101	(70%-133%)		
Perfluorodecanesulfonic acid (PFDS)	18.8		16.8	ng/L	89	(53%-142%)		
Perfluorodecanoic acid (PFDA)	19.5		18.0	ng/L	93	(62%-135%)		
Perfluorododecanoic acid (PFDoA)	19.5		19.5	ng/L	100	(66%-131%)		
Perfluoroheptanesulfonic acid (PFHpS)	18.5		18.1	ng/L	98	(66%-138%)		
Perfluoroheptanoic acid (PFHpA)	19.5		17.9	ng/L	92	(67%-135%)		

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QC Summary

Workorder: 490881		<u><u> </u></u>		<u></u>					Page 2 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated CompoundsBatch1921240									
Perfluorohexanesulfonic acid (PFHxS)	17.7		14.5	ng/L		82	(64%-137%)) JLS	10/02/19 06:05
Perfluorohexanoic acid (PFHxA)	19.5		18.9	ng/L		97	(67%-133%))	
Perfluorononanesulfonic acid (PFNS)	18.7		17.5	ng/L		93	(66%-130%))	
Perfluorononanoic acid (PFNA)	19.5		21.1	ng/L		108	(66%-134%))	
Perfluorooctanesulfonamide (PFOSA)	19.5		21.5	ng/L		111	(68%-137%)	
Perfluorooctanesulfonic acid (PFOS)	19.5		19.8	ng/L		102	(61%-131%))	
Perfluorooctanoic acid (PFOA)	19.5		18.8	ng/L		97	(63%-145%))	
Perfluoropentanesulfonic acid (PFPeS)	18.3		16.5	ng/L		90	(62%-139%)	
Perfluoropentanoic acid (PFPeA)	19.5		19.3	ng/L		99	(69%-132%)	
Perfluorotetradecanoic acid (PFTeDA)	19.5		22.5	ng/L		115	(65%-143%))	
Perfluoroundecanoic acid (PFUdA)	19.5		19.1	ng/L		98	(65%-134%))	
QC1204391615 LCSD 2,3,3,3-Tetrafluoro-2- (1,1,2,2,3,3,3- heptafluoropropoxy)-propanoic	18.8		18.1	ng/L	5	96	(0%-30%))	10/02/19 06:14
acid (PFPrOPrA) Fluorotelomer sulfonate 4:2 (4:2 FTS)	17.6		20.5	ng/L	26	116	(0%-35%))	
Fluorotelomer sulfonate 6:2 (6:2 FTS)	17.9		17.6	ng/L	14	98	(0%-36%))	
Fluorotelomer sulfonate 8:2 (8:2 FTS)	18.1		19.9	ng/L	13	110	(0%-39%))	

Page 7 of 20 SDG: 490881 Rev1

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QC Summary

Workorder: 490881			·	<u></u>				Page 3 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Date Time
Perfluorinated CompoundsBatch1921240								
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)	18.8		20.1	ng/L	. 4	107	(0%-25%) JLS	10/02/19 06:14
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)	18.8		21.9	ng/L	. 2	116	(0%-26%)	
Perfluorobutanesulfonic acid (PFBS)	16.7		17.2	ng/L	. 4	103	(0%-30%)	
Perfluorobutyric acid (PFBA)	18.8		19.3	ng/L	. 2	102	(0%-30%)	
Perfluorodecanesulfonic acid (PFDS)	18.2		17.2	ng/L	. 3	95	(0%-28%)	
Perfluorodecanoic acid (PFDA)	18.8		21.1	ng/L	. 16	112	(0%-29%)	
Perfluorododecanoic acid (PFDoA)	18.8		19.0	ng/L	. 3	101	(0%-30%)	
Perfluoroheptanesulfonic acid (PFHpS)	17.9		17.7	ng/L	. 2	99	(0%-30%)	
Perfluoroheptanoic acid (PFHpA)	18.8		19.6	ng/L	. 9	104	(0%-30%)	
Perfluorohexanesulfonic acid (PFHxS)	17.2		16.8	ng/L	. 15	98	(0%-30%)	
Perfluorohexanoic acid (PFHxA)	18.8		20.9	ng/L	. 10	111	(0%-23%)	
Perfluorononanesulfonic acid (PFNS)	18.1		18.2	ng/L	. 4	101	(0%-27%)	
Perfluorononanoic acid (PFNA)	18.8		18.7	ng/L	. 12	99	(0%-27%)	
Perfluorooctanesulfonamide (PFOSA)	18.8		20.2	ng/L	. 6	107	(0%-30%)	
Perfluorooctanesulfonic acid (PFOS)	18.8		19.9	ng/L	. 1	106	(0%-27%)	

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QC Summary

Workorder: 490881								
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range Anlst	Page 4 of 7 Date Time
Perfluorinated Compounds Batch 1921240		Bumpie Quai	<u><u>v</u>v</u>	Cintis	R D D / 0	REC /U	Kunge Amst	Dute Time
Perfluorooctanoic acid (PFOA)	18.8		18.9	ng/L	0	100	(0%-30%) JLS	10/02/19 06:14
Perfluoropentanesulfonic acid (PFPeS)	17.7		17.3	ng/L	4	98	(0%-29%)	
Perfluoropentanoic acid (PFPeA)	18.8		20.0	ng/L	3	106	(0%-30%)	
Perfluorotetradecanoic acid (PFTeDA)	18.8		20.6	ng/L	9	109	(0%-30%)	
Perfluoroundecanoic acid (PFUdA)	18.8		21.2	ng/L	10	112	(0%-28%)	
QC1204391613 MB 2,3,3,3-Tetrafluoro-2- (1,1,2,2,3,3,3- heptafluoropropoxy)-propanoic		U	ND	ng/L				10/02/19 05:56
acid (PFPrOPrA) Fluorotelomer sulfonate 4:2 (4:2 FTS)		U	ND	ng/L				
Fluorotelomer sulfonate 6:2 (6:2 FTS)		U	ND	ng/L				
Fluorotelomer sulfonate 8:2 (8:2 FTS)		U	ND	ng/L				
N-ethylperfluoro-1- octanesulfonamidoacetic acid (N- EtFOSAA)		U	ND	ng/L				
N-methylperfluoro-1- octanesulfonamidoacetic acid (N- MeFOSAA)		U	ND	ng/L				
Perfluorobutanesulfonic acid (PFBS)		U	ND	ng/L				
Perfluorobutyric acid (PFBA)		U	ND	ng/L				
Perfluorodecanesulfonic acid (PFDS)		U	ND	ng/L				
Perfluorodecanoic acid (PFDA)		U	ND	ng/L				

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QC Summary

			JIIIIIaI	<u> </u>					
Workorder: 490881									Page 5 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
Perfluorinated CompoundsBatch1921240									
Perfluorododecanoic acid (PFDoA)		U	ND	ng/L				JLS	10/02/19 05:56
Perfluoroheptanesulfonic acid (PFHpS)		U	ND	ng/L					
Perfluoroheptanoic acid (PFHpA)		U	ND	ng/L					
Perfluorohexanesulfonic acid (PFHxS)		U	ND	ng/L					
Perfluorohexanoic acid (PFHxA)		U	ND	ng/L					
Perfluorononanesulfonic acid (PFNS)		U	ND	ng/L					
Perfluorononanoic acid (PFNA)		U	ND	ng/L					
Perfluorooctanesulfonamide (PFOSA)		U	ND	ng/L					
Perfluorooctanesulfonic acid (PFOS)		U	ND	ng/L					
Perfluorooctanoic acid (PFOA)		U	ND	ng/L					
Perfluoropentanesulfonic acid (PFPeS)		U	ND	ng/L					
Perfluoropentanoic acid (PFPeA)		U	ND	ng/L					
Perfluorotetradecanoic acid (PFTeDA)		U	ND	ng/L					
Perfluoroundecanoic acid (PFUdA)		U	ND	ng/L					

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QC Summary

Workorder: 490881									Page 6	of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Ti	me
Semi-Volatile-GC/MS Batch 1919444										
QC1204387349 LCS **1,4-Dioxane-d8	4.00		3.55	ug/L		89	(70%-130%)	JMB3	09/24/19 1	2:24
QC1204387350 LCSD **1,4-Dioxane-d8	4.00		3.18	ug/L		79	(70%-130%)	I	09/24/19 1	2:49
QC1204387348 MB 1,4-Dioxane		U	ND	ug/L					09/24/19 1	1:59
**1,4-Dioxane-d8	4.00		3.05	ug/L		76	(70%-130%)	I		

Notes:

The Qualifiers in this report are defined as follows:

** Analyte is a surrogate compound

.....

- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- $N\!/\!A$ $\,$ RPD or %Recovery limits do not apply.
- N1 See case narrative
- ND Analyte concentration is not detected above the detection limit
- NJ Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier
- P Organics--The concentrations between the primary and confirmation columns/detectors is >40% different. For HPLC, the difference is >70%.
- Q One or more quality control criteria have not been met. Refer to the applicable narrative or DER.

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QC Summary

Workorder: 490881									Page 7 of 7
Parmname	NOM	Sample Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date Time
R Sample results are rejected									

U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

UJ Compound cannot be extracted

X Consult Case Narrative, Data Summary package, or Project Manager concerning this qualifier

Y QC Samples were not spiked with this compound

^ RPD of sample and duplicate evaluated using +/-RL. Concentrations are <5X the RL. Qualifier Not Applicable for Radiochemistry.

h Preparation or preservation holding time was exceeded

N/A indicates that spike recovery limits do not apply when sample concentration exceeds spike conc. by a factor of 4 or more or %RPD not applicable. ^ The Relative Percent Difference (RPD) obtained from the sample duplicate (DUP) is evaluated against the acceptance criteria when the sample is greater than five times (5X) the contract required detection limit (RL). In cases where either the sample or duplicate value is less than 5X the RL, a control limit of +/- the RL is used to evaluate the DUP result.

* Indicates that a Quality Control parameter was not within specifications.

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Technical Case Narrative NWRA - Carolinas Chapter SDG #: 490881

GC/MS Semivolatile

<u>Product:</u> Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1919444

Preparation Method: SW846 3535A **Preparation Procedure:** GL-OA-E-073 REV# 2 **Preparation Batch:** 1919441

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
490881002	8202-1
1204387348	Method Blank (MB)
1204387349	Laboratory Control Sample (LCS)
1204387350	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Surrogate Recoveries

Sample (See Below) did not meet surrogate recovery acceptance criteria. The sample was analyzed at a dilution. As a result, one or more surrogates were diluted out of the acceptance limits.

Sample	Analyte	Value
490881002 (8202-1)	1, 4-Dioxane-d8	69* (70%-130%)

Laboratory Control Sample Duplicate (LCSD)

An LCSD was used in place of matrix QC due to limited sample volume.

Technical Information

Sample Dilutions Sample 490881002 (8202-1) was diluted due to the presence of one or more over-range target analytes.

Electronic Filing: Received, Clerk's Office 11/23/2022 <u>LCMSMS-Misc</u>

<u>Product:</u> The Extraction and Analysis of Per and Polyfluroalkyl Substances Using LCMSMS <u>Analytical Method:</u> EPA 537.1 Mod, PFAS, Compliant with QSM Table B-15 <u>Analytical Procedure:</u> GL-OA-E-076 REV# 7 <u>Analytical Batches:</u> 1921240 and 1921239

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	<u>Client Sample Identification</u>
490881001	8202-1
1204391613	Method Blank (MB)
1204391614	Laboratory Control Sample (LCS)
1204391615	Laboratory Control Sample Duplicate (LCSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Technical Information

Sample Dilutions

The following samples were diluted to bring the over range concentrations within the calibration range and/or due to matrix interference that caused internal standards recoveries to fall outside the acceptance range. 490881001 (8202-1).

Analysis	490881
Analyte	001
2,3,3,3-Tetrafluoro-2-(1,1,2,2,3,3,3-heptafluoropropoxy)-propanoic acid (PFPrOPrA)	50X
Fluorotelomer sulfonate 6:2 (6:2 FTS)	100X
Fluorotelomer sulfonate 8:2 (8:2 FTS)	50X
Perfluorobutanesulfonate (PFBS)	50X
Perfluorobutyric acid (PFBA)	50X
Perfluoroheptanoic acid (PFHpA)	50X
Perfluorohexanoic acid (PFHxA)	50X
Perfluoropentanoic acid (PFPeA)	100X
Perfluorotetradecanoic acid (PFTeDA)	50X
Perfluorotridecanoic acid (PFTrDA)	50X

Miscellaneous Information

Additional Comments

Additional sample volume was not provided for matrix QC. Also, reduced sample volumes were used for all samples except 490876002 (7607-EB) due to elevated concentrations of target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

		GEL Laboratories, LLC 2040 Survey David
	atty Analytics	2040 Savage Koad Charleston SC 29407
Chain of		Phone: (843) 556-8171
OIK Urder Number		Fax: (843) 766-1178
NWRA c/o Hart & Hickman, PC Phone # 919-847-4241	Sample Analysis Requested (5) (Fill in the mumb	(Fill in the number of containare for each true)
nty Disposal, LLC Fax # 704-586-0007		
T	ainer'	C Preservative Type (6)
Patrick Stevens Send Results To:Genna Olson golson@harthickman.com		A 537 mod Comments
	ard:	Note: avtro comulo :
*Date Collected *Time *Time *Time *Time *Enter collected Collected <thcollected< th=""> Collected Co</thcollected<>	00051016 haz	required for sample is specific QC
N ML	1	
	Note that you will	Note that you will receive a separate conter -
	9/20/19 with bottle	9/20/19 with bottles for 1,4-dioxane analysis-
	for this site. It is (for this site. It is OK to include those in the -
	same lab report as these samples.	s these samples.
Chain of Custod	TAT Requested: Normal: X Rush: Snorifi-	
Time Received by (signed)	[V] MA	
18-18-19 1630 10 XXX 9111 1 85CD		1 Clave I [] Clave] []
	Additional Remarks:	1 1 1 1 1 1 1
> For sumple shipping and delivery details, see Sumule Receipt & Review Gram (CDD)	ng Use Only:	[] No Cooler Temp:C
1.1 Chain of Custody Number = Client Determined	ne: [X] Eastern [] Pacific []] Central []] Mountain	[] Other:
2.) QC Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Duplicate Sumple, G = Grah, C = Composite	Sumple, $\mathbf{G} = G$ mb, $\mathbf{C} = C$ omposite	
 Field Filtered: For fiquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered. Mentic Column Methods and a supplementation of the sample was field filtered or - N - for sample was not field filtered. 		
	sludge, SS=Solid Waste, O=Oil, F≖Filter, P≃Wipe, U=Urine, F=Fecal, N=Nar	
	d - 1).	
A start of the source, set = assertion pyerostice, SA = Southere Acid, HX = Hexane, ST = Souther Anazords [Characteristic Hazards] [Listed Waste	m Thiosuffate, If no preservative is added = leave field blank Ottoor	
E Listed Waste (F.K.P and U-listed wastes.))ther / Unknown Igh/low pH, asbestos, bervllitum, irritants, other	Please provide any additional details below regarding handling and/or disposal concorrection of a concernent
		of site collected from, odd matrices, etc.)
TSCA Regulated PCB = Polychlorinated		
KCKA metals		

Page 16 of 20 SDG: 490881 Rev1

AIIACHMENID

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	or cach test)	< Preservative Type (6)	Comments	Note: extra sample is	required for sampled specific OC	troi	hic	Fi	ling): F	Red	cei	ve	Service and the service of the servi	Cle	(Subject to Surcharge	s (] Level 3 [] Level	ice	Cooler Temp:C	/23/2	202	2		Please provide any additional details	Detow regarang nanang unavor apposa concerns. (i.e.: Origin of sample(s), type	of site collected from, odd matrices, etc.)		
GEL Laboratories, LLC 2040 Savage Road Charleston, SC 29407 Phone: (843) 556-8171 Fax: (843) 766-1178	(Fill in the number of containers for each test)															Specify:(] level 1 [] Level 2 [[]No			N=Nasal		Please provide	concerns. (i.e.:	of site collected		
GEL 2040 Charl Phono	879 25 EE EE EE															KRush: S		[] QC Summary [] le		Custody Seal Intact? [] Yes			=Wipe, U=Urine, F=Fecal,	cave field blank		ium, irritants, other			
ialty Analytics 1	Sample Analysis Requested (5)	50	2	5 Q	€88 t	ry ry X	 									TAT Requested: Normal:	[] Yes MNo	C of A	smarks:		, [av, e = composite	l Waste, O=Oil, F≃Filter, P	no preservative is added =	Inknown	(i.e.: High/low pH, asbestos, beryllium, irritants, other	iazards, etc.)		
Laboratories LLC Chemistry I Radiobioassey I Specialty Analytics Custody and Analytical Request GEL Project Manuger: Julie Redu			sample he considered:	srds r	avitystoll olquv avit olqu aiqu onvon¥ assd aldis assd aldis	000 (L) 022 9/J										TAT Re	Fax Results: [] Yes	Select Deliverable: []	5 Additional Remarks:	For Lab Receiving Use Only.		a rupiteae sampie, er – ur	Liquid, SO=Soil, SD=Sediment, SL=Sludge, SS=Solid Waste, O=Oil, F=Filter, P=Wipe, U=Urine, F=Fecal, N=Nasal	(0108/7470A - 1). T = Sodium Thiosulfate, If	Other OT= Other / I Inknown	(i.e.: High/lo	misc. health hazards, etc.) Description:	•	
	202-27-84			Joran Brahideway	Field Sample	²³ Piftered ¹³ Matrix ⁴⁵	2										Time	1, 1700	1119 8:53	Samela Callaction		aupte, mete - mauto option ield filtered.	Liquid, SO=Soil, SD=Sedi	d for each (i.e. 8260B - 3, 6 rbic Acid, HX = Hexane, S	e Waste	(F,K,P and U-listed wastes.)); ·		
	Phone # and - Pod- a7 84		2853	0125	*Time Collected (Millary) OC	(Hiteran) Code										res	(signed) Date	25 9-19-1	1444 - 9121		and the state of t	and, was - manth spike at r - N - for sample was not fi	∕ater, W=Water, ML=Misc	mber of containers provide = Sulfuric Acid, AA = Asco	Listed Wast		Waste code(s):		
HADSS Conter Number:	Her Hickmen R.C.	spesal, LLC	lise burne, A	Send Results To: Genna	*Date Collected	(1-16-10)										Chain of Custody Signatures	Received by (signed)	1 Teele	2 M. G	3 5		sample was field filtered or	face Water, WW≃Waste W	0 B , 6010 B /7470.A) and nui = Sodium Hydroxide, SA =	Characteristic Hazards ET = Flammahle/fonitable	CO = Corrosive	RE = Reactive	TSCA Regulated	s = rolyculorinated biphenyls
		3	Hur J	7		tare and stop date time										Chain 6	Date Time	9-19-19 1720		and describe and County	ermined	trip mains, r.v rieu uur cate with a - Y - for yes the	W=Groundwater, SW=Sur.	l method requested (i.e. 826 Acid, NI = Nitric Acid, SH			RE	ISI	
0. <u>14</u> NUZA NUZA NUZA 13. UIA	NWRA C	Jame: Saw Par	Attress 7434 Roscherro Hund, Roscherro,	y: P. Stevens	Sample ID	* For compositos - indicate start and stop date time											Conquished By (Signed) Da	ueth 9		3 3 3 3 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	 A subject supplies and according are number and equilations (1) that (1) younger concertor sums seems 1.1 however, 1.1 when 1.1. Chain of Custody Number = Client Determined 1.2. Chain of Custody Number = Client Determined 	 Q. COGS: N = NOTHAL SATIPLE, 10 - 117 DIAIN, FU - FREU DUPRCAUE, ED - Equipment DIAIN, 143 - MARTIN SPIRE SATIPLE, 1412. Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not field filtered. 	 Matrix Codes: DW=Drinking Water, GW=Groundwater, SW=Surface Water, WW=Waste Water, W=Water, ML=Misc 	 5.) Sample Analysis Requested: Analytical method requested (i.e. 82608, 60108/7470A) and number of containers provided for each (i.e. 8260B - 3, 6010B/7470A - 1). 6.) Preservative Type: HA = Hydrochloric Acid, NI = Nitric Acid, SH = Sodium Hydroxide, SA = Sulfuric Acid, AA = Ascorbic Acid, HX = Hexane, ST = Sodium Thiosulfate, If no preservative is added = leave field blank 	 Are there any known or possible hazards acconited with these cumber? 	a waa mese sumpres:	ils Mercury		um Ag= Silver ium MR= Miscellaneous RCRA metals
Parte: Parte:	Cont Name:	Poect/Site N	Address: 71	Geted By	81 R	ev1					АT	TA	CI	HN	E	١т	Davisher	1 rela	2	3	L.) Chain of Cus	 2.) QU UOGES: 1 3.) Field Filtered 	4.) Matrix Code	 Sample Anal Preservative 	7.) Are there i	noornie	RCRA Metals As = Asenic	Ba = Barium	Cr = Cadmum Cr = Chromium Pb = Lead

CEE Laboratories up	;	тħ		SAMPLE RECEIPT & REVIEW FORM								
Client: NWBA		<u>- a</u>	Ist	DG/AR/COC/Work Order: HAAV CANA HADRA								
Received By: ArA			1	ate Received: 9/19/19								
				Circle Applicable:								
Carrier and Tracking Number			11 12	Fedex Express FedEx Ground UPS Field Services Courier Other 4762 7563 2308 -1, 7762 7563 3418-1° 4762 7563 2764 -1°, 7762 7563 3290 -1°								
spected Hazard Information	Yes	z	*1f	Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation								
Shipped as a DOT Hazardous?		1		zard Class Shipped: UN#: UN2910, Is the Radioactive Shipment Survey Compliant? YesNo								
B) Did the client designate the samples are to be ceceived as radioactive?				DC notation or radioactive stickers on containers equal client designation.								
C) Did the RSO classify the samples as adioactive?			Ma Clr	aximum Net Counts Observed* (Observed Counts - Area Background Counts):CPM / mR/Hr lassified as: Rad 1 Rad 2 Rad 3								
Did the client designate samples are zardous?		\checkmark	1	C notation or hazard labels on containers equal client designation.								
Did the RSO identify possible hazards?		\mathbb{V}	PC	D or E is yes, select Hazards below. B's Flammable Foreign Soil RCRA Asbestos Beryllium Other:								
Sample Receipt Criteria	Ycs	ž	ź	Comments/Qualifiers (Required for Non-Conforming Items)								
Shipping containers received intact and sealed?	\bigvee			Circle Applicable: Seals broken Damaged container Leaking container Other (describe)								
Chain of custody documents included with shipment?	\bigvee			Circle Applicable: Client contacted and provided COC COC created upon receipt								
Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?*	\bigvee	PEEE		Preservation Method: Wet to Ice Packs Dry ice None Other: •all temperatures are recorded in Celsius TEMP:								
Daily check performed and passed on IR temperature gun?	\square			Temperature Device Serial #: <u>TB4-L6</u> Secondary Temperature Device Serial # (If Applicable): Circle Applicable: Seals broken Damaged containe Leaking container Othyr (describe)								
Sample containers intact and sealed?			\checkmark	6204-L (Lbottle) cap received cracked Sample ID's and Containers Affected;								
Samples requiring chemical preservation at proper pH?				If Preservation added, Lot#								
Do any samples require Volatile Analysis?			√.	II Yes, are Encores or Soil Kits present for solids? YesNoNA(If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? YesNoNA(If unknown, select No) Are liquid VOA vials free of headspace? YesNoNA Sample ID's and containers affected:								
Samples received within holding time?	$\overline{\Lambda}$			ID's and tests affected:								
Sample ID's on COC match ID's on bottles?	\langle			ID's and containers affected:								
Date & time on COC match date & time on bottles?	\bigvee	<u>,</u>		Circle Applicable: No dates on containers No times on containers COC missing info Other (describe)								
Number of containers received match number indicated on COC?	\checkmark			Circle Applicable: No container count on COC Other (describe)								
Are sample containers identifiable as GEL provided?	$ \Lambda$											
COC form is properly signed in relinquished/received sections?			\checkmark	Circle Applicable: Not relinquished A Other (describe)								
nments (Use Continuation Form if needed):												
PM (or PM	(A) re	view	: Ini	tials VOY Date 125 Page of								
-				GL-CHL-SR-001 Rev 6								

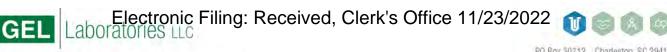
Page 18 of 20 SDG: 490881 Rev1

	\mathcal{T}	R.		SAMPLE RECEIPT & REVIEW FORM
Client: MAA	<u></u> _	<u>v</u>	Isi	DG/AR/COC/Work Orger: 40 DS 8
Received By: Art				ate Received: 9120119
Carrier and Tracking Number				Circle Applicable: FedEx Express FedEx Ground UPS Field Services Courier Other FFGZ 8638 8788
		<u> </u>		7767 8638 8034
uspected Hazard Information	Yes	Ž	_ ,	f Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigation
Shipped as a DOT Hazardous?		1		zard Class Shipped: UN#: UN2910, Is the Radioactive Shipment Survey Compliant? Yes No
Did the client designate the samples are to b ceived as radioactive?	.c	/	/_co	DC notation or radioactive stickers on containers equal elient designation.
Did the RSO classify the samples as dioactive?			Ma: Cla	Eximum Net Counts Observed* (Observed Counts - Area Background Counts):CPM / mR/Hr
) Did the client designate samples are zardous?			со	C notation or hazard labels on containers equal elient designation.
Did the RSO identify possible hazards?		V	If D PCt	D or E is yes, select Hazards below. B's Flammable Foreign Soil RCRA Asbestos Beryllium Other:
Sample Receipt Criteria	Yes	ź	^o Z	Comments/Qualifiers (Required for Non-Conforming Items)
Shipping containers received intact and sealed?	7			Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
Chain of custody documents included with shipment?	1			Circle Applicable: Client contacted and provided COC COC created upon receipt
Samples requiring cold preservation within $(0 \le 6 \text{ deg. C})$?*		1		Preservation Method, Wet lee) tee Packs Dry ice None Other: *all temperatures are recorded in Celsius TEMP:
Daily check performed and passed on I temperature gun?	R 🗸	10.00		Temperature Device Serial #: <u>TB4-L6</u> Secondary Temperature Device Serial # (If Applicable):
Sample containers intact and sealed?				Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
Samples requiring chemical preservation at proper pH?	n	J		Sample ID's and Containers Affected:
Do any samples require Volatile Analysis?			1	If Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) Are liquid VOA vials free of headspace? Yes No NA Sample ID's and containers affected:
Samples received within holding time?	17			ID's and tests affected:
Sample ID's on COC match ID's on bottles?	Ţ/			ID's and containers affected:
Date & time or COC match date & time on bottles?	1	19995	1	Circle Applicable: No dates on containers No times on containers COC missing info Other (describe)
Number of containers received match number indicated on COC?	Л		ľ	Circle Applicable: No container count on COC Other (describe)
Are sample containers identifiable as GEL provided?	\square			
COC form is properly signed in relinquished/received sections? AA 9	X		1	Circle Applicable: Not relinquished Other (describe)

~

State	Certification
Alaska	17-018
Alaska Drinking Water	SC00012
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330–15–00283, P330–15–00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water Louisiana NELAP	LA024
	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68–00485
Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 08 November 2019



a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

October 14, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for Great Oak Landfill Work Order: 491597

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on October 01, 2019. This original data report has been prepared and reviewed in accordance with GEL's standard operating procedures.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 491597 GEL Work Order: 491597

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecarbor Alectronic Filing Elecarbor Al

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

a ant Data Ostahar 14 2010

								Report Da	te: October	14, 2019
	Company : Address :		olinas Chapter Drive, Suite 804							
	Contact: Project:	Mr. Jim Rile	irginia 22202 y rGreat Oak Landf	fill						
	Client Sample ID Sample ID: Matrix:	491597001 Misc Liquid					roject: lient ID:	NWRA00119 NWRA001		
	Collect Date: Receive Date: Collector:	30-SEP-19 0 01-OCT-19 Client	9:55							
Parameter	Qual	ifier Result		DL	RL	Units	PF DF	Analyst Date	Time Batch	Metho
Semi-Volat	ile-GC/MS									
1,4-Dioxane	0 SIM 1,4-Dioxan	U ND		0.100	0.400	ug/L	0.020 1	JMB3 10/08/19	1130 1924252	. 1
The followi Method	ng Prep Methods v	A			A	Data	T!	Dran Datah		
SW846 3535A		cription 270E SIM Prep 1,4-	Dioxane		Analyst SJ	Date 10/07/19	Time 1230	Prep Batch	l	
	ing Analytical Met	-			55	10/07/12	, 1250	1724231		
Method	• •	ription	inca.				Analyst Cor	nments		
1		6 3535A/8270E SI	А				7 maryst Col	milents		
Surrogate/T	racer Recovery	Test				Result	Nominal	Recovery%	Acceptable L	imits
1,4-Dioxane-d		SW846 8270 SIM Received"	,4-Dioxane in Liquid	"As		3.43 ug/L	4.00	86	(70%-130%)	
Notes:										
DF: Dilutio DL: Detect			Lc/LC: Critical PF: Prep Factor RL: Reporting	r						

MDA: Minimum Detectable Activity MDC: Minimum Detectable Concentration RL: Reporting Limit SQL: Sample Quantitation Limit

Electronic Filing Elecarbor Alectronic Electronic Files

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: October 1/ 2019

								Report Da	te: October	14, 2019
	Company : Address :		olinas Chapter Drive, Suite 804							
		Arlington, Vi	irginia 22202							
	Contact:	Mr. Jim Rile	/							
	Project:	Analytical fo	rGreat Oak Landfi	11						
	Client Sample ID:	7607-2				Pr	oject:	NWRA00119		
	Sample ID:	491597002				C	lient ID:	NWRA001		
	Matrix:	Misc Liquid								
	Collect Date:	30-SEP-19 1	0:35							
	Receive Date:	01-OCT-19								
	Collector:	Client								
	0.1					TT				
Parameter	Qual	ifier Result		DL	RL	Units	PF DF	7 Analyst Date	Time Batch	Method
Semi-Volat										
	70 SIM 1,4-Dioxane	-		•••	10.0	~				
1,4-Dioxane		469		20.0	40.0	ug/L	0.200 20	JMB3 10/08/19	1154 1924252	2 1
	ing Prep Methods w	<u> </u>								
Method		ription			Analyst	Date	Tim	-	1	
SW846 3535A		70E SIM Prep 1,4-			SJ	10/07/19	1230	1924251		
	ing Analytical Met	nods were perfo	ormed:							
Method		iption					Analyst Co	mments		
1	SW840	5 3535A/8270E SIN	А							
Surrogate/T	Fracer Recovery	Test				Result	Nominal	Recovery%	Acceptable L	imits
1,4-Dioxane-d	18	SW846 8270 SIM 1 Received"	,4-Dioxane in Liquid "	As	2	47.1 ug/L	40.0	118	(70%-130%)
Notes:										
Column he	aders are defined as	follows:								
DF: Dilutio			Lc/LC: Critical	Level						
DL: Detect			PF: Prep Factor							
MDA: Min	nimum Detectable A	ctivity	RL: Reporting L	imit						

MDA: Minimum Detectable Activity MDC: Minimum Detectable Concentration

RL: Reporting Limit SQL: Sample Quantitation Limit

2040 Savage Road Charleston, SC 29407 - (843) 556-8171 - www.gel.com

QC Summary

Report Date: October 14, 2019

Page 1 of 2

NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia Contact: Mr. Jim Riley

Workorder: 491597

Parmname			NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
Semi-Volatile-GC/M Batch 192	S 4252												
QC1204398479 **1,4-Dioxane-d8	LCS		4.00			3.61	ug/L		90	(70%-130%)	JMB3	10/08/1	9 11:05
QC1204398478 1,4-Dioxane	MB				U	ND	ug/L					10/08/1	9 10:40
**1,4-Dioxane-d8			4.00			4.22	ug/L		105	(70%-130%)			
QC1204398483 **1,4-Dioxane-d8	491597002	MS	40.0	47.1		42.2	ug/L		106	(70%-130%)		10/08/1	19 12:19
QC1204398484 **1,4-Dioxane-d8	491597002	MSD	40.0	47.1		35.1	ug/L		88	(70%-130%)		10/08/1	19 12:44

Notes:

The Qualifiers in this report are defined as follows:

- ** Analyte is a surrogate compound
- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor

Page 5 of 11 SDG: 491597

2040 Savage Road Charleston, SC 29407 - (843) 556-8171 - www.gel.com

QC Summary

***	• • • • • • • •		-	QC DI	4111111a	<u> </u>						
Worko	rder: 491597										Pag	e 2 of 2
Parmna	me	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
N/A	RPD or %Recovery limits do	not apply.										
N1	See case narrative											
ND	Analyte concentration is not o	letected above the	detection lin	nit								
NJ	Consult Case Narrative, Data	Summary packag	e, or Project	Manager c	concerning	this qualif	ier					
Р	OrganicsThe concentrations	between the prin	nary and conf	irmation c	olumns/det	ectors is >	40% different	. For HPLC	c, the differ	ence is >7	70%.	
Q	One or more quality control c	riteria have not be	en met. Refe	r to the ap	plicable na	rrative or 1	DER.					
R	Sample results are rejected											
U	Analyte was analyzed for, but	not detected abo	ve the MDL,	MDA, MI	DC or LOD							
UJ	Compound cannot be extracted	d										
Х	Consult Case Narrative, Data	Summary packag	e, or Project	Manager c	concerning	this qualif	ier					
Y	QC Samples were not spiked	with this compou	nd									
٨	RPD of sample and duplicate	evaluated using +	-/-RL. Conce	ntrations a	are <5X the	RL. Qua	lifier Not App	licable for F	Radiochem	istry.		
h	Preparation or preservation he	olding time was e	kceeded									
^ The R five tim RL is us	licates that spike recovery limit telative Percent Difference (RP ues (5X) the contract required du sed to evaluate the DUP result. ates that a Quality Control para	D) obtained from etection limit (RL	the sample du). In cases wh	uplicate (I here either	DUP) is eva	aluated aga	ainst the accep	otance criteri	ia when the	e sample is	s greater	
			-rieue									

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Electronic Filing: Received, Clerk's Office 11/23/2022 GC/MS Semivolatile Technical Case Narrative NWRA - Carolinas Chapter SDG #: 491597

Product: Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1924252

<u>Preparation Method:</u> SW846 3535A <u>Preparation Procedure:</u> GL-OA-E-073 REV# 2 <u>Preparation Batch:</u> 1924251

The following samples were analyzed using the above methods and analytical procedure(s).

<u>GEL Sample ID#</u>	Client Sample Identification
491597001	7607-EB
491597002	7607-2
1204398478	Method Blank (MB)
1204398479	Laboratory Control Sample (LCS)
1204398483	491597002(7607-2) Matrix Spike (MS)
1204398484	491597002(7607-2) Matrix Spike Duplicate (MSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Spike Recovery Statement

The MS and MSD (See Below) spike recoveries were not within the acceptance limits. There was a detected presence of 1,4-Dioxane above the reporting limits in the un-spike parent sample that caused a biased calculated spike recovery result in the MS and MSD. The data results have been reported.

Sample	Analyte	Value
1204398483 (7607-2MS)	1, 4-Dioxane	0* (70%-130%)
1204398484 (7607-2MSD)	1, 4-Dioxane	0* (70%-130%)

Technical Information

Sample Dilutions

Samples 1204398483 (7607-2MS), 1204398484 (7607-2MSD) and 491597002 (7607-2) were diluted due to the presence of non-target analytes. The data from the dilutions are reported. Samples 1204398483 (7607-2MS), 1204398484 (7607-2MSD) and 491597002 (7607-2) were diluted due to the presence of one or more over-range

target analytes.

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

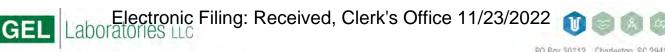
Holdunoon Photoman Photoman <t< th=""></t<>
NL D43174 Supple be consistent Collection (initiary) Collection (initiary) Supple be consistent AC ML ASS EA NH Provide interval (initiary) AA Provide interval (initiary) AA Provide interval (initiary) AA Provide interval (initiary) Provide interval Provide interval Provide interval Provide interval Provide interval Provide interval
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med) Date Tat Requested: Normal: Specify: med) Date Tat Requested: Normal: Specify: med) Date Time Fax Results: [] Yes Moi Modificional Renarks: Select Deliverable: [] C of A [] QC Summary [] level 1
Table Table Image: Specify: Table Image: Specify: Specify: Table Image: Specify: Specify: Table Image: Specify: Specify: Specify: Table Image: Specify: Specify
med) Date Time Fax Results: [] Yes [MNo] Def [] [] QC Summary [] [] [] [] Cof A [] QC Summary [] [] [] [] [] [] Additional Remarks:

6					SAMPLE RECEIPT & REVIEW FORM						
				s	DG/AR/COC/Work Order:						
R	eccived By:			D	hate Received: 10-1-19						
Carrier and Tracking Number					FedEx Express FedEx Ground UPS Field Services Courier Other						
u.	spected Hazard Information	n n	T		7763 8929 7266						
		Yes	2	+-	f Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigati						
	Shipped as a DOT Hazardous?		li		azard Class Shipped: UN#: UN2910, Is the Radioactive Shipment Survey Compliant? YesNo						
)	Did the client designate the samples are to be eived as radioactive?		L	Ka	OC notation or radioactive stickers on containers equal client designation.						
ld	Did the RSO classify the samples as ioactive?		l	Ma Ci	aximum Net Counts Observed* (Observed Counts - Area Background Counts):CPM / mR/Hr assified as: Rad 1 Rad 2 Rad 3						
') az	Did the client designate samples are ardous?		~		OC notation of hazard labels on containers equal client designation.						
)[Did the RSO identify possible hazards?		~	PC	D or E is yes, select Hazards below. B's Flammable Foreign Soil RCRA Asbestos Beryllium Other:						
	Sample Receipt Criteria	Yes	Ň	z	Continents (Ventices (Non-Conforming Iteres)						
	Shipping containers received intact and sealed?	V		0.000	Circle Applicable: Seals broken Damaged container Leaking container Other (describe)						
!	Chain of custody documents included with shipment?	И	11.2		Circle Applicable: Client contacted and provided COC COC created upon receipt						
;	Samples requiring cold preservation within $(0 \le 6 \text{ deg. } C)$?*	1	PI.		Preservation Method, Wet Ice Ice Packs Dry ice None Other: "all temperatures are recorded in Celsius TEMP:						
;	Daily check performed and passed on IR temperature gun?	V			Temperature Device Serial #						
-	Sample containers intact and sealed?		調査が		Circle Applicable: Seals broken Damaged container Leaking container Other (describe)						
	Samples requiring chemical preservation at proper pH?		5	/	Sample ID's and Containers Affected: If Preservation added, Lot#						
	Do any samples require Volatile				If Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer)						
	Analysis?		ANTER L	V	Sample ID's and containers affected:						
4	Samples received within holding time?	A	信息が		ID's and tests affected:						
_	Sample ID's on COC match ID's on bottles?	1	10.01		ID's and containers affected:						
ļ	Date & time on COC match date & time on bottles?	P	7		Circle Applicable: No dates on containers No times on containers COC missing info Other (describe)						
1	Number of containers received match number indicated on COC?	4		_	Circle Applicable: No container count on COC Other (describe)						
1	Are sample containers identifiable as GEL provided?		調約	J							
ľ	COC form is properly signed in relinquished/received sections?		羽		Circle Applicable: Not relinquished Other (describe)						

PM (or PMA) review: Initials 5/4 Date 10/2/19 Page ____ of ____

	incations as of 14 October 2019
State	Certification
Alaska	17–018
Arkansas	88-0651
CLIA	42D0904046
California	2940
Colorado	SC00012
Connecticut	PH-0169
DoD ELAP/ ISO17025 A2LA	2567.01
Florida NELAP	E87156
Foreign Soils Permit	P330-15-00283, P330-15-00253
Georgia	SC00012
Georgia SDWA	967
Hawaii	SC00012
Idaho	SC00012
Illinois NELAP	200029
Indiana	C-SC-01
Kansas NELAP	E-10332
Kentucky SDWA	90129
Kentucky Wastewater	90129
Louisiana Drinking Water	LA024
Louisiana NELAP	03046 (AI33904)
Maine	2019020
Maryland	270
Massachusetts	M-SC012
Massachusetts PFAS Approv	Letter
Michigan	9976
Mississippi	SC00012
Nebraska	NE-OS-26-13
Nevada	SC000122020-1
New Hampshire NELAP	2054
New Jersey NELAP	SC002
New Mexico	SC00012
New York NELAP	11501
North Carolina	233
North Carolina SDWA	45709
North Dakota	R-158
Oklahoma	2019–165
Pennsylvania NELAP	68-00485
Pennsylvania NELAP Puerto Rico	SC00012
S. Carolina Radiochem	10120002
Sanitation Districts of L	
	9255651
South Carolina Chemistry	10120001 TNL02024
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–28
Vermont	VT87156
Virginia NELAP	460202
Washington	C780

List of current GEL Certifications as of 14 October 2019



a member of The GEL Group INC

PO Box 30712 Charleston, SC 29417 2040 Savage Road Charleston, SC 29407 P 843,556,8171 F 843,766,1178

gel.com

December 19, 2019

Mr. Jim Riley NWRA - Carolinas Chapter 1550 Crystal Drive, Suite 804 Arlington, Virginia 22202

Re: Analytical for CMS Landfill Work Order: 498420

Dear Mr. Riley:

GEL Laboratories, LLC (GEL) appreciates the opportunity to provide the enclosed analytical results for the sample(s) we received on December 05, 2019. This original data report has been prepared and reviewed in accordance with GEL's standard operating procedures.

Test results for NELAP or ISO 17025 accredited tests are verified to meet the requirements of those standards, with any exceptions noted. The results reported relate only to the items tested and to the sample as received by the laboratory. These results may not be reproduced except as full reports without approval by the laboratory. Copies of GEL's accreditations and certifications can be found on our website at www.gel.com.

Our policy is to provide high quality, personalized analytical services to enable you to meet your analytical needs on time every time. We trust that you will find everything in order and to your satisfaction. If you have any questions, please do not hesitate to call me at (843) 556-8171, ext. 4289.

Sincerely,

Julie Roberson

Julie Robinson Project Manager

Purchase Order: GELP19-0905 Enclosures

GEL LABORATORIES LLC

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis Report for

NWRA001 NWRA – Carolinas Chapter

Client SDG: 498420 GEL Work Order: 498420

The Qualifiers in this report are defined as follows:

- * A quality control analyte recovery is outside of specified acceptance criteria
- ** Analyte is a Tracer compound
- ** Analyte is a surrogate compound
- U Analyte was analyzed for, but not detected above the MDL, MDA, MDC or LOD.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the Certificate of Analysis.

The designation ND, if present, appears in the result column when the analyte concentration is not detected above the limit as defined in the 'U' qualifier above.

This data report has been prepared and reviewed in accordance with GEL Laboratories LLC standard operating procedures. Please direct any questions to your Project Manager, Julie Robinson.

Julie Roberson

Reviewed by

Electronic Filing Elecarbor Alectrice C1/23/2022

2040 Savage Road Charleston SC 29407 - (843) 556-8171 - www.gel.com

Certificate of Analysis

Report Date: December 19, 2019

								Report Du	ac. December	17, 2017
	Company :		rolinas Chapter							
	Address :	1550 Crystal	Drive, Suite 804							
		Arlington, V	irginia 22202							
	Contact:	Mr. Jim Rile	y							
	Project:	Analytical fo	orCMS Landfill							
	Client Sample II	D: 1, 1A, 2, 2A				Pr	oject:	NWRA00119		
	Sample ID:	498420001				Cl	ient ID:	NWRA001		
	Matrix:	Water								
	Collect Date:	04-DEC-19	13:30							
	Receive Date:	05-DEC-19								
	Collector:	Client								
Parameter	Qua	lifier Result		DL	RL	Units	PF DF	Analyst Date	Time Batch	Method
Semi-Volat	ile-GC/MS									
	0 SIM 1,4-Dioxa	ne in Liquid "As	Received"							
1,4-Dioxane		214		4.00	8.00	ug/L	0.200 4	JMB3 12/11/19	0925 1947214	1
	ing Prep Methods	were performed								
Method		cription			Analyst	Date	Tim	-	1	
SW846 3535A	SW8	270E SIM Prep 1,4	Dioxane		SJ	12/10/19	0800	1947213		
The follow	ing Analytical Me	thods were perfe	ormed:							
Method	Des	cription					Analyst Co	mments		
1	SW8	46 3535A/8270E SI	М							
Surrogate/T	Tracer Recovery	Test				Result	Nominal	Recovery%	Acceptable L	imits
1,4-Dioxane-d	8	SW846 8270 SIM Received"	1,4-Dioxane in Liquid "	As	· · · · · · · · · · · · · · · · · · ·	25.3 ug/L	40.0	63*	(70%-130%)
Notes:										
Column he	aders are defined	as follows:								
DF: Dilutio			Lc/LC: Critical							
DL: Detect	ion Limit		PF: Prep Factor							

MDA: Minimum Detectable Activity MDC: Minimum Detectable Concentration CC/LC: Critical Level PF: Prep Factor RL: Reporting Limit SQL: Sample Quantitation Limit

2040 Savage Road Charleston, SC 29407 - (843) 556-8171 - www.gel.com

QC Summary

Report Date: December 19, 2019

NWRA - Carolinas Chapter Page 1 of 2 1550 Crystal Drive, Suite 804 Arlington, Virginia **Contact:** Mr. Jim Riley Workorder: 498420 Parmname NOM Sample Qual QC Units RPD/D% **REC%** Range Anlst Date Time Semi-Volatile-GC/MS 1947214 Batch OC1204451621 LCS **1,4-Dioxane-d8 4.00 3.18 79 (70%-130%) JMB3 12/10/19 15:57 ug/L OC1204451620 MB U ND 12/10/19 15:33 1.4-Dioxane ug/L 4.00 **1,4-Dioxane-d8 3.48 ug/L 87 (70%-130%) QC1204451622 498420001 MS 25.3 25.0 **1,4-Dioxane-d8 40.0 ug/L 63* (70% - 130%)12/11/19 09:50 QC1204451623 498420001 MSD 25.3 40.0 28.0 ug/L 12/11/19 10:13 **1,4-Dioxane-d8 70 (70% - 130%)

Notes:

The Qualifiers in this report are defined as follows:

- ** Analyte is a surrogate compound
- < Result is less than value reported
- > Result is greater than value reported
- A The TIC is a suspected aldol-condensation product
- B The target analyte was detected in the associated blank.
- C Analyte has been confirmed by GC/MS analysis
- D Results are reported from a diluted aliquot of the sample
- E Concentration of the target analyte exceeds the instrument calibration range
- H Analytical holding time was exceeded
- J See case narrative for an explanation
- J Value is estimated
- JNX Non Calibrated Compound
- N Organics--Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor
- N Presumptive evidence based on mass spectral library search to make a tentative identification of the analyte (TIC). Quantitation is based on nearest internal standard response factor

Page 4 of 10 SDG: 498420

Electronic Filing: Received, Clerk's Office 11/23/2022 GEL LABORATORIES LLC

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QC Summary

Workor			~ -									e 2 of 2
Parmnai	me	NOM	Sample	Qual	QC	Units	RPD/D%	REC%	Range	Anlst	Date	Time
N/A	RPD or %Recovery limits d	o not apply.										
N1	See case narrative											
ND	Analyte concentration is not	detected above the	detection lim	it								
NJ	Consult Case Narrative, Dat	a Summary package	e, or Project M	Manager	concerning t	his qualif	ier					
Р	OrganicsThe concentration	ns between the prim	ary and confi	rmation	columns/dete	ectors is >	40% different	. For HPLC	C, the differ	ence is >7	0%.	
Q	One or more quality control	criteria have not be	en met. Refe	to the a	pplicable nai	rative or 1	DER.					
R	Sample results are rejected											
U	Analyte was analyzed for, b	ut not detected abov	ve the MDL,	MDA, M	DC or LOD							
UJ	Compound cannot be extrac	ted										
Х	Consult Case Narrative, Dat	a Summary package	e, or Project I	Manager	concerning t	his qualif	ier					
Y	QC Samples were not spike	d with this compour	nd									
^	RPD of sample and duplicat	e evaluated using +	/-RL. Conce	ntrations	are <5X the	RL. Qua	lifier Not App	licable for H	Radiochemi	istry.		
h	Preparation or preservation	holding time was ex	ceeded									
^ The Re five time RL is us	icates that spike recovery lim elative Percent Difference (R es (5X) the contract required sed to evaluate the DUP resul tes that a Quality Control par	PD) obtained from detection limit (RL)	the sample du). In cases wh	plicate ere eithe	(DUP) is eva	luated ag	ainst the accep	otance criter	ia when the	e sample is	greater	

For PS, PSD, and SDILT results, the values listed are the measured amounts, not final concentrations.

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless qualified on the QC Summary.

Electronic Filing: Received, Clerk's Office 11/23/2022 GC/MS Semivolatile Technical Case Narrative NWRA - Carolinas Chapter SDG #: 498420

Product: Analysis of 1,4-Dioxane in Drinking Water by Solid Phase Extraction (SPE) and Gas Chromatography/Mass Spectrometry <u>Analytical Method:</u> SW846 3535A/8270E SIM <u>Analytical Procedure:</u> GL-OA-E-073 REV# 2 <u>Analytical Batch:</u> 1947214

<u>Preparation Method:</u> SW846 3535A <u>Preparation Procedure:</u> GL-OA-E-073 REV# 2 <u>Preparation Batch:</u> 1947213

The following samples were analyzed using the above methods and analytical procedure(s).

GEL Sample ID#	Client Sample Identification
498420001	1, 1A, 2, 2A
1204451620	Method Blank (MB)
1204451621	Laboratory Control Sample (LCS)
1204451622	498420001(1, 1A, 2, 2A) Matrix Spike (MS)
1204451623	498420001(1, 1A, 2, 2A) Matrix Spike Duplicate (MSD)

The samples in this SDG were analyzed on an "as received" basis.

Data Summary:

All sample data provided in this report met the acceptance criteria specified in the analytical methods and procedures for initial calibration, continuing calibration, instrument controls and process controls where applicable, with the following exceptions.

Quality Control (QC) Information

Surrogate Recoveries

Samples (See Below) did not meet surrogate recovery acceptance criteria. Since the parent sample and associated MS/MSD pair displayed similar recoveries, the failures were attributed to matrix interference and the data results are reported.

Sample	Analyte	Value
1204451622 (1, 1A, 2, 2AMS)	1, 4-Dioxane-d8	63* (70%-130%)
498420001 (1, 1A, 2, 2A)	1, 4-Dioxane-d8	63* (70%-130%)

Spike Recovery Statement

The MS or MSD (See Below) recovered spiked analytes outside of the established acceptance limits. As similar recoveries were displayed in the MS and MSD, the failures were attributed to sample matrix interference and the data were reported.

Sample	Analyte	Value
--------	---------	-------

1204451622 (1, 1A, 2, 2AMS)	1, 4-Dioxane	0* (70%-130%)
1204451623 (1, 1A, 2, 2AMSD)	1, 4-Dioxane	30* (70%-130%)

Technical Information

Sample Dilutions

Samples 1204451622 (1, 1A, 2, 2AMS), 1204451623 (1, 1A, 2, 2AMSD) and 498420001 (1, 1A, 2, 2A) were diluted due to the presence of one or more over-range target analytes.

Miscellaneous Information

Manual Integrations

Sample (See Below) required manual integration in order to properly identify one or more peaks and/or to correctly position the baseline as set in the calibration standard injections.

Sample	Analyte	Value
498420001 (1, 1A, 2, 2A)	Tetrahydrofuran-d8	Result 400ug/L

Certification Statement

Where the analytical method has been performed under NELAP certification, the analysis has met all of the requirements of the NELAC standard unless otherwise noted in the analytical case narrative.

Page: and a string to an and a			GEL Laboratories, LLC
Treast I want want a want of the second seco		u I Snariaitu Analidire	2040 Savage Road Charleston SC 20407
CAC Number ⁽¹⁾ .	Chain of Custody and Analytical Request	contraint the	Phone: (843) 556-8171
PANumber:	GEL Work Order Number: GEL Project Manuger:	0 Ch 2	Fax: (843) 766-1178
Cient Name:	Phone #	Sample Analysis Requested ⁽⁵⁾	(Fill in the number of containers for each test)
Post/Site Name:	Fax #	Shoeld this	Preservative Type (6)
Athress:		sample be considered:	
Cellected By:	Send Results To:	spa	Comments Note: evtra samula is
0 Sample ID * Exercise indicate and and horizon	*	avitasoihs: viaqu: avoi ovon yo nvon yo naced aidouo tasimua tato wolly yo wolly yo wolly yo	Electric of all sectors and a sector of a
T	12.04-19 13:30	d () () () () () () () () () () () () ()	røn
LA	N 	2 2 2	
R			ilir
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A			R
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Α.			eiv
СН			ed
M			C
ΞN			ler
	Chain of Custody Signatures	TAT Requested: Normal: Rush:	Specify: (Subject to Surcharge)-
Hunquished By (Signed) Date Time	Received by (signed) Date Time	Fax Results: [] Yes [] No	0
	1 a. alperer 12/5/19 8:5	Select Deliverable: [] C of A [] QC Summary [[]level1 []Level2 []Level3 []Level 📆
2	2	Additional Remarks:	e
3	3	For Lab Receiving Use Only: Custody Seal Intact? [] Yes [] No Cooler Temp:
> For sumple shipping and defivery details, see Sample Receipt & Review form (SRR, 1) Chain of Custody Number = Client Determined	unple Receipt & Review form (SRR.) Sample Collection Time Zone.	[] Eastern [] Pacific [] Central	/23/
2.) QC Codes: N = Normal Sample, $TB = Trip Blank$, $FD = Fiel$	2.) QC Codes: N = Normal Sample, TB = Trip Blank, FD = Field Duplicate, EB = Equipment Blank, MS = Matrix Spike Sample, MSD = Matrix Spike Duplicate Sample, G = Grab, C = Composite	ce Duplicate Sample, $\mathbf{G} = \mathbf{Grab}, \mathbf{C} = \mathbf{Composite}$	20
 Field Filtered: For liquid matrices, indicate with a - Y - for yes the sample was field filtered or - N - for sample was not filed to the sample was not filed. M. Merry ColumNUM-Translation Water CWU-Committeness CWU-Committeness With the sample was not filed. 			
 Sample Analysis Requested: Analytical method requested (i.e 		Liquu, 5O-504, 5D-504, 6D-50408, 5D-50404 Waste, O=04, F=F1464, F=Wipe, U=Urne, F=Fecal, N=Nasa i for each (i.e. 8260B - 3, 6010B/470A - 1).	
 6.) Preservative Type: HA = Hydrochloric Acid, NI = Nitric Acid 6. Acid 	6.) Preservative Type: HA = Hydrochloric Acid, NI = Nitric Acid, SH = Sodium Hydroxide, SA = Suchfaric Acid, HX = Hexane, ST = Sodium Thiosultate, If no preservative is added = leave field blank	5T = Sodium Thiosulfate, If no preservative is added = leave field blank	
(1) Are there any known or possible hazards associated with these samples?	Characteristic Hazards Listed Waste FL = Flammable/Ignitable LW= Listed Waste CO = Corrosive (FK P and TLitelod wireles)	Other OT = Other / Unknown G = Hick/Journ H. ackerse, kommun. Action 24	Please pro below rega
RCRA Metals As = Arsenic He = Mercury		(i.e. 118 www.pri, woosios, oerynnam, in nams, om merse health hazards, etc.) Descrimter	of site collected from, odd matrices, etc.)
Se= Selenium A== Silver	TSCA Regulated PCB = Polychlorinated		
1 MR= Miscellaneous	biphenyls		

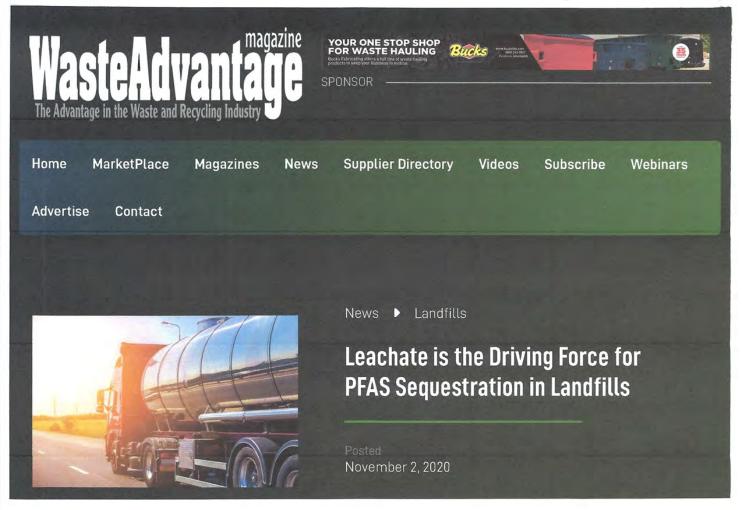
-	ient: NWAA			. Is	SAMPLE RECEIPT & REVIEW FORM 494 420
R	eceived By: ATA				Date Received: 12/5/19
Carrier and Tracking Number			Circle Applicable: FedEx Express FedEx Ground UPS Field Services Courier Other		
					Lane USU Say
Sus	pected Hazard Information	Yes	ź	2 *I	If Net Counts > 100cpm on samples not marked "radioactive", contact the Radiation Safety Group for further investigatio
	hipped as a DOT Hazardous?	_	1	и II	azard Class Shipped: UN#: UN2910, Is the Radioactive Shipment Survey Compliant? YesNo
ece	Did the client designate the samples are to be ived as radioactive?		V		OC notation or radioactive stickers on containers equal client designation.
adi	Did the RSO classify the samples as pactive?		/		aximum Net Counts Observed* (Observed Counts - Area Background Counts): CPM / mR/Hr assified as: Rad 1 Rad 2 Rad 3
א (נ aza	Did the client designate samples are rdous?		V	1.	DC notation or hazard labels on containers equal client designation.
.) C	id the RSO identify possible hazards?			PC	B's Flammable Foreign Soil RCRA Asbestos Beryllium Other:
	Sample Receipt Criteria	Yes	ŚŻ	2 [°]	
1	Shipping containers received intact and sealed?				Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
2	Chain of custody documents included with shipment?				Circle Applicable: Client contacted and provided COC <coc created="" receipt<="" td="" upon=""></coc>
_	Samples requiring cold preservation within $(0 \le 6 \text{ deg. } C)$?*				Preservation Method, Wet Ice Jce Packs Dry ice None Other: *all temperatures are recorded in Celsius TEMP:
	Daily check performed and passed on IR temperature gun?	\square			Temperature Device Serial #: <u>T.B.4 – LC</u> Secondary Temperature Device Serial # (If Applicable):
5	Sample containers intact and sealed?	$ \Lambda$			Circle Applicable: Seals broken Damaged container Leaking container Other (describe)
;	Samples requiring chemical preservation it proper pH?		\langle	C.	Sample ID's and Containers Affected: If Preservation added, Lot#:
	Do any samples require Volatile Analysis?			1	If Yes, are Encores or Soil Kits present for solids? Yes No NA (If yes, take to VOA Freezer) Do liquid VOA vials contain acid preservation? Yes No NA (If unknown, select No) Are liquid VOA vials free of headspace? Yes No NA Sample ID's and containers affected:
S	amples received within holding time?	7			ID's and tests affected:
	ample ID's on COC match ID's on ottles?	$\overline{\Lambda}$			ID's and containers affected:
С 0	ate & time on COC match date & time n bottles?				Circle Applicable: No dates on containers No times on containers COC missing info Other (describe)
In	umber of containers received match umber indicated on COC?	1			Circle Applicable: No container count on COC Other (describe)
G	re sample containers identifiable as EL provided?	7			
re	OC form is properly signed in linquished/received sections?	7			Circle Applicable: Not relinquished Other (describe)
m	ents (Use Continuation Form if needed):	/	88 .		

PM (or PMA) review: Initials 517 Date 12/6/19 Page 1 of

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State	Certification
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DoD ELAP/ ISO17025 A2LA	2567.01
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Sanitation Districts of L	9255651
South Carolina Chemistry	10120001
Tennessee	TN 02934
Texas NELAP	T104704235-19-15
Utah NELAP	SC000122019–29
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Potential for sequestering PFAS shown through mass balance approach. By Arie Kremen, PhD

Landfill leachate is the major pathway by which per- and polyfluoroalkyl substances (PFAS) exit the containment of modern Subtitle D municipal solid waste landfills. PFAS concentrations in leachate vary over time and can be much greater than those found in sanitary wastewaters. The leachate-borne PFAS contribution to the mass loading of publicly owned treatment works (POTW) that accept leachate can equal that contributed by sanitary and industrial wastewaters. This observation contributes to the perception that landfills are PFAS sources.

However, the wastewater treatment centered perception does not correctly reflect the overall role landfills play in the PFAS cycle. To properly determine this role, we have conducted a mass balance

analysis, quantitatively accounting for inputs to and outputs from landfills that convey PFAS. The goal is to determine whether landfills are PFAS sources—as generally perceived—or if they are sequestering PFAS. The high-level analysis is a nationwide mass balance using published research and studies. The results show that the bulk of the landfilled PFAS is sequestered and effectively removed from the environment. The data does have limitations and more work is needed to validate and refine the findings. However, this work serves as a starting point for establishing data-driven PFAS policies and practices. While landfills are shown to sequester PFAS, leachate is the predominant pathway for PFAS out of landfills. Reducing leachate generation is expected to lower the output while reducing operating costs.

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PFAS in Landfills Becomes Mobile

SEE HOW

PFAS are a group of about 4,000 synthetic chemicals used to make fluoropolymer coatings and products that are resilient to physical, chemical and biological degradation. Products are treated with PFAS to imbue heat, stain, grease and water repellency properties to a wide variety of consumer products, including clothing, furniture, adhesives, food packaging, non-stick cooking surfaces and personal care products. In the environment, PFAS are highly mobile and can bioaccumulate in flora and fauna.

Products at the end of their useful life are discarded in landfills, where mechanical breakdown causes PFAS to detach and become mobile. Mobilized PFAS, together with other constituents, can be carried by landfill liquids and landfill gas (LFG). Modern landfills are designed and operated to remove liquids and landfill gas to ensure stability, minimize nuisances, and avoid the creation of safety hazards and adverse environmental conditions. Liquids are generally treated to remove contaminants, while landfill gas is either flared, used for energy generation, or converted to renewable natural gas.

PFAS are generally resilient to biological and chemical processes and are typically unaffected by conventional leachate and wastewater treatment. Data shows that PFAS from leachate and sanitary wastewaters accumulate in biosolids generated in biological wastewater treatment. Some jurisdictions allow the land application of biosolids for soil conditioning purposes. Others are restricting land application in favor of other disposal alternatives, including incineration and landfilling. Among other constituents, landfilled biosolids introduce PFAS into the landfill from leachate and other sources.

The mass balance approach evaluates changes over time in the amount of a constituent within a system

and can provide insight into the relative strengths of inputs and outputs. When applied to landfills, the main inputs include waste, cover material and precipitation, while the outputs are mostly landfill gas and leachate.



We applied the mass balance approach to evaluate if landfills sequester PFAS. In other words, do landfills retain more PFAS than they release to the environment? The estimate is performed on a national level for municipal solid waste landfills. Figure 1 shows a map of landfilled waste by counties in the U.S.

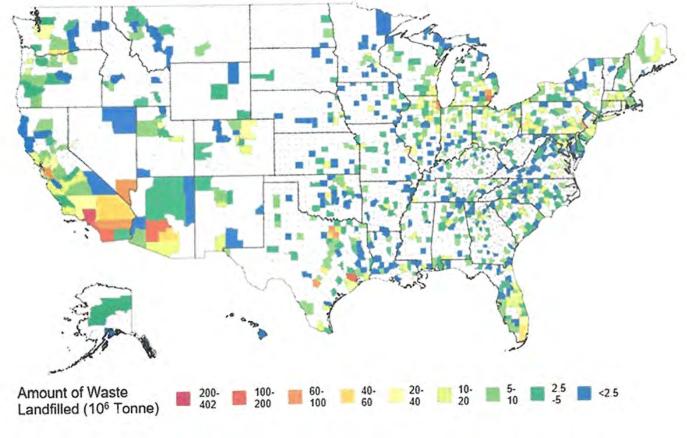


Figure 1

Map of landfilled waste by counties in the U.S. Credit: Waste Informatics: Establishing Characteristics of Contemporary U.S. Landfill Quantities and Practices, September 2016, Environmental Science & Technology, by Jon Powell, José Carlos Pons, and Marian Ruth Chertow, Yale

Mass balance relies on the laws of conservation, one of the most basic tools in scientific investigation. It is routinely applied to a range of static, dynamic, electric, nuclear and chemical systems. Financial budgeting is the application of the law to the world of finance. Mass conservation is generally applied to a well-defined domain (known as the control volume), and accounts for mass entering, leaving or accumulating in the control volume. Mass balance analyses may also consider the production or consumption of a constituent within the control volume in chemical or biological processes. Mass conservation can be expressed as:

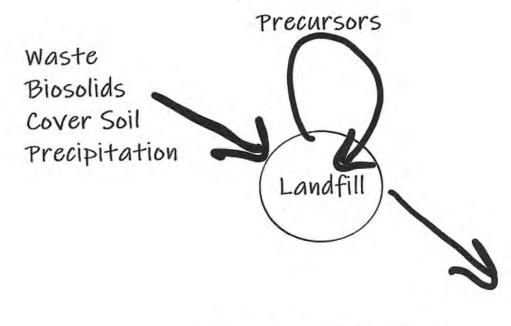
 $M0 + (min - mout + Rnet)\Delta t = M0 + \Delta t$,

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where M0 and M0+∆t is the mass contained within the control volume at the start and end of the calculation period, respectively. The terms min and mout are the mass transfer fluxes into and out of the control volume during the investigated time period, and Rnet is the net rate of production and consumption within the domain. If Rnet is positive, mass is generated at a greater pace than it is being consumed. The terms can be quantified using physical, chemical or biological models or data derived from measurements. Figure 2 shows a generalized schematic of the mass balance approach.



Leachate Landfill Gas

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Figure 2: Schematic of mass balance approach. Image courtesy of Tetra Tech, Inc.

Inputs to and Outputs from a Typical Landfill

Municipal Solid Waste

PFAS content of municipal solid waste (MSW) varies widely. There is no standard methodology for obtaining representative MSW samples and establishing their PFAS content. Values for individual fractions range from 0 to more than 1,000 nanogram PFAS per gram of sample (ng/g). A value of 10 ng/g is considered a representative figure characterizing the overall MSW PFAS content. According to the EPA, about 52.1 percent (2017) of municipal solid waste is landfilled, representing about 137.7 million tons per year. Based on these figures, the annual PFAS disposal rate is calculated to be 2,755 pounds (lbs) per year (1,250 kilograms (kg)/year).

Biosolids

SEE HOW

EPA estimated that biosolid production from wastewater treatment is 7.18 million tons per year (6.51 million kg/year). About 60 percent is land-applied, with 20 percent each being landfilled and incinerated. Landfilling biosolids contributes between 1,030 and 1,295 lbs of PFAS per year (470 to 590 kg/year) (Venkatesan and Halden, 2013).

ATTACHMENT E

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Cover Soils

Many facilities use offsite materials for daily and intermediate soil cover. Soils used may contain contaminants that make the soils unsuitable for off-landfill applications. Little information is documented on PFAS content in such soils. However, Sepulvado et al. (2011) have evaluated the PFAS content, specifically PFOA and PFOS, in agricultural soils to which biosolids were land-applied. Data from this research is used for a conservative estimate of PFAS input with cover soils. PFAS leached from cover soils is assumed to be accounted for in leachate (see below).

Cover soil use is estimated to be 20 percent of the landfilled waste mass, of which land-applied agricultural soils are assumed to be 1 percent. Based on Sepulvado et al., biosolid land-application can result in PFOS and PFOA concentrations of 200 ng/g and 25 ng/g in agricultural soils, respectively. This results in a total of 225 ng PFAS per gram of soil. From these data, the PFAS input with cover soils is calculated to be 123.9 lbs. per year (56.2 kg/year). Work done by McLachlan et al. (2019) on soil/water partitioning of PFAS found that the majority of leaching occurred within 49 to 120 days, with a partitioning coefficient of approximately 0.5. In other words, PFAS are split 50/50 between soil and water, which means cover soils add about 62.0 lbs. per year to the PFAS mass balance.

Precursors

In addition to the inputs discussed previously, landfilled waste contains precursor compounds that are converted to perfluoroalkyl acids (PFAA). These are considered PFAS. Unfortunately, a lack of analytical standards limits the quantification of PFAA. In the proprietary total oxidizable precursor assay (TOPA), chemical oxidation is applied to a sample converting precursor compounds to terminal PFAA. When applied to landfill leachate, results indicate that precursor compounds can amount to approximately 50 percent.

Precipitation

In the past, rainwater was not considered to be a PFAS transport route. However, research conducted at the National Atmospheric Deposition Program at the University of Wisconsin-Madison detected PFAS in all 37 samples collected. Most samples contain less than 1 ng PFAS per liter (ng/l). The highest concentration was nearly 5.5 ng/l, with a mode concentration of less than 1 ng/l. A separate study by the North Carolina Department of Environmental Quality, Division of Air Quality, found 500 ng/l in samples near a PFAS-producing facility.



Precipitation is the dominant source for leachate generation. For estimation purposes, the annual rainwater infiltration rate equals the leachate generation rate. Lang et al. (2019) provide an estimate of 16,180 million gallons of leachate per year. Conservatively, the rainwater PFAS concentration is assumed to be 10 ng/l. The PFAS input from rainwater is calculated to be 1.35 lbs. per year (0.61 kg/year).

Outputs

Leachate

Lang et al. (2017) developed an estimate for the PFAS mass in leachate, across three climatic regions of the U.S. The model estimates the annual leachate volume and extrapolates PFAS mass from a limited number of samples using a Monte-Carlo analysis. Results indicate that the 90th percentile range for PFAS carried by leachate ranges from 1,240 to 1405 lbs per year (563 to 638 kg/year). The leachate generation rate is estimated to be 16,140 million gallons per year (61.1 million m3/year).

Landfill Gas

Monitoring data indicates that PFAS are dry-deposited in areas downwind of landfills, indicating that fugitive and point-source emissions could be sources. Flaring of landfill gas (LFG) is believed to incompletely destroy PFAS. Tian (2018) directly measured PFAS content in landfill gas and found that concentrations ranged from 650 to 850 pg/m3 of LFG. This mass balance analysis uses a value of 1,000 pg/m3 for a conservative estimate.

Applying the EPA LandGEM model to estimate LFG generation from landfilled MSW with a methane generation rate of k=0.05 1/year and a specific methane generation capacity of L0=100 m3/Mg, the LFG generation in 2020 is estimated to be 771,900 million scfm per year (21,858 m3/year). Based on these estimates, the PFAS content of the annual LFG generation is calculated to be 0.05 lbs. per year (0.02 kg/year). In comparison to the other sources, this amount is negligible. For purposes of this analysis, PFAS contained in fugitive emissions or in flared landfill gas are not considered.

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Summing it Up: Most Landfilled PFAS is Sequestered

Table 1 summarizes the PFAS mass loadings of the inputs and outputs considered. Annual PFAS inputs are

approximately 2.2 tons, of which MSW contributes about 60 percent. Biosolids and precursors represent 25 percent and 15 percent, respectively. Agricultural cover soils contribute approximately 1 percent. Nationally, the contribution from rainwater is negligible. This is also true for landfills situated in areas with high concentrations in rainwater.

These results show that the majority of the PFAS output is associated with leachate—the amount in landfill gas is negligible. The results also show that less than one third (27.7 percent to 29.2 percent) of PFAS landfilled is collected with leachate. In other words, the bulk of the landfilled PFAS is sequestered and effectively removed from the environment. This amounts to about 3,234 lbs. to 3,415 lbs. per year (1,467 kg to 1,549 kg/year).

More Work Needed to Build on High Level Mass Balance Data

It should be no surprise that landfills retain more PFAS than they are releasing. Modern landfills are designed, constructed, and operated to eliminate uncontrolled discharges and reduce leachate generation. It is also not surprising that leachate is the major pathway for PFAS leaving the landfill environment as they are substantially non-volatile, which means only very low quantities are present in LFG. Landfilled biosolids account for between 83 percent and 93 percent of the PFAS discharged with leachate, indicating that PFAS mass exchange between wastewater treatment and landfill is nearly balanced (see Figure 3).



Figure 3: Leachate truck. Photo courtesy of Getty Images.



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Component	Inp	uts	Outputs		
	Min	Max	Min	Max	
MSW	1,:	250			
Biosolids	470	590			
Cover Soil	28.	.1			
Rainwater	0.0	61			
Precursors ¹	282	319			
Leachate			563	638	
Landfill Gas			0.	.02	
Totals	2,030.7	2,187.7	563.02	638.02	

Table 1: PFAS Mass Balance Summary Table (kg/year).Table courtesy of Tetra Tech, Inc.

While this approach can identify overall trends and provide estimates for the average PFAS mass cycle, it does not account for regional variation, nor does is provide site-specific guidance. Two major factors affect the accuracy/usefulness of the evaluation:

• The mass balance is a meta-analysis built upon research conducted by other multi-disciplinary groups. It uses data covering approximately two decades, from the early 2000s to 2019, and there have been delays in data gathering, analysis and publication. Year-over-year changes are likely to be small and their effect on the overall outcome are not expected to change the outcome significantly. Of note is the EPA PFOA Stewardship Program, under which eight major companies agreed to a 95 percent reduction in the

manufacture and use of PFOA and its precursors. While the program took effect during the period considered in this research, its effects are offset by the time until such products are landfilled and the import of products from regions that have not joined the program.

• The variety of analytical methods, definitions and decisions by scientists in the PFAS research relied upon in this analysis increases the uncertainty of the data presented. For example, Lang et al. considered 19 substances while Venkatesan and Halden accounted for 13. However, these datasets and others agree that perfluorooctanesulfonic acid (PFOS) and perfluorooctanoic acid (PFOA) account for the majority of the considered PFAS.

The mass balance also relies upon landfilling practice assumptions to estimate certain components. These assumptions are appropriate for this type of high-level analysis, and efforts were made to err on the side of caution.

With these limitations in mind, the mass balance indicates a few areas that will have an impact on future PFAS cycle policies and practices. We may expect that landfilling rates for biosolids will increase, as a reaction of jurisdictions to limit land-application. This is likely to reduce the introduction of PFAS to groundwater sources but increase PFAS disposal at landfills. On a local level, PFAS discharged with leachate to a POTW has been shown to exceed the headworks mass loading from sanitary wastewater in some cases. As effluent from POTW is discharged to streams it can enter the potable water supply. Leachate treatment for PFAS is challenging, due to the nature of leachate and PFAS. Until technological solutions are developed and economically feasible, a larger amount of PFAS can be sequestered by landfilling biosolids. | WA

Arie Kremen, PhD, is a civil and environmental engineer at Tetra Tech, Inc. (Pasadena, CA) with more than 25 years of experience in solid waste engineering and water resources, with a professional specialization on leachate management and disposal. His academic background is in beneficial reuse of reclaimed wastewater, including biological nutrient recovery. He has worked abroad and in the U.S. on wastewater and leachate treatment; landfill design and construction; and landfill remediation/closure projects. Dr. Kremen is the vice-Chair of the SWANA technical committee on landfill liquids, where he is leading the organization's effort into building a PFAS wiki for the solid waste industry. He can be reached at **Arie.Kremen@tetratech.com**.



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Note

1. Estimated as 50 percent of the leachate PFAS mass loading.



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Best Practices for Optimizing **PFAS ANALYSIS**



Per- and polyfluoroalkyl substances (PFAS) are currently of great public health and environmental concern. Because **PFAS are ubiquitous and commonly used in materials routinely employed for chemical analysis**, laboratories are in need of streamlined protocols to minimize background contamination from these chemicals and quickly generate accurate data. This ebook outlines best practices, from the field to the bench, for achieving those goals.

Collecting Samples

Personal Gear

Sampling for PFAS without contaminating the samples can be challenging due to the prevalence of these chemicals in many consumer products and standard sampling equipment. To avoid the possibility of cross-contamination, lab and field personnel should select field clothing and personal protective equipment (PPE) carefully when collecting or preparing samples for PFAS analysis.

	Items to AVOID During Sampling	Items RECOMMENDED During Sampling
	Water-resistant, waterproof or stain- treated clothing, boots and/or rain gear made from materials containing PFAS.	Rain gear made from polyurethane or wax-coated materials. Boots made with polyurethane and polyvinylchloride (PVC).
	Clothing with fabric softener or suspected of containing PFAS. Some items labeled as "PFOA-free" contain replacement PFAS.	Cotton clothing is recommended and should be well washed before use due to possible contamination from PFAS-related treatments.
Sunscreens, moisturizers, hand cream or other related products.		Avoid using any personal care products.
		14/20 30



During collection, well-washed cotton clothing and outer gear made from polyurethane or wax-coated materials is recommended.



Field Equipment and Sampling Bottles

Potential sources for PFAS cross-contamination include many items commonly found in the sampling equipment, such as items directly involved in the sample collection (e.g., automatic samplers, dippers and tubing) and other accessories. To ensure an accurate assessment of PFAS, sampling personnel should take precautions when collecting samples.

Due to potential adsorption of analytes onto glass, lab and field personnel should use polypropylene containers for all standard, sample and extraction preparations. Polypropylene bottles fitted with polypropylene screw caps allow for PFAS sampling without the risk of cross-contamination. Sample bottles must be discarded after use to prevent contamination from previous sampling procedures.

Items to AVOID During Sampling	Items RECOMMENDED During Sampling
Any items with a non-stick coating containing PFAS, including containers, tubing or any other waterproofed items (e.g., notebooks).	High-density polyethylene (HDPE) or polypropylene containers with HDPE or polypropylene caps.
Plastic materials potentially containing PFAS.	HDPE or silicone tubing materials.





Background Contamination

In order to check for residual PFAS on sampling equipment and overall contribution from different sources during the sampling event, equipment and field blanks should be collected prior to and during sampling. When collecting samples, personnel should use new nitrile gloves and replace them frequently to avoid cross-contamination.

Standard precautions for sample collection (e.g., bottle cap should not be placed on any other surface, avoid contact with inside of cap or bottle) should be strictly followed. After the sample is collected and capped, the sample bottle(s) should be placed in a resealable plastic bag separate from all other sample bottles.

Avoid reusing sampling equipment as previous uses may have involved PFAS-containing materials. Maintain separate supplies for PFAS sampling and for other contaminants. Before using new equipment, test for the presence of PFAS.

When reuse of materials and sampling equipment is necessary, lab and field personnel should follow standard decontamination procedures (as described later in this ebook) and confirm the absence of PFAS before reusing the equipment. It is also recommended to avoid the use of any materials listed on pages 3 and 4.

Lab Equipment Cleaning & Decontamination

PFAS can be present in the water and/or cleaning agents used in decontamination processes. When cleaning sampling equipment, lab personnel should avoid using decontamination soaps containing fluorosurfactants such as Decon 90. Water from an on-site well is also a potential source of contamination.

Alconox[®] and/or Liquinox[®] are recommended for decontamination processes as well as potable water from a municipal drinking water supply. Sampling equipment should be scrubbed using a polyethylene or PVC brush and flushed with water before the next use. Water should be always verified as "PFAS-free" before it is used for field and decontamination blanks and decontamination processes.

Food & Beverages

Standard safety protocols do not allow the presence of food and drinks in laboratories and areas where sampling is occurring. During the PFAS analysis, this safety protocol is even more relevant as food packaging, wrappers and containers may contain PFAS and can cause cross-contamination. Drinks and food should be kept nearby (e.g., staging area for sampling) to ensure personnel's safety.



Preparing Samples

LABORATORY MATERIALS

Preparation and Storage of Stock Solutions and Standards

Stock solutions should be prepared and stored in PFAS-free high-density polyethylene (HDPE) or polypropylene (PP) containers with lined or unlined HDPE or polypropylene caps. Do not store samples in containers made of glass or low-density polyethylene (LDPE) materials. PFAS can adsorb to glass, especially when the chemicals are stored in a glass container for long periods of time.

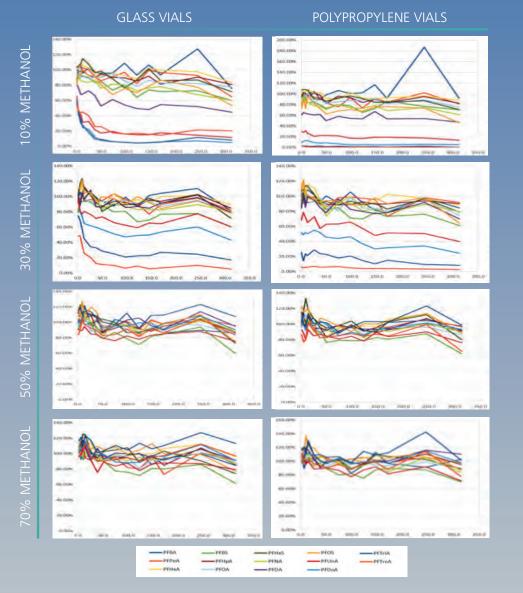
Stability of the standards solutions for a predetermined interval of time when stored under recommended conditions is a relevant parameter for ensuring the quality of the analysis. As shown in Figure 1 (see next page), 50% methanol in water (same mixture as that used in ASTM D7979) is the optimal solution for dissolving PFAS and maintaining them in solution.





Figure 1: Potential adsorption of PFAS on the vial surface

Plots of PFAS recovery against shelf life (time/hour) for the various solvents in glass and polypropylene LC vials.



Mixtures with lower concentrations of methanol (10% and 30%) show larger losses of PFAS due to the insolubility of PFAS in the solvent used. The recovery results for 90% methanol are similar to that of 70% methanol. However, the **higher methanol content evaporates faster and causes changes in the sample volume**.

The PFAS concentration in the vial may change after the vial cap is pierced as the organic solvent (e.g., methanol:water solution) and/or PFAS compound can be lost through the puncture. If calibration standards are to be used multiple times, it is recommended to use an amber glass vial with sealed replaceable caps. Sealing the vials immediately after injection may reduce the loss of PFAS.

The use of LC propylene vials is commonly recommended for the analysis of PFAS. Shimadzu scientists compared LC propylene vials to amber glass vials (used in the majority of general applications and more easily resealed) to determine the potential adsorption of PFAS on the vial surface. Similar recovery and quantitation were observed for both types of materials, as shown in Figure 1.

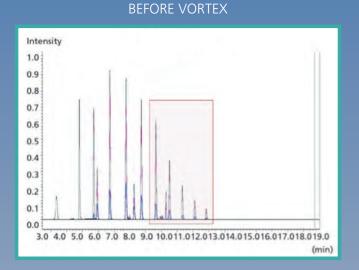
Sample Preparation and Injection

Some currently published methods (EPA 537, EPA 537.1) require a step of sample pre-concentration by solid phase extraction (SPE). Materials used in the manufacturing of supplies for preparing the samples by SPE may also contain PFAS. To avoid pre-concentrating the background PFAS during this step of the analysis, all new SPE cartridges, solvents and vials for collecting samples must be tested for PFAS prior to the first use.

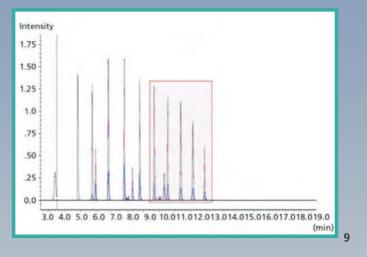
PFAS-free tubing should be used for loading samples into the cartridges. If automatic sample extractors are employed for this step of the analysis, checking with the manufacturer is strongly recommended to identify all components made of PFTE and replace them when feasible.

Once samples are pre-concentrated and ready for injection in the LC-MS/MS or samples are prepared accordingly to methods that allow for large volume injection (ASTM D7979), they may sit in the autosampler tray for extended periods of time. In these situations, some PFAS compounds may settle, precipitate or adsorb on the surface. It is important to remember to mix the extract/sample before (re)injection. Vortexing the solution before injection ensures a homogenous solution and optimum results. Figure 2 shows the chromatogram of the PFAS compounds before and after vortexing a 50 ng/L standard allowed to sit for 24 hours. The recovery of the long-chain PFAS is considerably lower before vortex.

Figure 2: PFAS compounds before and after vortexing a 50 ng/L standard allowed to sit for 24 hours



AFTER VORTEX



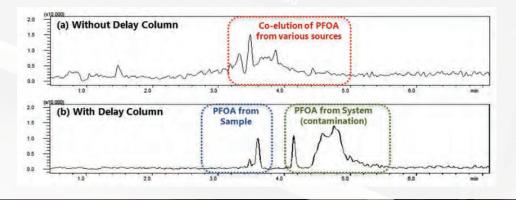


Instrumentation

It is recommended to use a solvent delay column (installed after the mixer and before the autosampler) to delay the elution of PFAS originating from solvent bottles and other parts of the liquid chromatography system (e.g., pumps and tubing). As shown in Figure 3 below, using the delay column enables the detection of PFOA originating solely from the sample.

Additionally, bypassing the degasser when possible is recommended as well as replacing any PTFE-containing tubing and parts in the LC.

Figure 3: Chromatogram of PFOA: (a) without delay column and (b) with delay column



10

Shimadzu's team of service engineers can help you set up the exact LC configuration (including solvent lines, tubing, bypassing of solvent lines and more) that is proven to deliver contamination-free results. For more information, please contact a Shimadzu expert at **800-477-1227** or visit **www.OneLabOneEarth.com**.

In collaboration with EPA and ASTM International, Shimadzu is working to advance research and technical knowledge related to PFAS exposure and contamination. Using Shimadzu LC-MS/MS instruments, they have vetted standardized methods for analyzing PFAS compounds in a diverse type of samples. Designed with proprietary ultrafast technologies and patented ion focusing technology, Shimadzu's LC-MS/MS systems deliver fast, high-quality results for PFAS analysis.



To learn more about Shimadzu's solutions for PFAS analysis, visit **www.OneLabOneEarth.com**





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An Equipment Manufacturer's Perspective on Regulatory Guidance and Ambiguity on PFAS in Groundwater Sampling

Sandy Britt, PG, CHG

QED Environmental Systems Inc. sbritt@qedenv.com



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What about PFAS? Addressing the materials issues

- There is concern that sampling for PFAS using sampling equipment manufactured from fluoropolymers (e.g., Teflon, PTFE, ETFE, FEP) could result in sample contamination
- Recommendations or requirements in regulatory guidance documents, SOPs and "fact sheets" from industry organizations to avoid the use of all fluoropolymers have been based on an abundance of caution, and research continues to determine which materials can be safely used
- Manufacturers of sampling equipment and components such as plastic tubing are <u>challenged with finding alternate materials</u> that can meet performance requirements while meeting needs for both PFAS sampling and other organic compounds



Some examples...

RESEARCH ARTICLE

WILEY

Evaluating PFAS cross contamination issues

Samuel A. Bartlett | Katherine L. Davis

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Abstract

Avoiding cross contamination from per- and polyfluoroalkyl substances (PFAS) that may occur during sampling of environmental media is the key to ensure reliable analytical results during a PFAS sampling program. Due to the ubiquitous nature of PFAS in commonly used sampling materials and personal protective equipment, mitigating the risk of cross contamination is a challenge that requires a conservative approach when planning and executing a PFAS sampling program. This article describes a conservative approach to PFAS sampling and includes a case study that evaluated three insect repellent products to determine their suitability for use during PFAS investigation. The three products were verified to be PFAS-free for the 17 PFAS included in the analysis and, therefore, these products are suitable for use during PFAS sampling activities without concern for cross contamination.

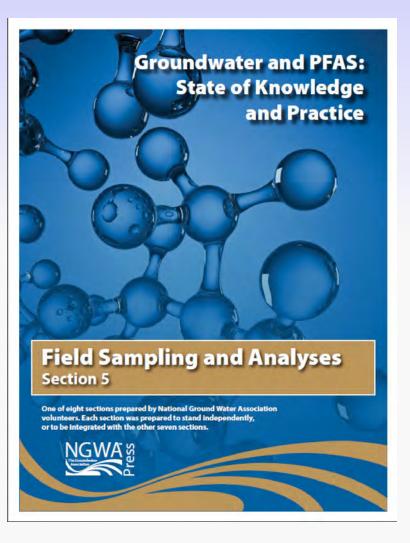
"A common trend in many PFAS sampling documents is to completely prohibit the use or even the presence of suspected items on a project site undergoing PFAS sampling."

"A conservative PFAS sampling guidance should include testing procedures to evaluate whether a material suspected of containing PFAS presents a risk of cross contamination."



Some examples, continued

"The materials of construction.... should be free from polytetrafluorethylene (PTFE) or ethylene tetrafluoroethylene (ETFE) to the maximum extent practicable.





From NGWA, March 2018

Per- and Polyfluoroalkyl Substances (PFAS) Sampling Guidelines

CALIFORNIA STATE WATER QUALITY CONTROL BOARD DIVISION OF WATER QUALITY



March 20, 2019

3.1 SAMPLING EQUIPMENT

The actual list of PFAS-containing materials potentially encountered onsite will change based on the specific sampled media and site-specific sampling conditions. Allowable materials include high-density polyethylene (HDPE), polypropylene, silicone, stainless steel, nylon, PVC, acetate, and cotton. Do not use any equipment that contains any known fluoropolymers including, but not limited to:

- Polytetrafluoroethylene (PTFE), including the trademark Teflon® and Hostaflon®, which can be found in many items, including but not limited to ball check-valves on certain bailers, the lining of some hoses and tubing, some wiring, certain kinds of gears, lubricant, and some objects that require the sliding action of parts.
- Polyvinylidene fluoride (PVDF), including the trademark Kynar[®], which can be found in many items, including but not limited to tubing, films/coatings on aluminum, galvanized or aluminized steel, wire insulators, and lithium-ion batteries.
- Polychlorotrifluoroethylene (PCTFE), including the trademark Neoflon[®], which can be found in many items, including but not limited to valves, seals, gaskets, and food packaging.
- Ethylene-tetrafluoro-ethylene (ETFE), including the trademark Tefzel[®], which can be found in many items, including but not limited to wire and cable insulation and covers, films for roofing and siding, liners in pipes, and some cable tie wraps.
- Fluorinated ethylene propylene (FEP), including the trademarks Teflon[®] FEP and Hostaflon[®] FEP, and may also include Neoflon[®], which can be found in many items, including but not limited to wire and cable insulation and covers, pipe linings, and some labware.
- Low density polyethylene (LDPE) should not be used for any items that will come into direct contact with the sample media. LDPE can be found in many items, including but not limited to containers and bottles, plastic bags, and tubing.

From Per- and Polyfluoroalkyl Substances (PFAS) Sampling Guidelines, CALIFORNIA STATE WATER QUALITY CONTROL BOARD, March 2019. https://www.waterboards.ca.gov/water_issues/programs/pfas/

What are my options?

- Examples of alternate materials offered in guidance documents all have some limitations:
 - HDPE isn't as strong and flexible as PTFE and FEP cycle life testing on HDPE bladders showed 1,500 – 3,000 cycles to failure, equal to 1-2 years of use for dedicated pumps (PTFE = 200K cycles, 100-200 years use)
 - Polypropylene is rather inflexible and tends to take a set when used for materials such as tubing, making it difficult to impossible uncoil, especially in cold weather
 - Silicone rubber is flexible but has a high capacity for sorption of organics
 - Vinyl (Tygon or flexible polyvinyl chloride) is made flexible through the use of phthalate plasticizers that will leach into samples, also absorbs organics
 - Alternatives to Viton (FKM), such as nitrile rubber, often leach other organic compounds - QED testing of nitrile showed up to 10,000 µg/l carbon disulfide



Is there actually PFAS in my Teflon?

- Not all fluoropolymers will leach PFAS into groundwater samples
- The only way to be certain that sampling equipment is PFAS-free is through material testing and analysis
- QED testing has shown that PTFE pump bladders and seals and FEP tubing have tested to be free of PFAS based on the lowest available laboratory reporting limits
- Manufacturers of sampling equipment and components such as plastic tubing are challenged with finding alternate PFAS-free materials that can meet engineering performance requirements while also meeting sampling program needs for other organic compounds such as fuels and solvents (VOCs and SVOCs) without sample bias or contamination
- Portable and dedicated sampling pumps and passive sampling systems are available that are entirely PFAS-free and Teflon-free



Some early research studies of common commercial and consumer products show PTFE thread tape and "pipe dope" as likely sources of PFAS

Table 6-1. Comparison of source strengths for total amount of PFCA (TPFCA) in a hypothetical, "typical" American home a

Group ID	Article category	TPFCA in article	Article quantity ^b	TPFCA in home (mg)
A	Pre-treated carpeting ^c	48.4 ng/cm^2	150 m^2	72.6
В	Commercial carpet-care liquids	12000 ng/g	6 kg ^d	71.8
С	Household carpet/fabric-care liquids and foams	953 ng/g	1 kg	0.95
	Treated apparel	198 ng/g	2 kg	0.40
D E F	Treated home textile and upholstery	336 ng/g	5 kg	1.68
F	Treated non-woven medical garments	795 ng/g	0 kg	0
G	Treated floor waxes and stone/tile/wood sealants	2430 ng/g	1 kg	2.42
H	Treated food contact paper	3100 ng/g	0.01 kg	0.03
I	Membranes for apparel	124 ng/g	1 kg	0.12
J	Thread seal tapes and pastes	603 ng/g	0.02 kg	0.01
Κ	Non-stick cookware	0.028 ng/cm ²	1 m^2	0.0003
L	Dental floss and plaque removers	31.3 ng/g	0.005 kg	0.0002
М	Miscellaneous	69.5 ng/g	0	0

^a The average, single-family home size in the U.S. in 2004 was 2330 ft² (http://www.nahb.org/). ^b The quantities of articles are rough estimates. ^c Assuming 70% of floor area is carpet; conversion factors for total PFCA are given in supporting information. ^d For one application; dilution factor is considered.



From Perfluorocarboxylic Acid Content in 116 Articles of Commerce, EPA/600/R-09/033, March 2009

Peristaltic Pumps

- Fits any well diameter, including small direct-push wells and multi-level systems
- Suction lift limited to 20 26 (6 8m) feet water depth, including drawdown
- Flexible elastomeric tubing, such as silicone, is required at pump head but can be attached to other nonfluoropolymer tubing materials such as HDPE & LDPE
- While peristaltic pumps are often cited as less accurate for gas sensitive parameters (e.g., VOCs, metals), PFAS are not volatile and quite stable in water, so no sample bias is expected



Battery-powered peristaltic pump



AC-powered peristaltic pump



Electric Submersible Pumps

- Fit into 2-inch (50mm) well casings
- Sampling depths up to 275 feet (84m) for AC-voltage pumps and 50 – 200 feet (15m - 60m) for DC-voltage pumps
- Greater depths for DC pumps using drop tube inlet where water depth <150'
- May not work where guidance or GWSAP for PFAS sampling prohibit use of Teflon (fluoropolymers) - many electric pumps have PTFE motor seals, PTFE wear parts and ETFE-coated motor cable
- Testing for PFAS in Grundfos Redi-Flo2 (DiGuiseppi, et al., 2014) showed PFBA detection (>100 ng/L) – most likely source is ETFE (Tefzel®) wire insulation. QED testing of ETFE tubing detected PFBA at 750 ng/L

AC-voltage pump, control box and generator





ETFE Tubing, 24 hour minimum soak test

Perfluorinated Sulfonic Acids and Perfluorinated Carboxylic Acids by HPLC/MS

Analysis Method: PFC/537M Prep Method: EPA 3535A

Analyte Name	Result	MRL	Dil.	Date Analyzed	Date Extracted	Q
HFPO-DA	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorobutanoic Acid	750	10	1	09/29/16 12:39	9/29/16	*
Perfluoropentanoic Acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorobutane Sulfonate	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorohexanoic Acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluoroheptanoic Acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorohexane Sulfonate	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorooctanoic Acid	ND U	2.5	1	09/24/16 09:07	8/26/16	
Perfluorononanoic Acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorooctane Sulfonate	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorodecanoic Acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluoroundecanoic Acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorodecane Sulfonate	ND U	5.0	1	09/29/16 12:39	9/29/16	*
Perfluorododecanoic Acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluorooctylsulfonamide	ND U	5.0	1	09/29/16 12:39	9/29/16	*
Perfluoro-n-tridecanoic acid	ND U	6.3	1	09/24/16 09:07	8/26/16	
Perfluoro-n-tetradecanoic acid	ND U	5.0	1	09/29/16 12:39	9/29/16	*
Perfluoroheptane sulfonate	ND U	6.3	1	09/24/16 09:07	8/26/16	
N-ethylperfluoro-1-octanesulfonamide	ND U	5.0	1	09/29/16 12:39	9/29/16	*
N-methylperfluoro-1-octanesulfonamide	ND U	6.3	1	09/24/16 09:07	8/26/16	
2-(N-ethylperfluoro-1-octanesulfonamido)- ethanol	ND U	6.3	1	09/24/16 09:07	8/26/16	8
2-(N-methylperfluoro-1-octanesulfonamido) -ethanol	ND U	6.3	1	09/24/16 09:07	8/26/16	
6:2 Fluorotelomer sulfonate	ND U	6.3	1	09/24/16 09:07	8/26/16	
8:2 Fluorotelomer sulfonate	ND U	6.3	1	09/24/16 09:07	8/26/16	

Air-Powered Bladder Pumps

- Designs are available to fit well as small as 0.5" well casing and multilevel tubing wells
- Sampling depths to 1,000' (300 m) lift, even greater depths with drop tube inlets
- Wide range of material choices (PVC, stainless steel, poly) to match contaminant chemistry and background water quality – BUT – dedicated pumps historically use PTFE bladders, which can't be used under some sampling plans
- Portable and dedicated pumps are available with HDPE & LDPE bladders, but these often don't have the long bladder life typical of PTFE bladders and are designed to be replaced frequently, which defeats the purpose of a dedicated system



Dedicated Bladder Pumps



Portable Bladder Pumps



Electronic Filing: Received, Clerk's Office 11/23/2022

QED Sample Pro[®] PFAS-Free/Teflon-free Portable Bladder Pump Sampling Systems

The most reliable portable sampling pump is PFAS-Free

Sample Pro

The Original PFC-Free Bladder Pump

The Sample Propump and Tubing are and have Always Been PFAS-Free

WELL WIZARD [®] ZeroTM and ClearTM



- Well Wizard Zero models are constructed entirely from non-fluoropolymer plastics that have been tested and certified to be PFAS-free
- Well Wizard Clear models will use the same components but retain the PTFE bladder for very low level organic testing also tested PFAS-free
- QED's industry-first HDPE twin bonded tubing meets all PFAS sampling program requirements and has been tested for PFAS, VOCs and SVOCs
- Models available to sample to 600 feet depth (300 PSI pressure) and can sample to nearly unlimited depths using drop tube inlet systems
- Available November December 2019



Passive and No-Purge Samplers

- Much simpler to design without any fluoropolymers few to no moving parts
- Polyethylene Diffusion Bag (PDB) won't work for PFAS will not equilibrate
- Whole water samplers can work if sample volume requirements are met
- Some available without any fluoropolymers, but testing is still recommended to ensure that no PFAS can leach from materials used



PDB Sampler

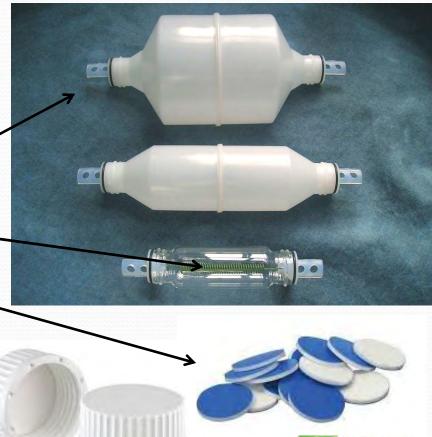
Snap Sampler[®]

Electronic Filing: Received, Clerk's Office 11/23/2022

SNAP SAMPLER ZeroTM

- All components tested for PFAS
- Molded acetal "snap caps" with EPDM ⁻ O-ring seals
- Passivated stainless steel center springs
- Distinctive white HDPE liner bottle caps for 125 mL and 350 mL poly bottles and white/blue septa caps for 40 mL VOA vials sealed in separate packaging
- Available November December 2019





Sampling Equipment Recommendations

- Follow a common sense approach to the use of any materials and supplies – look for studies on PFAS content in materials and, when in doubt, either test your system or eliminate suspect materials
- For new dedicated pump systems, portable pump systems and passive samplers, equipment blank testing can determine if they're PFAS-free, or obtain certification from the manufacturer that the equipment and tubing has been tested and is PFAS-free
- For existing dedicated sampling systems, test in place for absence or presence of PFAS in samples before replacing any components
 - Where results are ND in all wells, systems can be used (unless GWSAP or regulatory restrictions on existing materials exist)
 - Where PFAS is detected in some or all wells, those wells can be sampled again using a known PFAS-free system to determine if source is the sampling system or if PFAS existing in the water
 - When a sampling system shows PFAS, look for sources such as PTFE thread tape, gaskets or seals that could be eliminated or replaced with alternate materials





Questions?

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